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CHARACTERIZATION OF FUNGAL CONTAMINANTS IN B20 BIODIESEL
STORAGE TANKS AND THEIR EFFECT ON FUEL COMPOSITION

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CHARACTERIZATION OF FUNGAL CONTAMINANTS IN B20 BIODIESEL
STORAGE TANKS AND THEIR EFFECT ON FUEL COMPOSITION

A THESIS APPROVED FOR THE
DEPARTMENT OF MICROBIOLOGY AND PLANT BIOLOGY

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“Science is more than a body of knowledge. It is a way of thinking; a way of skeptically interrogating the universe with a fine understanding of human fallibility.”

-Carl Sagan

To Matias. You are my candle in the dark.

To Sebastian and Adrian. My biggest dream is to be part of your adventures and accomplishments, as you are being part of mine.

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Abstract

The liquid transportation fuel B20 biodiesel is an 80:20 blend of petroleum-derived ultra-low sulfur diesel (ULSD) and biodiesel. Although B20 biodiesel represents a fungible fuel with a reduced carbon footprint compared to petroleum diesel, it is more susceptible to microbial contamination and biodegradation. The research described in this thesis characterized the numerically abundant fungi responsible for fouling in B20 biodiesel storage tanks. This work also investigated the effect of microbial contamination and proliferation on B20 biodiesel composition. Fungi from the genera *Wickerhamomyces* and *Byssochlamys* were abundant in the B20 storage tanks that were monitored in this study. Members of the yeast *Wickerhamomyces anomalus* SE3 and the filamentous fungus *Byssochlamys sp.* SW2 that represent the major taxa in B20 storage tanks were isolated and characterized for their ability to degrade components of B20 biodiesel. Both *Wickerhamomyces anomalus* SE3 and *Byssochlamys sp.* SW2 were able to use B20 biodiesel as sole carbon and energy source. We show that the presence of *Byssochlamys sp.* SW2 can alter the composition in B20 biodiesel in storage tanks, and we offer a model for predicting the severity of biodegradation. *Byssochlamys sp.* SW2 preferentially degraded palmitic and linoleic acid methyl esters, and our *in situ* model supports the hypothesis that palmitic and linoleic acid methyl esters are the most susceptible components to biodegradation. We suggest the use of alternative feedstocks containing less palmitic and linoleic acid for B20 biodiesel production to increase fuel stability in storage tanks.

Chapter 1: Problem Statement

The energy demand of a globalized, increasing world population is rising quickly. Worldwide energy consumption is projected to increase 1.4% per year (Sieminski, 2014) from 549 quadrillion British Thermal Unit (BTUs) in 2012 to 815 quadrillion BTUs in 2040. Fossil fuels, including petroleum, coal and natural gas, account for more than three-quarters of the world's total energy consumption (DOE, 2016). The United States (U.S.) is the largest consumer of petroleum using approximately 19 million barrels per day in 2014, and projected to increase 1% per year (EIA, 2016a).

The Department of Defense (DoD) is the largest single consumer of energy in the U.S., with the Armed Forces purchasing 32.0 billion gallons of petroleum at a cost of \$107.2 billion, from 2007 to 2014 (GAO, 2015). This represents approximately 1.9% of total annual U.S. petroleum consumption (EIA, 2016a), with the Air Force (USAF) accounting for 48% of the total DoD energy consumption (USAF, 2013). The reliance on fossil fuel has led to concerns mainly related with energy security (Leiby, 2007) and environmental hazards (Kharaka & Dorsey, 2005). The DoD has focused on the use of alternatives to petroleum-based liquid transportation fuels in order to increase its use of renewable, reliable, and clean energy sources (Congress, 2005).

The DoD has established the technical, economic and environmental requirements for any alternative fuel to be included in their portfolio (Blakeley, 2012).

Technically, the fuel must be fungible or “drop-in”, meaning that it requires no modification to existing engines and infrastructure. Any alternative fuel must also be cost-competitive with petroleum fuels, as well as environmentally sustainable, derived from feedstocks that do not affect the food market, and fulfill regulatory initiatives to reduce greenhouse emissions. The USAF has selected Ethanol E85 (85% ethanol and 15% gasoline) and B20 biodiesel (20% biodiesel and 80% Ultra Low Sulfur Diesel) for displacing gasoline and diesel, respectively (DOD, 2007). Both biofuels are used in non-tactical vehicles as part of a USAF goal for reducing in 2% the annual petroleum consumption for vehicles through 2020 (DOD, 2011).

The majority of the USAF biofuel consumption was B20,, and by year 2011, more than 60 bases were dispensing it (DOD, 2011). Biodiesel is a “fuel comprised of mono-alkyl esters of long chain fatty acids derived from vegetable oils or animal fats” (ASTM-D7467-15, 2015). Biodiesel is produced by converting a triglyceride feedstock to fatty acid methyl esters (FAME) through a base-catalyzed transesterification reaction (Fig. 1.1) (Knothe, Krahl, & Van Gerpen, 2015). During this process, byproducts are produced including glycerol, soaps and water which are removed to obtain neat biodiesel (B100) (Hoekman, Broch, Robbins, Ceniceros, & Natarajan, 2012). Soybeans represent the feedstock most commonly used in 90% of U.S. production followed by corn oil, canola, and animal fats (EIA, 2016b).

However, there is an increased interest in non-food feedstocks like jatropha (Sarin, Sharma, Sinharay, & Malhotra, 2007) and microalgae (Ahmad, Yasin, Derek, & Lim, 2011; Chen et al., 2015) for future commercialization. The triglyceride

feedstocks represent renewable sources of energy and the fuel produced with them biodegradable (Jakeria, Fazal, & Haseeb, 2014).

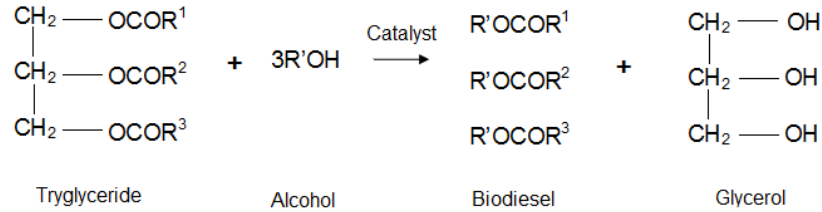


Figure 1.1 Transesterification reaction scheme to produce biodiesel (Adapted from Knothe, 2008) R1, R2, R3 represent the hydrocarbon chains of the parental triglyceride. R' present the alkyl radical of the used alcohol.

Neat biodiesel can be blended in any proportion to petroleum diesel (Weiksner, Crump, & White, 2008), but B20 is the most commonly used blend in the U.S. (AFDC, 2016) due to several advantages (AFDC, 2016). Specifically, B20 is compatible with existing diesel engines and infrastructure, while contributing with the engine power and increasing the fuel efficiency (Lahane & Subramanian, 2015). B20 has better tolerance to cold weather than higher blends (Knothe et al., 2015), and presents a good balance between gas emissions and costs (NREL, 2009).

Despite its advantages, the stability of biodiesel and, therefore B20 presents several technical problems (Jakeria et al., 2014; Pullen & Saeed, 2012). FAMES in biodiesel increase its oxygen content and hygroscopic nature compared to the hydrocarbons in petroleum based ULSD. This makes biodiesel oxidatively unstable (Pullen & Saeed, 2012) and susceptible to microbial contamination especially during mid and long term storage (Knothe et al., 2015; Lee, Ray, & Little, 2010; Zuleta, Baena, Rios, &

Calderón, 2012). Uncontrollable microbial growth in B20 fuel systems increases the presence of flocculent material along with sludge formation and fouling of fuel probes (Chao, Liu, Zhang, & Chen, 2010; Passman, 2003). Consequences of microbial metabolism of FAMES (Jakeria et al., 2014; Schleicher, Werkmeister, Russ, & Meyer-Pittroff, 2009) include water formation and biodegradation of the fuel (Knothe et al., 2015). Ultimately, the properties and quality of the fuel are altered.

We have studied the B20 fuels from underground storage tanks at several Air Force Bases, both with and without reported issues with fuel quality (color, clarity, particulates, Fig 1.2) reported by operators (Dr. Wendy J. Crookes-Goodson, personal communication). Fuels of compromised quality from two different AFBs (SE and SW) had substantial microbial contamination believed to be the root cause of reported issues. Molecular characterization of the microbial assemblages showed that these fuels harbored an abundance of biomass from the fungal classes Eurotiomycetes (genus *Byssochlamys*) and Saccharomycetes (genus *Wickerhamomyces*).



