To: AIChE National Review Committee

From: Group 1737

Subject: Completion of Neural Stem Cell Manufacturing Facility

To whom it may concern:

Group 1737 was organized in an attempt to design a manufacturing facility that produces neural stem cells. We are pleased to announce the completion of the design along with a detailed economic and safety analysis. The group recommends pursuing the construction of the manufacturing facility with plans to expand internationally when the economic opportunity is present.

We are looking forward to the review from your committee. With this transmittal, the group gives the committee permission for public review on the facility. The quality input will be used as a tool for improving the facility as the group looks to move into detailed design. We look forward to the honest and useful feedback from our work.

Design for Neural Stem Cell Commercial Manufacturing Facility

Group Number

9 March 2016

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Abstract:

Over the past ten years, advances in stem cell therapy have progressed from low yield treatments on small organisms, to effective, targeted treatments on humans. One of the largest markets for stem cell therapy is treatment of bodily injury, particularly in the nervous system. Research has shown that induced pluripotent stem cells can expand and differentiate into neural stem cells, which has been clinically shown to relieve pain caused by spinal cord injuries.¹⁶ Currently, over 250,000 Americans suffer from spinal cord injuries, with a 5% increase predicted over the next five years. Stem cell therapy is a rapidly developing field to provide treatment of these injuries, and with the technology available, a facility can be design to produce enough neural stem cells to treat the entire market.

Specifically, induced pluripotent stem cells have shown to grow rapidly with relatively high viability in vitro. By selectively differentiating induced pluripotent stem cells into neural stem cells, a large amount of neural stem cells can be provided to victims of spinal cord injury. ¹⁶ The facility outlined in this report is designed to take a batch of 100,000 induced pluripotent stem cells and create enough neural stem cells to treat the predicted American population over the next five years. The total amount of neural stem cells needed to treat the entire market is approximately 1.2 trillion cells per year. Batch processes, especially pharmaceutical processes, have relatively low service factors. A service factor of 0.75 was determine in order to account for the significant down time associated with batch processes because sterilization is key to maintaining the integrity of the facility and product.

Economically, the viability of the project has been evaluated over a 5 year time period with a minimum rate of return of 50%. It is very likely that international expansion will present an economic opportunity in a relatively short time frame. To account for potential expansion, the facility has the ability to incorporate overseas markets when the opportunity arises. It is recommended that the company moves forward with detailed design of the facility immediately to capitalize on the current financial opportunity.

Introduction:

Stem cell therapy is at the frontier of modern treatment methods. Research has shown that the efficiency of stem cell cultures has increased exponentially over the last five years; however, this advance is met with an increasing market size. Stem cell treatment has been validated on a small scale, but there are currently no large scale treatment options are available in the United States due to restrictions by the FDA.²⁷ On the other hand, other countries that implement stem cell therapy have produced promising results. Spinal cord injuries that are treated outside of the United States typically consist of a series of 12-16 injections, with each injection consisting of around three million stem cells¹⁹. The total number of people affected by spinal cord injuries in the United States over the next five years is predicted to be 310,000. In order to make enough stem cells to treat the entirety of the affected population in the United States, a facility needs to be able to have a high viability of stem cells, with extreme precision and efficiency within the process. Furthermore, the facility needs to have a high throughput in order to meet the market demand, but costs need to be minimized to ensure the economic viability of the project. Overall, recent stem cell research has shown potential for production of high throughput processes that allow the industry to reach a larger market. However, no current facility has been designed to provide therapy to the entire population.

Process Flow Diagram:

In the Process Flow Diagrams seen across the next few pages, the volumes given are the total volume for one batch. In particular, the media is changed more than once per batch; however, the replaced media is not of equivalent volume to the original media. Since cell concentration must be maintained within a certain range, a larger volume of replacement media is needed than the original volume due to an increased number of cells. Therefore, it is more applicable to display the volume per batch instead of the volume added when materials need to be replaced. The process is illustrated over three drawings for clarity and simplicity.



Figure 1-PFD of Neural Stem Cell Production Facility





Figure 2-PFD of Neural Stem Cell Production Facility

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Figure 3-PFD of Neural Stem Cell Production Facility

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Stream number 1 2 3 4 5 6 7 8 9 10 11	.2 13
Phase Solid/Liquid Liquid Gas Liquid Solid/Liquid Solid/Liquid Solid/Liquid Solid/Liquid Gas Liquid Solid/Liquid Solid/Liq	Liquid Liquid
Pressure (psig) 0.00 0.00 15.00 0.00 0.00 0.00 0.00 0.0	00 0.00
Temperature (°C) 37.00 37.00 20.00 37.00	.00 37.00
Heat Capacity (J/gC) 3.45 4.3 1.0005 4.3 4.3 3.45 4.3 3.45 1.0005 4.3 4.3 3.45	45 4.3
Actual Density (kg/m ³) 1125.00 1100.00 1.225 1100.00 1100.00 1125.00 1100.00 1125.00 1.225 1100.00 1100.00 11	5.00 1100.00
Total Volumetric Flow Rate (mL/batch)	
iPSCs (mL) 6.30E-04 7.56E-03 7.56E-03 9.08E-02 9.08E-02 1.09E+00 1	09 -
Expansion Media (mL) - 3.20 - 24.00 1.20 - 12.00 288.00 144.00	- 3456
Air (mL/min) 0.31 0.14	
Differentiation Media (mL)	
NSCs (mL)	
Stream Summary Table - Expansion and Differentiation of Induced Pluripotent Stem Cells to Neural Stem Cells	
Stream number 14 15 16 17 18 19 20 21 22 23 24	.5 26
Phase Gas Liquid Solid/Liquid Solid/Liquid Liquid Liquid Solid/Liquid Gas Solids/Liquids Gas Solids/Liquids Gas	Liquid Liquid
Pressure (psig) 15.00 0.00 0.00 0.00 0.00 0.00 0.00 0.0	00 0.00
Temperature (°C) 20.00 37.00 37.00 37.00 37.00 37.00 37.00 37.00 37.00 37.00 37.00 37.00 127.00 3	.00 37.00
Heat Capacity (J/gC) 1.0005 4.3 4.3 4.3 4.3 4.3 4.3 3.45 4.3 2.19 4.299 2.19	45 4.3
Actual Density (kg/m ³) 1.23 1100.00 1100.00 1100.00 1100.00 1100.00 1125.000 1100.00 1.58 1100.00 1.58 11	5.00 1100.00
Total Volumetric Flow Rate (mL/batch)	
iPSCs (mL) 13.05 13.05 13.4 19.80 - 1	.80 -
Expansion Media (mL) - 315.2 1730 1730 1730 2.07E+04 2.07E+04 -	- 2.07E+04
Air (mL/min) 173.00 2.07E+03	
Differentiation Media (mL)	
NSCs (mL)	
Steam (L) 41400 - 20760	
Stream Summary Table - Expansion and Differentiation of Induced Divinetant Stem Calle to Neural Stem Calle	
Stream number 27 28 29 30 31 32 33 34 35 36 37	8 30
Bbase Liquid Solid Cas Solid/Liquid Solid/Liquid Solid/Liquid Liquid Cas Solid/Liquid Solid/Liquid Liquid Cas Solid/Liquid Solid/Liquid Solid/Liquid Liquid Cas Solid/Liquid Solid/Liquid Liquid Cas Solid/Liquid Solid/Liquid Liquid Cas Solid/Liquid Liquid Cas Solid/Liquid Liquid Cas Solid/Liquid Solid/Liquid Liquid Cas Solid/Liquid Liquid Cas Solid/Liquid Liquid Cas Solid/Liquid Liquid Liquid Liquid Liquid Cas Solid/Liquid Liquid Liquid Liquid Liquid Cas Solid/Liquid Liquid Liquid Liquid Liquid Liquid Liquid Liquid <thliq< th=""> <thliq< thr=""> Liq</thliq<></thliq<>	uid Liquid
$\begin{array}{ c c c c c c c c c c c c c c c c c c c$	
Temperature (°C) 37.00 37.00 20.00 134.00 37.00 25.00	.00 13.00
Heat Capacity (I/gC) A 3 3 45 1 0005 2 19 A 28 A 28 A 184 1 96 A 184 A 184 A 184	.00 37.00
Actual Density (kg/m^3) 1100 00 1125 00 1.22 1.00 00 1100 00 1100 00 1050 00 <t< td=""><td>0.00 1050.00</td></t<>	0.00 1050.00
Total Volumetric Flow Pate (ml /batch)	5.00 1050.00
iPSCs (ml)	
Expansion Media (ml) 1.84E+05 -<	
Air (ml/min) - - 5230 -	
Ditterentiation Media (ml) - - - - 52330 52331 - - - - -	
Differentiation Media (mL) - - - 52330 52331 -	
Differentiation Media (mL) - - - 52330 52331 -	 L/hr 4.871/hr
Differentiation Media (mL) - - - 52330 52331 -	 L/hr 4.87 L/hr -

Table 1-Stream Summary Table for Neural Stem Cell Production Facility

Material Balances:

To verify the system meets constraints material and energy balances were performed. The validity was determined by stating that what enters a unit must be equals what comes out. In order to perform and accurate material balance, cell and media densities need to be approximated. The material balances were done on a mass basis and densities were assumed to be uniform. The values for the density of media, stem cells, and air are 1100 kg/m³, 1125 kg/m³, and 0.0012 kg/m³, respectively⁸. If the difference between mass that enters and mass that exits is approximately zero then the system was considered balanced. Each expansion vessel as well as the differentiation reactor were balanced. A summary is seen in Table 1 and in the Appendix for all of the units.

CP-101				
Stream	Mass (g)			
1	0.000709			
2	1.1			
5	1.10071			
In-Out	-1E-06			
Stream	Energy (J)			
1	0.06150575			
2-4	118.25			
5	118.326325			
In-Out	-0.01481925			

The first expansion material balance was performed as follows. There are 7.1×10^{-4} g of cells entering the system and 1.1g of fresh media entering. 1.10071g of material leaves after 86 hours. This yields 1.0×10^{-6} as the difference between what comes in and what exits. This is close enough to zero; therefore, this system is balanced.

Similarly, the second expansion has 8.51×10^{-3} g of cells entering with 13.2 g of fresh media. 13.2021 g of spent media and expanded cells leaves the system after 84 hours. The difference between in and out was found to be 6.4 x 10⁻³. This confirms the system is balanced.

The third expansion has 0.102g of cells entering. There is 158.4 g of media entering as well. With this system, air is fed to this expander and the total oxygen consumed is 0.867g per batch. The amount of material out from this batch is 159.3g. The difference between in and out of the system is 0.0721. This is small enough to consider the system balanced

Moreover, the fourth expansion has 1.23g of cells entering. The media entering is 1901g and the oxygen amount entering is 0.867g. The final mass of the leaving mixture is 1901.95g. The difference in material in and out is 0.95g. In comparison to the overall magnitude of the feed streams this difference is small enough to be considered zero. This is the last expansion carried out in the laminar flow cabinet.

Lastly, the fifth expansion is carried out in a 40L bioreactor with air sparged through the media. The mass of cells entering is 14.68g and the mass of media added is 22,814g. The amount of oxygen added is 2.19g. This make the overall difference 0.92g. This is low enough to qualify as balanced.

The differentiation is the last step in the process. The amount of cells entering is 22.4g. The amount of media charged is 57,560g. The amount of oxygen delivered is 63.2g. The final amount out of the system is 57,607g. The difference then from in and out is 42.11g. This is a little large compared to the other units but compared to the overall system this is within acceptable range. A possible explanation of this is the lack of information of growth rates and substrate consumption rate of the cells during differentiation. Overall though this unit was considered balanced taking into account the uncertainty of the kinetic data.

Process Description:

Stem cell culturing is a delicate process and caution must be used throughout the process. The process begins with a batch of 100,000 induced pluripotent stem cells stored in a 1mL plastic vial in the vapor phase of a liquid nitrogen tank. Specifically, a temperature indicator and low temperature alarm are set to notify operators of a significant decrease in the temperature of the tank. The stem cells need to be kept at cryogenic conditions until the facility is ready to operate in order to properly preserve the raw materials. Premature thawing of the cells significantly reduces the overall production¹⁷. Once the facility is ready to begin a batch, the induced pluripotent stem cells are transferred from cryogenic storage to a thaw bath for two minutes to fluidize the cells. Water at 40°C is used to thaw the cells at an appropriate rate. After two minutes, the stem cells are then transferred by an operator to a 35mm square culture plate.

For all expansion chambers, the expansion media chosen is Essential 8 by Gibco. For the first expansion, 1.6mL is used each change in conjunction with the cell culture. The media in the culture is changed every three days and the cells stay in the culture plate for 85 hours. The cells are separated from the media via centrifuge and transferred to a 25mL T-Flask with fresh media for 85 hours. The cell culture goes through five expansion chambers in total, each with the same residence time. The size of the final three expansion chambers are 150mL, 2L, and 40L, respectively. At the end of the expansion phase, the number of cells is expected to grow from 100,000 to around 3.1 billion over the course of 15 days. In order to detach the cell aggregates, trypsin is added before the cells are removed. Moreover, all expansion chambers are kept at 37°C via electrical heating by a hot plate. Before transferring the process into a larger expansion chamber, the process is concentrated and water for injection is used to fluidize the cells to allow for transportation. After the 5th expansion chamber, the cells are sent to a centrifuge to isolate

the stem cells from the media where they are then fed to a bioreactor. For each expansion chamber, an oxygen consumption rate of 15.4 pmol/hr is used to determine the amount of oxygen needed to be supplied for each batch. Similarly, a media consumption rate of 8.51 pmol/hr is used to determine how often the media needs to be changed in the permanent expansion chamber²⁶.

In order to determine how long to keep the induced pluripotent stem cells, solver was used to maximize the operating time throughout the year by changing the total residence time of the cells in the expansion phase by varying the number of batches per year with the constraint of meeting the need for the market. The result is the ideal residence time of the cells in the expansion phase. However, nothing regarding the number of chambers, size, or media volume is included. Based on research, the media needs to be changed every other day to prevent unviable conditions of the cells. Cell cultures have shown significant a significant drop off in viability at concentrations over 1.2 million cells/mL because cells will compete against each other for nutrients¹⁷. To prevent this, cells are transferred to a larger expansion chamber with a higher volume of media to dilute the concentration of the cells. To produce the number of cells that need to be produced for each batch and maintain the appropriate concentrations, five expansion chambers are used with increasing size to maximize the efficiency of the process. Also, multiple articles have shown the doubling time of induced pluripotent stem cells to be around 24 hours. This produces a logarithmic relationship between growth factor and residence time. A graph of this relationship can be seen in Figure 4.



Figure 4-Graph of Residence Time versus Growth Factor

Additionally, number of cells versus residence time has an exponential relationship, which is consistent with standard cell growth. Similarly, this relationship is depicted graphically in Figure 5. Approximately three billion cells are sent to the reactor, which after differentiation produces around 58 billion neural stem cells per batch. This factor is specific to the neural cell media specification for growth factors. To meet market demand, 19 batches of 58 billion induced pluripotent stem cells are required.



Figure 5-Graph of Number of Cells versus Residence Time

As stated previously, centrifuges are used after each expansion step to separate stem cells from media. For this process, centrifuges that handle very small volumes are placed in a vial and are separated in a closed system while large volumes are separated using an open flow system. The first three expansion chambers all use the same centrifuge because it can handle the small range of volumes. However, the other two centrifuges are open systems. For open system centrifuges, fluid is sent through the system, where stem cells stick to the wall, and then removed and transferred to the next unit. Each open system centrifuge is constructed from stainless steel to prevent contamination of the cells.

The bioreactor has a 200L reactor volume and the liquid level in the reactor is expected to range from 22 liters to 53 liters over the course of seven days. The bioreactor needs to be able to handle one trillion neural stem cells each year to meet market demands. Neural induction media is added to the process to initiate differentiation to neural stem cells. The media is changed every

other day to maintain the proper process conditions in the bioreactor. A low shear agitator is used to maintain a well-mixed composition to ensure maximum yield. Low shear is critical to ensure that the stem cells are not damaged during the differentiation process. Furthermore, three process variable are critical to control in the reactor: pH, temperature, and oxygen concentration⁵. A change in pH is directly associated with consumption of media. As the nutrients in the media are consumed, waste is produced and the pH will deviate from ideal conditions. Once the pH reaches unacceptable conditions, the media is changed to minimize cell death²². Controls systems are installed to notify operators when the media needs to be changed to prevent undesirable conditions. Likewise, temperature is controlled by pulling a slip stream off of the boiler to provide hot water to maintain a temperature of 37°C inside the vessel¹⁷. Since the bioreactor is insulated, a minimal amount of heat is lost from the system. However, an automated temperature controller is necessary due to the economic consequences; the hot water feed is essential to maintaining the viability of the cells and needs to be controlled as precisely as possible. Lastly, the oxygen concentration is critical to the viability of the stem cells. Mitosis requires glucose and amino acids that the media provides as well as oxygen from a sparger to maximize the efficiency of cell division. Since oxygen concentration of the process is vital, an automated controls system is added to more precisely maintain the ideal operating conditions. Oxygen concentrations are set at 20% through a flow controlled gas sparger. After seven days in the bioreactor, the cells are sent to an inactivation chamber, where 37% gaseous formaldehyde is added at 1.31 liters per second for two hours. This step removes any viruses in the process, but does not harm the neural stem cells¹⁰. Gaseous formaldehyde targets the protein coat of the virus to inactivate it.

In order to separate the differentiated stem cells from the undifferentiated stems cells, a magnetic automated cell separator is used to sort the cells. Antigens with metal attached are injected into the process that bond to the neural stem cells and leaves the undifferentiated cells untouched. The batch stays in the separator for two hours to allow for complete separation. After two hours, the undifferentiated stem cells pass through the unit and the neural stem cells are removed from the separator by using water for injection. The cells are concentrated down to three million cells per milliliter using water for injection. In order to protect the cells, DMSO is added as a cryopreservative and the finished product is sent to cryogenic storage that is kept at - 148°C using liquid nitrogen¹⁵. Again, temperature control is critical to the viability of the product. The neural stem cells are kept in storage until it is shipped to the formulation group to prepare the product for packaging.

Energy Balance and Utility Requirements:

A large amount of operating costs each year is spent on energy consumption in the form of heating and cooling. Balances on all of the units was necessary to confirm the validity of the design calculations. The balance was performed in similar fashion to the mass balance by finding the energy into the unit and subtracting the energy out of the system and adding the energy either transferred into or out of the system. This sum should add to be zero if the system is truly in balance. Due to uncontrollable losses, however, none of the energy balance summed to exactly zero. The physical data of the system can be seen below in Table 3. These heat capacity values were estimated based on medium properties as well as cell properties. The heat of reaction for the cell growth and differentiation were determined to be negligible due to the low amount of heat needed for temperature control as well as the heat transferred in the laminar flow cabinet as the cabinet has such a controlled environment. All energy balances are found in the Appendix. A sample is located at the end of this section.

For Streams 1, 2, and 5 the energy was calculated by Equation 1 below. Where C_p is the heat capacity above and T is the temperature of the stream entering.

$$h = mC_pT \tag{1}$$

Where:

h = enthalpy of the material (J)

m = Mass of the Material (g)

Cp = Heat Capcity of the Material (J/gC)

T = Temperature of the Material (C)

The energy into the system was calculated as 0.062 Joules (J) for Stream 1 using Equation 1. For Stream 2 the amount of heat entering the first expansion is calculated as 118.25 J. The heat exiting the system in Stream 5 then is calculated as 118.33 J. Finding the difference then becomes -0.0148 J: small enough to be considered balanced.

For TF-100, the same process was applied by using Equation 1 across all of the streams. For Stream 6 the amount of energy entering is found to be 0.738 J. For Stream 4 the energy is 2100.1 J for the media entering. For the exiting mixture in Stream 7, the heat is 2100.46 J. This leaves the difference of the in and out as 0.40 J with neglecting heat transfer from the box. The value of the difference is low enough to be accepted.

The first stirred flask, SF-100, has 8.857 J entering with the cells, 25201.4 J with the fresh media. This is the first expansion to require air so the energy entering with Stream 9 is 21.8 J. The final mixture exits with 25226 J of energy. This leaves a difference between energy in and out at 5.8 J. This is small enough when compared to the overall amount of energy in the system.

This step is still contained in a laminar flow box so the energy transferred by the system is considered negligible.

The second stirred flask, SF-101, the cells bring in 106.7 J in Stream 12. The media is Stream 13 brings in 302,417 J. The air brings in 21.7 J in stream 14. The exit mixture has an energy amount of 302,529 J. These values give a balance of 16.4 J. This is a close enough value in comparison with the stream values to justify a balance closing.

The fifth expansion chamber is the first vessel to be outside of a laminar flow cabinet. This required a jacketed to maintain the temperature in the vessel. To calculate the amount of heat needed to maintain the internal temperature, the vessel dimensions of length 0.587 m and diameter 0.147 m were found by using Excel Solver to find the L and D that satisfy an L/D ratio of four.²⁵ The heat transfer area was estimated as the surface area of a cylinder with the dimensions given above. This gives a surface area of 0.271 m^2 for the inner surface the outer was found by adding a half inch of insulation to the outer thickness of the vessel. The outer surface area was found to be 0.277 m². The heat transfer rate could then be found by using the resistance method detailed in Perry's Chemical Engineers Handbook²⁴. The heat transfer rate to the environment was found to be 1.19 Watts or 4.056 BTU/hr. This amount although small is nonnegligible. Using this rate and a hot water stream the flow rate of water needed to keep the internal temperature was calculated to be 1.85 kg/hr. This ensures there is no significant temperature change in the reactor. To perform the energy balance, Stream 19, the fresh media feed was found to carry 3.63 MJ. The cells carry 1,880 J of energy and the air has 55 J of energy. The heating from the water over the cycle introduces 6,260 J of energy. The heat loss to the environment is 6,260 J The final amount of energy out of the system is found to be 3.63 MJ. This gives a difference of 1,092 J. This is close enough to zero given the magnitude of the other streams.

The differentiator was sized in a similar manner to the final expander with an L/D of 4 and a final vessel size of 108 liters. The heat loss with a half inch of insulation was found to be -3.1 Watts. This means that water at a flow rate of 4.85 kg/hr at 1 degree temperature drop. This gives a total amount of heat needed to be transferred at 1.87 MJ. The cells bring in 19.4 kJ from Stream 28. The media brings 9.1583 MJ in stream 27 and the air brings in 1,590 J with Stream 29. The final amount of heat coming out of the system is found to be 9.16 MJ. This gives a total difference of 1,940 J. This is low enough when compared to the stream magnitudes.