Figure 1.2 Images of B20 fuel samples obtained from a storage tank at a USAF facility. Yellow and bright fuel obtained prior exposure to the tanks (left) and fuel obtained from the bottom of a tank experiencing water and fouling problems (right). Flocculent material present in the interface between fuel and water (right).

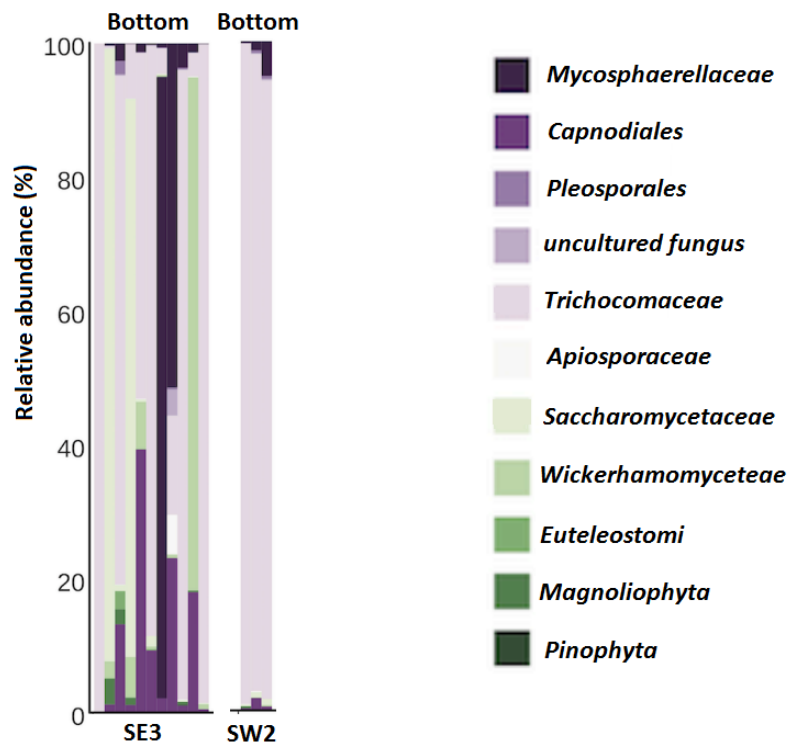


Figure 1.3 Fungal communities detected at the bottom of storage tanks (Adapted from Stamps, 2016). Relative abundance of fungal families in B20 storage tanks at two different locations (SE and SW) based on amplified 18S rRNA gene libraries.

Members of the fungal genera *Byssochlamys* and *Wickerhamomyces* are ubiquitous (Passoth, Fredlund, Druvefors, & Schnürer, 2006; Samson, Houbraken, Varga, & Frisvad, 2009). Members from the genus *Byssochlamys* are commonly found in fruits and soil (Kotzekidou, 1999), and are usually associated with spoilage of heat-processed foods (Samson, Houbraken, Varga, & Frisvad, 2009). Members from the genus *Wickerhamomyces* exhibit wide metabolic and physiological diversity (Passoth, Fredlund, Druvefors, & Schnürer, 2006) and have been frequently isolated in natural habitats like plants, fruits and insects (Walker, 2011).

Members from both fungal genera have been detected in fuel systems (Bücker et al., 2011; Gassen et al., 2015; Rauch et al., 2006), although they have not been extensively characterized. A detailed study of individual fuel contaminants is only recommended when the microbial organism is abundant and its analysis can provide insight on the sources of contamination (Hill & Hill, 2008) and/or the susceptibility of the fuel to its growth (Cazarolli et al., 2014; Sheridan, Nelson, & Tan, 1971).

The present research sought to identify and characterize the abundant fungal contaminants from B20 fuel storage tanks at two Air Force bases, and determine whether their presence was correlated with fuel degradation. In Chapter 2, the isolation and detailed characterization of the members of the genera *Byssochlamys* and *Wickerhamomyces* is described. Also, their ability to degrade B20 fuel as a sole carbon and energy source is analyzed. This information is used to discuss their ecological predominance in the tanks. In Chapter 3, the chemical composition of

B20 fuel samples obtained from the contaminated storage tanks is analyzed. The exploration of patterns of fungal biodegradation in the samples is discussed. This information was then used to evaluate the susceptibility of the B20 biodiesel to fungal contamination. Moreover, these results were applied to the development of a statistical tool useful to monitor biodegradation of the fuel.

Chapter 2: Characterization of Fungal Contaminants from B20

Storage Tanks

2.1 Introduction

Biodiesel is composed of fatty acid methyl esters (FAMES) converted from the fats and oils of various plant and animal feedstocks (ASTM-D7467-15, 2015). Blends of biodiesel, such as B20, which is composed of 20% FAMES and 80% Ultra Low Sulfur Diesel (ULSD), have the advantage of retaining most of the properties of petroleum diesel but at the same time having a lower carbon footprint (Knothe et al., 2015). Due to its composition, biodiesel is much more biodegradable than petroleum diesel (Mariano, Tomasella, De Oliveira, Contiero, & De Angelis, 2008; Peterson & Moller, 2005). This can be an advantage when there is a spill of biodiesel, but it is also a drawback when microorganisms that can degrade biodiesel colonize storage tanks (Passman, 2013). All grades of fuels are susceptible to microbial growth (Dodos, Konstantakos, Longinos, & Zannikos, 2012; Hill & Hill, 2008; Leja & Broda, 2009; Rauch et al., 2006), but with the introduction of unleaded gasoline, ultra-low sulfur diesel and alternative fuels, microbial contamination and proliferation during storage has become more frequent and severe (Hill & Hill, 2008).

The susceptibility of a fuel to microbial contamination depends on its composition, especially when they provide a source of macro and micronutrients needed for

growth (Gaylarde, Bento, & Kelley, 1999). The FAMES in biodiesel are a readily available source of carbon and energy (Prince, Haitmanek, & Lee, 2008). Coupled with the hygroscopic nature of the biodiesel (Fregolente, Fregolente, & Wolf Maciel, 2012) and the sub-inhibitory concentrations of sulfur in ULSD (Ali, Ghaloum, & Hauser, 2006; Londry & Suflita, 1998), B20 biodiesel blends are particularly prone to contamination and proliferation compared to fossil fuels (Bücker et al., 2011; Dodos, Konstantakos, Longinos, & Zannikos, 2012; Zimmer et al., 2013).

Microorganism can be introduced to the fuel distribution systems soon after the refinery processes (ASTM:D6469, 2003). During normal operations, fuel transfers out of storage tanks increase the risk of microbial contamination (Engelen, 2009; Passman, 2013) because air/water vapor is pulled from outside the tanks, through vents, to compensate for the vacuum caused by the fuel removal. Microorganisms are easily dispersed through atmospheric circulation (Nemergut et al., 2013) and contaminate the storage tanks when attached to particles, dust, and pollen from proximal soils (Rauch et al., 2006).

Any B20 storage tank contains multiple habitats, each with unique selective pressures (Passman, 2003). At the fuel-water interface, aerobic and facultative microorganism will thrive (Engelen, 2009). In this zone, water, nutrients and oxygen are available to support biofilm formation (Lee et al., 2010), both in the fluid as well as on tank walls (Passman, 2003). Fuel degraders will metabolize

FAMEs (Bücker et al., 2011; Prince et al., 2008) through enzymatic hydrolysis and subsequent beta-oxidation. (Jakeria et al., 2014; Kumari & Gupta, 2014) Organic acids and other byproducts can serve as nutrient sources for other microorganisms (Hill & Hill, 2008). The co-metabolism of hydrocarbon in biodiesel/petroleum diesel blends is also possible (Pasqualino, Montane, & Salvado, 2006; Zhang, Peterson, Reece, Haws, & Möller, 1998). Anaerobic microorganisms will be present in the bottom of the tanks, where oxygen is usually depleted. The anaerobic metabolism of FAME will include hydrolysis (Lapinskienė & Martinkus, 2007) and β -oxidation (Sousa, Smidt, Alves, & Stams, 2009). Under these conditions, the weak organic acids along with electro-potential gradients can accelerate the rate of biocorrosion (Aktas et al., 2010).

We have monitored B20 biodiesel storage tanks at two USAF facilities since 2014. These installations have experienced recurrent filter clogging (fouling), presence of water, particulates, and increased rates of corrosion (Stamps, 2016). Microscopy, molecular analyses, and cultivation experiments indicated that the fungal organisms across multiple tanks and locations were members of the families *Trichocomaceae* and *Saccharomycetaceae*. The research presented here aimed to study the fungal organisms that were predominant in these biofilms. The numerically dominant OTU from these systems was a member of the family *Trichocomaceae* and the genus *Byssochlamys*. The numerically abundant *Saccharomycetaceae* was a member of the genus *Wickerhamomyces* (Stamps, 2016).

B20 storage tanks are adequate places to support fungal growth since moisture, nutrient availability, and surface area are available (Bücker et al., 2011). Fungi are heterotrophs showing great phenotypic plasticity to different environments (Barnett & Barnett, 2011; Hawker, 2016). During B20 fuel storage, uncontrollable proliferation of fungal hyphae and single-celled yeasts can contribute to flocs and filter plugging (Passman, 2003). Most fungi are capable of aerobic and fermentative assimilation of many carbon substrates (C. Kurtzman, Fell, & Boekhout, 2011) including biologically available fuel components in B20 biodiesel, which can originate from changes in fuel properties as result of deterioration (Bücker et al., 2011; Gassen et al., 2015). Finally, organic acids and water are released as byproducts of fuel metabolism, creating favorable conditions for corrosion of metal surfaces (Little & Ray, 2002).

The role of members of the genera *Byssochlamys* and *Wickerhamomyces* as fuel contaminants in storage tanks across the USAF is still under investigation, but their abundance suggests that they play a role both in biofilm formation, bio-deterioration, and perhaps an increased risk of microbially influenced corrosion. This study aimed to isolate, identify, and characterize these fungal organisms from the contaminated storage tanks. We hypothesized that *Byssochlamys* and *Wickerhamomyces* are fuel degraders able to grow using B20 as sole carbon and energy source. To test this hypothesis, we quantified the loss of B20 components using GC-MS as part of growth experiments. We also investigated morphological

and physiological traits of the isolates as baseline information for future studies regarding the risks of fungal contamination and their role in B20 storage tanks.

2.2 Materials and Methods

2.2.1 Sample Collection and Cultivation

Fuel samples were collected from underground storage tanks at two USAF facilities located in the southwest (SW) and southeast (SE) of the continental United States. Samples from SE were obtained during the months of August 2014 and April 2015. Samples from SW were collected in September 2014.

At SE, 1 L samples were collected from the bottom of 3 underground B20 biodiesel storage tanks using a sample thief (“Bacon Bomb”; Koehler Instrument Company; Holtsville, New York). Each sample was transferred into sterile HDPE bottles, shipped at room temperature, and processed within 24-72 h of collection. Biomass was recovered by filtration using a Stericup® bottle top filter unit (120 mm dia, 0.22 µm pore size, PES filter; EMD Millipore, Billerica, MA) attached to a sterile 1 L glass bottle. After filtration, the filter was cut into quarters with a sterile disposable scalpel and placed into Hestrin Schramm (HS) broth (Hestrin & Schramm, 1954), which contains (per L); 20 g glucose, 5 g yeast extract, 5 g polypeptone, 2.7 g NH_2PO_4 , and 1.15 g of citric acid. The pH of the medium was adjusted to 5.5 with 0.1 M HCl prior to autoclaving. Inoculated cultures were incubated at 25 °C for 48 h and shaking at 250 rpm. Aliquots (100 µL) of each culture were subsequently spread onto solid HS medium containing 1.5% w/v agar to facilitate isolation of individual clonal populations.

At SW, biofilms were sampled from the surface of steel witness coupons suspended in three storage tanks (Stamps, 2016) using nylon flocked swabs (Therapak Corp, Los Angeles, CA). Enrichment cultures of the liquid media Sabouraud Dextrose (SAB), Potato Dextrose (PD), Malt Extract (ME) (Becton, Dickinson and Company; Franklin Lakes, NJ), and HS were inoculated on site, shipped to the laboratory, and spread (100 μ L) onto solid agar medium of the same composition within 24-72 hours of collection. Isolated fungi were obtained in pure culture by repeated sub-culturing on solid agar. For all purposes, cultures were incubated aerobically at room temperature for 5-7 days and shaking at 250 rpm. Stocks of pure cultures were stored in 10% glycerol at -80 °C.

2.2.2 Molecular Identification of Fungal Isolates

Total genomic DNA was extracted from biomass of each isolate using the UltraClean Microbial DNA Isolation Kit (MoBio Lab. Inc., Carlsbad, CA) following manufacturer's instructions. For molecular identification at the genus level, a fragment of the 18S rRNA gene including the V4 and V5 regions was amplified using PCR with universal primers 566F (5'-CAGCAGCCGCGGTAATTCC-3') and 1200R (5'-CCCGTGTTGAGTCAAATTAAGC-3') (Hadziavdic et al., 2014). For identification at a potentially higher taxonomic level, the divergent D1/D2 domain of large subunit (26S) ribosomal RNA was amplified using primers NL-1 (5'-

GCATATCAATAAGCGGAGGAAAAG-3`) and NL-4 (5`GGTCCGTGTTTCAAGACGG-3`) (C. P. Kurtzman & Robnett, 1998). Gene fragments were amplified using 20 µL of 5 Prime Hot master mix (5 Prime, Inc., Gaithersburg, MD) and 0.2 µM of each primer in a total volume of 50 µL. Thermal cycling for the 18S rRNA PCR was carried out in a Techne, TC-512 thermal cycler consisting for 35 cycles of 94 °C for 45 s, 60 °C for 45 s and 72 °C for 1 m, and held at 4 °C, while the reaction for the amplification of the 26S rRNA fragments consisted of 36 cycles, 94 °C for 1 m, 52 °C for 45 s and 72 °C for 2 m, and holding at 4 °C.

Amplified fragments were purified using Agencourt AMPure XP paramagnetic beads (Beckman Coulter Inc., Indianapolis, IL) following the manufacturer's recommendations. Fragments were sequenced in an automatic DNA sequencer (3130xl Genetic Analyzer, Applied Biosystems / Thermo Scientific, Carlsbad, CA) using the Big Dye terminator cycle sequencing kit (ver. 3.1). Sequences were aligned using ClustalX (Higgins, Thompson, & Gibson, 1997) against their closest matches according to the NCBI and SILVA refseq database and phylogenetic trees were constructed using the Maximum Likelihood method with 500 bootstrap replicates in the program MEGA version v6.0 (Kumar, Nei, Dudley, & Tamura, 2008).

2.2.3 Phenotypic and Biochemical Characterization

The fungal isolates of interest were tested for utilization of various compounds as energy and/or carbon sources, as well as growth in vitamin-free medium, high osmotic pressure, different temperatures and pH under aerobic conditions (C. P. Kurtzman, Fell, Boekhout, & Robert, 2011). Growth was either monitored by optical density (OD) at 600 nm (Spectronic 20D, Milton Roy, DE) or by dry weight (Analytical Balance Metler Toledo AL104). Morphology was described using scanning electron microscope (ZEISS NEON 40 EsB, Samuels Roberts Noble Microscopy Laboratory, at the University of Oklahoma).

For each growth experiment, an inoculum was prepared by growing the isolates on YM agar, which contains (per L); 3 g of yeast extract, 3 g of malt extract, 5 g of peptone and 10 g of glucose. These cultures were incubated for 24-48 h at room temperature and shaking at 250 rpm before inoculum preparation (C. P. Kurtzman et al., 2011). For the yeasts, the cells were transferred to Yeast Nitrogen Base (Sigma Aldrich) liquid medium with 1% glucose for 48 h at room temperature and shaking at 250 rpm. Each experimental tube was inoculated with a yeast suspension to achieve an initial OD of 0.02. For experiments with filamentous fungi, a spore suspension was prepared by removing mycelia from YM agar with sterile water, filtration (8.0 μm pore size) and washing (3 times) by filtration, followed by centrifugation and resuspension of spores in sterile water. Each experimental tube was inoculated with 1×10^6 spores ml^{-1} .

To test the ability of the fungal isolates to grow in vitamin-free medium, 5 mL of Artificial Sump Water (ASW, per L; 15 mg, 35 mg NaF, 2 mg CaCl₂, 18 mg KNO₃, 10 mg Na₂SO₄, 15 mg (NH₄)₂SO₄, and 17 mg K₂HPO₄) (McNamara et al., 2005) was used, supplemented with 10 g of glucose. After 1 week of incubation at 25 °C, an aliquot was transferred to a new set of test tubes and growth was evaluated after another week of incubation. To test the ability to grow at high osmotic pressure, the fungi were grown on 50% glucose agar plates containing per 0.1 L 13 g of agar, 500 g of glucose and 1 g yeast extract, and in liquid medium containing per L 100 g of sodium chloride, 50 g of glucose, and 6.7 g of Yeast Nitrogen Base (Sigma Aldrich, USA) (C. P. Kurtzman et al., 2011). Growth at various temperatures was determined in Yeast Nitrogen Base liquid medium supplemented with 5% glucose for the yeast and 10 g/L of agar added for solid medium for the filamentous fungi. Temperatures tested included 9, 15, 20, 25, 30, 37, 40, 45 and 50 °C. The range and optimal pH for both organisms was determined by growth at various pH (3.0-10.0 with increments of 1.0 pH units) at 25 °C and shaking at 250 rpm in HS medium adjusting the pH with an appropriate buffer (citric acid-sodium solution, MES, MOPs, Tris base, sodium carbonate-sodium bicarbonate solution, Sigma Aldrich, Buffer Reference Center).