The final unit is the refrigeration unit for final storage. This unit needs a large amount of energy due to the large temperature drop the cells require to remain viable long term. The energy for the system was found using Equation 2 below.

$$Q = mC_p (T - T_f) + m\lambda + mC_{ps} (T - T_{final})$$
⁽²⁾

Where:

Q = heat transfer needed

m = total mass being chilled (kg)

- T = initial temperature of mixture (°C)
- T_f = fusion temperature (°C)
- λ = heat of fusion of mixture (J)

 C_{ps} = heat capacity of the solid mixture (J/kg °C)

 T_{final} = final temperature of the mixture (°C)

Most of the energy needed to chill comes from the freezing process as would be expected from any phase change. The constant values are seen below in Table 4. Using Equation 2 with Page 21 of 136 the constants below the energy needing to be transferred was found to be -12,867 kJ. This is handled by Asymptotes VIA freeze QUAD system. The cycle time for a batch was found to be 117 minutes with a maximum cooling rate of 1 C/min.

Energy Requirements for Freezing			
Mass In (g)	25645		
Cp (J/gC)	4.545		
Heat of Fusion (J/g)	333.55		
Cps (J/gC)	1.4		
Initial Temperature (C)	25		
Freezing Temperature (C)	0		
Final Temperature (C)	-80		
Energy Required (kJ)	14340		

Table 4-Refrigeration Constants

CP-101				
Stream	Energy (J)			
1	0.06150575			
2	118.25			
5	118.326325			
In-Out	-0.01481925			

Table 5-Example Energy Balance

Once the energy balances confirmed the validity of the equipment design, utility costing was performed. The overall utility costs were relatively low due the process consisting of batch reactors with residence times under three minutes. Additionally, much of the equipment did not require utilities as the glucose within the cellular media provided the energy for the reactions to occur as opposed to heat or high pressures. It is important to note that the process is a batch process so the cost of utilities per year is determined by calculating the utility cost per batch and multiplying the cost per batch by the total number of batches. The majority of units which

required utility input were support systems for the process itself, such as the boiler and laminar flow cabinets.

The utility costs for predesigned equipment was determined by converting the set voltage and electrical currents to power in kilowatts and using the provided costing information of \$0.05 per kW-hr. This method was applied to R-300, the laminar flow cabinet, the air filtration systems, and the centrifuges: C-100, C-200, C-300. A sample of this calculation can be seen below. All other calculations of this method can be found in the Appendix.

Purifier Utilities				
Volts V	230			
Amps A	5.4			
Power kW	1.242			
Energy Used per Batch				
kW-hr	10335.92			
Price of Energy Per				
Batch \$/kW-hr	0.05			
Price to Run \$/Year	9818.59			

Table 6-Purchased Equipment Utility Costs

For all other units, hand calculations using data and engineering heuristics were used to approximate energy consumption. One such unit this for which this method was used is the boiler unit. Both the costs of electricity to heat the coils and the boiler feed water were calculated using by hand calculation found in Analysis, Synthesis, and Design of Chemical Processes.²⁵ An example of the costing process can be seen below equations for the sample calculation can be found in the Appendix.

Operational Costing			
Boiler Feed Water \$/kg	0.00245		
Electricity \$/kW-hr	0.05		
Boiler Feed Water \$	0.010995		

Electricity \$	586.8408
Total \$	586.8518
Total/year	11150.18

Table 7-Boiler Utility Costs

Water proved to be an important utility in the process because it is needed for sterilization in between batches.²¹ For both the final expansion chamber, BR-200, and the differentiator, BR-300, water is needed to heat the jacketed reactors to isothermal conditions. Instead of installing a heat exchanger to heat the water before feeding it to the reactor jacket, excess boiler feed water is fed to the reactor. Boiler feed water is used because the low cost of the utility makes slightly oversizing the boiler more economic than purchasing two new heat exchangers.²⁵ Additionally, water for injection for the final product vials is needed to prepare the stem cells for injection into a patient. The total weight of the water for injection need was calculated by determining the necessary concentration for therapy in the vials with the cells and replacing the media with water for injection.²⁰ Finally, boiler feed water was also used in the warm water bath to thaw the frozen iPSCs after cryostorage. Boiler feed water was selected to warm the cells due to the low cost and because the boiler feed water is treated, making it less harsh on the cell vials. An example calculation of the energy calculations for the warm water bath can be seen in Table 8.

Water Heat Calculations					
Heat Required by Water (J)	0.85				
Water Heat Capacity (J/kgK)	4.108				
ΔT water (K)	1.11				
Mass of Water (kg)	0.19				
Volume of Water (L)	0.19				
Volume of Water (gal)	0.05				
Volume of Tank (gal)	0.10				
Time in Thawing (min)	2				
Energy Transferred (mJ)	7.10437875				

Table 8- Energy Transfer for Thawing of Stem Cells

Equipment List and Unit Descriptions:

Boiler:

To prevent contamination, the process equipment is sterilized between each batch. Due to its low operating cost, steam-in-place was chosen as the method of sterilization. A boiler is used to generate the necessary 132 °C saturated steam for steam-in sterilization. 132 °C saturated steam was chosen for sterilization due to its common use in CDC sterilization techniques.²¹ The boiler must generate enough steam to contact the all of the surface area of the process vessels with steam for three minutes. The total volume of all the process vessels is 0.272 m³; however, the boiler was sized to produce ten times the volume of steam necessary to both meet industry standards and ensure enough steam is produced to fill the process vessels for the full 3 minute sterilization time.²¹ In order to generate 2.72 m³ of steam, 4.30 kg of boiler feed water is fed to the boiler for each steam-in process. The boiler has a duty of 62 kW to convert the boiler feed water to saturated steam. Due to the relatively small duty of the boiler and lack of corrosive or caustic materials, plain carbon steel was used as the MOC to avoid unnecessary cost. After use,

the steam vents to atmosphere so a full boiler system is not needed. This was done because the steam generated is so small, it is more cost effective to generate more steam than to install an entire boiler system. Heating coils were used to generate steam because electricity is a cheaper utility than fuel gas.²⁵

Cryogenic Storage/Preparation:

Both the purchased adult iPSCs and the final NSCs must be kept in cryogenic storage until use or shipping, respectively.³ Liquid nitrogen is used to keep the stem cells at the desired temperature. The stem cells are stored in a 24K Cryostorage System by MidSci.¹ This system was chosen due to the low cost compared to other products and its large storage capacity of 24050 2 mL vials. The storage units use SS as a material of construction and have a liquid nitrogen capacity of 365 L. Three storage units were selected to account for large hold up of the inlet adult iPSCs; the NSCs and iPSCs are stored in separate storage units to avoid cross contamination of stem cells. Due to a significant evaporation rate, excess liquid nitrogen for the storage units is kept in an exterior tank. The exterior tank was sized based on the need to refill the Cryostorage tanks every 52 days.¹ This is a common heuristic used for liquid nitrogen storage. This calculation led to a necessary size of 1500 L; however, nitrogen storage tanks have nominal sizing so a tank with a volume of 1893 L (3000 gal) was used. The exterior tank was assumed to be pressurized to 200 psig, based on literature values.¹¹ Due to the incredibly low temperature of the nitrogen, insulated stainless steel was used for the MOC. A refrigeration vessel was also incorporated to prepare the NSCs for cryogenic storage. The CryoMed Controlled Rate Freezer was used because of its FDA certification and stainless steel construction. The freezer has a max liquid hold up of 34 L which is the smallest unit than can

hold the entirety of the product in one batch. The freezer also can full chill to cryogenic conditions at -150° C and has a small electrical requirement of 120 V and 60Hz.

Viral Inactivation Tank:

A major issue with pharmaceutical manufacturing is the potential for the presence of viruses in the medical treatment. To reduce this potential, a viral inactivation tank was placed before the cell separation. The tank uses a sparger to distribute gaseous formaldehyde throughout the NSCs to kill off any viruses. The tank maintains a formaldehyde concentration 0.2 g/L for 2 hours to ensure complete virus inactivation. A sparger was necessary to ensure even distribution of formaldehyde throughout the media due to the natural mixing inherent to a sparger. Also, a sparger avoids having to spend extra money on an agitator for the tank. In order to reduce batch time, the inactivation tank was sized identically as differentiator to a volume of 108 L so the entirety of the 52 L batch could fit in the tank and only half way fill the tank. The tank was assumed to be at atmospheric pressure and is constructed from stainless steel to prevent corrosion and contamination.

Pumps:

Eight pumps were added throughout the process to ensure consistent volumetric flow rates. A major issue with the use of pumps is the shear stress caused by the impeller damaging the stem cells. To avoid cell damage, peristaltic pumps were used in place of traditional centrifugal pumps. The low shear of peristaltic pumps make them ideal for pumping cell media through the process. Since peristaltic pumps do not significantly increase the pressure of the working fluid, sizing was based entirely off the volumetric flow rate of 1.75 lpm. The Vector 2004 model peristaltic pump is used for all 8 pumps.

Cell Separator:

While the cell differentiator provides nearly 100% conversion from iPSCs to NSCs, separation of undifferentiated cells from NSCs is still necessary. A CliniMACS Prodigy cell separator is used to purify the cell product stream before being chilled. The cell separator uses magnetic-activated cell sorting to differentiate the neural stem cells. This is done by using magnetic antigens with metal attached to cause to allow a magnetic field to separate the NSCs from other cells. The device has a MOC of HDPE (high density polyethylene) and is design in accordance with the Miltenyl Biotec company standards.

Expansion Chambers:

Due to the limited feed stock, multiple expansion chambers are needed to produce the requisite number of cells before differentiation. The expansion is done in 5 parts with 4 expansions done on a bench top scale and the final expansion done in a large tank. Each expansion chamber was sized to start expansion with an initial cellular density of 100,000 cells/mL and end with a cellular density of 1.2 million cells/mL.⁸ These densities were chosen based to coincide with the original starting density and the maximum cellular density possible while still ensuring cellular vitality.⁸ Residence time in each chamber was determined by using a logarithmic function to model the time needed for the cell density to change from the starting density to the final density. The residence time for each expansion chamber can be seen in the Appendix. The original minimum number of chambers was only 4; however a 5th expansion

chamber was added to ensure none of first four chambers would be undersized and to account for future international market expansion. The first four expansion chambers are placed in laminar flow cabinet, along with a single bench-scale centrifuge for separating the concentrated stem cells from used media. The final expansion chamber feeds to an automated centrifuge which concentrates and separates the expanded stem cells from the media.

Bioreactor:

The GE XDR-200 Bioreactor Complete by GE was chosen as the cell differentiator for the process. This reactor type was chosen because of the known cost of the vessel and that it satisfactorily meets requirements to differentiate iPSCs into NSCs. A reactor size of 200L was chosen because it was the smallest nominal reactor volume that was still at least double the size of the process fluid.

It was deemed appropriate that the reactor be at least double the 52L volume of the cells and media to ensure over-pressure events and backflow through the vents or reactors could be avoided. The differentiator is a batch reactor which uses an agitator to stir the suspended cells in the differentiation media and operates at 37°C and atmospheric pressure respectively. The batch reaction has a residence time of seven days based on media specifications which illustrate the time necessary for iPSCs to differentiate into NSCs.

Centrifuges:

Three centrifuges were placed in the process to concentrate and separate cells from media. The first centrifuge is a bench top-scale size with a volume of 0.03m³ and has individual vials in which the media and cells can be placed during centrifugation. The second and third

centrifuges follow the final expansion chamber and the bioreactor, respectively. Centrifuges were sized to not operate over 3000 RPM and 300 RCF to prevent damage to cells. The requirement of low shear is satisfied at these conditions. All centrifuges were sized by assuming an L/D of 4 and using solver by changing settling velocity to determine an overall volume and a distance away from the center of the expansion chamber. The final volumes of the Centrifuges 2 and 3 are 0.05 m³ and 0.08m³, respectively.

Equipment Specification Sheets:

Equipment Specification Sheets-Vessels						
Equipment Tag	Category	Description	Extreme Temperature	Max Pressure	Volume/Surface Area	MOC
		Culture Plate				
		First Expansion				
		Takes 1.0*10 ⁵				
		Cells to 1.2*10 ⁶				
CP-101	Expansion Chamber	Cells	37 °C	1.01 Bar	962.11mm	Plastic
		T Flask Second				
		Expansion Takes				
		1.2*10 ⁶ Cells to				
TF-100	Expansion Chamber	1.44*10 ⁷ Cells	37 °C	1.01 Bar	19mL	Plastic
		Spinner Flask				
		Third Expansion				
		Takes 1.44*10 ⁷				
		Cells to 1.73*10 ⁸				
SF-100	Expansion Chamber	Cells	37 °C	1.01 Bar	250mL	Plastic
		Spinner Flask				
		Fourth Expansion				
		Takes 1.73*10 ⁸				
		Cells to 2.07*10 ⁹				
SF-101	Expansion Chamber	Cells	37 °C	1.01 Bar	3L	Plastic
		Spinner Flask				
		Fifth Expansion				
		Takes 2.07*10 ⁹				
		Cells to 3.14*10 ⁹				
BR-200	Expansion Chamber	Cells	37 °C	1.01 Bar	40L	SS
		Converts iPSCs to				
BR-300	Differentiator	NSCs	37°C	1.01 Bar	200L	SS
		Kills Viruses				
		without				
V-300	Inactivation Tank	damaging cells	37°C	1.01 Bar	200L	SS
		Prepares Cells for				
R-300	Media Prep Tank	Cryostorage	-150°C	1.01 Bar	20L	SS
		Thaws iPSCs out	45000	1.01.5	-	
WWB	warm Water Bath	of cryostorage	-150°C	1.01 Bar	5L	<u>>></u>
		SURTS OUT				
CS-300	Cell Sorter	cells from NSCs	25 °C	1 01 Bar	251	55
V-300 R-300 WWB CS-300	Inactivation Tank Media Prep Tank Warm Water Bath Cell Sorter	damaging cells Prepares Cells for Cryostorage Thaws iPSCs out of cryostorage Sorts out undifferentiated cells from NSCs	37°C -150°C -150°C 25 °C	1.01 Bar 1.01 Bar 1.01 Bar 1.01 Bar	200L 20L 5L 25L	SS SS SS

Table 9-Vessel Equipment Specification Sheets

Equipment Specification Sheets-Pumps						
Equipment Tag	Category	Description	Pressure Rise	Flow Rate	мос	
P-200 Pump		Peristaltic Pump Feeds concentrated cells to BR-200	~0 psi	1.75 L/min	SS	
P-201	Pump	Peristaltic Pump Feeds media to BR- 201	~0 psi	1.75 L/min	ss	
P-203	Pump	Peristaltic Pump Feeds cells and media to C-201	~0 psi	1.75 L/min	SS	
P-300	Pump	Peristaltic Pump feeds concentrated cells to BR-300	~0 psi	1.75 L/min	SS	
P-301	Pump	Peristaltic Pump feeds differentiated cells to V-301	~0 psi	1.75 L/min	SS	
P-302	Pump	Peristaltic Pump feeds cells to CS-300	~0 psi	1.75 L/min	SS	
P-303	Pump	Peristaltic Pump feeds sorted cells to R-300	~0 psi	1.75 L/min	SS	
P-304	Pump	Peristaltic Pump feeds media to BR-300	~0 psi	1.75 L/min	SS	

Table 10-Pump Equipment Specification Sheet

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Equipment Specification Sheets-Centrifuges								
Equipment Tag	Category	Description	Diameter	RPM	мос			
C-100	Centrifuge	Peristaltic Pump Feeds concentrated cells to BR-200	0.05	1976	SS			
C-200	Centrifuge	Peristaltic Pump Feeds media to BR-201	0.1	1329	SS			
C-201	Centrifuge	Peristaltic Pump Feeds cells and media to C-201	0.13	1159	SS			

Table 11-Pump Equipment Specification Sheet

Equipment Specification Sheets-Support Systems								
Equipment Tag	Category	Description	Temperature	Pressure	MOC			
LFC	Sanitation	Provides clean	25°C	1.01bara	SS			
		air for first 4						
		expansion						
		chambers						
Boiler	Sanitation	Boils water for	121°C	2.05bara				
		steam-in			SS			
		sanitation						
HEPA Filter	Sanitation	Provides clean	25°C	1.01bara	SS			
		air for						
		manufacturing						
		facility						
Cyrostorage	Storage	Keeps NSC	-150°C	2.02bara				
		product cold			SS			
		until shipping						
Refrigerator	Storage	Cools NSCs for	-150°C	2.02bara	cc			
		Cryostorage			33			

Table 12-Pump Equipment Specification Sheet

Equipment Cost Summary:

Generally, biomedical facilities have smaller capacities than standard chemical plants, but the price per area is significantly more expensive due to the need for highly specialized equipment and tight physical constraints on the product.⁷ In particular, the neural stem cell production facility must be consistent with the current good manufacturing practices as outlined by the FDA.^{9,18} Maintenance of the facility is paramount not only to manufacture a quality product, but to ensure the health and safety of the employees on site. A distribution of the fixed capital investments can be seen in Figure 6.



Figure 6-Distribution of Fixed Capital Investment

Figure 6 does not include the capital investment for the boiler unit or the cost of facility construction due to the magnitude of each item. Proper cleanliness/manufacturing of the facility is vital to maintain the integrity of product because a significant portion of the process is open to atmosphere. For example, high efficiency particulate air resistance (HEPA) systems are used throughout the facility in order to keep the air quality consistent with current good manufacturing Page 34 of 136

practices. Multiple HEPA standard air filters are installed in the air lines to ensure that the air does not contaminate the process. However, compared to the construction cost of the facility, the cost of the ventilation system is relatively insignificant. The cost of the air purifiers is estimated at \$14,400 for the entire system.

Similarly, the most important piece of equipment in the process is the bioreactor because it is the source of revenue generation. The reactor takes induced pluripotent stem cells and differentiates them into neural stem cells using a neural induction medium. Tight parameters such as pH, oxygen, and temperature must be stable in order to ensure maximum viability of the stem cells.²³ These process variables are maintained by installing control systems. Furthermore, the reactor needs to be constructed from stainless steel to provide enhanced corrosion and fouling protection. Altogether, the reactor estimate is \$259,600.

Although the product of the bioreactor is what ultimately makes the process economical, the expansion chambers allow the process to generate the volume of stem cells needed to meet market demands. The process starts at 100,000 cells and expands to over three billion cells based on the doubling time of the cells.¹⁶ This process is time consuming and needs to be done as efficiently as possible. However, a single expansion chamber cannot be used because the process cannot be controlled appropriately. Therefore, five expansion chambers are used to slowly grow the process from a volume of 1 mL to 25 L. The first four of these chambers are disposable and range in size from a 35mm cell culture plate to a 3 L plastic vessel. Since the chambers are disposable, it is more economical to replace them after each batch rather than adding downtime to steam treat the chambers. Each year it is estimated that the disposable expansion chambers costs \$7600. However it is not economical to replace the 40L expansion chamber after every batch. Therefore, a permanent vessel is installed to minimize the capital

investment of the project. However, the vessel needs to be steam treated after each batch. Stainless steel is used in place of plastic to provide a more structurally sound and durable vessel. The last expansion chamber is expected to cost \$153,000.

Subsequently, centrifuges provide separation of materials with different densities. For pharmaceutical processes, centrifuges are invaluable because they allow for fine separation between the product and other materials. With regard to neural stem cells production, the unit consists of three centrifuges that vary in size to handle different volumes of fluid. The first centrifuge is a closed system, where the process is put into a vial compatible with the centrifuge. The last two are open systems where the stem cells stick to the outside of the unit while the waste passes through. The cells are then washed to force them into the next step of the process. The cost of the three centrifuges combined is estimated at \$38,700.

The kill tank can be modeled as a vessel even though it reacts with protein coating on viruses to inactivate them. Formaldehyde is bubbled through the vessel using a sparger to ensure the stem cells do not have viruses in solution. The viral inactivation chamber is made of stainless steel for quality and durability and is estimated at \$40,100.

After viral inactivation, the product goes to a cell separation unit that removes undifferentiated stem cells and excess media from the neural stem cells. This unit is estimated at \$76,700 and gives a high purity product.

Lastly, the cells are sent to a vessel that adds dimethylsulfoxide to ensure proper cryopreservation of the neural stems cells.¹⁵ These cells are then refrigerated in a different unit to bring them to cryogenic storage conditions. The combined cost of both of these units is estimated to be \$42,000.
Storage conditions are key to maintaining the cells in cryogenic state and liquid nitrogen is used to keep the cells at these conditions. Storage tanks are utilized to store the cells in the vapor phase of liquid nitrogen.¹⁵ The initial storage tank holds 100,000 induced pluripotent stem cells. The final storage tank holds 365L of liquid nitrogen to preserve the final volume of cells before shipping. Also, a large liquid nitrogen storage tank needs to be present on site to provide fresh liquid nitrogen to account for heat losses in the Dewar. Altogether, there are three liquid nitrogen storage tanks that have been estimated to cost \$101,000.

Steam needs to be generated to sterilize all non-disposable equipment and hot water is needed to maintain a constant temperature in the bioreactor. To do this, a boiler must be installed to produce steam. The estimated cost of the boiler has been estimated to be \$16,800,000. Lastly, there needs to be work done on the process in order for the fluid to be transferred for automated parts of the facility. Eight pumps are placed in locations where material needs to be transferred from one location to another where manual transfer is either impractical or inefficient. Each pump provides the same volumetric flow rate and the same discharge pressure; therefore, the cost of each pump is the same. In total, the pumps are estimated to cost \$122,000.

The total equipment cost is approximately \$18 million. This accounts for units in the process, and all units needed to maintain the integrity of the process as well as utilities.

Fixed Capital Investment Summary:

The most expensive part to the entire facility is the building in which the facility is enclosed. It is also the largest portion of the capital investment. For this particular project, the building must be incompliance with Class 1000 GMP Production Facility protocol to meet standards.⁸ The price of the lab can be estimated based on an average laboratory size with above average cost per gross square foot (gsf). In 2014, the average cost per gross square foot was \$725-834 and facility size ranged from 100,000-200,000gsf. By choosing a 150,000gsf facility that costs \$800/gsf, the facility can be estimated to cost around \$100 million.⁸ The relatively high estimation stems from two key factors: the large space required for the pharmaceutical process and the high standard for cleanliness. In comparison, the cleanliness of the facility is that of a hospital operating room. This increases the cost of the facility considerably.