2.2.4 Chemotaxonomic Characterization

Biomass for fatty acid analysis was collected from a plate of HS agar final pH 5.5 after 48 hours incubation for the yeast and 5 days incubation for the filamentous fungi, both at room temperature. Fatty acid methyl esters were extracted using the Sherlock Microbial Identification System (MIDI; version 6.1) according to manufacturer's protocol (MIDI, Newark, Delaware USA). Fatty acids were identified using an Agilent Technologies 6890N gas chromatograph (Patel et al., 2015). The results were expressed in the form of percentages using the QTSA peak naming database.

2.2.5 B20 Biodegradation Experiments with Fungal Isolates

Fungal biodegradation was evaluated by direct measure of fungal growth and consumption of the fuel components. Two independent experiments were designed: a 15-day incubation experiment to measure fungal growth and evaluate the ability to degrade B20 biodiesel, and an 80-day experiment to determine preferential consumption of B20 components and detect biodegradation metabolites. All biodegradation experiments were incubated aerobically at 25 °C and shaking at 250 rpm. Culture tubes for both growth experiments contained filter-sterilized B20 as the sole carbon source and ASW liquid medium (pH 5.5) in a 1:100 ratio. For the longer experiment, caps w/PTFE a liner were used to avoid evaporation of the fuel. For the yeast, the inocula were from cultures transferred multiple times in ASW

amended with B20. An equal amount of yeast inoculum was added to each test tube. The filamentous fungi were inoculated with a conidia suspension in a final concentration of 1×10^6 spores mL^{-1} in each test tube. Growth was measured as CFU mL^{-1} for the yeast and dry weight for the filamentous fungi. The fuel phase of these cultures was extracted with hexane (Sigma Aldrich CHROMASOLV®, for HPLC, $\geq 97.0\%$ (GC)) and analyzed by Gas Chromatography/Mass Spectrometry (GC/MS) (See protocol, Section 3.2). The experiment was carried out with numerous replicates, allowing for destructive sampling of triplicates at each time point. The degradation of compounds was evaluated as the amount (%) of the peak area remaining relative to day 0 of the negative control. The aqueous-phase from triplicate cultures was filtered, extracted and analyzed for metabolites. Ethyl acetate (3 mL) was added to 1 mL of aqueous culture, separated from the aqueous phase, and dried, under N_2 gas to a final volume of 100 μL . These extractions were derivatized by addition of a 1:1 volume of BSTFA (N, O-Bis (trimethylsilyl) trifluoroacetamide, Sigma Aldrich) and incubation at 75 °C for 15 m. A 50- μL aliquot of the derivatized extraction was diluted with hexane to a final volume of 1 mL and analyzed by GC/MS (Shimadzu QP2010-SE, University of Oklahoma). For all experiments, negative controls (un-inoculated) were included to evaluate contamination risks and assess abiotic degradation. Un-amended controls (inoculated but with no B20) were also included to evaluate nutrient carryover from the initial inocula.

2.3 Results

2.3.1 Isolation

A total of 10 fungal organisms were isolated under aerobic conditions, the majority of them recovered from HS agar (Table 2.1). Strain differentiation was first approached by alignment of the 18S rRNA gene sequences against the refseq database. From the swabs taken at SW, the isolates were related to the genera *Byssochlamys*, *Rhodotorula* and *Rhodospiridium*. From the fuel at SE, the isolates corresponded to the genera *Aspergillus*, *Aureobasidium*, *Galactomyces*, *Hypopichia*, *Meyerozyma*, *Rhizopus* and *Wickerhamomyces*. Of the 10 isolates, the genus *Wickerhamomyces* and *Byssochlamys* represented the genera detected in greatest abundance in the storage tanks (Stamps, 2016), and were chosen for further characterization.

2.3.2 Characterization of the Filamentous Fungus *Byssochlamys* (strain ID: SW2)

An isolate, SW2, that was likely a member of the genus *Byssochlamys* was further characterized in order to more precisely determine its taxonomic identity, its physiological properties and metabolic capabilities. Colonies of SW2 on HS were 15-20 mm in diameter, consisting of cinnamon-brown mycelia and cream-colored fuzzy edges (Figure 2.1 a), with brown mycelia covering the whole plate after 5 days. Cellular morphology included ellipsoidal conidia with a flattened base and

average length of 4.32 x 2.70 μm (Fig. 2.1 b). A 511 bp sequence from the D1/D2 domain of large subunit (28S) ribosomal RNA had a similarity of 99% with members of the genus *Byssochlamys*. Phylogenetic analysis based on this sequence fragment suggested that it might be a new species of *Byssochlamys* (Figure 2.3). Isolate SW2 was able to assimilate 44 carbon sources from 96 tested. It grew fast (<72 h) on sources like glucose, fructose, succinic acid, galactose, salicilin; growth was slow (> 4 days) in carbon sources like tween 80 and putrescine; and negative in compounds like glycerol (Table 2.3). Isolate SW2 grew in vitamin-free medium as well as in medium with high osmotic pressure (Table 2.3). The optimal growth temperature was 30°C. Growth at 37 °C and at 40°C was detectable after 5 days of incubation. SW2 was also able to grow at temperatures as low as 10°C in HS agar after 6 days of incubation (Figure 2.11). Growth was observed at pH range from 3 to 8 (Figure 2.12). The major fatty acids in its cell wall were C18:2 $\omega_{6,9c}$ /C18:0 ANTE (Table 2.3).

2.3.3 Characterization of the Yeast *Wickerhamomyces* (strain ID: SE3)

Sequence (491 bp) from the D1/D2 domain of the large subunit (28S) ribosomal RNA of isolate SE3 was identical to that of members of the genus *Wickerhamomyces* and was clustered with the yeast *Wickerhamomyces anomalus* (Figure 2.2). Taxonomic identification of isolate SE3 was conducted by characterization of its morphological and physiological properties. Colonies of SE3 were white and smooth with an entire margin after 48 h at 25 °C on HS agar (Fig.

2.1 d). The cellular morphology of this yeast consisted of spherical-elongated cells with multilateral budding (Fig. 2.1 e). Isolate SE3 was able to assimilate 34 carbon sources from the 96 that were tested. It grew quickly (<72 h) on sources like glucose, fructose, succinic acid, and glycerol; growth was slow (> 4 days) in carbon sources like tween, salicilin and serine; and negative in compounds like tween 80 (Table 2.2). Isolate SE3 grew in vitamin-free medium as well as in medium with high osmotic pressure (Table 2.2). Isolate SE3 grew optimally at 30 °C (Figure 2.9). Growth was also noticeable after at 37 °C after 30 hours incubation and at 40 °C after 4 days. The lowest temperature where growth was determined was 9 °C in HS agar after 4 days of incubation. Isolate SE3 grew at a pH of 3 to 9 (Figure 2.10), and its major fatty acids were C18:2_{ω6,9c}/C18:0_{ANTE} (Table 2.2).

2.3.4 Fungal Biodegradation of B20 biodiesel

The isolates SW2 and SE3 were grown with B20 as sole carbon an energy source. *Wickerhamomyces anomalus* SE3 was able to use components of the B20 fuel to grow. Exponential growth of *Wickerhamomyces anomalus* SE3 was observed after 4 days of lag phase. The yeast reached biomass maximum of 10⁷ CFU ml⁻¹ after 15 days of incubation when the initial concentration was 10³ CFU ml⁻¹ (Fig. 2.6 c, Fig. 2.1 f). Analysis of the B20 component of the medium at day 7 showed a reduction in the peak areas of all the detected fatty acid methyl esters as well as many of the hydrocarbon components of the fuel. Compared to the negative control at day 7, the FAMES had a reduction of more than 50% from their original intensity by biotic

mechanisms (Fig. 2.4 a). Hydrocarbons including alkanes, branched alkanes such as octyl- cyclo hexane, and even some aromatics such as octyl-benzene also experienced a considerable reduction (Fig. 2.5).

Byssochlamys sp. SW2 was able to use B20 fuel components to grow. Spore germination of the fungus *Byssochlamys sp.* SW2 was observed after 4 days. After 15 days of incubation, biomass reached 2.6 ± 0.5 mg (dry weight). Following inoculation with a suspension of conidia (1×10^6 spores mL⁻¹) at day 0, analysis of the B20 biodiesel component of the medium at day 7 did not show any reduction of the fuel components compared to the negative control at day 0 (Fig. 2.4). After 30 days of incubation; however, a reduction of $21.0 \pm 9.4\%$ of the peak area for methyl palmitate was observed relative to the negative control at day 0 (Fig. 2.7). After 80 days of incubation, an analysis of the aqueous phase for metabolites detected palmitic, linoleic, oleic and stearic acid (Figure 2.8). The spectra showed palmitic acid as the most abundant fatty acid detected in the water phase. Phase contrast microscopy of samples at day 80 revealed the presence of chlamydospores and hyphae (Fig. 2.1 c).

In all of the biodegradation experiments, no growth was seen in the un-amended controls (no B20 added). Analysis of the negative controls after 7 days of incubation, showed evaporation of all fuel components (~15-20% reduction in the peak areas). All long-term incubations were sealed with Teflon-lined caps and;

therefore, the negative controls showed no abiotic losses. After 80 days incubation, no metabolites were detected in the negative controls.

2.4 Discussion

In the present study, members representing ten different genera of fungi were isolated from B20 storage tanks at two USAF facilities. Prior molecular analyses based on 18S rRNA gene libraries indicated that representatives of the genera *Byssochlamys* and *Wickerhamomyces* were persistent in these tanks and numerically abundant (Stamps, 2016). The isolates identified as *Byssochlamys* sp. SW2 and *Wickerhamomyces anomalus* SE3 were of particular interest in this study because previous studies have implicated them in fouling and increased risk of microbially influenced corrosion (Stamps, 2016). Most of the organisms isolated in this study have been reported as contaminants of fuel systems, especially *Aspergillus* sp., *Rhodotorula* sp., and *Aureobasidium* sp. (Bücker et al., 2011; Rauch et al., 2006).

The genus *Byssochlamys* (anamorph *Paecilomyces*) is composed of a group of mitosporic filamentous fungi that are widely distributed in nature (Zawadneak et al., 2015). Members of this genus have been implicated in fuel contamination and have been isolated from petroleum diesel tanks (Gassen et al., 2015; Lee et al., 2010), jet fuel (Rauch et al., 2006) and biodiesel storage tanks (Bento & Gaylarde, 2001). Its presence has been considered an important problem in the food industry, particularly in pasteurized and canned food (Banner, Mattick, & Splittstoesser, 1979; Houbraken, Varga, Rico-Munoz, Johnson, & Samson, 2008) due to its heat resistant spores (Samson et al., 2009). In the tanks under study, members of the genus *Byssochlamys* have proven to be a numerically abundant organisms in

biofilms causing fouling (Stamps, 2016). We isolated the strain *Byssochlamys sp.* SW2 from biofilms attached to the metal surface of coupons placed inside B20 biodiesel storage tanks.

Phylogenetic analysis of sequences from portions of the small subunit (SSU) and large subunit (LSU) rRNA genes showed that strain SW2 clustered within the genus *Byssochlamys*. However, there was not close similarity with any referenced species. We attempted to amplify the ITS region with the primers ITS1 and ITS4; ITS1-F and ITS4 (White, Bruns, Lee, & Taylor, 1990) and use this sequence data to more closely identify/differentiate SW2 from the other known taxa. The ITS region of the rRNA represents a highly variable region that is useful for species identification (Martin & Rygiewicz, 2005); however, successful amplification with these primers can vary (Schoch et al., 2012). Amplification of the ITS with these primers for our isolate was not successful. Based on phylogenetic analysis of the sequence for the LSU rRNA gene, our isolate might be a novel species, but this remains to be demonstrated.

Byssochlamys sp. SW2 grew with B20 as the sole carbon and energy source using Teflon-lined caps, producing a biofilm. Members of the genus *Byssochlamys*, are known for their ability to grow under low oxygen tensions (Taniwaki, 1995).

Degradation of the fuel was observed at day 30 with a preferential consumption for palmitic acid methyl ester. This result is in agreement with other studies where the closest relative *Pichia variotti* has been reported as a major contaminant of crude

palm oil (Campinha, Machado, & Araújo, 2007). Similar biodegradation experiments demonstrated a preferential consumption for palmitic and oleic acid ME by *Pichia variotii* (Bücker et al., 2011).

The components of B20 biodiesel that are preferentially degraded can also be deduced by the detection of their metabolite intermediates (Aktas et al., 2010; Parisi et al., 2009). After 80 days of incubation, the major FAMES of the B20 were substantially reduced in abundance and their fatty acids were detectable in the aqueous phase. Palmitic acid ME was reduced more than the other FAMES and palmitic acid was more abundant than linoleic, oleic and stearic acid. Overall, this could suggest that these FAMES were initially hydrolyzed to methanol and their representative fatty acids. This process can be catalyzed by lipases able to break down the FAME structure (Jakeria et al., 2014), just as members of the genus *Paecilomyces* have been shown to do (Fernandes, Valério, Feltrin, & Sand, 2012). Fatty acids of smaller chain lengths by two C would have suggested subsequent β -oxidation was occurring; however, these metabolites were either not present or in amounts not detectable by our methods.

Abiotic oxidation of FAMES to fatty acids is also possible (Jakeria et al., 2014). Nevertheless, these acidic compounds were not detected in any of the replicates for the negative controls. The cell walls of fungi also contain C16:0, C18:2, C18:0 and C18:1 as major fatty acids, which could have been extracted and detected. However, the biodegradation of the fatty acid methyl esters peaks by *Byssochlamys sp.* SW2 is

in agreement with the correspondent detected fatty acids (Figure 2.8). Moreover, the aqueous phase of the culture was filtered prior to extraction in order to limit the concentration of fatty acids proceeding from the cell walls.

An interesting observation of the fungal growth during the long-term growth experiments was the formation of chlamyospores in ASW with B20 as sole carbon and energy source. Chlamyospores are thick-walled resting spores that are produced under unfavorable conditions, and due to their perennation nature, these cultures still harbored plenty of viable organisms (Barnett & Barnett, 2011; Hawker, 2016). Chlamyospores can occur when the fungus competes with other microorganisms for substrates or when nutrients are deficient (Lockwood & Filonow, 1981). In the long-term experiments, there were plenty of carbon substrates but other macronutrients were limited and waste products were definitely accumulating.

Members of the genus *Wickerhamomyces* are yeasts considered to be ubiquitous in natural environments (C. P. Kurtzman, 2011; Walker, 2011) such as soil (Hesham et al., 2006), plants (Sláviková, Vadkertiová, & Vránová, 2007), and associated with human (Murphy et al., 1986) and animal (Ricci et al., 2010) hosts. The study described here was the first in which an isolate of *Wickerhamomyces anomalus* was obtained from a B20 fuel storage tank. The ubiquity of members of this genus, in particular *Wickerhamomyces anomalus*, is due to their tolerance to different and often stressful conditions (Passoth et al., 2006; Walker, 2011). Our isolate SE3 was

able to grow under rather acidic and basic pH (3-9), high salinity (5% NaCl), high osmolarity (50% Glucose), over a wide range of temperatures (9-40°C), as well as in vitamin-free medium. The isolate SE3 differs from the *Wickerhamomyces anomalus* type strain (NRRL-Y-366) in that SE3 cannot grow at pH 10, cannot assimilate galactose, and is able to grow at 37 °C.

Wickerhamomyces anomalus SE3 was able to grow using B20 as sole carbon source degrading FAMES and hydrocarbons of various chain lengths. Aerobic experiments posed a specific technical challenge with evaporation of fuel components (Prince et al., 2008). Although some of the loss of fuel components might have been due to evaporation, our results are in agreement with those of other studies. For example, *Wickerhamomyces anomalus* strains AEH and 2.2540 isolated from oil-contaminated soil are fuel degraders able to metabolize aromatic hydrocarbons (Hesham et al., 2006; Pan, Yang, Zhang, Zhang, & Yang, 2004). We found that *Wickerhamomyces anomalus* SE3 was able to degrade octyl-benzene but not naphthalene. *Wickerhamomyces sp.* are closely related to members of the genus *Candida*, which are frequently implicated as capable of degrading the hydrocarbons in fuels (Bento & Gaylarde, 2001; Miranda et al., 2007; Rauch et al., 2006). The yeast *Candida silvicola* has a teleomorph named *Pichia hosltii* that can grow aerobically in B20 reaching a maximum biomass of 10^8 CFU mL⁻¹ after 7 days of incubation (Bücker et al., 2011). In this study, *Wickerhamomyces anomalus* SE3 reached a maximum biomass of 10^7 CFU mL⁻¹ after 7 days that was the basis for proposing the organism was capable of degrading B20 biodiesel.