Moreover, the equipment listed above describes the bare modulus cost. However, there are additional costs to consider that are independent of the cost of production. Turton et. al. generalizes these costs into a "contingencies and fees" category. Included in this category are costs for shipping, installation, and labor that arise after the piece of equipment is built.²⁵ The result of these additional costs is the total installed cost. An 18% rate of contingencies and fees is assumed for this project in accordance with heuristics in Turton et al. The total installed cost is a function of the bare modulus cost as seen in Equation 3.

$$C_{TM} = 1.18C_{BM} \tag{3}$$

Where:

 $C_{TM} = Total Installed Cost$

 $C_{BM} = Bare Modulus Cost$

Stainless steel was chosen for this process because the higher quality metal provides the corrosion resistance needed to keep the process clean while still being relatively inexpensive. Moreover, carbon steel introduces rust into the process over time; therefore, a higher quality metal is installed in piping to keep the process clean. Other exotic metals would perform better, but stainless steel is compatible with the system and the most economical option. Since piping is in contact with the process, stainless steel is used to prevent contamination. The cost is not estimated in this report because not enough information is known regarding the physical layout of the facility, but an upgrade in piping metallurgy needs to be taken into economic consideration when the detailed design phase begins. A summary of the fixed capital investment can be seen in Table 13. The costs included in the table are the total installed cost of each piece of equipment.

Fixed Capital Cost Summary										
Bioreactor	\$	259,600								
Permanent Expansion Chamber	\$	153,400								
Small Centrifuge	\$	12,900								
Medium Centrifuge	\$	15,300								
Big Centrifuge	\$	17,500								
Kill Tank	\$	40,100								
Cryopreparation Tank	\$	17,500								
Liquid Nitrogen Storage Tank	\$	84,300								
Boiler Unit	\$	19,722,000								
Refrigeration Unit	\$	24,780								
Cryo Storage Unit	\$	16,190								
Cell Counter	\$	153,400								
Laminar Flow Cabinet	\$	14,285								
HEPA Filters	\$	19,580								
Warm Water Tank	\$	3,500								
Peristaltic Pumps	\$	122,000								
Cell Sorter	\$	76,700								
Facility Construction	\$	101,422,000								

Table 13-Summary of Fixed Capital Cost

When batch size is small the process needs to be controlled manually, but when the batch is large enough, it can be controlled more precisely through automation, introducing an extra cost in the fixed capital. Similarly, automated parts of the process are controlled via a distributed control system. Specifically, the bioreactor and the final expansion chamber have control systems designed to keep the process at optimal conditions. These conditions are very sensitive and directly affect the viability of the stem cells. Also, error is increased when operators have more interaction with the process. However, the process cannot be fully automated due to the small volume of the early processes. Automating the process would cause some material to not be transferred, causing a significant decrease in the yield of a batch. On the same hand, extreme delicacy is required for the early stages of the batch process, which cannot be accomplished with automation. This becomes favorable when the batch is larger because human control becomes more difficult as the capacity increases.

To estimate the cost of the control systems for the facility, a standard estimation was used and added onto the installed cost of the equipment. This includes the control valve, logic controller, measurement device, and the electrical wiring. Automated controls are critical on the bioreactor, where temperature, oxygen concentration, and pH need to be tightly controlled. Ultimately, significant extra expenses apart from the cost of equipment need to be considered when determining the economic viability of a project.

Safety, Health, and Environmental Considerations:

A unique set of parameters are required for facilities with biological components are required to satisfy OSHA and FDA requirements. Keeping all employees safe and protecting the environment is the most important priority for the facility. Failure to do either of these results is unacceptable in operation. Material that is not sent to the packaging and formulation group needs to be treated properly in order to comply with environmental regulations. From an environmental standpoint, biological waste needs to be handled properly to prevent widespread contamination, but non-biological waste is still treated with the normal precautions. Biological material needs to be inactivated before other disposal methods can be implemented. However, there are some materials that are not compatible with the sewer system regardless of the presence of biological organisms.⁹ Specifically, DMSO is not compatible with the sewer system and needs to be disposed in an organic waste container. Ultimately, an operating procedure detailing the disposal methods for each material in the process is created to prevent incidents in the facility.

Health of employees is maintained through proper operating and safety procedures. Employees handling hazardous chemicals will wear the proper personal protective equipment as outlined by the materials safety and data sheet. Proper handling of biological waste is critical because it presents a unique hazard. All biological waste is inactivated via formaldehyde or steam treatment. A table outlining the material properties present in the facility can be seen below. Most materials do not present high risk hazard and are handled by the process design. A list of key properties of chemicals used in the process is seen in Table 14.

Material	Molecular Weight (g/mol)	Normal Boiling Point (°C)	Flammability Limit (%)	Flash Point (°C)
Induced Pluripotent Stems Cells	-	-	-	-
Neural Stem Cells	-	-	-	-
Expansion Media	-	-	-	-
Reactor Media	-	-	-	-
Trypsin	23.3 kDa	Denatures at 50°C	N/A	N/A
Dimethylsulfoxide	78.13	189	2.6-3.5	50
Formaldehyde	30.03	-19	7.0-71.0	60
Nitrogen	28.02	-195.8	None	N/A
Water for Injection	18.02	0	None	None
EDTA	292.24	614.2	None	93.3

Material	Autoignition Temperature (°C)	Liquid Density (kg/m ³)	Reactivity with Water	Toxicity Limits (ppm)
Induced Pluripotent Stems Cells	-	1125	-	-
Neural Stem Cells	-	1125	-	-
Expansion Media	-	1100	-	-
Reactor Media	-	1100	-	-
Trypsin	N/A	N/A	Soluble	None
Dimethylsulfoxide	N/A	1101	Hygroscopic	7920
Formaldehyde	430	1080	Soluble	42
Nitrogen	None	807	None	N/A
Water for Injection	None	1000	N/A	N/A
EDTA	N/A	860	Soluble	N/A

Table 14-Material Properties Summary

Since very low temperatures are used for cell preservation, proper protective equipment is used to prevent skin damage due to contact from liquid nitrogen. Likewise, DMSO is an incredibly miscible solvent and diffuses through skin upon contact. Other hazards originate from low pressure steam and other cleaning procedures. Low pressure steam generated from the boiler unit causes burns when exposed to open skin. Therefore, procedures are placed around the stream outlet. All materials safety and data sheets are onsite and readily available in an effort to keep all operators safe and knowledgeable. The material safety and data sheets are also included in the Appendix.

Moreover, the process is designed to keep all material inside of the pipes through units that are closed systems. The controls system is designed to keep the process variables at the desired set points, prevent contamination, and maintain the integrity of the facility. However, some parts of the process have to be performed manually. Operators and lab technicians will be trained on how to perform the critical, manual steps in the process as well as emergency shutdown procedures. The NOAA provides an interaction matrix where a hazards associated with chemicals can be simulated. Table 15 shows the results of the interaction matrix.

DIMETHYL SULFOXIDE	Caution Potentially hazardous	DIMETHYL SULFOXIDE		
NITROGEN, REFRIGERATED LIQUID (CRYOGENIC LIQUID)	Caution 🗖	Compatible 🗖	NITROGEN, REFRIGERATED LIQUID (CRYOGENIC LIQUID)	
EDTA	Incompatible Generates gas Generates heat Intense or explosive reaction Polymerization hazard	Compatible 🗖	Compatible 🗖	EDTA
WATER	Caution Polymerization hazard	Compatible 🗖	Compatible 🗖	Caution Corrosive
P = Potentially self polymerizable				

Table 15-Interaction Matrix Depicting Material Compatibility

The results indicate that the only incompatible materials are formaldehyde and EDTA; therefore, these materials need to be separate to prevent incompatible mixing of chemicals. EDTA is used in conjunction with trypsin to remove the cells from the bioreactor to send it to the inactivation tank, which has formaldehyde. However, the reaction does not occur at the concentrations present in the tank. If the concentrations are ever increased, EDTA is removed from the system in order to keep the process safe.

A key part of maintaining the scheduled production rate is the downtime, where operators need to sterilize and clean equipment efficiently to minimize the shutdown time. To ensure minimal downtime, all personnel are trained on FDA cGMP practices and facility procedures prior to starting work.⁴ In addition to the procedures in place to prevent a workplace injury, inherent safety considerations have been included in the design to minimize the potential for an accident. A summary of these inherent safety procedures can be seen in Table 16. The hazards

seen below address designs within the system and outline the action needed to safely deal with each hazard in the facility.

Hazard	Inherent Safety Concept	Action
Extreme storage temperatures	Simplification	Implement concise operating procedure
Biological material	Attentuation	Maximize automation of process/dilute material before treatment
Toxic Cryopreservative	Minimization	Automated Injection Volume
Low Pressure Steam	Moderation	Calculate volume to sterilize equipment
Toxic Waste	Attenuation	Dilute waste before disposal
Atmospheric Contamination	Substitution	Utilize high purity air/water instead of tap water/ambient air
Mechanical	Simplification	Sufficient training in operation of centrifuges/pumps

Table 16-Summary of Inherent Safety Designs

Particularly, pinch hazards arise when maintenance is performed on centrifuges and peristaltic pumps. Therefore, sufficient training in maintenance and repairs is needed to simplify the process. Furthermore, the steam used for cleaning is an open system and extra protective layers are installed to keep operators safe. Hot work gloves are worn at all times to prevent burns to the skin in addition to standard PPE. Implementing inherent safety concepts within the design of the facility minimizes the risk of recordable injuries.

All cleaning procedures are in compliance with FDA standards and provide the company with a safe working environment as well as sterile biological products. Any vessel that contains biological material is steam cleaned for 3 minutes as per FDA guidelines²¹. Any material sent to the sewage system is subject to the constraints set by the city. Regardless of regulations, no living organisms can be sent to the sewer. Therefore, living waste is treated by inactivation via a detergent before it is properly disposed. Table 17 below summarizes the potential consequences associated with failing to prevent hazards present in the process. The consequences are assessed on a "Low-Medium-High" Scale.

Hazard	Equipment Damage	Environmental Compliance	Loss of Life	Disruption of Other Business Units	Legal/PR	Community Impact
Biological Waste in Sewer	-	Medium	-	-	High	High
DMSO Release	-	Low	Low	Low	Low	-
Formadehyde Release	-	-	Low	Medium	Medium	-
Contamination of Process	-	Medium	-	High	-	-
Steam	-	-	Low	-	Medium	-

Table 17-Potential Consequences Summary Table

The first measure that needs to be taken in order to comply with current good manufacturing practices is HEPA quality air within the facility. Air circulating in the facility is required to be medical grade to prevent stem cell contamination and ensure maximum growth. Furthermore, water for injection is required for any application where water comes in contact with cells to prevent contamination as well. The water needs to be as pure as possible since it will be injected into patients with the stems cells. FDA standards require an increase in the purity of water from that of tap water⁴. Moreover, existing safeguards are used in order to minimize the risk within the process. Stainless steel is used in place of carbon steel for all permanent process equipment to prevent corrosion or contamination and plastic or glass is used for all disposable parts. Also, operators are required to wear special personal protective equipment to prevent contact with the cells. This prevents contamination of both the operator as well as the stem cells. All of these materials are the result of an inherently safe facility⁹.

In conclusion, there is no potential for project termination based solely on safety concerns. However, the process needs to be approved in the United States in order for the project to begin. Currently, any treatment involving the usage of stem cells is not permitted by the FDA. Operator training is critical because of the delicacy of the process and must be significant. As a whole, the facility present no significant safety dangers to the employees or surrounding community. Overall, proper safety and environmental procedures will be followed to ensure a safe, efficient work environment.

Other Important Considerations:

In order to meet production needs multiple batches a year are needed. This requires finding the proper amount of batches as well as the scheduling of each batch. To do this, each step in the batch must be timed and the next step ready as soon as the previous is done. The trick in scheduling is finding when to start the next batch. To accomplish this task, Intelligen SuperPro Designer is used. The software allows for input of process and time steps in a batch operation. The chart tools can then find a schedule that allows for the calculated number of batches to be achieved. After building a simulation a Gantt chart was generated using SuperPro. This chart is seen below in Figure 6. From Figure 6, it can be seen that there is an overlap of 169 hours that was determined by the software. This accounts for the time to set up and clean the vessels from the previous batch as well as start the next. There is also a service factor of 0.75 accounted for in this batch timing for FDA inspections and equipment maintenance during the year to comply with cGMPs. Figure 7 below also shows the time steps for each process in the batch with time amounts in hours next to each step. The longest by far is the differentiation but the early expansions take a considerate amount of time of the batch as well.

Another important decision made during evaluation of this project was the choice of expansion medium and differentiation medium. The expansion medium was chosen to be Essential 8 from Thermo Fisher. This media was selected because of the success in growing large numbers of pluripotent stem cells through multiple passages with no karyotype issues found. In conjunction this media does not require extra CO₂ for the buffering system. The differentiation media was chosen to be Gibco PSC Neural Induction Medium. This mediam was selected for availability and functionality of turning pluripotent cells into neural stem cells. This media is widely available for purchase from Thermo Fisher in large amounts that are needed.

This media also has the functionality that is needed to induce PSCs to neural stem cells. This makes it a top choice for induction. Also there is no CO_2 needed for buffering thereby reducing the cost of operation. Another reason is that while the cells are being differentiated they are still able to expand by a factor of 20 with this media over the course of 7 days. This is also a shorter time span than other mediums used for induction. The properties described above make the two choices clear when compared to other forms of media.

			Duration	Start Time	End Time		10 20	30	40 :	50 60	70	80	90	100	110 12	0 13	30 1	40 1	50 160	170	180	190	200	210 220	230	240	250	260	270	280	290	300	310	320	330 day
		Task	(h)	(h)	(h)	Desc	224 448	672	896 112	0 1344 1	568 17	92 201	6 2240	2464	2688 29	12 31	136 33	360 35	84 3808	4032	4256 4	480 47	704 49	28 5152 5	376 56	00 582	4 6048	8 6272	6496	6720	6944	7168	7392	7616 7	840 h
1		+ Complete Recipe	537.68 (0.00	537.68		, . 	<u> </u>							<u> </u>					 T	T				T	T	Ī					<u> </u>			
53		Complete Recipe (Batch #2.)	537.68	368 39	906.07																														
10	5	Complete Recipe (Batch #2)	537.69	736.79	1274.46																														
15	7	Complete Recipe (Batch #4.)	537.68 1	1105.17	1642.85																														
20	0	Complete Recipe (Batch #5)	537.68 1	1473 57	2011.24																														
26	1	E Complete Recipe (Batch #6.)	537.68 1	1841.96	2379.64																														
31	3	Complete Recipe (Batch #7)	537.68	2210.35	2748.03																														
36	5	Complete Recipe (Batch #8.)	537.68	2578 74	3116.42																														
41	7	Complete Recipe (Batch #9)	537.68	2947 13	3484 81																														
46	0	Complete Recipe (Batch #10)	537.68	3315 52	3853.20														_																
52	1	Complete Recipe (Batch #11)	537.68	3683.92	4221 59															-															
57	3	Complete Recipe (Batch #12)	537.68	4052.31	4589.99																		1												
62	5	Complete Recipe (Batch #13)	537.68 4	4420 70	4958 38															'			,												
67	7	Complete Recipe (Batch #14)	537.68 4	4789.09	5326.77																		ŕ												
72	9	Complete Recipe (Batch #15)	537.68	5157.48	569516																		T		<u>_</u>										
78	1	Complete Recipe (Batch #16)	537.68	5525.87	6063 55																							-							
83	3	Complete Recipe (Batch #17)	537.68	5894.27	6431.94																							-							
88	5	Complete Recipe (Batch #18)	537.68 6	6262.66	6800.33																														
•						4																													

Figure 6-Gantt chart for Yearly Production

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Duration	Start Time	End Time		1	2	3		4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23
(h)	(h)	(h)		16	32 48	64	80	96	112 1	28 144	160 17	6 192	208 2	24 240	256 2	72 288	304 3	20 336	352 368	384	400 41	16 432	448 40	54 480	496 5	12 528	544
537.68	0.00	537.68							1	1			1	1	1	1	1	1				1			1	1	
84.75	0.00	84.75	T-Flask																								
363.39	0.00	363.39	Media Prep																								
88.75	84.75	173.50	T-Flask							1																	
0.03	173.25	173.28	Centrifugation																								
86.75	173.28	260.03	Shake Flask																								
0.08	259.78	259.87	Centrifugation																								
86.76	259.87	346.62	Shake Flask																								
0.08	346.62	346.71	Centrifugation																								
15.49	346.62	362.12	Seed Cell Culture																								
0.08	362.12	362.20	C1																								
168.93	362.20	531.13	Cell Culture																						1		
4.00	531.13	535.13	Cell Sorting																								
2.55	535.13	537.68	Freeze-Thaw Module																								

Figure 7-Batch process time sheet

Manufacturing Costs:

Cost of manufacturing can be divided into three categories: direct costs, fixed costs, and general expenses. Direct costs and fixed costs make up operating expenses that change with the rate of production and costs that are independent of changes in production rate, respectively. While general expenses represent the overhead required to perform everyday business function.

Total costs can be estimated as the sum between the three types of cost seen above. The formula for calculating the total costs is given by Turton et. al. is seen in Equation 4^{25}

 $Total Cost = C_{RM} + C_{WT} + C_{UT} + 2.215C_{OL} + 0.19COM + 0.146FCI + depreciation$ (4)

Where:

 $C_{RM} = Cost of Raw Materials$

 $C_{WT} = Cost of Waste Treatment$

 $C_{UT} = Cost of Utilities$

 $C_{OL} = Cost$ of Operating Labor

COM = Cost of Manufacturing

FCI = Fixed Capital Income

By estimating total manufacturing costs, the day to day plant operation costs can be estimated.

Operating labor provides a significant capital and personal investment. Highly skilled operators are required for this process, which makes their services more expensive. Simply having the skills is not enough; operators must be trained specifically to the process. The number of operators can be calculated from Equation 5.

$$N_{OL} = (6.29 + 31.7P^2 + 0.23N_{NP})^{0.5}$$

Where:

 N_{OL} = Number of Operators per shift

P = Number of Processing Steps that Involve Handling of Particulate Solids

 N_{NP} = Number of Nonparticulate Processing steps

The pieces of equipment can be summed and plugged into Equation () to show a need for four operators per shift. For the stem cell manufacturing facility, there are no steps that involve handling of particulate solids. Moreover, pumps and vessel are not included in account for N_{NP} in Equation 5.

Operators for this process need to be extensively trained and highly skilled in order to operate the process efficiently and precisely. Therefore, it is assumed that their average salary is significantly higher than that of an average operator. A salary of \$110,000 per year is used in order to estimate the total cost of operating labor. Lastly, Equation 5 produces a need for four operator on site at all times. Turton et. al²⁵ uses Equation 6 to estimate the total number of operators needed to be hired for the manufacturing facility.

$$Total Hired Operators = 4.5 N_{OL}$$
(6)

Eighteen operators will be hired to run the facility at all times. Therefore, the total cost of operators per year is estimated to be \$2,000,000. As seen in Figure 8, the cost of operating labor is a large percentage of the total manufacturing cost associated with a facility.

The utilities costing was done with information given in the AIChE National Design competition memo. Utilities for the neural stem cell manufacturing consist of boiler feed water,

water for injection, electricity, sewer water, and liquid nitrogen. Boiler feed water is used to provide low pressure steam to sterilize all non-disposable equipment after each batch as well as maintain isothermal conditions in the bioreactor and expansion chamber. Electricity is used to power the boiler unit, peristaltic pumps, centrifuges, and agitators in the bioreactors. Furthermore, water for injection is added to the stem cells in order to store them at the appropriate concentration. Usage of the sewer is used for disposal of waste that is compatible with city sewer systems; other waste must be treated before other disposal method are incorporated. Lastly, liquid nitrogen is used to keep stem cells at cryogenic storage conditions. A summary of the utilities can be seen in Table 18 below.

Utility	Cost Data	Tot	tal Yearly Cost
Electricity (per kWhr)	\$ 5.00	\$	10,678
Sewer (per 1000 gal)	\$ 5.00	\$	5.22
Water for Injection (per 1000 liters)	\$ 1,000	\$	7,600
Liquid Nitrogen (per liter)	\$ 0.25	\$	1,734
Boiler Feed Water (per 1000 kg)	\$ 2.45	\$	0.20

Table 38-Estimated Yearly Utility Costs

All yearly costs are in 2016 dollars and account for all utilities needed to perform the process. Utilities that are used outside of the process are not included due to lack of knowledge. For example, the electricity used for lighting in the facility is not included because not enough information is known presently. Ultimately, utility costs make up a relatively small amount of the total yearly manufacturing costs.

Pharmaceutical manufacturing facilities tend to have more raw materials than standard chemical plant due to the number of additive for the process. Although the induced pluripotent stem cells are relatively cheap, other raw materials are significantly more expensive such as media. Table 19 below summarizes the yearly cost for raw materials.

Raw Material	,	Yearly Cost						
Induced Pluripotent Stem Cells	\$	1,753						
Expansion Media	\$	328,028						
Differentiation Media	\$	4,600,620						
Trypsin w/ EDTA	\$	1,313						
Formaldehyde	\$	33,526						
Dimethylsulfoxide	\$	180,804						

Table 19-Summary of Yearly Raw Material Costs

The other factor taken into consideration in Equation 4 includes estimations based on the cost of manufacture and fixed capital investment. These play a significant role in the manufacturing cost of the process, and have been calculated in previous sections. The Figure below illustrates the proportions of the three types of cost present in the facility.



Figure 8-Distribution of Manufacturing Costs

Even though depreciation is included in the formulation of these costs, Figure 8 does not

include depreciation because it changes independent of escalation.

Economic Analysis:

The driving force behind this project is the economic benefit to the company undertaking the risk of pursuing it. The minimum rate of return given that the project must pass to be profitable is 50%. This represents the minimum amount of return that the project could support before outside investment would be more profitable to pursue.

The following section details the capital estimation of the equipment that could not be directly cost from the AICHE GE spreadsheet. The methods used are consistent with Turton et. al.²⁵ for costing equipment. A list of equipment not costed from GE are seen below in Table 20 with the classification for sizing with key sizing parameter.