During long-term incubation experiments, cultures and controls were sealed with Teflon-lined caps to avoid issues with evaporation of the more volatile fuel components. Despite a 4:1 volume ratio of air to medium, *Wickerhamomyces anomalus* SE3 was not able to degrade components of the B20 biodiesel and did not exhibit obvious growth. These incubations still yielded viable cells, even after 45 days of incubation. One explanation could rely on a metabolic characteristic found in some members of the order Saccharomycetaceae (Dashko, Zhou, Compagno, & Piškur, 2014). In an anoxic environment, cell synthesis stops in *Saccharomyces cerevisiae* and *Candida albicans* after a few generations if the anaerobic growth factors ergosterol, unsaturated fatty acids, and nicotinic acid are not present (Fornairon-Bonnefond, Demaretz, Rosenfeld, & Salmon, 2002). We considered the possibility that oxygen would be rapidly consumed by the cells inside the sealed tubes, but wanted to simulate what might happen in a B20 biodiesel storage tank that became anoxic. In a minimal medium like ASW with B20, these organisms would not have access to these anaerobic growth factors. From these results, we hypothesized that *Wickerhamomyces anomalus* SE3 needs unknown anaerobic factors to continue to grow and degrade B20 in ASW once O₂ is depleted. This remains to be tested.

Fungi have various adaptations that allow them to colonize and survive in diverse niches (Leducq, 2014). The isolates SW2 and SE3 have the morphological and physiological potential to grow as numerically abundant organism in the B20 storage tanks. *Byssochlamys* sp. SW2 and *Wickerhamomyces anomalus* SE3 are fuel

degraders, which is a characteristic that provides a nearly unlimited source of carbon and energy inside the tanks. *Wickerhamomyces anomalus* SE3 also had the ability to assimilate glycerol, which would be an advantage if the fuel contains traces of glycerol, a by-product of the transesterification reaction (Gandhi & Wille, 2013). The morphological trait of *Byssochlamys sp.* SW2 to form mycelium networks can be beneficial at the fuel-water interface where it will compete for limited nutrients with other organisms (Buzzini & Margesin, 2014). Both isolates *Byssochlamys sp.* SW2 and *Wickerhamomyces anomalus* SE3 showed a wide range of adaptation to temperature and pH, the ability to grow under high osmotic pressure and without added vitamins. The tolerance to temperature and pH is beneficial in B20 biodiesel, where temperatures can fluctuate depending on the environmental conditions (Passman, 2003), and where bottom water can range from 6.8-8.5 (Passman, 2003), but can be acidify due to the presence of organic acids from microbial metabolism (Bücker et al., 2011) or can become very basic due to the presence of atmospheric corrosion products like iron hydroxides (Morcillo, Fuente, Díaz Ocaña, & Cano, 2011). Their tolerance to high osmotic pressure can provide more resistance to the attack of antimicrobials added to the fuel as mitigation strategy (Passman, 2003) and would serve them well in the low water environment of the fuel. Finally, the isolates showed that they can synthesize all the vitamins that they require (Madan & Thind, 1998), which would be beneficial in the micronutrient-limited environment of fuel storage tanks.

The results of this research have lead us to conclude that *Byssochlamys sp.* SW2 and *Wickerhamomyces anomalus* SE3 were the fungal flora dwelling in the B20 storage tanks. Understanding the flora of the fuel system represents an essential baseline for a comprehensive analysis of the problems occurring with the fuel and the infrastructure. The identification of the prominent fungi in contaminated B20 biodiesel storage systems and their physiological properties should allow operators to better monitor, understand, and prevent contamination and proliferation.

Table 2.1 Identity and provenance of isolated fungi.

Isolate^a	Sample/Incubation Description^b
<i>Wickerhamomyces sp.</i>	SE, 3, F, HS
<i>Rhizopus sp.</i>	SE, 3, F, HS
<i>Aureobasidium sp.</i>	SE, 3, F, HS
<i>Galactomyces sp.</i>	SE, 3, F, HS
<i>Hypopichia sp.</i>	SE, 3, F, HS
<i>Aspergillus sp.</i>	SE, 3, F, HS
<i>Meyerozyma sp.</i>	SE, 3, F, HS
<i>Byssochlamys sp.</i>	SW, 2, C, HS
<i>Rhodosporidium sp.</i>	SW, 3, C, SAB
<i>Rhodotorula sp.</i>	SW, 3, C, SAB

^a Isolate identity based on 18S SSU rRNA gene sequence identity (100%) with nearest cultivated representative.

^b Sample descriptors include the Southeast (SE) or Southwest (SW) AFB, tank (2 or 3), fuel (F) or coupon surface (C) as inoculum, and initial isolation on either Hestrin-Schramm (HS), Malt Extract (ME), or Sabouraud (SAB) medium.

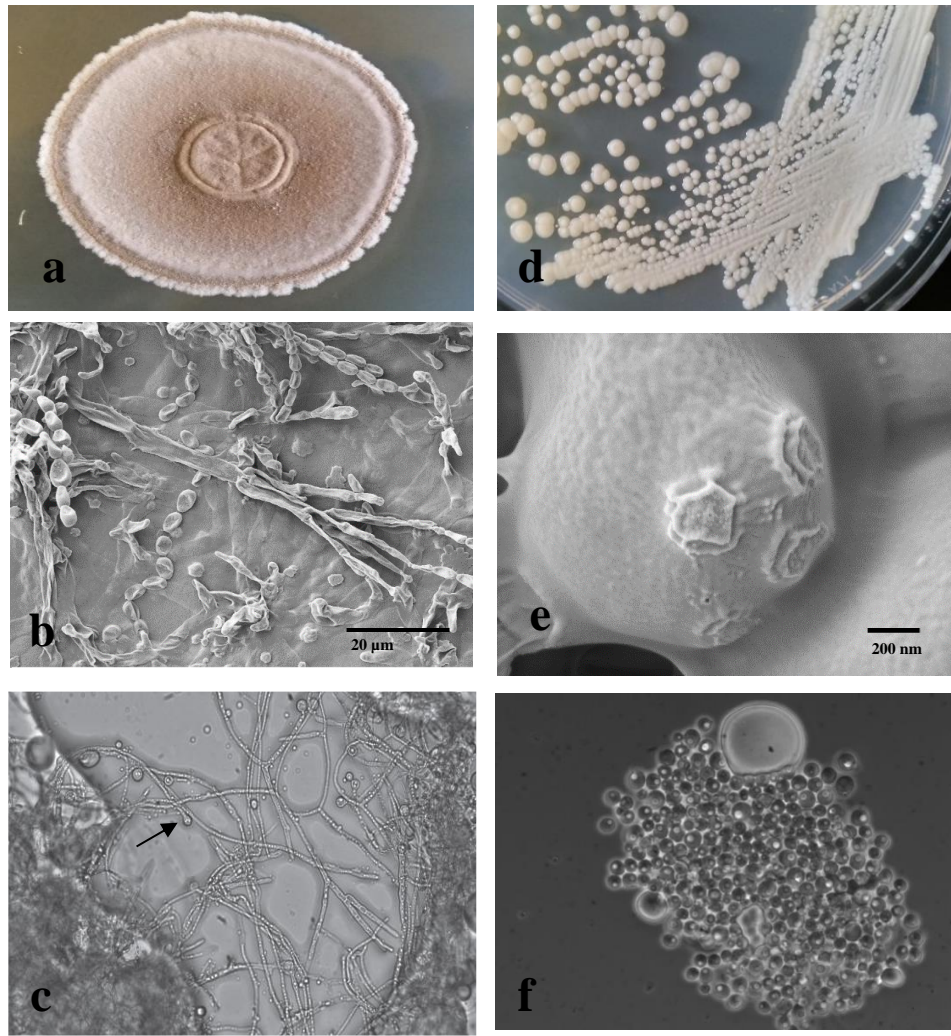


Figure 2.1 Morphology of *Byssochlamys sp. SW2* and *Wickerhamomyces anomalus SE3*. *Byssochlamys sp. SW2* (a) colony morphology agar and (b) conidia and conidiophore morphology (1.00 kX mag) after 5 days growth on HS are shown. Panel (c) shows terminal chlamydospores (arrow) and hyphal morphology using phase contrast microscopy (400x mag) after incubation (80 d) in B20 minimal medium. *Wickerhamomyces anomalus SE3* (d) colony morphology and (e) cellular morphology with budding scars (24.08 kX mag) after 3 days of growth on HS agar. Panel (f) shows cellular morphology using phase contrast microscopy (1000 X mag) after 15 days of growth in B20 minimal medium.