Equipment	Classification	Key Parameter
Boiler	Electric Heater	kW
Peristaltic Pumps	Positive displacement pumps	kW

Table 20-4EquipmentCcosted from Turton et. al.

For costing these pieces of equipment, Equation 7 below was used from Appendix A.1 in $Turton^{25}$. This equation calculates the equipment free onboard cost or the manufacture cost without shipping included in 2001. To move from 2001 to 2016 dollars the CEPCI was used. The values for the CEPCI is seen below as well. The CEPCI is used by dividing the current by the past and multiplying this ratio by the C_p amount.

Equipment	K1	K2	K3	CEPCI 2001	CEPCI 2016
Centrifuges	4.7681	0.974	0.024	397	560
Electric Heater	6.9617	-1.48	0.3161		

Table 215-Costing factors for equipment

$$\log_{10}(C_p) = K_1 + K_2 \log_{10}(A) + K_3 (\log_{10}(A))^2$$

(7)

Where:

 $C_p = Bare Bones Cost$

 $K_i = Cost Factor$

A = Key Sizing Parameter

The next step to cost these pieces was to account for the MOC and the pressure requirements of the equipment. Since pressure was low for both pieces of equipment the pressure factor was determined to be 1. The costing equation for total installed cost is seen below for both pieces. Equation 8 is used for centrifuge sizing while Equations 9 and 10 is for the steam boiler. The boiler has special considerations as the steam superheat temperature is critical to cost of manufacture. The bare module cost takes into account the piping, controls, and instrumentation needed for these pieces of equipment as well as the cost of the unit and shipping.

$$C_{bm} = C_p F_{bm} \tag{8}$$

$$C_{bm} = C_p F_{bm} F_{\Delta T} \tag{9}$$

Where:

 $C_{bm} = bare module cost$

 $F_{bm} = costing \ factor \ for \ C_{bm}$

 $F_{\Delta T}$ = degree of overheat factor from equation 9

and

$$F_{\Delta T} = 1 + 0.00184\Delta T - 0.00000335(\Delta T)^2 \tag{10}$$

Where:

 ΔT = amount of super heating in C^o

Using these equations the three centrifuges and boiler unit were costed. The boiler was the most expensive by far due to the nature of a boiler unit and that it needed to be a certain size over required to meet regulations. The final cost used in discounted cash flow analysis was calculated from C_{bm} and is detailed below in Equation 11. This is the total cost with contingencies and fees. This takes into account a buffer of 15% overages on construction or cost of materials and 3% of cost of the project in fees. These numbers were chosen as a representative sample of normal project planning guidelines discussed in Turton et. al²⁵. This gives a total of 18% for the over estimation of the overall cost of each piece of equipment.

$$C_{tm} = 1.18C_{bm} \tag{11}$$

Where:

 $C_{tm} = total cost with contingency and fees$

 $C_{bm} = bare module cost$

The application of these equations to three centrifuges and the steam boiler is seen below in Table 22. The centrifuges are of average to small size for bioprocesses. The first one is a small bench top model that uses disposable vials. The second is a larger bench top model that uses larger half liter to liter centrifuge vials but is still manual. The third centrifuge process the cells produced by the fifth expansion chamber and is made of stainless steel and is not disposable. The MOC for the boiler is carbon steel and the thickness is consistent with the pressure. The C_p, C_{bm}, C_{tm}, and key costing parameter are seen below in Table 22.

Equipment	Ср	Cbm	Ctm	Key Paramter
C-100	\$12,561.00	\$19,721.00	\$32,855.00	Diameter= .2 m
C-200	\$5,967.00	\$9,369.00	\$15,608.00	Diameter= .09 m
C-201	\$16,088.00	\$25,258.00	\$42,079.00	Diameter= .26
Boiler	\$3,949,624.00	\$11,848,871.00	\$19,722,252.00	Duty (kW)=26349

Table 22-Cost of Equipment

Incorporating cost calculated with the costs sourced from the GE data gives a total capital estimate of \$21,260,000; this does not include the building cost. The boiler was the largest individual cost at \$19,720,000. This can be attributed to the cost of all of the controls and piping needed for this system. A summary of equipment costs can be seen in Table 23 below; the items highlighted are costed from Turton et al²⁵. All others were costed using GE data as well as internet data.

Fixed Capital Cost Summary				
Bioreactor	\$	259,600		
Permanent Expansion Chamber	\$	153,400		
Small Centrifuge	\$	12,900		
Medium Centrifuge	\$	15,300		
Big Centrifuge	\$	17,500		
Kill Tank	\$	40,100		
Cryopreparation Tank	\$	17,500		
Liquid Nitrogen Storage Tank	\$	84,300		
Boiler Unit	\$	19,722,000		
Refrigeration Unit	\$	24,780		
Cryo Storage Unit	\$	16,190		
Cell Counter	\$	153,400		
Laminar Flow Cabinet	\$	14,285		
HEPA Filters	\$	19,580		
Warm Water Tank	\$	3,500		
Peristaltic Pumps	\$	122,000		
Cell Sorter	\$	76,700		
Facility Construction	\$	101,422,000		

Table 236-Total Equipment cost

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It is worth noting that no equipment was spared because all equipment is easily ordered with quick lead times on shipping.

Revenue:

The revenue for this project comes from the sale of the stem cells as therapy. An assumption made is that the stem cells can be sold at a bulk price not seen in the current market. This means that the current rate of \$520.00 is much higher than the therapy cells would sell for. Using the current selling price as a reference a final sale price of \$250 per million cells to ensure a reasonably affordable treatment option for patients. Multiplying this by the number of million cells produced a year yields a sales revenue of \$230,000,000.

In order to determine the minimum sale price of the neural stem cells to achieve a rate of return of 50%, a breakeven analysis was performed. The result of this analysis was a sale price of \$150 per million neural stem cells. This confirms that the breakeven analysis is accurate since the minimum rate of return given in the memo is 50%.

DCFROR:

Combining the revenue and utilities cost with capital investment gives a basis to perform a discounted cash flow rate of return analysis (DCFROR). This analysis takes into account the tax rate of 40% normally seen as well as depreciation and escalation of equipment and operating costs. An escalation rate of 0.86% was determined from the Department of the Treasury, using factors published each month. The depreciation rate was done using MACRS for a 5 year depreciation life with a half year assumed in year 1.

The major factors that affect the DCFROR are building cost, and boiler cost. These two costs contribute the most to the initial capital investment that in turn has a large effect on the rate

of return and net present value. These costs were minimized then to increase the viability of the project. To determine how viable the project is, the DCFROR was compared to the given i* of 50%. Any DCFROR below this point was deemed unviable for continuation. The project was determined to have a NPV of \$95,730,000. This gives a DCFROR of 97.8% which is greater than i* therefore the project is viable.

The cash flow for the project is seen below in Table 25. The company situation was assumed to be stand alone as the statement stated that this is being pursued by a small to midsized biomedical company. This means that there is loss forward in the event of negative taxable income.

A return on investment comes when the project starts returning money and no longer paying off debts. There are two commonly reported times, which are undiscounted and discounted payback periods. The only difference is the discount to account for the time value of money. The payback time for the discounted period is found to be 1.27 years while the undiscounted is 0.995 years. These both confirm that the project is economically attractive to pursue as the return on investment comes before the project is over. A summary of this data is shown below in Table 24.

Estimated Payback Periods			
Discounted Payback Period 0.995			
Undiscounted Payback Period	1.270		

Table 24-7Payback Periods

Project Title:	Neural Stem Cell Therapy							
Corporate Financial Situation	Stand Alone							
i*	0.5							
Other Relevant Info	MACRS	*De	preciation					
1=\$1	1/1/2017		1/1/2018	3	1/1/2019	1/1/2020	1/1/2021	1/1/2022
End of Year	0		1	-	2	3	4	5
Sales Revenue	0	\$	232,180,000	\$	5 234,180,000	\$ 236,190,000	\$ 238,220,000	\$ 240,270,000
+Salvage Value	0		0		0	0	0	0
-Royalties	0		0		0	0	0	0
Net Revenue	0		232,180,000		234,180,000	236,190,000	238,220,000	240,270,000
- Expansion Media Cost	0	\$	(331,000)	\$	6 (334,000)	\$ (337,000)	\$ (339,000)	\$ (342,000)
- Reactor Media Cost	0	\$	(4,633,000)	\$	\$ (4,673,000)	\$ (4,713,000)	\$ (4,753,000)	\$ (4,794,000)
- DMSO	0	\$	(182,000)	\$	5 (184,000)	\$ (186,000)	\$ (187,000)	\$ (189,000)
- Formaldehyde	0	\$	(33,800)	\$	\$ (34,100)	\$ (34,400)	\$ (34,700)	\$ (35,000)
- Induced Pluripotent Stem Cell	0	\$	(1,800)	\$	5 (1,800)	\$ (1,800)	\$ (1,800)	\$ (1,800)
- Liquid Nitrogen	0	\$	(6,700)	\$	5 (6,800)	\$ (8,100)	\$ (6,900)	\$ (6,900)
- 0.05% Trypsin w/ EDTA	0	\$	(21,720)	\$	5 (21,900)	\$ (22,140)	\$ (22,280)	\$ (22,470)
- Disposable Expansion Chambers	0	\$	(9,157)	\$	5 (9,239)	\$ (9,322)	\$ (9,393)	\$ (9,475)
-Cell Sorting Kits	0	\$	(616,292)	\$	621,592)	\$ (626,938)	\$ (632,329)	\$ (637,767)
- Boiler Feed Water	0	\$	(10,800)	\$	5 (10,900)	\$ (11,000)	\$ (11,100)	\$ (11,100)
- Operating Labor	0	\$	(2,420,640)	\$	5 (2,441,458)	\$ (2,462,454)	\$ (2,483,631)	\$ (2,504,990)
- Manufacturing Costs	0	\$	(22,690,000)	\$	5 (22,890,000)	\$ (23,080,000)	\$ (23,280,000)	\$ (23,480,000)
- Utilities	0	\$	(19,800)	\$	5 (19,900)	\$ (20,100)	\$ (20,300)	\$ (20,500)
Building Utility Costs	0	\$	(229,961)	\$	6 (231,938)	\$ (233,933)	\$ (235,945)	\$ (237,974)
- Depreciation	0	\$	(6,849,313)	\$	6 (9,401,047)	\$ (6,679,197)	\$ (5,046,087)	\$ (5,046,087)
- Amortization	0		0		0	0	0	 0
- Depletion	0		0)	0	0	0	 0
- Loss Forward	0		0)	0	0	0	0
- Writeoff	0		0)	0	0	0	\$ (89,665,477)
Taxable Income	\$-	\$	194,120,000	\$	5 193,300,000	\$ 197,760,000	\$ 201,160,000	\$ 113,270,000
- Tax at 40%		\$	77,648,000	\$	5 77,320,000	\$ 79,104,000	\$ 80,464,000	\$ 45,308,000
Net Income	\$ -	\$	116,472,000	\$	5 115,980,000	\$ 118,656,000	\$ 120,696,000	\$ 67,962,000
+ Depreciation		\$	6,849,313	\$	9,401,047	\$ 6,679,197	\$ 5,046,087	\$ 5,046,087
+ Writeoff			0)	0	0	0	\$ 89,665,477
-Fixed Capital	\$ (122,687,208)		0		0	0	0	0
Cash Flow	(122,687,208)		123,321,313		125,381,047	125,335,197	125,742,087	162,673,565
Discount Factor (P/F _{i*,n})	1.000		0.667		0.444	0.296	0.198	0.132
Discounted Cash Flow	(122,687,208)		82,214,208		55,724,910	37,136,355	24,837,943	21,422,033
NPV @ i*=	98,648,241							
DCFROR =	99.1%							

Table 258-Cash Flow Table for Neural Stem Cell Project Life

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Sensitivity:

To determine the stability of the project a sensitivity analysis on capital investment, operation cost, and selling price. These parameters have the most variability and have a large impact on DCFROR. The initial capital investment was varied $\pm 40\%$. This variance was chosen as many of the pieces of equipment are highly variable in cost depending on where it is sourced. The operational cost was varied at $\pm 15\%$. This was chosen as many of the materials are sourced from limited vendors with limited stocks. Finally the revenue was varied $\pm 25\%$. This was chosen to be representative of the market as more stem cell therapies are approved so competition will increase.

To graphically represent this analysis tornado charts are used. They show the variation in both the NPV and DCFROR of each situation. Overall this tool is useful to gage the variability in the profitability of the project in a dynamic world. Figures 9 and 10 show the NPV and DCFROR Tornado Charts, respectively.



Figure 9-NPV Tornado Chart Page 61 of 136

The revenue variance has the largest effect on the overall NPV of the project. This is normal as revenue is the source of positive cash flow in the project. The capital expense has an odd shape around the center, because as at higher estimations the building cost dominates the overall expense.



Figure 10-DCFROR Tornado Chart

The DCFROR tornado shows a much larger dependence on capital expense as DCFROR is sensitive to large changes in costs or revenues occurring early in the project. Overall, the minimization and negotiation of operating costs and fixed capital investment is key to keeping the project profitable moving forward into the future.

Conclusions:

Based on the information given in the report it can be concluded that the neural stem cell manufacturing facility is economically attractive. Currently, utilization of stem cells as a type of therapy is not allowed in the United States; the only legal application is biomedical research²⁷.

Legalization of stem cells therapy will inevitably introduce competition into the domestic market. Proper training is critical to the success of the facility, especially for operator training. Although this presents a challenge, the project is economically attractive. Sensitivity analyses and uncertainty calculations produce a rate of return greater than the minimum rate of return. Moreover, due to the nature of the process, downtime plays a significant factor in the production of the facility. In order to satisfy market demand, optimization of downtime is critical.

As with any project, increasing capacity creates an opportunity to capitalize financially. The memo states that the facility is looking to expand into European and Asian markets in the future, but meeting the demands of the United States is a primary objective. As designed, the facility can meet the demands of the United States over the project evaluation lifetime. It has also been concluded that international expansion in the next 10-15 years is going to produce a significant financial opportunity. However, the facility is designed to handle the additional needs of foreign markets. An example of this is the over sizing of the bioreactor. The unit has the ability to handle double the current capacity, while additional expansion can be performed without disturbing the original process.

Recommendations:

Based on the previous information detailed in the report, it is in the best interests of the company to pursue the project. Small scale testing needs to be done to validate the safety and applicability of the project to the Food and Drug Administration. Also, the company needs to wait for legal rulings before significant capital is invested in the project. Once the patents are obtained for the process, the company needs to act quickly to maximize the 20 year life on the patent. Based on the success of the company, an increase in competition is expected once the

therapy is approved. Lastly, significant upfront cost needs to be avoided prior to approval of the project by the court. Based on the company's size, it cannot afford to invest on a project with a significant up front cost. Overall, proceeding with the project is highly recommended; however, detailed design of the facility needs to occur after the therapy is legalized.

Additionally, the manufacturing facility will most likely need to be scaled up to keep up with market demand. This will call for an increase in capacity, whether that comes from modification of the current process or construction of a new facility. Location of the facility also plays a significant factor in the viability of the project. Regardless of the legal status on the project, location provides the company with an economic edge. Research into the most profitable location is highly recommended in order to maximize the viability of cells during shipping. Moreover, legal issues will decrease when the project is expanded into foreign markets. Stem cell therapy is legal is countries across the world and expansion into these particular markets is highly recommended.

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Appendix

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SuperPro Simulation:



Figure 11-Super Pro Simulation Setup

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Material Balances/Energy Balances:

CP-101					
Stream	Mass (g)				
1	0.00				
2	1.10				
5	1.10				
In-Out	0.00				
Stream	Energy (J)				
1	0.06				
2	118.25				
5	118.33				
In-Out	-0.01				

TF-100					
Stream	n Mass (g)				
6	0.0085				
4	13.2000				
7	13.2021				
In-Out	0.0064				
Stream	Energy (J)				
6	0.7382				
4	2100.1200				
7	2100.4555				
In-Out	0.4027				

SF-100					
Stream	Mass (g)				
8	0.102				
10	158.400				
9	0.867				
11	159.297				
In-Out	0.072				
Stream	Energy (J)				
9	8.857				
10	25201.440				
13	21.783				
11	25226.273				
In-Out	5.808				

SF-101					
Stream	Mass (g)				
12	1.230				
13	1900.800				
14	0.867				
16	1901.947				
In-Out	0.950				
Stream	Energy (J)				
12	106.703				
13	302417.280				
14	21.783				
16	302529.396				
In-Out	16.370				

BR-200					
Stream	Mass (g)				
19	22814.00				
20	14.68				
21	2.19				
23	22829.95				
In-Out	0.92				
Stream	Energy (J)				
19	3629707.40				
20	1884.77				
21	55.02				
23	3631569.28				
Heat Lost	62640.00				
Heat added	62640.00				
In-Out	77.91				

BR-300					
Stream	Mass (g)				
27	57563.00				
28	22.40				
29	63.20				
31	57606.49				
In-Out	42.11				
Stream	Energy (J)				
27	9158273.30				
28	1943.20				
29	1587.90				
31	9161569.11				
Heat Lost	-1874880.00				
Heat added	1874880.00				
In-Out	235.29				

R-300					
Stream	Energy (kJ)				
Heat	-12867				
Mass in (g)	25645				
Cp (J/gC)	4.545				
Heat of Fusion (J/g)	333.55				
Cp,s (J/gC)	1.4				
Т (С)	25				
Tf (C)	0				
Final T (C)	-80				
Rate of Cooling (C/min)	1				
Time to cool (min)	117				
Power needed (kW)	-1.83				





Figure 12-Basic Control Scheme of Stem Cell Manufacturing Facility






P-300 BR-300 Feed Pump	BR-300 Differentiation Bioreactor	P-301 BR-300 Transfer out pump	V-300 Viral inactivation) Ce n tank conce	CS-300 Il Sorting and entration system	R-300 Cell freezing unit	R-300
M-300 BR-300 agitator motor	F-300 Vent filter for BR-300	F-301 Vent Filter for V-300	P-302 Transfer Pump For CS- 300	P-303 Transfer pump to R- 300	P-304 Expansion media transfer in pump		35

Figure 63Basic Control Scheme of Manufacturing Facility



Costing Information: (All Total Manufacturing	Costs are in 2016 Dollars)
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		Siz	zed Equipment Cos	ting Summar	у		
Final Expansion Centrifuge			Small Scale Centrifuge			Pre-differentiation Centrifuge	
Ср	\$12,561		Ср	\$5,967		Ср	\$16,088
К1	4.7681		К1	4.7681		К1	4.7681
К2	0.974		К2	0.974		К2	0.974
КЗ	0.024		К3	0.024		КЗ	0.024
A	0.2		A	0.09		А	0.26
Fp	1		Fp	1		Fp	1
Fbm	1.57		Fbm	1.57		Fbm	1.57
Cbm	\$19,721		Cbm	\$9,369		Cbm	\$25,258
Ctm	\$32,855		Ctm	\$15,608		Ctm	\$42,079
Warm Wat	er Tank		Inactivation Tank			Liquid Nitrogen Storage Tank	
Ср	\$2,081		Ср	\$2,251		Ср	\$2,985
К1	3.5565		К1	3.7957		К1	3.4974
К2	0.3776		К2	0.4593		К2	0.4485
КЗ	0.0905		КЗ	0.016		КЗ	0.1074
A	0.0004		A	0.1		А	0.89
Fp	1		Fp	1		Fp	2.2
Fm	1		Fm	3.1		Fm	3.6
Cbm	\$2,081		Cbm	\$17,766		Cbm	\$49,747
Ctm	\$3,468		Ctm	\$29,597		Ctm	\$82,877
			Boiler Capital	Costing			
			Cp°	\$3,949,624			
			К1	6.9617			
			К2	-1.48			
			КЗ	0.3161			
			A	26348			
			CDM	\$11,848,871			
			rum Ctm 2001	3 612 001 CC0			
			Ctm 2001	\$10 777 757			
			0.111 2010	, <i>, , , , , , , , , , , , , , , , , , </i>	J		
l							

Purchased Equipment Costing Summary				
Equipment	Cost	Quantity		
Bioreactor	\$259,600	1		
Permanent Expansion Chamber	\$153,400	1		
Disposable Expansion Chambers	\$9,074	4		
Cryopreparation Tank	\$17,500	1		
Refrigeration Unit	\$24,780	1		
Cryo Storage Unit	\$16,190	1		
Liquid Nitrogen Dewer	\$325	2		
Cell Counter	\$153,400	1		
Laminar Flow Cabinet	\$14,285	2		
HEPA Filters	\$19,580	2		
Cell Sorter	\$76,700	1		



Utilities Summary				
Equipment	Utility Consumed	Price per Year		
Cryogenic Storage	Liquid Nitrogen	\$6,642		
Refrigeration System	Electricity	\$26		
Boiler Unit	Boiler Feed Water	\$4		
Boiler Unit	Electricity	\$1,317		
HEPA Filter	Electricity	\$1,034		
Laminar Flow Cabinet	Electricity	\$5,742		
Agitator	Electricity	\$64		

Materials Cost	
Induced Pluripotent Stem Cells	\$1,820.20
Trypsin w/ EDTA	\$21,531
Expansion Media	\$328,028
Differentiation Media	\$4,600,620
DMSO	\$180,804
Oxygen	\$0

Utilities Consumed:

Utilities Summary				
Equipment	Utility Consumed	Price per Year		
Cryogenic Storage	Liquid Nitrogen	\$6,642		
Refrigeration System	Electricity	\$26		
Boiler Unit	Boiler Feed Water	\$4		
Boiler Unit	Electricity	\$1,317		
HEPA Filter	Electricity	\$1,034		
Laminar Flow Cabinet	Electricity	\$5,742		
Agitator	Electricity	\$64		

Utilities System Capital Investment				
Equipment	Utility Consumed	Amount Consumed		
Cryogenic Storage	Liquid Nitrogen	2562 L		
Refrigeration System	Electricity	529 kW hr		
Boiler Unit	Boiler Feed Water	34455 kg		
Boiler Unit	Electricity	26340 kW hr		
HEPA Filter	Electricity	10335 kW hr		
Laminar Flow Cabinet	Electricity	57421 kW hr		
Agitator	Electricity	1277 kW hr		

Boiler Operational Costing*		
Boiler Feed Water \$/kg	0.00245	
Electricity \$/kW-hr	\$0.05	
Boiler Feed Water \$	\$4.44	
Electricity \$	\$1,317.00	
Total \$	\$1,321.44	
Total/year	\$25,107.41	
*Per Steamout		

Process Summary Table:

Basic Process Information			
Initial Number of Cells	100000		
Initial Batch Size (mL)	1.00		
Initial Concentration(Cells/mL)	100000		
Growth Factor	31163		
Maximum Cell Concentration (Cells/mL)	1.2 x 10 ⁶		
Expansion Batch Size (Cells)	3.14 x 10 ⁹		
Total Expansion Residence Time (hrs)	358.25		
Individual Expansion Residence Time (hrs)	84		
Neural Stem Cells (Cells/Batch)	6.27 x 10 ¹⁰		
Batches per Year	18.3		
Total Neural Stem Cells (Cells/Year)	1.151 x 10 ¹²		

Market Demand	
Dose Size (mL)	0.3
Dosage Concentration (Cells/mL)	3.0 x 10 ⁶
Doses per Patient	16
People Treated per Year	62000
Total Cell Demand (Cells/Year)	1.151 x 10 ¹²

Batch Processing Information		
Manual Error %	3.0	
Service Factor	0.75	
Total Working Days Available	273.75	
Stem Cell Viability %	80	

Expansion Phase Summary Table						
Expansion	Expansion Cells Volume (mL) Residence Time (hrs)		Total ResidenceTime (hrs)			
1	1.20 x 10 ⁶	1	86.0	86.0		
2	1.44 x 10 ⁷	12	84.0	174.0		
3	1.73 x 10 ⁸	144	84.0	260.0		
4	2.07 x 10 ⁹	1728	84.0	346.0		
5	3.14 x 10 ⁹	20736	14.5	360.5		

Expansion Phase Components						
Phase 1	Volume (mL)	Temperature (°C)	Pressure (bar)			
Air (Oxygen)	N/A	20	1			
Media	1	37	1			
Trypsin	0.5	20	1			
Phase 2	Volume (mL)	Temperature (°C)	Pressure (bar)			
Air (Oxygen)	N/A	20	1			
Media	12	37	1			
Trypsin	6	20	1			
Phase 3	Volume (mL)	Temperature (°C)	Pressure (bar)			
Air (Oxygen)	0.14 mL/min	20	1			
Media	144	37	1			
Trypsin	72	20	1			
Phase 4	Volume (mL)	Temperature (°C)	Pressure (bar)			
Air (Oxygen)	173 mL/min	20	1			
Media	1728	37	1			
Trypsin	1000	20	1			
Phase 5	Volume (mL)	Temperature (°C)	Pressure (bar)			
Air (Oxygen)	2.07 L/min	20	1			
Media	20736	37	1			
Trypsin	10000	20	1			

Reactor Summary Table				
Cells In	3.14 x 10 ⁹			
Neural Stem Cells Out	6.27 x 10 ¹⁰			
Conversion	0.8			
Total Volume (L)	200			
Residence Time (hrs)	168			
Glucose Consumption Rate (g/(cell*hr))	2.15 x 10 ⁻¹⁰			
Hot Water Flow Rate (L/hr)	4.87			
Material of Contruction	Stainless Steel			
Other Features:	Low Shear Agitator			
	Pressure Relief Valve			

Reactor Process Variables				
Temperature (°C)	37			
Pressure (bar)	1			
O ₂ Concentration (%)	20			
Liquid Volume (L)	52.725			
рН	7.2			

Sparger Sizing Table	
Total Sparger Contact Area (mm ²)	33.35
Number of Holes	30
Diameter per Hole (mm)	1.190
Diamter of Sparger (m)	0.133

Inactivation Chamber Specification Table				
Diameter (m)	0.233			
Height (m)	0.934			
Liquid Level Fraction	0.625			
Volume of Tank (m ³)	0.1			
Liquid Volume (L)	52.7			
Material of Construction	Stainless Steel			

Inactivation Chamber Process Variables				
Temperature (°C)	37			
Pressure (bar)	1.01			
Formaldehyde Concentration (g/L)	0.2			
Residence Time (hrs)	2.0			
Liquid Density (kg/m ³)	1200			
Gas Density (kg/m³)	1.38			
Formaldehyde Amount (L)	39.25			

Cryostorage Sizing Table				
Product Cryostorage Container	MIDSCI 24K Cryostorage			
Capacity (1 mL vials)	24050			
Liquid N ₂ Capacity(I)	365			
Holding time (days)	52			
Diameter (m)	0.956			
Height (m)	1.11			
Refigerator Storage Volume (m ³)	45.31			
NSC CryoPrep Volume (m ³)	0.21			

Liquid Nitrogen Storage Tank				
Diameter (m)	1.22			
Height (m)	3.05			
Pressure (barg)	13.79			
Volume (liters)	1892.71			
Dewar Volume (liters)	1.00			

BATCH SHEET NO.	
PRODUCT	Main simulation
SPD Case File	H:\Spring Design\Main simulation.spf
Created By	CEAT ITS

OPERATING INSTRUCTIONS

The following procedures/operations must be performed in order.

P-4 (in TFR-101)

CHARGE-1: INSTRUCTIONS FOR MANUFACTURING	DATE	START (h)	END (h)	SIGN	C H E C K
Charge 0.00 kg (or 0.00 L) via Port S-103 to TFR-101.		0.00	0.25		
CHARGE-2: INSTRUCTIONS FOR MANUFACTURING	DATE	START (h)	END (h)	SIGN	C H E C K

Charge 0.00 kg (or 0.00 L) via Port S-106 to TFR-101.					
Comments		0.25	0.50		
REACT-1: INSTRUCTIONS FOR MANUFACTURING	DATE	START (h)	END (h)	SIGN	C H E C K
Allow the mixture to react in TFR-101. The exit temperature should not exceed 37.0 °C and the exit pressure 1.0 bar. Make sure that the final working vessel volume does not exceed the maximum allowable working to vessel volume ratio (90.0 %).		0.50	84.50		
TRANSFER-OUT-1: INSTRUCTIONS FOR MANUFACTURING	DATE	START (h)	END (h)	SIGN	C H E C K
Transfer 0.00 kg (0.00 L) of material out of TFR-101 into TFR-102 at a rate of 600.00 kg/h (604.16 L/h). Make sure that the final working vessel volume does not exceed the maximum allowable working to vessel volume ratio (90.0 %).		84.50	84.75		

P-2 (in BR-102)

CHARGE-1.					С
	DATE	START (h)	END (h)	SIGN	H E C
					K
Charge 20.81 kg (or 20.78 L) via Port Expanision Media					
Feed to BR-102.					
		0.00	0.21		
		0.00	0.51		
Comments					
HEAT-1:		STADT	END		С
	DATE	(h)	(h)	SIGN	н Е
INSTRUCTIONS FOR MANUFACTURING					C K
Heat the contents of BR-102 for 30.98 min to a final					
temperature of 37.00 °C.					
		0.31	0.91		
Comments					
TRANSFER-OUT-1:			END		C
	DATE	(h)	END (h)	SIGN	H E
INSTRUCTIONS FOR MANUFACTURING				~~~~	C
Transfer 20.81 kg (21.01 L) of material out of BR-102 into					Λ
SBR-101 at a rate of 297.09 kg/h (300.00 L/h). Make sure					
that the final working vessel volume does not exceed the					
maximum allowable working to vessel volume ratio (90.0					
%).		0.91	1 23		
		0.51	1.25		
Comments					
CHARGE-2:		START	END		C H
	DATE	(h)	(h)	SIGN	E
					C K

Charge 52.33 kg (or 52.24 L) via Port Diff Media to BR- 102. Comments		362.11	362.4 5		
HEAT-2:	DATE	START (h)	END (h)	SIGN	C H E C K
Heat the contents of BR-102 for 30.99 min to a final temperature of 37.00 °C.		362.45	363.0 5		
TRANSFER-OUT-2: INSTRUCTIONS FOR MANUFACTURING	DATE	START (h)	END (h)	SIGN	C H E C K
Transfer 52.33 kg (52.84 L) of material out of BR-102 into BR-101 at a rate of 600.00 kg/h (605.85 L/h). Make sure that the final working vessel volume does not exceed the maximum allowable working to vessel volume ratio (90.0 %). Comments		363.05	363.3 9		

P-5 (in TFR-102)

CHARGE-1: INSTRUCTIONS FOR MANUFACTURING	DATE	START (h)	END (h)	SIGN	C H E C K
Charge 0.01 kg (or 0.01 L) via Port S-107 to TFR-102.		84.75	85.00		
PULL-IN-1: INSTRUCTIONS FOR MANUFACTURING	DATE	START (h)	END (h)	SIGN	C H E C K
Transfer 0.00 kg (0.00 L) of material into TFR-102 from Source Equipment via at a rate of 600.00 kg/h (604.16 L/h). Make sure that the final working vessel volume does not exceed the maximum allowable working to vessel volume ratio (90.0 %).		85.00	85.25		
REACT-1: INSTRUCTIONS FOR MANUFACTURING	DATE	START (h)	END (h)	SIGN	C H E C K
Allow the mixture to react in TFR-102. The exit temperature should not exceed 37.0 °C and the exit pressure 1.2 bar. Make sure that the final working vessel volume does not exceed the maximum allowable working to vessel volume ratio (90.0 %).		85.25	173.2 5		
TRANSFER-OUT-1: INSTRUCTIONS FOR MANUFACTURING	DATE	START (h)	END (h)	SIGN	C H E C K

Transfer 0.01 kg (0.01 L) of material out of TFR-102 into CF-102 at a rate of 600.00 kg/h (605.17 L/h). Make sure that the final working vessel volume does not exceed the			
maximum allowable working to vessel volume ratio (90.0 %). Comments	173.25	173.5 0	

P-7 (in CF-102)

Repeat the following operations 1 time(s).

CENTRIFUGE-1: INSTRUCTIONS FOR MANUFACTURING	DATE	START (h)	END (h)	SIGN	C H E C K
Centrifuge in CF-102 at a rate of 0.41 L/h for 2.00 min.		173.25	173.2 8		

P-8 (in SFR-101)

0111205.4	1				C
CHARGE-1: INSTRUCTIONS FOR MANUFACTURING	DATE	START (h)	END (h)	SIGN	H E C K
Charge 0.14 kg (or 0.14 L) via Port Ex media to SFR-101.					
Comments		173.28	173.5 3		
PULL-IN-1:					С
INSTRUCTIONS FOR MANUFACTURING	DATE	START (h)	END (h)	SIGN	H E C K
Transfer 0.00 kg (0.00 L) of material into SFR-101 from Source Equipment via at a rate of 600.00 kg/h (592.53 L/h). Make sure that the final working vessel volume does not exceed the maximum allowable working to vessel volume ratio (90.0 %).		173.53	173.7 8		
REACT-1: INSTRUCTIONS FOR MANUFACTURING	DATE	START (h)	END (h)	SIGN	C H E C K
Feed SFR-101 via Port Stream Name from Source Equipment. Allow the mixture to react. The exit temperature should not exceed 37.00 °C and the exit pressure 1.01 bar. Make sure that the vessel volume does not exceed 7.27 % of the total vessel volume. Comments		173.78	259.7 8		
TRANSFER-OUT-1:					C
INSTRUCTIONS FOR MANUFACTURING	DATE	(h)	END (h)	SIGN	H E C K

Transfer 0.14 kg (0.14 L) of material out of SFR-101 into CF-103 at a rate of 600.00 kg/h (605.55 L/h). Make sure that the final working vessel volume does not exceed the			
maximum allowable working to vessel volume ratio (90.0			
%).	259.78	260.0	
		3	1
Comments			
comments			1

P-9 (in CF-103)

Repeat the following operations 1 time(s).

CENTRIFUGE-1: INSTRUCTIONS FOR MANUFACTURING	DATE	START (h)	END (h)	SIGN	C H E C K
Centrifuge in CF-103 at a rate of 1.75 L/h for 5.00 min.		259.78	259.8 6		

P-10 (in SFR-102)

0111205.4	1				C
CHARGE-1: INSTRUCTIONS FOR MANUFACTURING	DATE	START (h)	END (h)	SIGN	H E C K
Charge 1.72 kg (or 1.73 L) via Port Ex media 2 to SFR-102.					
Comments		259.86	260.1 1		
PULL-IN-1:					С
INSTRUCTIONS FOR MANUFACTURING	DATE	START (h)	END (h)	SIGN	H E C K
Transfer 0.00 kg (0.00 L) of material into SFR-102 from Source Equipment via at a rate of 600.00 kg/h (599.75 L/h). Make sure that the final working vessel volume does not exceed the maximum allowable working to vessel volume ratio (90.0 %).		260.11	260.3 6		
REACT-1: INSTRUCTIONS FOR MANUFACTURING	DATE	START (h)	END (h)	SIGN	C H E C K
Feed SFR-102 via Port Stream Name from Source Equipment. Allow the mixture to react. The exit temperature should not exceed 37.00 °C and the exit pressure 1.79 bar. Make sure that the vessel volume does not exceed 43.61 % of the total vessel volume. Comments		260.36	346.3 6		
TRANSFER-OUT-1:					C
INSTRUCTIONS FOR MANUFACTURING	DATE	(h)	END (h)	SIGN	H E C K

Transfer 1.98 kg (1.99 L) of material out of SFR-102 into CF-104 at a rate of 600.00 kg/h (605.61 L/h). Make sure that the final working vessel volume does not exceed the			
maximum allowable working to vessel volume ratio (90.0			
%).	346.36	346.6	
		2	
Comments			

P-11 (in CF-104)

Repeat the following operations 1 time(s).

CENTRIFUGE-1: INSTRUCTIONS FOR MANUFACTURING	DATE	START (h)	END (h)	SIGN	C H E C K
Centrifuge in CF-104 at a rate of 23.98 L/h for 5.00 min.		346.62	346.7 0		

P-1 (in SBR-101)

				1	С
PULL-IN-1:		START	END		н
INSTRUCTIONS FOR MANUEACTURING	DATE	(h)	(h)	SIGN	E
					C K
Transfer 20.81 kg (21.01 L) of material into SBR-101 from Source Equipment via at a rate of 600.00 kg/h (605.85 L/h). Make sure that the final working vessel volume does not exceed the maximum allowable working to vessel volume ratio (50.0 %).		346.62	346.9 0		
PULL-IN-2: INSTRUCTIONS FOR MANUFACTURING	DATE	START (h)	END (h)	SIGN	C H E C K
Transfer 0.09 kg (0.09 L) of material into SBR-101 from Source Equipment via at a rate of 300.00 kg/h (300.52 L/h). Make sure that the final working vessel volume does not exceed the maximum allowable working to vessel volume ratio (50.0 %).		346.90	347.1 5		
REACT-1: INSTRUCTIONS FOR MANUFACTURING	DATE	START (h)	END (h)	SIGN	C H E C K
Feed SBR-101 via Port Stream Name from Source Equipment. Allow the mixture to react. The exit temperature should not exceed 37.00 °C and the exit pressure 1.01 bar. Make sure that the vessel volume does not exceed 49.98 % of the total vessel volume. Comments		347.15	361.6 5		
TRANSFER-OUT-1:	DATE	START (h)	END (h)	SIGN	C H E C K

Transfer 20.92 kg (21.12 L) of material out of SBR-101 into CF-101 at a rate of 100.00 kg/h (100.97 L/h). Make			
exceed the maximum allowable working to vessel volume ratio (90.0 %).	361.65	362.1 1	

P-3 (in CF-101)

Repeat the following operations 1 time(s).

CENTRIFUGE-1: INSTRUCTIONS FOR MANUFACTURING	DATE	START (h)	END (h)	SIGN	C H E C K
Centrifuge in CF-101 at a rate of 84.49 L/h for 5.00 min.		362.11	362.2 0		

P-6 (in BR-101)

					C
PULL-IN-1:	DATE	START (h)	END (h)	SIGN	H E C
Transfer 52.33 kg (52.84 L) of material into BR-101 from Source Equipment via at a rate of 600.00 kg/h (605.85 L/h). Make sure that the final working vessel volume does not exceed the maximum allowable working to vessel volume ratio (90.0 %).		362.20	362.5 3		K
PULL-IN-2: INSTRUCTIONS FOR MANUFACTURING	DATE	START (h)	END (h)	SIGN	C H E C K
Transfer 0.24 kg (0.24 L) of material into BR-101 from Source Equipment via at a rate of 600.00 kg/h (603.02 L/h). Make sure that the final working vessel volume does not exceed the maximum allowable working to vessel volume ratio (50.0 %).		362.53	362.7 8		
REACT-1: INSTRUCTIONS FOR MANUFACTURING	DATE	START (h)	END (h)	SIGN	C H E C K
Feed BR-101 via Port Stream Name from Source Equipment. Allow the mixture to react. The exit temperature should not exceed 37.00 °C and the exit pressure 1.01 bar. Make sure that the vessel volume does not exceed 49.88 % of the total vessel volume.		362.78	530.7 8		
TRANSFER-OUT-1:	DATE	START (h)	END (h)	SIGN	C H E C K

Transfer 52.72 kg (53.21 L) of material out of BR-101 into CS-105 at a rate of 600.00 kg/h (605.59 L/h). Make sure that the final working vessel volume does not exceed the			
maximum allowable working to vessel volume ratio (90.0 %)		531 1	
, , , , , , , , , , , , , , , , , , ,	530.78	2	
Comments			

P-12 (in CS-105)

Repeat the following operations 1 time(s).

CENTRIFUGE-1: INSTRUCTIONS FOR MANUFACTURING	DATE	START (h)	END (h)	SIGN	C H E C K
Centrifuge in CS-105 at a rate of 13.30 L/h for 240.00 min. Comments		531.12	535.1 2		

P-13 (in FT-101)

PULL-IN-1:		START	END		C H
INSTRUCTIONS FOR MANUFACTURING	DATE	(h)	(h)	SIGN	E C K
Transfer 5.29 kg (5.32 L) of material into FT-101 from Source Equipment via at a rate of 600.00 kg/h (603.22 L/h). Make sure that the final working vessel volume does not exceed the maximum allowable working to vessel volume ratio (99.0 %).		535.12	535.3 8		
FREEZE-1: INSTRUCTIONS FOR MANUFACTURING	DATE	START (h)	END (h)	SIGN	C H E C K
Freeze the contents of FT-101 for 117.00 min to a final temperature of -80.00 °C.		535.38	537.4 1		
TRANSFER-OUT-1: INSTRUCTIONS FOR MANUFACTURING	DATE	START (h)	END (h)	SIGN	C H E C K
Transfer 5.29 kg (5.10 L) of material out of FT-101 into Target Equipment at a rate of 600.00 kg/h (578.50 L/h). Make sure that the final working vessel volume does not exceed the maximum allowable working to vessel volume ratio (99.0 %).		537.41	537.6 7		

Key Equations

Material and Energy Balance

$$In - Out + Generation = Accumulation$$
$$Q = mC_p \Delta T$$
$$Q = m\lambda$$

Tanks Sizing

$$V = \frac{\pi}{4} d^2 l$$

Heat Transfer

$$Q = \frac{\Delta T}{\sum R}$$
$$R = \frac{\ln(\frac{R_2}{R_1})}{2\pi R_1 k l}$$

Costing

$$C_p^o = 10^{K_1 + K_2 \log(A) + K_3 (\log(A))^2}$$

$$C_{Bm} = C_p^o (B_1 + B_2 F_m F_p)$$

$$F_p = \frac{\frac{(P+1)D}{2[850 - 0.6(P+1)]} + 0.00315}{0.0063}$$

$$F_T = 1 + 0.00184\Delta T - 0.00000335(\Delta T)^2$$

$$C_{Tm} = 1.18C_{Bm}$$

Boiler

$$Q = mC_p\Delta T + m\lambda$$
$$\dot{m} = \rho \dot{Q}$$

Inactivation Chamber:

$$A_o = 22.3(\frac{V}{.61})\sqrt{\frac{\rho}{\Delta P}}$$

Where:

$$A_o = Total Area of Holes (m2)$$

$\Delta P = Pressure Drop (Pa)$

- V = Velocity of the Fluid (m/s)
- ρ = Density of Fluid (kg/m³)

$$\Delta P = (\frac{4000f(.15)(.35)}{1.1d} - 1)E_k$$

Where:

- f = Friction Factor
- D = Diameter (m)
- E_k = Kinetic Energy (J)
- ΔP = Pressure Drop (Pa)

$$E_k = \frac{810(1.1)\rho V^2}{d^4}$$

Where:

- ρ = Density of Fluid (kg/m³)
- V = Velocity of the Fluid (m/s)
- D = Diameter (m)

 E_k = Kinetic Energy (J)

$$R_e = \frac{\rho V D}{\mu}$$

Where:

- ρ = Density of Fluid (kg/m³)
- V = Velocity of the Fluid (m/s)
- D = Diameter of the Pipe (m)
- μ = Dynamic Viscosity of the Fluid (kg/s/m)

Re = Reynolds Number

Biological Growth

$$N = N_o e^{0.0289t}$$

Where:

N = Number of Cells

 $N_o =$ Initial Number of Cells

t = Residence time (hours)

Material Safety Data Sheet

1. PRODUCT AND COMPANY IDENTIFICATION

Product Name Formaldehyde 37% Solution

Cat No. 9300-1, 9300-5, 9300-55, 9311, 9315, C4320, C4320-5 Synonyms No information available.

Recommended Use Laboratory chemicals

2. HAZARDS IDENTIFICATION

Target Organs Gastrointestinal tract (GI), Central nervous system (CNS), Eyes, Respiratory system, Skin, Optic nerve, Liver, Kidney, spleen, Blood **Potential Health Effects**

Acute Effects

Principle Routes of Exposure

Eyes Causes burns.

Skin Toxic in contact with skin. Causes burns. May produce an allergic reaction. Inhalation Vapor harmful. Toxic by inhalation. Causes burns. Ingestion May be fatal or cause blindness if swallowed. Ingestion may cause gastrointestinal irritation, nausea, vomiting and diarrhea. May cause burns to the digestive tract. Chronic Effects May cause cancer. Tumorigenic effects have been reported in experimental animals.. Experiments have shown reproductive toxicity effects on laboratory animals. May cause adverse liver effects. May cause adverse kidney effects. Repeated contact may cause allergic reactions in very susceptible persons. Danger of very serious irreversible effects. See Section 11 for additional Toxicological information.