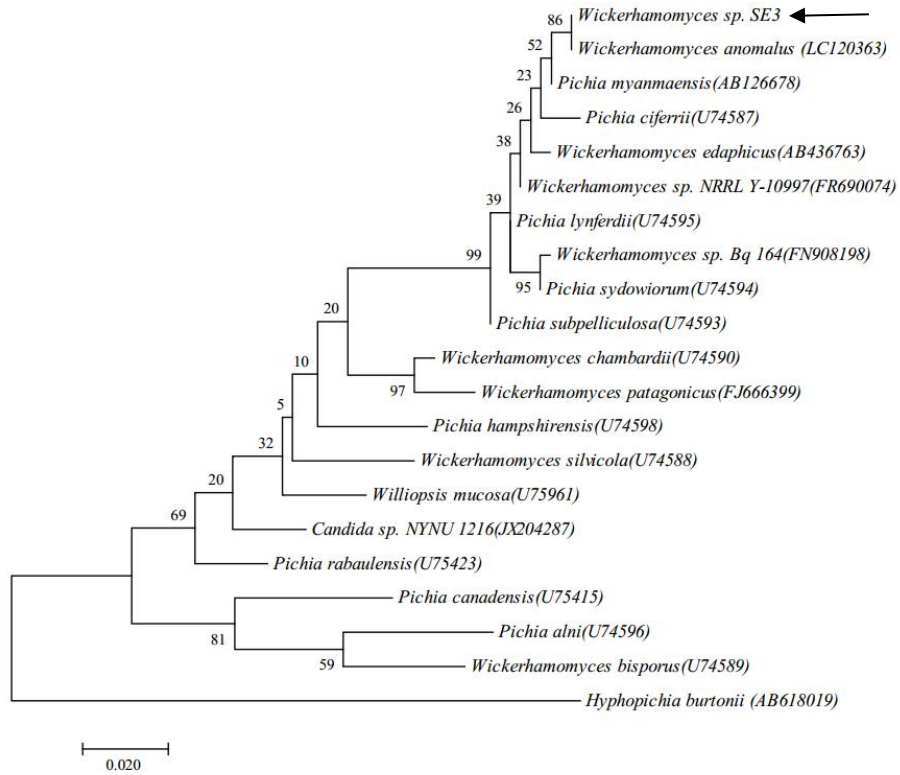


Figure 2.2 Maximum Likelihood tree based on 28S rRNA sequence phylogeny of the yeast *Wickerhamomyces anomalous* SE3 and its close relatives. Phylogeny is based on sequences of the D1/D2 domains of the 28S rRNA genes for the isolate (arrow) and selected closest relatives. Sequence for *Hyphopichia burtonii* was used as the outgroup, values of a bootstraps analysis >50% (500 replicates) are displayed at each node, and the scale bar represents 0.02 changes per nucleotide position.

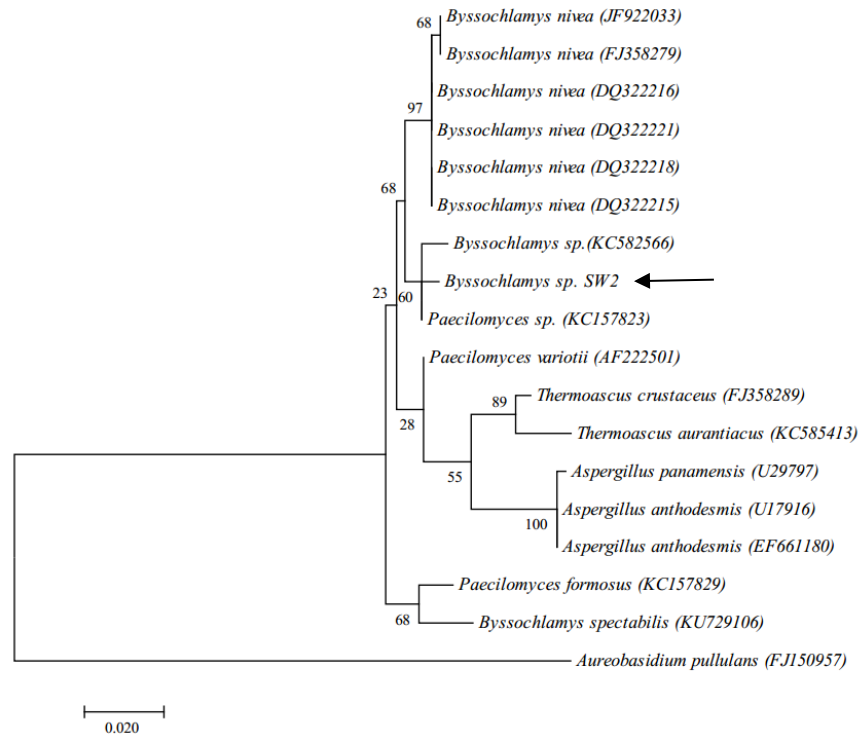


Figure 2.3 Maximum Likelihood tree based on 28S rRNA sequence phylogeny of the filamentous fungus *Byssochlamys sp. SW2* and its close relatives.

Phylogeny is based on sequences of the D1/D2 domains of the 28S rRNA genes for the isolate (arrow) and selected closest relatives. Sequence for *Aureobasidium pullulans* was included as the outgroup. Bootstrap values >50% (1000 replicates) are displayed on supported nodes. The error bar represents 0.02 changes per nucleotide position.

Table 2.2 Physiological characteristics of *Wickerhamomyces anomalus* SE3.

Tween 80	-	Salicin	w
N-Acetyl-DGalactosamine	-	Sedoheptulosan	-
N-Acetyl-DGlucosamine	-	D-Sorbitol	-
N-Acetyl-DMannosamine	-	L-Sorbose	-
Adonitol	-	Stachyose	-
Amygdalin	-	Sucrose	+
D-Arabinose	-	D-Tagatose	-
L-Arabinose	-	D-Trehalose	-
D-Arabitol	+	Turanose	+
Arbutin	+	Xylitol	-
D-Cellobiose	+	D-Xylose	-
α -Cyclodextrin	-	γ -Amino-butyric Acid	+
β -Cyclodextrin	-	Bromosuccinic Acid	-
Dextrin	-	Fumaric Acid	+
i-Erythritol	+	β -Hydroxy-butyric Acid	-
D-Fructose	+	γ -Hydroxy-butyric Acid	+
L-Fucose	-	p-Hydroxyphenylacetic Acid	-
D-Galactose	-	α -Keto-glutaric Acid	+
D-Galacturonic Acid	-	D-Lactic Acid Methyl Ester	+
Gentiobiose	w	L-Lactic Acid	+
D-Gluconic Acid	-	D-Malic Acid	+
D-Glucosamine	-	L-Malic Acid	+
α -D-Glucose	+	Quinic Acid	+
Glucose-1-Phosphate	+	D-Saccharic Acid	-
Glucuronamide	-	Sebacic Acid	-
D-Glucuronic Acid	-	Succinamic Acid	+
Glycerol	+	Succinic Acid	+
Glycogen	-	Succinic Acid Mono-Methyl Ester	-
m-Inositol	-	N-Acetyl-L-Glutamic Acid	-
2-Keto-D-Gluconic Acid	-	Alaninamide	-
α -D-Lactose	-	L-Alanine	+
Lactulose	-	L-Alanyl-Glycine	-
Maltitol	+	L-Asparagine	+
Maltose	+	L-Aspartic Acid	+
Maltotriose	+	L-Glutamic Acid	+
D-Mannitol	+	Glycyl-L-Glutamic Acid	-
D-Mannose	w	L-Ornithine	w
D-Melezitose	+	L-Phenylalanine	w
D-Melibiose	-	L-Proline	+
α -Methyl-DGalactoside	-	L-Pyroglutamic Acid	-
β -Methyl-DGalactoside	-	L-Serine	w
α -Methyl-D-Glucoside	+	L-Threonine	-
β -Methyl-D-Glucoside	+	2-Amino Ethanol	-
Palatinose	+	Putrescine	-
D-Psicose	-	Adenosine	-
D-Raffinose	-	Uridine	-
L-Rhamnose	-	Adenosine-5'-Monophosphate	-
D-Ribose	-		
Growth on/at:			
Vitamin-free medium	+	pH 3	+
9°C	+	pH 10	-
37°C	+	10% NaCl 5% Glucose	+
40°C	+	50% Glucose	+
45°C	-		
Major Fatty Acids :			
	C16:0	15.50%	
	C18:2 _{ω6,9c} /C18:0 _{ANTE}	35.21%	
	C18:1 _{ω9c}	24.94%	
	C18:1 _{ω7c}	15.55%	

Symbols denote assimilation +, Positive; -, Negative; w, Weak.