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Emergency Telephone Number

Chemtrec US: (800) 424-9300 Chemtrec EU: (202) 483-7616

Emergency Overview

Revision Number 1

Flammable liquid and vapor. Cancer hazard. Poison, may be fatal or cause blindness if swallowed. Cannot be made nonpoisonous.

Vapor harmful. Toxic by inhalation, in contact with skin and if swallowed. Causes burns by all exposure routes. May cause an allergic skin reaction. Danger of very serious irreversible effects. Creation Date 17-Mar-2010 Revision Date 17-Mar-2010

Company

Richard Allan Scientific A Subsidiary of Thermo Fisher Scientific 4481 Campus Drive Kalamazoo, MI 49008 Tel: (800) 522-7270 DANGER!

Appearance Colorless Physical State Liquid odor pungent

Aggravated Medical Conditions Central nervous system disorders. Gastrointestinal tract. Preexisting eye disorders. Skin

disorders.

3. COMPOSITION/INFORMATION ON INGREDIENTS

Haz/Non-haz **Component CAS-No Weight %** Water 7732-18-5 48 - 53 Buffers NA -Formaldehyde 50-00-0 37 - 38 Methyl alcohol 67-56-1 10 - 15

4. FIRST AID MEASURES

Eye Contact Rinse immediately with plenty of water, also under the eyelids, for at least 15 minutes. Immediate medical attention is required.

Skin Contact Wash off immediately with plenty of water for at least 15 minutes. Immediate medical attention is required.

Inhalation Move to fresh air. If breathing is difficult, give oxygen. Do not use mouth-to-mouth resuscitation if victim ingested or inhaled the substance; induce artificial respiration with a respiratory medical device. Immediate medical attention is required.

Ingestion Do not induce vomiting. Call a physician or Poison Control Center immediately.

Notes to Physician Treat symptomatically.

5. FIRE-FIGHTING MEASURES

Flash Point 60°C / 140°F Method No information available. Autoignition Temperature 430°C / 806°F Explosion Limits Upper 73 vol % Lower 7 vol % Suitable Extinguishing Media Use water spray, alcohol-resistant foam, dry chemical or carbon dioxide. Unsuitable Extinguishing Media No information available. Hazardous Combustion Products No information available. Sensitivity to mechanical impact No information available. Sensitivity to static discharge No information available. Specific Hazards Arising from the Chemical Flammable. Risk of ignition. Vapors may form explosive mixtures with air. Vapors may travel to source of ignition and flash back.

Containers may explode when heated.

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Protective Equipment and Precautions for Firefighters

As in any fire, wear self-contained breathing apparatus pressure-demand, MSHA/NIOSH (approved or equivalent) and full protective

gear. Thermal decomposition can lead to release of irritating gases and vapors.

6. ACCIDENTAL RELEASE MEASURES

Personal Precautions Use personal protective equipment. Remove all sources of ignition. Take precautionary measures against static discharges. Do not get in eyes, on skin, or on clothing.

Environmental Precautions Should not be released into the environment.

Methods for Containment and Clean

Up

Remove all sources of ignition. Soak up with inert absorbent material. Take precautionary measures against static discharges. Keep in suitable and closed containers for disposal.

7. HANDLING AND STORAGE

Handling Use only under a chemical fume hood. Use explosion-proof equipment. Wear personal protective equipment. Keep away from open flames, hot surfaces and sources of ignition. Take precautionary measures against static discharges. Do not breathe vapors or spray mist. Do not get in eyes, on skin, or on clothing.

Storage Keep containers tightly closed in a dry, cool and well-ventilated place. Keep away from heat and sources of ignition. Flammables area.

8. EXPOSURE CONTROLS / PERSONAL PROTECTION

Engineering Measures Use only under a chemical fume hood. Use explosion-proof electrical/ventilating/lighting/equipment. Ensure that eyewash stations and safety showers are close to the workstation location.

Exposure Guidelines

Component ACGIH TLV OSHA PEL NIOSH IDLH

Formaldehyde Ceiling: 0.3 ppm (Vacated) TWA: 3 ppm

(Vacated) STEL: 10 ppm (Vacated) Ceiling: 5 ppm TWA: 0.75 ppm STEL: 2 ppm IDLH: 20 ppm TWA: 0.016 ppm Ceiling: 0.1 ppm Methyl alcohol TWA: 200 ppm STEL: 250 ppm Skin (Vacated) TWA: 200 ppm (Vacated) TWA: 260 mg/m3 (Vacated) STEL: 325 mg/m₃ (Vacated) STEL: 250 ppm Skin TWA: 200 ppm TWA: 260 mg/m3 IDLH: 6000 ppm TWA: 200 ppm TWA: 260 ma/m₃ STEL: 250 ppm STEL: 325 mg/m3 Component Quebec Mexico OEL (TWA) Ontario TWAEV Formaldehyde Ceiling: 3 mg/m3 Ceiling: 2 ppm Peak: 3 mg/m₃ Peak: 2 ppm STEL: 1.0 ppm CEV: 1.5 ppm

Page 3 / 10 Revision Date 17-Mar-2010 NFPA Health 3 Flammability 2 Thermo Fisher Scientific - Formaldehyde 37% Solution Instability 0 Physical hazards N/A

Component Quebec Mexico OEL (TWA) Ontario TWAEV Methyl alcohol TWA: 200 ppm TWA: 262 mg/m₃ STEL: 328 mg/m3 STEL: 250 ppm Skin TWA: 200 ppm TWA: 260 mg/m3 STEL: 250 ppm STEL: 310 mg/m3 TWA: 200 ppm TWA: 260 mg/m3 STEL: 325 mg/m3 STEL: 250 ppm Skin NIOSH IDLH: Immediately Dangerous to Life or Health **Personal Protective Equipment** Eye/face Protection Wear appropriate protective eyeglasses or chemical safety goggles as described by OSHA's

eye and face protection regulations in 29 CFR 1910.133 or European Standard EN166 Skin and body protection Wear appropriate protective gloves and clothing to prevent skin exposure Reprint for Protection Follow the OSHA respirator regulations found in 29 CFR 1910.134 or European Standard

Respiratory Protection Follow the OSHA respirator regulations found in 29 CFR 1910.134 or European Standard EN

149. Use a NIOSH/MSHA or European Standard EN 149 approved respirator if exposure limits are exceeded or if irritation or other symptoms are experienced

9. PHYSICAL AND CHEMICAL PROPERTIES

Physical State Liquid

Appearance Colorless odor pungent Odor Threshold No information available. pH 2.8 - 4.0 Vapor Pressure 6768 mmHg @ 20 °C Vapor Density 1.01 Viscosity No information available. Boiling Point/Range 96.1 - 101°C / 205 - 213.8°F Melting Point/Range 0°C / 32°F Decomposition temperature °C No information available. Flash Point 60°C / 140°F Evaporation Rate No information available. Specific Gravity 1.0749 - 1.2020 Solubility Soluble in water log Pow No data available

10. STABILITY AND REACTIVITY

Stability Stable under normal conditions.
Conditions to Avoid Incompatible products. Heat, flames and sparks.
Incompatible Materials Strong oxidizing agents, Strong bases, Acids, Acid anhydrides, Acid chlorides, Metals, Peroxides
Hazardous Decomposition Products Carbon monoxide (CO), Formaldehyde, Methanol Hazardous Polymerization Hazardous polymerization does not occur Hazardous Reactions . None under normal processing.

11. TOXICOLOGICAL INFORMATION

Acute Toxicity

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Product Information No acute toxicity information is available for this product **Component Information** Component LD50 Oral LD50 Dermal LC50 Inhalation Water 90 mL/kg (Rat) Not listed Not listed Formaldehyde 500 mg/kg (Rat) Not listed 0.578 mg/L (Rat) 4 h Methyl alcohol 5628 mg/kg (Rat) 15800 mg/kg (Rabbit) 64000 ppm (Rat) 4 h 83.2 mg/L (Rat) 4 h Irritation Causes burns by all exposure routes **Toxicologically Synergistic** Products No information available. Chronic Toxicity Carcinogenicity The table below indicates whether each agency has listed any ingredient as a carcinogen. Component ACGIH IARC NTP OSHA Mexico Formaldehyde A2 Group 1 Reasonably Anticipated X Not listed ACGIH: (American Conference of Governmental Industrial Hygienists) A1 - Known Human Carcinogen A2 - Suspected Human Carcinogen A3 - Animal Carcinogen ACGIH: (American Conference of Governmental Industrial Hygienists) IARC: (International Agency for Research on Cancer) IARC: (International Agency for Research on Cancer) Group 1 - Carcinogenic to Humans Group 2A - Probably Carcinogenic to Humans Group 2B - Possibly Carcinogenic to Humans NTP: (National Toxicity Program) NTP: (National Toxicity Program) Known - Known Carcinogen Reasonably Anticipated - Reasonably Anticipated to be a Human Carcinogen Sensitization May cause sensitization by skin contact

Mutagenic Effects Mutagenic effects have occurred in humans.

Reproductive Effects Experiments have shown reproductive toxicity effects on laboratory animals.

Developmental Effects Developmental effects have occurred in experimental animals.

Teratogenicity Teratogenic effects have occurred in experimental animals..

Other Adverse Effects Tumorigenic effects have been reported in experimental animals.. See actual entry in RTECS

for complete information.

Endocrine Disruptor Information No information available

12. ECOLOGICAL INFORMATION

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12. ECOLOGICAL INFORMATION

Ecotoxicity

Component Freshwater Algae Freshwater Fish Microtox Water Flea Formaldehyde Not listed Leuciscus idus: LC50 = 15mg/L 96h Not listed EC50 = 20 mg/L 96h EC50 = 2 mg/L 48h Methyl alcohol Not listed Pimephales promelas: LC50> 10000 mg/L 96h EC50 = 39000 mg/L 25 min EC50 = 43000 mg/L 15 min EC50 > 10000 mg/L 5 min EC50 > 10000 mg/L 24h

Persistence and Degradability No information available

Bioaccumulation/ Accumulation No information available Mobility .

Component log Pow Water -1.87 Formaldehyde 0.35 Methyl alcohol -0.74

13. DISPOSAL CONSIDERATIONS

Waste Disposal Methods Chemical waste generators must determine whether a discarded chemical is classified as a

hazardous waste. Chemical waste generators must also consult local, regional, and national hazardous waste regulations to ensure complete and accurate classification **Component RCRA - U Series Wastes RCRA - P Series Wastes** Formaldehyde - 50-00-0 U122 -Methyl alcohol - 67-56-1 U154 -

14. TRANSPORT INFORMATION

DOT UN-No UN1198 Proper Shipping Name FORMALDEHYDE, SOLUTIONS, FLAMMABLE Hazard Class 3 Subsidiary Hazard Class 8 Packing Group III TDG UN-No UN1198 Proper Shipping Name FORMALDEHYDE, SOLUTIONS, FLAMMABLE Hazard Class 3 Subsidiary Hazard Class 8 Packing Group III

14. TRANSPORT INFORMATION

ΙΑΤΑ

UN-No UN1198 Proper Shipping Name FORMALDEHYDE, SOLUTIONS, FLAMMABLE Hazard Class 3 Subsidiary Hazard Class 8 Packing Group III IMDG/IMO UN-No UN1198 Proper Shipping Name FORMALDEHYDE, SOLUTIONS, FLAMMABLE Hazard Class 3 Subsidiary Hazard Class 8 Packing Group III

15. REGULATORY INFORMATION

International Inventories Component TSCA DSL NDSL EINECS ELINCS NLP PICCS ENCS AICS CHINA KECL Water X X - 231-791-2 - X - X X KE-35400 Х Formaldehyde X X - 200-001-8 - X X X X KE-17074 Х Methyl alcohol X X - 200-659-6 - X X X X KE-23193 Х Legend: X - Listed E - Indicates a substance that is the subject of a Section 5(e) Consent order under TSCA. F - Indicates a substance that is the subject of a Section 5(f) Rule under TSCA. N - Indicates a polymeric substance containing no free-radical initiator in its inventory name but is considered to cover the designated polymer made with any free-radical initiator regardless of the amount used. P - Indicates a commenced PMN substance R - Indicates a substance that is the subject of a Section 6 risk management rule under TSCA. S - Indicates a substance that is identified in a proposed or final Significant New Use Rule T - Indicates a substance that is the subject of a Section 4 test rule under TSCA. XU - Indicates a substance exempt from reporting under the Inventory Update Rule, i.e. Partial Updating of the TSCA **Inventory Data Base** Production and Site Reports (40 CFR 710(B). Y1 - Indicates an exempt polymer that has a number-average molecular weight of 1,000 or greater. Y2 - Indicates an exempt polymer that is a polyester and is made only from reactants included in a specified list of low concern reactants that comprises one of the eligibility criteria for the exemption rule. **U.S. Federal Regulations** TSCA 12(b) Not applicable **SARA 313**

Component CAS-No Weight % SARA 313 - Threshold Values % Formaldehyde 50-00-0 37 - 38 0.1 Methyl alcohol 67-56-1 10 - 15 1.0 SARA 311/312 Hazardous Categorization Acute Health Hazard No Chronic Health Hazard No Fire Hazard Yes Sudden Release of Pressure Hazard No Reactive Hazard No **Clean Water Act Component CWA - Hazardous** Substances **CWA - Reportable** Quantities **CWA - Toxic Pollutants CWA - Priority Pollutants** Formaldehyde X 100 lb - -**Clean Air Act Component HAPS Data Class 1 Ozone Depletors Class 2 Ozone Depletors** Formaldehyde X -Methyl alcohol X -OSHA **Component Specifically Regulated Chemicals Highly Hazardous Chemicals** Formaldehyde 0.5 ppm Action Level 0.75 ppm TWA 2 ppm STEL TQ: 1000 lb CERCLA This material, as supplied, contains one or more substances regulated as a hazardous substance under the Comprehensive Environmental Response Compensation and Liability Act (CERCLA) (40 CFR 302) **Component** Hazardous Substances RQs CERCLA EHS RQs Formaldehyde 100 lb 100 lb Methyl alcohol 5000 lb -**California Proposition 65** This product contains the following Proposition 65 chemicals: Component CAS-No California Prop. 65 Prop 65 NSRL Formaldehyde 50-00-0 Carcinogen 40 µg/day State Right-to-Know Component Massachusetts New Jersey Pennsylvania Illinois Rhode Island Formaldehyde X X X X X

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Component Massachusetts New Jersey Pennsylvania Illinois Rhode Island Methyl alcohol X X X X X U.S. Department of Transportation Reportable Quantity (RQ): Y DOT Marine Pollutant N DOT Severe Marine Pollutant N U.S. Department of Homeland Security This product contains the following DHS chemicals: Component DHS Chemical Facility Anti-Terrorism Standard Formaldehyde 11250 lb STQ (solution) Other International Regulations Mexico - Grade Moderate risk, Grade 2

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Canada This product has been classified in accordance with the hazard criteria of the Controlled Products Regulations (CPR) and the MSDS contains all the information required by the CPR. WHMIS Hazard Class B3 Combustible liquid D1B Toxic materials D2A Very toxic materials D2B Toxic materials E Corrosive material

16. OTHER INFORMATION

Prepared By Regulatory Affairs Thermo Fisher Scientific Tel: (412) 490-8929 Creation Date 17-Mar-2010 Print Date 17-Mar-2010 Revision Summary "***", and red text indicates revision

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Disclaimer

The information provided on this Safety Data Sheet is correct to the best of our knowledge, information and belief at the date

of its publication. The information given is designed only as a guide for safe handling, use, processing, storage,

transportation, disposal and release and is not to be considered as a warranty or quality specification. The information

relates only to the specific material designated and may not be valid for such material used in combination with any other

material or in any process, unless specified in the text. End of MSDS

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Safety Data Sheet

GHS - Classification

Life Technologies 5791 Van Allen Way

Precautionary Statements

P210 - Keep away from heat/sparks/open flames/hot surfaces. - No smoking P280 - Wear protective gloves/protective clothing/eye protection/face protection P370 + P378 - In case of fire, use water/water spray/water jet/carbon dioxide/sand/foam/alcohol resistant foam/chemical powder for extinction Health hazards Not Hazardous Signal Word WARNING **Hazard Statements** H227 - Combustible liquid **SECTION 2: Hazards identification** Product code D12345 Company/undertaking identification **Physical hazards** Country specific Emergency Number (if available): SECTION 1: Identification of the substance/mixture and of the company/undertaking GHS Physical Hazard 1 Flammable liquids Product name DMSO (dimethylsulfoxide), anhydrous GHS Physical Hazard Category Number Category 4 For research use only. Not intended for human or animal diagnostic or therapeutic uses. D12345 Identification of the substance or mixture **Revision date** Product name DMSO (dimethylsulfoxide), anhydrous **Page** 1 / 6 24 hour Emergency Response: 866-536-0631 301-431-8585 Outside of the U.S. +1-301-431-8585 31-Jul-2015 +(55)-2139581449 (português) www.lifetechnologies.com Life Technologies 5250 Mainway Drive Burlington, ONT CANADA L7L 6A4 800/263-6236 Product code
PO Box 6482 Carlsbad, CA 92008 +1 760 603 7200 CHEMTREC Brazil (Rio De Janeiro) inhalation Remove to fresh air. If symptoms persist, call a physician. If not breathing, give artificial respiration. Notes to Physician Treat symptomatically. eyes

SECTION 5: Firefighting measures

Target Organ Effects No known effects under normal use conditions.

Suitable extinguishing media Water spray. Carbon dioxide (CO2). Foam. Dry chemical.

Mild eye irritation.

Special protective equipment for firefighters Wear self-contained breathing apparatus and protective suit.

Carcinogenic effects None.

SECTION 6: Accidental release measures

Principle Routes of Exposure Potential Health Effects HMIS

Personal precautions ELIMINATE all ignition sources (no smoking, flares, sparks or flames in immediate area). Use personal protection equipment.

Methods for cleaning up Soak up with inert absorbent material. Sweep up and shovel into suitable containers for disposal. After cleaning, flush away traces with water.

Mutagenic effects None.

Skin

Reproductive toxicity

SECTION 4: First aid measures

None.

Mild skin irritation. Components of the product may be absorbed into the body through the skin.

Skin contact Wash off immediately with plenty of water. If symptoms occur, obtain medical advice.

Sensitization

Eye contact Rinse immediately with plenty of water, also under the eyelids, for at least 15 minutes. Consult a physician if necessary.

No sensitization responses were observed.

Ingestion Never give anything by mouth to an unconscious person. If symptoms persist, call a physician. Do not induce vomiting without medical advice.

Specific effects

Health 1

Chemical Name CAS-No EINECS-No Weight % Reactivity 0 dimethylsulfoxide 67-68-5 D12345 200-664-3 95-100 Revision date Product name DMSO (dimethylsulfoxide), anhydrous Page 2 / 6 We recommend handling all chemicals with caution. 31-Jul-2015

SECTION 3: Composition/information on ingredients

www.lifetechnologies.com Product code Flammability 2

Environmental precautions

Odor No data available

Odor Threshold No data available

Storage

Respiratory protection In case of insufficient ventilation, wear suitable respiratory equipment. Keep away from heat, sparks, flame and other sources of ignition (i.e., pilot lights, electric motors and static electricity). Do not store near combustible materials.

Keep in properly labeled containers.

Chemical Name OSHA PEL OSHA PEL (Ceiling) ACGIH OEL (TWA) ACGIH OEL (STEL)

Hand protection Impervious butyl rubber gloves. Nitrile gloves are not recommended. Some brands of Nitrile gloves have breakthrough times of five minutes.

SECTION 7: Handling and storage

See Section 12 for more information. Eve protection Wear safety glasses with side shields (or goggles). dimethylsulfoxide None Skin and Body Protection Lightweight protective clothing. None None Hygiene measures Handle in accordance with good industrial hygiene and safety practice. None Oxidizing properties No information available Water solubility miscible **Environmental exposure** controls Upper explosion limit 61% - 64% Prevent product from entering drains. Lower explosion limit 2.4% - 2.8% **SECTION 8: Exposure controls/personal protection Engineering measures** Ensure adequate ventilation, especially in confined areas. **SECTION 9: Physical and chemical properties** Handling

General information

Personal Protective Equipment

Avoid contact with skin and eyes. Always wear reccommended Personal Protective Equipment. **Exposure Limits** Form Liquid Personal Protective Equipment requirements are dependent on the user institution's risk assessment and are specific to the risk assessment for each laboratory where this material may be used. Prevent further leakage or spillage if safe to do so. D12345 Appearance No information available Revision date °C 188 - 190 Product name DMSO (dimethylsulfoxide), anhydrous °F 64.4 - 66.2 Page 3 / 6 °F 417.2 - 420.8 Melting point / melting range °C 18 - 19 **Autoignition Temperature** 31-Jul-2015 °C 87 - 89 Boiling point / boiling range www.lifetechnologies.com

°F 370 - 374 flash point Product code °C 214 - 216 °F 188.6 - 192.2 Hazardous decomposition products formed under fire conditions. Sulphur oxides. Carbon oxides. Sensitization No sensitization responses were observed. SECTION 10: Stability and reactivity polymerization Chemical Name LD50 (oral,rat/mouse) LD50 (dermal,rat/rabbit) **SECTION 12: Ecological information** LC50 (inhalation,rat/mouse) Hazardous polymerization does not occur. dimethylsulfoxide Ecotoxicity Contains no substances known to be hazardous to the environment or not degradable in waste water treatment plants. 14500 mg/kg Oral LD50 >40000 mg/kg bw Mobility completely soluble. >5000 mg/l Biodegradation Inherently biodegradable. Bioaccumulation Material does not bioaccumulate. **Principle Routes of Exposure Potential Health Effects Chemical Name Freshwater Algae** Data Water Flea Data Freshwater Fish **Species Data Microtox Data log Pow SECTION 11: Toxicological information** dimethylsulfoxide logPow-2.03 Viscosity eyes Mild eye irritation. Stability **SECTION 13: Disposal considerations** Skin Mild skin irritation. Components of the product may be absorbed into the body through the skin. Dispose of contents/containers in accordance with local regulations. Stable under normal conditions. inhalation

No information available. PH Range Acute Toxicity Ingestion No information available. Materials to avoid D12345 Strong acids. Strong oxidizing agents. Revision date Carcinogenic effects Product name DMSO (dimethylsulfoxide), anhydrous None. Page 4 / 6 6-8 Mutagenic effects 31-Jul-2015 None.

No data available www.lifetechnologies.com

Hazardous decomposition

products

Product code Reproductive toxicity None.

IATA

For research use only. Not intended for human or animal diagnostic or therapeutic uses. **Packing group** None

SECTION 14: Transport information

UN-No none Proper Shipping Name No dangerous good in sense of these transport regulations SECTION 16: Other information

Hazard Class None Subsidiary class Reason for revision SDS sections updated. None

TSCA

Clean Air Act, Section 112 Hazardous Air Pollutants (HAPs) (see 40 CFR 61) This product does not contains HAPs. California Proposition 65

This product does not contain any Proposition 65 chemicals.

US Federal Regulations

WHMIS Hazard Class

B3 - Combustible liquid

SECTION 15: Regulatory information

dimethylsulfoxide
This product has been classified in accordance with the hazard criteria of the Controlled Products Regulations (CPR)
and the
MSDS contains all the information required by the CPR.
US State Regulations
SARA 313
This product is not regulated by SARA.
Component
Chemical Name
Revision date
Massachusetts - RTK
Product name DMSO (dimethylsulfoxide), anhydrous
New Jersey - RTK
Page 5 / 6
Pennsylvania - RTK Illinois - RTK Rhode Island - RTK
31-Jul-2015
dimethylsulfoxide
www.lifetechnologies.com
Product code

"The above information was acquired by diligent search and/or investigation and the recommendations are based on

prudent application of professional judgment. The information shall not be taken as being all inclusive and is to be

used only as a guide. All materials and mixtures may present unknown hazards and should be used with caution.

Since the Company cannot control the actual methods, volumes, or conditions of use, the Company shall not be held

liable for any damages or losses resulting from the handling or from contact with the product as described herein.

THE INFORMATION IN THIS SDS DOES NOT CONSTITUTE A WARRENTY, EXPRESSED OR IMPLIED,

INCLUDING ANY IMPLIED WARRANTY OF MERCHANTABILITY OR FITNESS FOR ANY PARTICULAR PUPOSE" Revision date 31-Jul-2015 Product code Product name DMSO (dimethylsulfoxide), anhydrous www.lifetechnologies.com

End of Safety Data Sheet D12345

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Ethylenediamine Tetraacetic Acid, Disodium Salt

Glycine, N,N'-1,2-ethanediylbis[N-(carboxymethyl)-, disodium salt, dihydrate; Ethylenediaminetetraacetic acid disodium salt dihydrate; Ethylenediamine tetraacetic acid disodiumsalt dihydrate; Disodium edetate dihydrate; Glycine, N,N'-1,2-ethanedlylbis [N-(carboxymethyl), disodium salt, dihydrate; Disodium dihydrogen ethylenediamine-N,N, N',N'-tetraacetate, dihydrate; EDTA, disodium, dihydrate; Ethylene diamine tetraacetic acid, disodium salt; Disodium EDTA, dihydrate; Acetic acid, (ethylenedinitrilo)tetra-, disodium salt, dihydrate Powder.

CHEMTREC: 800.424.9300 Outside US: 703.527.3887

SAFETY DATA SHEET

GHS product identifier Other means of identification Product type Emergency telephone number (with hours of operation)

Section 1. Identification

Chemical name : Glycine, N,N'-1,2-ethanediylbis[N-(carboxymethyl)-, sodium salt, hydrate (1:2:2) Supplier's details : Thermo Fisher Scientific **Pierce Biotechnology** P.O. Box 117 Rockford, IL 61105 **United States** 815.968.0747 or 800.874.3723 7 AM - 5 PM Central Time (GMT -06:00) Ethylenediamine Tetraacetic Acid, Disodium Salt Relevant identified uses of the substance or mixture and uses advised against Not applicable. **Product code** SDS # CAS # **Chemical formula**

0017892 1879480 1890270 1896264 1896649 1336 C10-H14-N2-O8.2Na.2H2-O 6381-92-6

Section 2. Hazards identification

Classification of the Not classified. substance or mixture

Signal word : No signal word.

Hazard statements : No known significant effects or critical hazards.

Precautionary statements

Prevention : Not applicable.

Response : Not applicable.

Storage : Not applicable.

Disposal : Not applicable.

GHS label elements

OSHA/HCS status : While this material is not considered hazardous by the OSHA Hazard Communication Standard (29 CFR 1910.1200), this MSDS contains valuable information critical to the safe handling and proper use of the product. This MSDS should be retained and

available for employees and other users of this product.

Hazards not otherwise

classified

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: Handling and/or processing of this material may generate a dust which can cause

mechanical irritation of the eyes, skin, nose and throat.

Date of issue/Date of revision : 8/23/2013. Date of previous issue : No previous validation. Version : 1 1/10 Ethylenediamine Tetraacetic Acid, Disodium Salt

Section 3. Composition/information on ingredients

Glycine, N,N'-1,2-ethanediylbis[N-(carboxymethyl)-, sodium salt, hydrate (1:2: 2)

98 - 100 6381-92-6

Ingredient name % CAS number

There are no additional ingredients present which, within the current knowledge of the supplier and in the

concentrations applicable, are classified as hazardous to health or the environment and hence require reporting in

this section.

Chemical name : Glycine, N,N'-1,2-ethanediylbis[N-(carboxymethyl)-, sodium salt, hydrate (1:2:2) Other means of

identification

: Glycine, N,N'-1,2-ethanediylbis[N-(carboxymethyl)-, disodium salt, dihydrate; Ethylenediaminetetraacetic acid disodium salt dihydrate; Ethylenediamine tetraacetic acid disodiumsalt dihydrate; Disodium edetate dihydrate; Glycine, N,N'-1,2-ethanedlylbis [N-(carboxymethyl), disodium salt, dihydrate; Disodium dihydrogen ethylenediamine-N,N, N',N'-tetraacetate, dihydrate; EDTA, disodium, dihydrate; Ethylene diamine tetraacetic acid, disodium salt; Disodium EDTA, dihydrate; Acetic acid, (ethylenedinitrilo)tetra-, disodium salt, dihydrate

CAS number : 6381-92-6

Substance/mixture

CAS number/other identifiers

ε,

Occupational exposure limits, if available, are listed in Section 8.

Substance

Any concentration shown as a range is to protect confidentiality or is due to batch variation.

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Wash out mouth with water. Remove victim to fresh air and keep at rest in a position comfortable for breathing. If material has been swallowed and the exposed person is conscious, give small quantities of water to drink. Do not induce vomiting unless directed to do so by medical personnel. Get medical attention if symptoms occur.

Immediately flush eyes with plenty of water, occasionally lifting the upper and lower eyelids. Check for and remove any contact lenses. Get medical attention if irritation occurs.

Flush contaminated skin with plenty of water. Remove contaminated clothing and shoes. Get medical attention if symptoms occur.

Remove victim to fresh air and keep at rest in a position comfortable for breathing. Get medical attention if symptoms occur. In case of inhalation of decomposition products in a fire, symptoms may be delayed. The exposed person may need to be kept under medical surveillance for 48 hours.

Section 4. First aid measures

Eye contact Skin contact Inhalation Ingestion :

2

Description of necessary first aid measures

Most important symptoms/effects, acute and delayed

Inhalation : Exposure to airborne concentrations above statutory or recommended exposure limits may cause irritation of the nose, throat and lungs. Exposure to decomposition products may cause a health hazard. Serious effects may be delayed following exposure.

Ingestion : No known significant effects or critical hazards.

Skin contact : No known significant effects or critical hazards.

Exposure to airborne concentrations above statutory or recommended exposure limits may cause irritation of the eyes.

Eye contact :

Over-exposure signs/symptoms

Skin contact

Inhalation Adverse symptoms may include the following:

respiratory tract irritation coughing

No specific data.

Eye contact : Adverse symptoms may include the following: irritation redness Potential acute health effects Date of issue/Date of revision : 8/23/2013. Date of previous issue : No previous validation. Version : 1 2/10 Ethylenediamine Tetraacetic Acid, Disodium Salt

Section 4. First aid measures

Protection of first-aiders : No action shall be taken involving any personal risk or without suitable training. Notes to physician : In case of inhalation of decomposition products in a fire, symptoms may be delayed. The exposed person may need to be kept under medical surveillance for 48 hours. Specific treatments : No specific treatment.

Ingestion : No specific data.

See toxicological information (Section 11)

Indication of immediate medical attention and special treatment needed, if necessary

Section 5. Fire-fighting measures

Promptly isolate the scene by removing all persons from the vicinity of the incident if there is a fire. No action shall be taken involving any personal risk or without suitable

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training. Hazardous thermal decomposition products Specific hazards arising

from the chemical Decomposition products may include the following materials: carbon dioxide carbon monoxide nitrogen oxides metal oxide/oxides No specific fire or explosion hazard. Fire-fighters should wear appropriate protective equipment and self-contained breathing apparatus (SCBA) with a full face-piece operated in positive pressure mode. **Special protective** equipment for fire-fighters Use an extinguishing agent suitable for the surrounding fire. **Extinguishing media** Remark : May be combustible at high temperature. None known. Suitable extinguishing

media

. Unsuitable extinguishing media

Special protective actions for fire-fighters

Section 6. Accidental release measures

Environmental precautions

Personal precautions, protective equipment and emergency procedures

Move containers from spill area. Approach release from upwind. Prevent entry into sewers, water courses, basements or confined areas. Vacuum or sweep up material and place in a designated, labeled waste container. Avoid creating dusty conditions and prevent wind dispersal. Dispose of via a licensed waste disposal contractor. Note: see Section 1 for emergency contact information and Section 13 for waste disposal.

: No action shall be taken involving any personal risk or without suitable training. Evacuate surrounding areas. Keep unnecessary and unprotected personnel from entering. Do not touch or walk through spilled material. Avoid breathing dust. Put on appropriate personal protective equipment.

Avoid dispersal of spilled material and runoff and contact with soil, waterways, drains and sewers. Inform the relevant authorities if the product has caused environmental pollution (sewers, waterways, soil or air).

Large spill :

Move containers from spill area. Vacuum or sweep up material and place in a designated, labeled waste container. Dispose of via a licensed waste disposal contractor. Small spill :

Methods and materials for containment and cleaning up

For non-emergency personnel

For emergency responders : If specialised clothing is required to deal with the spillage, take note of any information in

Section 8 on suitable and unsuitable materials. See also the information in "For nonemergency

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personnel". Date of issue/Date of revision : 8/23/2013. Date of previous issue : No previous validation. Version : 1 3/10 Ethylenediamine Tetraacetic Acid, Disodium Salt

Section 7. Handling and storage

Advice on general occupational hygiene Conditions for safe storage, including any incompatibilities

incompatibilities

Eating, drinking and smoking should be prohibited in areas where this material is handled, stored and processed. Workers should wash hands and face before eating, drinking and smoking. Remove contaminated clothing and protective equipment before entering eating areas. See also Section 8 for additional information on hygiene measures. Store in accordance with local regulations. Store in original container protected from direct sunlight in a dry, cool and well-ventilated area, away from incompatible materials (see Section 10) and food and drink. Keep container tightly closed and sealed until ready for use. Containers that have been opened must be carefully resealed and kept upright to prevent leakage. Do not store in unlabeled containers. Use appropriate containment to avoid environmental contamination.

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Protective measures : Put on appropriate personal protective equipment (see Section 8). Avoid breathing dust.

Precautions for safe handling None.

Section 8. Exposure controls/personal protection

Hand protection

Use a properly fitted, particulate filter respirator complying with an approved standard if a risk assessment indicates this is necessary. Respirator selection must be based on known or anticipated exposure levels, the hazards of the product and the safe working limits of the selected respirator.

Chemical-resistant, impervious gloves complying with an approved standard should be worn at all times when handling chemical products if a risk assessment indicates this is necessary.

Safety eyewear complying with an approved standard should be used when a risk assessment indicates this is necessary to avoid exposure to liquid splashes, mists, gases or dusts. If contact is possible, the following protection should be worn, unless the assessment indicates a higher degree of protection: safety glasses with side-shields. If operating conditions cause high dust concentrations to be produced, use dust goggles. Eye/face protection

Respiratory protection :

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Body protection Personal protective equipment for the body should be selected based on the task being performed and the risks involved and should be approved by a specialist before handling this product.

Environmental exposure

controls

: Emissions from ventilation or work process equipment should be checked to ensure they comply with the requirements of environmental protection legislation. In some cases, fume scrubbers, filters or engineering modifications to the process equipment will be necessary to reduce emissions to acceptable levels.

Appropriate engineering

controls

: Use only with adequate ventilation. If user operations generate dust, fumes, gas, vapor or mist, use process enclosures, local exhaust ventilation or other engineering controls to

keep worker exposure to airborne contaminants below any recommended or statutory limits.

Wash hands, forearms and face thoroughly after handling chemical products, before eating, smoking and using the lavatory and at the end of the working period. Appropriate techniques should be used to remove potentially contaminated clothing. Wash contaminated clothing before reusing. Ensure that eyewash stations and safety showers are close to the workstation location.

Hygiene measures : Control parameters Individual protection measures Occupational exposure limits Skin protection

Other skin protection : Appropriate footwear and any additional skin protection measures should be selected based on the task being performed and the risks involved and should be approved by a specialist before handling this product.

Date of issue/Date of revision : 8/23/2013. Date of previous issue : No previous validation. Version : 1 4/10 Ethylenediamine Tetraacetic Acid, Disodium Salt

Section 9. Physical and chemical properties

Physical state Melting point Vapor pressure **Relative density** Vapor density Solubility Solid. [Crystalline powder.] 248°C (478.4°F) Not available. Not available. Not available. Soluble in the following materials: cold water and hot water. Odor Odorless. рΗ Color White. **Evaporation rate** Not available. **Auto-ignition temperature** Flash point Not available. Closed cup: >100°C (>212°F) Not available. 4 to 6 [Conc. (% w/w): 5%] Viscosity Not available. Odor threshold Not available. Partition coefficient: noctanol/ water 2 2 2 ż ÷

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Appearance

Boiling point : Not available.

Flammability (solid, gas) : Slightly flammable in the presence of the following materials or conditions: heat. Non-flammable in the presence of the following materials or conditions: shocks and mechanical impacts.

May be combustible at high temperature. Lower and upper explosive

(flammable) limits

: Not available. Burning rate : Not available. Burning time : Not available. SADT : Not available. Decomposition temperature : >240°C (>464°F) Solubility in water : 37.22 g/l

Section 10. Stability and reactivity

Hazardous decomposition

products Conditions to avoid No specific data. Under normal conditions of storage and use, hazardous decomposition products should not be produced. Chemical stability The product is stable. No specific data.

: Incompatible materials : Possibility of hazardous reactions

: Under normal conditions of storage and use, hazardous reactions will not occur. Reactivity : No specific test data related to reactivity available for this product or its ingredients.

Section 11. Toxicological information

Acute toxicity Information on toxicological effects Date of issue/Date of revision : 8/23/2013. Date of previous issue : No previous validation. Version : 1 5/10 Ethylenediamine Tetraacetic Acid, Disodium Salt

Section 11. Toxicological information

Glycine, N,N'-1 2-ethanedivlbis[N-(carboxymethyl)-, sodium salt, hydrate (1:2:2) LD50 Oral Rat >2000 mg/kg -Product/ingredient name Result Species Dose Exposure Carcinogenicity Not available. **Mutagenicity** Not available. **Teratogenicity** Not available. **Reproductive toxicity** Not available. Irritation/Corrosion Not available. Sensitization Not available.

Information on the likely

routes of exposure

Inhalation : Exposure to airborne concentrations above statutory or recommended exposure limits may cause irritation of the nose, throat and lungs. Exposure to decomposition products may cause a health hazard. Serious effects may be delayed following exposure. Ingestion : No known significant effects or critical hazards. Skin contact : No known significant effects or critical hazards. Exposure to airborne concentrations above statutory or recommended exposure limits may cause irritation of the eyes. Eye contact : Symptoms related to the physical, chemical and toxicological characteristics Skin contact Ingestion Inhalation Adverse symptoms may include the following: respiratory tract irritation coughing No specific data. No specific data. Eye contact : Adverse symptoms may include the following: irritation redness Specific target organ toxicity (single exposure) Specific target organ toxicity (repeated exposure) Not available. Not available. **Aspiration hazard** Not available. : Routes of entry anticipated: Oral, Inhalation. Potential acute health effects Classification Glycine, N,N'-1, 2-ethanedivlbis[N-(carboxymethyl)-, sodium salt, hydrate (1:2:2) None. - -Product/ingredient name OSHA IARC NTP Date of issue/Date of revision: 8/23/2013. Date of previous issue : No previous validation. Version : 1 6/10 Ethylenediamine Tetraacetic Acid, Disodium Salt Section 11. Toxicological information Not available. Conclusion/Summary : Exposure can cause stomach pains, vomiting and diarrhea. General: Repeated or prolonged inhalation of dust may lead to chronic respiratory irritation. Carcinogenicity : No known significant effects or critical hazards. Mutagenicity : No known significant effects or critical hazards. Teratogenicity : No known significant effects or critical hazards. Developmental effects : No known significant effects or critical hazards. Fertility effects : No known significant effects or critical hazards. Potential chronic health effects

Delayed and immediate effects and also chronic effects from short and long term exposure Numerical measures of toxicity Not available. Acute toxicity estimates Potential immediate effects : Not available. Short term exposure Potential delayed effects : Not available. Potential immediate effects : Not available. Long term exposure Potential delayed effects : Not available.

Section 12. Ecological information

Bioaccumulative potential Other adverse effects : No known significant effects or critical hazards. Not available. Toxicity Not available. Persistence and degradability Soil/water partition coefficient (Koc) : Not available. Mobility in soil Not available.

Section 13. Disposal considerations

The generation of waste should be avoided or minimized wherever possible. Disposal of this product, solutions and any by-products should at all times comply with the requirements of environmental protection and waste disposal legislation and any regional local authority requirements. Dispose of surplus and non-recyclable products via a licensed waste disposal contractor. Waste should not be disposed of untreated to the sewer unless fully compliant with the requirements of all authorities with jurisdiction. Waste packaging should be recycled. Incineration or landfill should only be considered when recycling is not feasible. This material and its container must be disposed of in a safe way. Empty containers or liners may retain some product residues. Avoid dispersal **Disposal methods :**

Date of issue/Date of revision : 8/23/2013. **Date of previous issue :** No previous validation. **Version :** 1 7/10 Ethylenediamine Tetraacetic Acid, Disodium Salt

Section 13. Disposal considerations

of spilled material and runoff and contact with soil, waterways, drains and sewers.

Section 14. Transport information

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-
-
Not regulated.
-
-
Not regulated.
-
-
DOT Classification IATA
UN number
UN proper
shipping name
Transport
hazard class(es)
Packing group
Additional
information
Environmental
```

hazards Special precautions for user Transport in bulk according to Annex II of MARPOL 73/78 and the IBC Code No. No.

Transport within user's premises: always transport in closed containers that are upright and secure. Ensure that persons transporting the product know what to do in the event of an accident or spillage. : Not available.

Section 15. Regulatory information

U.S. Federal regulations : Clean Air Act Section 112 (b) Hazardous Air **Pollutants (HAPs)** : Not listed **Clean Air Act Section 602 Class I Substances** : Not listed **Clean Air Act Section 602 Class II Substances** : Not listed **DEA List I Chemicals** (Precursor Chemicals) : Not listed **DEA List II Chemicals** (Essential Chemicals) : Not listed State regulations TSCA 8(a) CDR Exempt/Partial exemption: Not determined United States inventory (TSCA 8b): This material is listed or exempted. SARA 302/304 SARA 304 RQ : Not applicable. No products were found. **Composition/information on ingredients** SARA 311/312 Classification : Not applicable. No products were found. **Composition/information on ingredients** Date of issue/Date of revision : 8/23/2013. Date of previous issue : No previous validation. Version : 1 8/10 Ethylenediamine Tetraacetic Acid, Disodium Salt

Section 15. Regulatory information

Massachusetts : This material is not listed. New York : This material is not listed. New Jersey : This material is not listed. Pennsylvania : This material is not listed. Canada inventory : This material is listed or exempted. Australia inventory (AICS): This material is listed or exempted. China inventory (IECSC): This material is listed or exempted. Japan inventory: This material is listed or exempted. Korea inventory: This material is listed or exempted. Malaysia Inventory (EHS Register): Not determined. New Zealand Inventory of Chemicals (NZIoC): This material is listed or exempted. Taiwan inventory (CSNN): Not determined. International regulations International lists : **Chemical Weapons Convention List Schedule I Chemicals** : Not listed **Chemical Weapons Convention List Schedule II Chemicals** : Not listed **Chemical Weapons Convention List Schedule III Chemicals** : Not listed

Section 16. Other information

12/12/2013. **History** Date of printing : : 8/23/2013. Hazardous Material Information System (U.S.A.) 0 0 0 0 0 0 National Fire Protection Association (U.S.A.) Health Special Instability/Reactivity Flammability Health Flammability Caution: HMIS® ratings are based on a 0-4 rating scale, with 0 representing minimal hazards or risks, and 4 representing significant hazards or risks Although HMIS® ratings are not required on SDSs under 29 CFR 1910. 1200, the preparer may choose to provide them. HMIS® ratings are to be used with a fully implemented **HMIS**® program. HMIS® is a registered mark of the National Paint & Coatings Association (NPCA). HMIS® materials may be purchased exclusively from J. J. Keller (800) 327-6868. The customer is responsible for determining the PPE code for this material. Reprinted with permission from NFPA 704-2001, Identification of the Hazards of Materials for Emergency Response Copyright ©1997, National Fire Protection Association, Quincy, MA 02269. This reprinted material is not the complete and official position of the National Fire Protection Association, on the referenced subject which is represented only by the standard in its entirety. Copyright ©2001, National Fire Protection Association, Quincy, MA 02269. This warning system is intended to be interpreted and applied only by properly trained individuals to identify fire, health and reactivity hazards of chemicals. The user is referred to certain limited number of chemicals with recommended classifications in NFPA 49 and NFPA 325, which would be used as a guideline only. Whether the chemicals are classified by NFPA or not. anyone using the 704 systems to classify chemicals does so at their own risk. Page 123 of 136

Physical hazards

Chronic Health Hazard Date of issue/Date of revision : 8/23/2013. Date of previous issue : No previous validation. Version : 1 9/10 Ethylenediamine Tetraacetic Acid, Disodium Salt

Section 16. Other information

Date of issue/Date of

revision Version

To the best of our knowledge, the information contained herein is accurate. However, neither the above-named

supplier, nor any of its subsidiaries, assumes any liability whatsoever for the accuracy or completeness of the

information contained herein.

Final determination of suitability of any material is the sole responsibility of the user. All materials may present

unknown hazards and should be used with caution. Although certain hazards are described herein, we cannot

guarantee that these are the only hazards that exist.

Notice to reader

Date of previous issue

Indicates information that has changed from previously issued version.

References : Not available.

Key to abbreviations : ATE = Acute Toxicity Estimate

BCF = Bioconcentration Factor

GHS = Globally Harmonized System of Classification and Labelling of Chemicals

IATA = International Air Transport Association

IBC = Intermediate Bulk Container

IMDG = International Maritime Dangerous Goods

LogPow = logarithm of the octanol/water partition coefficient

MARPOL 73/78 = International Convention for the Prevention of Pollution From Ships,

1973 as modified by the Protocol of 1978. ("Marpol" = marine pollution)

UN = United Nations

No previous validation.

1 Prepared by : SDS Specialist

Date of issue/Date of revision : 8/23/2013. Date of previous issue : No previous validation. Version : 1 10/10

MATERIAL SAFETY DATA SHEET - "LIQUID NITROGEN" ROC Group of Companies Page 1 of 6 **1. PRODUCT AND COMPANY IDENTIFICATION** PRODUCT NAME: Nitrogen, refrigerated liquid CHEMICAL NAME: Nitrogen **CHEMICAL FAMILY**: Inert gas SYNONYMS: Cryogenic Liquid Nitrogen, Liquid Nitrogen, LIN **CHEMICAL** FORMULA: N2 USE: Medical purposes, Inerting, Safe storage of Food, Concrete strengthening, etc., NAME AND ADDRESS: Refrigeration & Oxygen Co. **Corporate Office** Area No 1, Block 21 C, Central Slaughter House Street Shuwaikh Industrial Area Kuwait. WEB ADDRESS: www.rockuwait.com; E-mail: info@rocq8.com TELEPHONE: (+965) 844 844 2. HAZARDS IDENTIFICATION **EMERGENCY OVERVIEW:** WARNING! Extremely cold liquid and gas under pressure.

Can cause rapid suffocation.

Can cause severe frostbite.

POTENTIAL HEALTH EFFECTS INFORMATION:

ROUTES OF EXPOSURE:

INHALATION: Simple Asphyxiant.

Nontoxic, but may cause suffocation by displacing the oxygen in air. Exposure to oxygendeficient

atmosphere (<19.5%) may cause dizziness, drowsiness, nausea, vomiting, excess salivation, diminished mental alertness, loss of consciousness and death. Exposure to atmospheres containing

8% to 10% or less oxygen will bring about unconsciousness without warning and so quickly that the individuals cannot help or protect themselves. Lack of sufficient oxygen may cause serious injury or death.

EYE CONTACT: Tissue freezing and severe cryogenic burns of eyes

SKIN CONTACT: Tissue freezing and severe cryogenic burns of skin

SKIN ABSORPTION: Not applicable

INGESTION: Not applicable

CHRONIC EFFECTS: None established

MEDICAL CONDITIONS AGGRAVATED BY OVEREXPOSURE: None

OTHER EFFECTS OF OVEREXPOSURE: None

CARCINOGENICITY: Not listed by NTP, OSHA, or IARC.

POTENTIAL ENVIRONMENTAL EFFECTS: No adverse ecological effects are expected.

3. COMPOSITION/INFORMATION ON INGREDIENTS

MATERIAL SAFETY DATA SHEET - "LIQUID NITROGEN"

ROC Group of Companies Page 2 of 6

INGREDIENT NAME: Nitrogen

PERCENTAGE >99%

CAS NUMBER 7727-37-9

4. FIRST AID MEASURES

FIRST AID PROCEDURES:

INHALATION: Persons suffering from lack of oxygen should be removed to fresh air. If victim is

not breathing, give artificial respiration. If breathing is difficult, give oxygen. Obtain prompt medical attention.

EYE CONTACT: In case of splash contamination, immediately flush eyes with water for at least 15 minutes. See a physician, preferably an ophthalmologist.

SKIN CONTACT: Remove any clothing that may restrict circulation to frozen area. Do not rub frozen parts as tissue damage may result. As soon as practical, place the affected area in a warm water bath which has a temperature not exceeding 105 °F (40°C). Never use dry heat. In case of massive exposure, remove clothing while showering with warm water. Call a physician as soon as

possible.

Frozen tissue is painless and appears waxy with a possible yellow color. It will become swollen, painful, and prone to infection when thawed. If the frozen part of the body has been thawed by the

time medical attention has been obtained, cover the area with dry sterile dressing with a large

bulky protective covering.

INGESTION: Not applicable

NOTES TO PHYSICIAN: None

5. FIREFIGHTING MEASURES

FLAMMABLE PROPERTIES: Nonflammable and does not support combustion.

EXTINGUISHING MEDIA: Use extinguishing media appropriate for the surrounding fire. **PROTECTION OF FIREFIGHTERS:**

SPECIFIC HAZARDS ARISING FROM THE CHEMICAL: When spilled the liquid will vaporize

rapidly forming an oxygen-deficient vapor cloud. Evacuate this vapor cloud area. Visibility may be

obscured in the vapor cloud. Pressure in a container can build up due to heat and it may rupture if pressure relief devices should fail to function. Contact with cold liquid or gas may cause frostbite.

PROTECTIVE EQUIPMENT AND PRECAUTIONS FOR FIREFIGHTERS: Simple asphyxiant.

If possible, remove containers from fire area or cool with water. Do not direct water spray at the container vent. Self contained breathing apparatus may be required for rescue workers. Evacuate this area.

SENSITIVITY TO STATIC DISCHARGE: Not Applicable

SENSITIVITY TO MECHANICAL IMPACT: None

6. ACCIDENTAL RELEASE MEASURES

MATERIAL SAFETY DATA SHEET - "LIQUID NITROGEN"

ROC Group of Companies Page 3 of 6

PERSONAL PRECAUTIONS: Use personal protection recommended in Section 8. Evacuate all personnel from the affected area. Ventilate area or remove containers to a well ventilated location.

To increase rate of vaporization, spray large amounts of water onto the spill from an upwind position. If leaking from container or its valve, contact your supplier.

ENVIRONMENTAL PRECAUTIONS: Not applicable.

METHODS FOR CONTAINMENT: Shut off source if possible without risk.

METHODS FOR CLEAN-UP: Not applicable.

OTHER INFORMATION: None.

7. HANDLING AND STORAGE

HANDLING: Never allow any unprotected part of the body to touch un-insulated pipes or vessels

that contain cryogenic fluids. The extremely cold metal will cause the flesh to stick fast and tear when one attempts to withdraw from it.

Use a suitable four-wheel hand truck for container movement. Cryogenic containers shall be handled and stored in an upright position. Do not drop or roll containers on their sides. If user experiences any difficulty operating container valve discontinue use and contact supplier. For additional precautions see Section 16, Other Information.

STORAGE: Store and use with adequate ventilation. Compressed gas cylinders shall be separated

from materials and conditions that present exposure hazards to or from each other. Do not store in

a confined space. Cryogenic containers are equipped with pressure relief devices to control internal

pressure. Under normal conditions these containers will periodically vent product. Some metals such as carbon steel may become brittle at low temperatures and will easily fracture. Prevent entrapment of liquid in closed systems or piping without pressure relief.

8. EXPOSURE CONTROLS/PERSONAL PROTECTION

EXPOSURE GUIDELINES:

OSHA PEL-TWA: None NIOSH IDLH: None

ACGIH TLV: Simple asphyxiant

ENGINEERING CONTROLS:

VENTILATION: Natural or mechanical to prevent oxygen-deficient atmospheres below 19.5% oxygen.

PERSONAL PROTECTIVE EQUIPMENT:

EYE/FACE PROTECTION: Full face shield and safety glasses are recommended.

SKIN PROTECTION: Loose fitting thermal insulated or leather gloves. Safety shoes are recommended when handling liquid containers. Long sleeve shirts and trousers without cuffs. RESPIRATORY PROTECTION (SPECIFY TYPE):

General Use: None required

Emergency Use: Self-contained breathing apparatus (SCBA) or positive pressure airline with mask

are to be used in oxygen-deficient atmosphere. Air purifying respirators will not function. 9. PHYSICAL AND CHEMICAL PROPERTIES

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APPEARANCE: Colorless

ODOR: Odorless

ODOR THRESHOLD: Not applicable

PHYSICAL STATE: Cryogenic liquid

pH: Not applicable

MELTING POINT: -345.8 °F (-209.9 °C) @ 1 atm

BOILING POINT: -320.4 °F (-195.8 °C) @ 1 atm

FLASH POINT: Not applicable

EVAPORATION RATE (Butyl Acetate=1): Not applicable

FLAMMABILITY: Nonflammable

FLAMMABLE LIMITS IN AIR BY VOLUME:

LOWER: Not applicable; UPPER: Not applicable

VAPOR PRESSURE (AT 20 °C): Not applicable

GAS DENSITY: 0.072 lb/ft3 (1.153 kg/m3) @ 70 of (21.1 °C) and 1 atm

SPECIFIC GRAVITY (Air =1): 0.967 @ 70 °F (21.1 °C) and 1 atm

SOLUBILITY IN WATER: Vol/Vol at 32 °F (0 °C): 0.023

COEFFICIENT OF WATER/OIL DISTRIBUTION: Not available

AUTOIGNITION: Nonflammable

DECOMPOSITION TEMPERATURE: Not applicable

MOLECULAR WEIGHT: 28.01 EXPANSION RATIO: (for liquid to gas) 70 °F (21.1 °C): 1 to 696.5 **10. STABILITY AND REACTIVITY CHEMICAL STABILITY: Stable** CONDITIONS TO AVOID: None **INCOMPATIBLE MATERIALS: None** HAZARDOUS DECOMPOSITION PRODUCTS: None POSSIBILITY OF HAZARDOUS REACTIONS: Will not occur **11. TOXICOLOGICAL INFORMATION** The product is simple asphyxiant. ACUTE DOSE EFFECTS: LD50: None LCso: None **REPEATED DOSE EFFECTS: None established IRRITATION:** None SENSITIZATION: None **GENETIC EFFECTS: None DEVELOPMENTAL EFFECTS: None TERATOGENICITY:** None SYNERGISTIC MATERIALS: None **REPRODUCTIVE EFFECTS:** None TARGET ORGAN EFFECTS: None MATERIAL SAFETY DATA SHEET - "LIQUID NITROGEN" ROC Group of Companies Page 5 of 6

MUTAGENICITY: None

12. ECOLOGICAL INFORMATION

ECOTOXICITY: No adverse ecological effects are expected. It does not contain any Class I or Class II ozone depleting chemicals (40 CFR Part 82). Not listed as a marine pollutant by DOT (49

CFR Part 171).

13. DISPOSAL CONSIDERATIONS

WASTE DISPOSAL METHOD: Do not attempt to dispose of residual or unused quantities. Contact your supplier.

For emergency disposal, discharge slowly to the atmosphere in a well ventilated area or outdoors.

14. TRANSPORT INFORMATION

Product Identification Number: 1977 BASIC SHIPPING DESCRIPTION: PROPER SHIPPING NAME: Nitrogen, refrigerated liquid HAZARD CLASS: 2.2 (Nonflammable Gas) IDENTIFICATION NUMBER: UN 1977 ADDITIONAL INFORMATION: PRODUCT RQ: Not applicable SHIPPING LABEL(s): Nonflammable gas PLACARD (When required): Nonflammable gas SPECIAL SHIPPING INFORMATION: Containers should be transported in a secure position, in a well ventilated vehicle. The transportation of compressed gas containers in automobiles or in closed-body vehicles can present serious safety hazards and should be discouraged. For air shipments, the "Cryogenic Liquid" handling label must be used in addition to the nonflammable gas (Division 2.2) hazard label on packages and over packs containing cryogenic liquids.

15. REGULATORY INFORMATION & OTHER INFORMATION

SPECIAL PRECAUTIONS: Use piping and equipment adequately designed to withstand pressures and temperatures to be encountered. Use a check valve or other protective apparatus in any line or piping from the container to prevent reverse flow. Cross contamination of gases, liquids, or both can also create a hazardous condition inside a cylinder, dewar, or vessel (e.g., flammable and oxidizing gases can create an explosive mixture), which may result in rupture. To prevent cryogenic liquids or cold gas from being trapped in piping between valves the piping shall

be equipped with pressure relief devices. Only transfer lines designed for cryogenic liquids shall be

used. It is recommended that all vents be piped to the exterior of the building.

Shipment of compressed gas containers that have not been filled with the owner's consent is a violation of Federal law (49 CFR Part 173.301.b)

HAZARD RATINGS AND RATING SYSTEMS:

NFPA RATINGS:

MATERIAL SAFETY DATA SHEET - "LIQUID NITROGEN"

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HEALTH: =3: FLAMMABILITY: =0: INSTABILITY: =0: SPECIAL: SA

STANDARD VALVE CONNECTIONS:

THREADED: CGA 295

PIN-INDEXED YOKE: Not applicable

ULTRA HIGH INTEGRITY: Not applicable

Use the proper connections; DO NOT USE ADAPTERS. DO NOT FORCE FIT CONNECTIONS.

The information and recommendations in this Material Safety Data Sheet relate only to the specific material mentioned herein and do

not relate to use otherwise ie., in combination with any other material or in any process.

The information and recommendations herein are taken from our extensive experiences and the data contained in recognized references and believed by us to be accurate. Refrigeration group

of companies make no warranties either expressed or implied with respect there to and assume no liability in connection with the use of such information and recommendation.

Safety Data Sheet

SECTION 2: Hazards identification Health hazards

GHS - Classification

Skin corrosion/irritation Category 2 Serious eye damage/eye irritation Category 2 Respiratory sensitization Category 1 Specific target organ systemic toxicity (single exposure) Category 3 Signal Word

DANGER Identification of the substance or mixture

24 hour Emergency Response: 866-536-0631

301-431-8585 Outside of the U.S. +1-301-431-8585 Product code Physical hazards 27250018 27250018 Company/undertaking identification For research use only. Not intended for human or animal diagnostic or therapeutic uses. Revision date Product name TRYPSIN, 1/250Trypsin (1:250) porcine parvovirus tested Not Hazardous Page 1 / 6

SECTION 1: Identification of the substance/mixture and of the company/undertaking

Product name 30-Jan-2015 TRYPSIN, 1/250Trypsin (1:250) porcine parvovirus tested www.lifetechnologies.com Product code Life Technologies 5250 Mainway Drive Burlington, ONT CANADA L7L 6A4 800/263-6236 Life Technologies 5791 Van Allen Way PO Box 6482 Carlsbad, CA 92008 +1 760 603 7200 Reproductive toxicity None Skin Sensitization R42 - May cause sensitization by inhalation Irritating to skin. Target Organ Effects None under normal use conditions inhalation Irritating to respiratory system.

SECTION 4: First aid measures

Ingestion

Skin contact Wash off immediately with plenty of water for at least 15 minutes. Take off all contaminated clothing and wash it before reuse. Immediate medical attention is required.

HMIS

May be harmful if swallowed. Ingestion may cause gastrointestinal irritation,

nausea, vomiting and diarrhea.

Hazard Statements

H316 - Causes mild skin irritation

H320 - Causes eye irritation

H335 - May cause respiratory irritation

H334 - May cause allergy or asthma symptoms or breathing difficulties if inhaled

Precautionary Statements

P261 - Avoid breathing dust/fume/gas/mist/vapors/spray P305 + P351 + P338 - IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing P342 + P311 - If experiencing respiratory symptoms: Call a POISON CENTER or doctor/physician P332 + P313 - If skin irritation occurs: Get medical advice/attention P304 + P312 - IF INHALED: Call a POISON CENTER or doctor if you feel unwell P312 - Call a POISON CENTER or doctor/physician if you feel unwell **Principle Routes of Exposure Potential Health Effects Specific effects** Carcinogenic effects None eyes Irritating to eyes. Mutagenic effects None

SECTION 3: Composition/information on ingredients

Flammability 0 Health 2 * Chronic Hazard 27250018 **Chemical Name CAS-No Revision date** EINECS-No Product name TRYPSIN, 1/250Trypsin (1:250) porcine parvovirus tested Weight % Page 2 / 6 Reactivity 0 Trypsin 9002-07-7 30-Jan-2015 232-650-8 60-100 www.lifetechnologies.com Product code We recommend handling all chemicals with caution. **Engineering measures** Ensure adequate ventilation, especially in confined areas. **Ingestion** Call a physician or poison control center immediately. Never give anything by mouth to an unconscious person. Do not induce vomiting without medical advice.

See Section 12 for more information.

SECTION 6: Accidental release measures

Personal Protective Equipment

Personal Protective Equipment requirements are dependent on the user institution's risk assessment and are specific

to the risk assessment for each laboratory where this material may be used.

Notes to Physician

SECTION 7: Handling and storage

Treat symptomatically.

Respiratory protection In case of insufficient ventilation, wear suitable respiratory equipment. Suitable extinguishing media

Personal precautions Ensure adequate ventilation. Avoid contact with skin, eyes or clothing. Use personal protection equipment.

Handling Always wear reccommended Personal Protective Equipment. No special handling advices are necessary.

Hand protection Impervious gloves.

Water spray. Carbon dioxide (CO2). Foam. Dry chemical.

Eye contact

Eye protection Wear safety glasses with side shields (or goggles).

Storage Keep in a dry, cool and well-ventilated place. Keep in properly labeled containers.

Skin and Body Protection Lightweight protective clothing.

Methods for cleaning up Take up mechanically, placing in appropriate containers for disposal.

Hygiene measures Handle in accordance with good industrial hygiene and safety practice.

Rinse immediately with plenty of water, also under the eyelids, for at least 15

minutes. Immediate medical attention is required.

Special protective equipment for firefighters

SECTION 8: Exposure controls/personal protection

Wear self-contained breathing apparatus and protective suit. **Exposure Limits Environmental precautions** inhalation 27250018 **Chemical Name OSHA PEL Revision date OSHA PEL (Ceiling)** Product name TRYPSIN, 1/250Trypsin (1:250) porcine parvovirus tested ACGIH OEL (TWA) Page 3 / 6 ACGIH OEL (STEL) Remove to fresh air. If not breathing, give artificial respiration. Call a physician or poison control center immediately. Trypsin 30-Jan-2015 None None www.lifetechnologies.com None None Product code Prevent further leakage or spillage if safe to do so. Prevent product from entering drains. **SECTION 5: Firefighting measures Principle Routes of Exposure Potential Health Effects** eyes Irritating to eyes **SECTION 9: Physical and chemical properties** Skin Irritating to skin **SECTION 10: Stability and reactivity** inhalation Irritating to respiratory system Ingestion May be harmful if swallowed Ingestion may cause gastrointestinal irritation, nausea, vomiting and diarrhea **Environmental exposure** controls Carcinogenic effects None Stability Stable under normal conditions. Mutagenic effects None Prevent product from entering drains. Do not allow material to contaminate ground

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water system. Reproductive toxicity None. Materials to avoid Strong oxidizing agents. Sensitization R42 - May cause sensitization by inhalation Form Hazardous decomposition products None under normal use conditions. Target Organ Effects None under normal use conditions solid polymerization None under normal processing. **SECTION 12: Ecological information** Appearance **SECTION 11: Toxicological information** Ecotoxicity No information available powder Acute Toxicity **General information** Oxidizing properties No information available Odor Chemical Name LD50 (oral,rat/mouse) LD50 (dermal,rat/rabbit) LC50 (inhalation,rat/mouse) Water solubility soluble Trypsin No data available No data available 27250018 No data available No information available **Revision date** °F No data available Product name TRYPSIN, 1/250Trypsin (1:250) porcine parvovirus tested °C No data available Page 4 / 6 °F No data available °F No data available Melting point / melting range °C No data available 30-Jan-2015 **Autoignition Temperature** °C No data available www.lifetechnologies.com Boiling point / boiling range °F No data available Product code flash point °C No data available **SECTION 13: Disposal considerations** Mobility ΙΑΤΑ No information available Proper shipping name No dangerous good in sense of these transport regulations Dispose of contents/containers in accordance with local regulations. Hazard Class None **Bioaccumulation** Subsidiary class None

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No information available **Packing group** None

SECTION 14: Transport information

UN-No none Biodegradation No information available SECTION 15: Regulatory information

Trypsin 9002-07-7 (60-100) This product has been classified in accordance with the hazard criteria of the Controlled Products Regulations (CPR) and the MSDS contains all the information required by the CPR. **US State Regulations** Listed **SARA 313** This product is not regulated by SARA.

Component Chemical Name Massachusetts - RTK New Jersey - RTK Pennsylvania - RTK

Illinois - RTK Rhode Island - RTK

Trypsin - Listed - - -27250018 Revision date Product name TRYPSIN, 1/250Trypsin (1:250) porcine parvovirus tested Page 5 / 6 TSCA

Clean Air Act, Section 112 Hazardous Air Pollutants (HAPs) (see 40 CFR 61)

This product does not contains HAPs.

California Proposition 65

This product does not contain any Proposition 65 chemicals. 30-Jan-2015

US Federal Regulations

www.lifetechnologies.com Product code

WHMIS Hazard Class

D2A - Very toxic materials D2B - Toxic materials

"The above information was acquired by diligent search and/or investigation and the recommendations are based on

prudent application of professional judgment. The information shall not be taken as being all inclusive and is to be

used only as a guide. All materials and mixtures may present unknown hazards and should be used with caution.

Since the Company cannot control the actual methods, volumes, or conditions of use, the Company shall not be held

liable for any damages or losses resulting from the handling or from contact with the product as described herein.

THE INFORMATION IN THIS SDS DOES NOT CONSTITUTE A WARRENTY, EXPRESSED OR IMPLIED,

INCLUDING ANY IMPLIED WARRANTY OF MERCHANTABILITY OR FITNESS FOR ANY PARTICULAR PUPOSE"

For research use only. Not intended for human or animal diagnostic or therapeutic uses. **Product name** TRYPSIN, 1/250Trypsin (1:250) porcine parvovirus tested

End of Safety Data Sheet

Page 6 / 6 SECTION 16: Other information

Reason for revision

30-Jan-2015 Product code www.lifetechnologies.com SDS sections updated. 27250018 Revision date