Table 2.3 Physiological characteristics^a of *Byssochlamys sp.* SW2.

Tween 80	w	Salicin	+
N-Acetyl-DGalactosamine	-	Sedoheptulosan	-
N-Acetyl-DGlucosamine	-	D-Sorbitol	+
N-Acetyl-DMannosamine	w	L-Sorbose	+
Adonitol	-	Stachyose	-
Amygdalin	+	Sucrose	+
D-Arabinose	-	D-Tagatose	-
L-Arabinose	+	D-Trehalose	+
D-Arabitol	`	Turanose	+
Arbutin	+	Xylitol	-
D-Cellobiose	+	D-Xylose	+
α -Cyclodextrin	-	γ -Amino-butyric Acid	+
β -Cyclodextrin	+	Bromosuccinic Acid	-
Dextrin	+	Fumaric Acid	-
i-Erythritol	+	β -Hydroxy-butyric Acid	-
D-Fructose	+	γ -Hydroxy-butyric Acid	-
L-Fucose	-	p-Hydroxyphenylacetic Acid	+
D-Galactose	+	α -Keto-glutaric Acid	-
D-Galacturonic Acid	+	D-Lactic Acid Methyl Ester	-
Gentiobiose	+	L-Lactic Acid	w
D-Gluconic Acid	+	D-Malic Acid	-
D-Glucosamine	w	L-Malic Acid	-
α -D-Glucose	+	Quinic Acid	+
Glucose-1-Phosphate	-	D-Saccharic Acid	-
Glucuronamide	-	Sebacic Acid	+
D-Glucuronic Acid	-	Succinamic Acid	-
Glycerol	-	Succinic Acid	+
Glycogen	+	Succinic Acid Mono-Methyl Ester	+
m-Inositol	+	N-Acetyl-L-Glutamic Acid	-
2-Keto-D-Gluconic Acid	+	Alaninamide	-
α -D-Lactose	+	L-Alanine	+
Lactulose	+	L-Alanyl-Glycine	w
Maltitol	+	L-Asparagine	-
Maltose	+	L-Aspartic Acid	-
Maltotriose	+	L-Glutamic Acid	+
D-Mannitol	+	Glycyl-L-Glutamic Acid	-
D-Mannose	+	L-Ornithine	-
D-Melezitose	-	L-Phenylalanine	-
D-Melibiose	-	L-Proline	+
α -Methyl-DGalactoside	-	L-Pyroglutamic Acid	-
β -Methyl-DGalactoside	-	L-Serine	-
α -Methyl-D-Glucoside	w	L-Threonine	-
β -Methyl-D-Glucoside	+	2-Amino Ethanol	-
Palatinose	+	Putrescine	w
D-Psicose	-	Adenosine	-
D-Raffinose	w	Uridine	-
L-Rhamnose	+	Adenosine-5'-Monophosphate	-
D-Ribose	-		
Growth on/at:			
Vitamin-free medium	+	pH 3	+
9°C	-	pH 9	-
37°C	+	10% NaCl 5% Glucose	+
40°C	+	50% Glucose	+
45°C	-		
Major Fatty Acids :			
	C16:0		11.34%
	C18:2 _{ω6,9} /C18:0 _{ANTE}		43.86%
	C18:1 _{ω9c}		21.99%

^a symbols denote assimilation +, Positive; -, Negative; w, Weak.

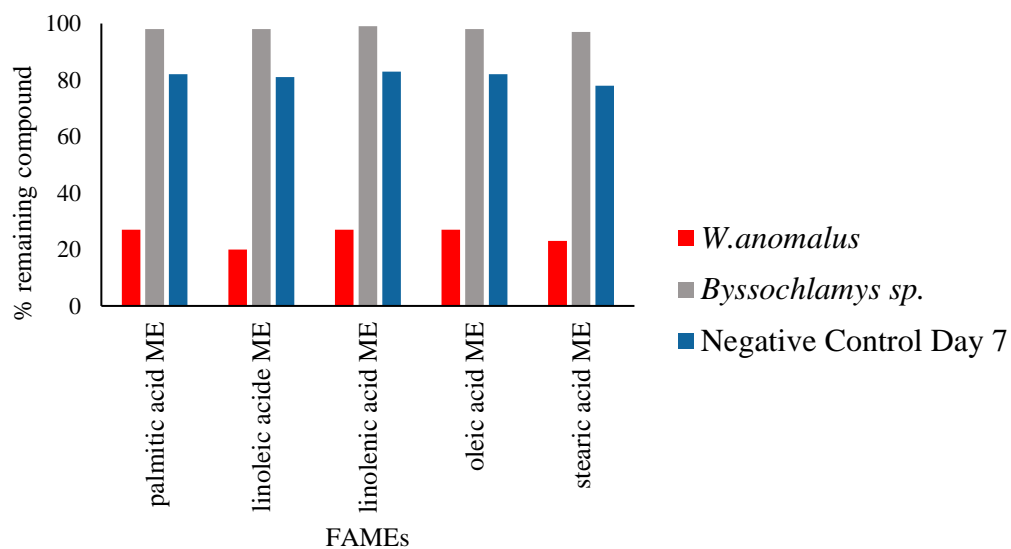


Figure 2.4 Degradation of FAME in B20 biodiesel after 7 days of incubation with isolates *Wickerhamomyces anomalus* SE3 and *Byssochlamys sp.* SW2. Colored bars represent the remaining percentage of compound after 7 days of incubation compared to negative control at day 0.

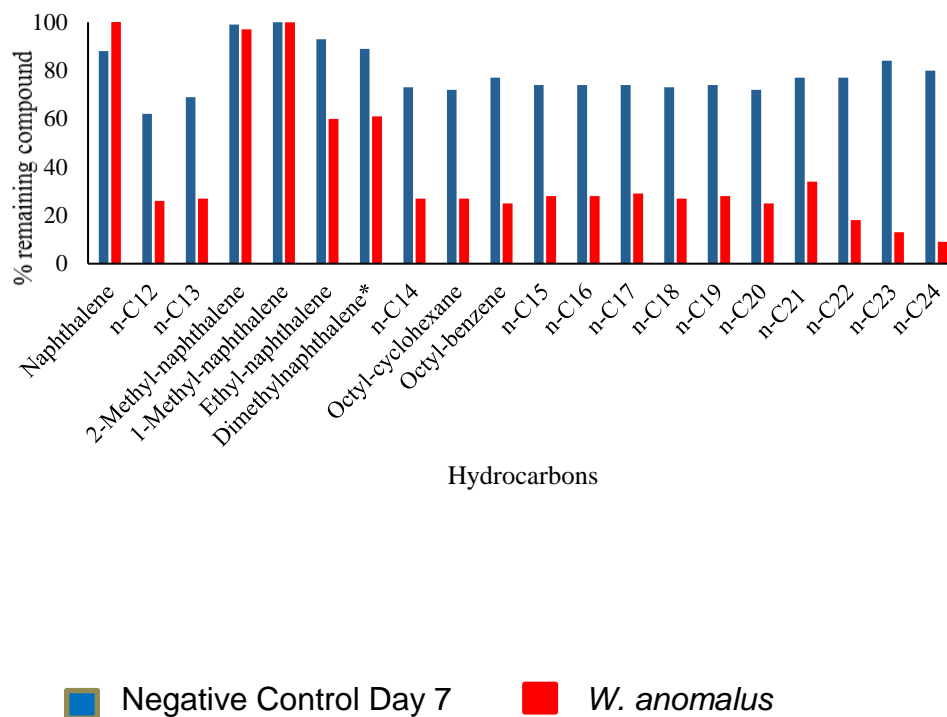


Figure 2.5 Degradation of hydrocarbons in B20 biodiesel after 7 days of incubation with isolate *Wickerhamomyces anomalus* SE3. Colored bars represent the remaining percentage of compound after 7 days of incubation compared to negative control at day 0.

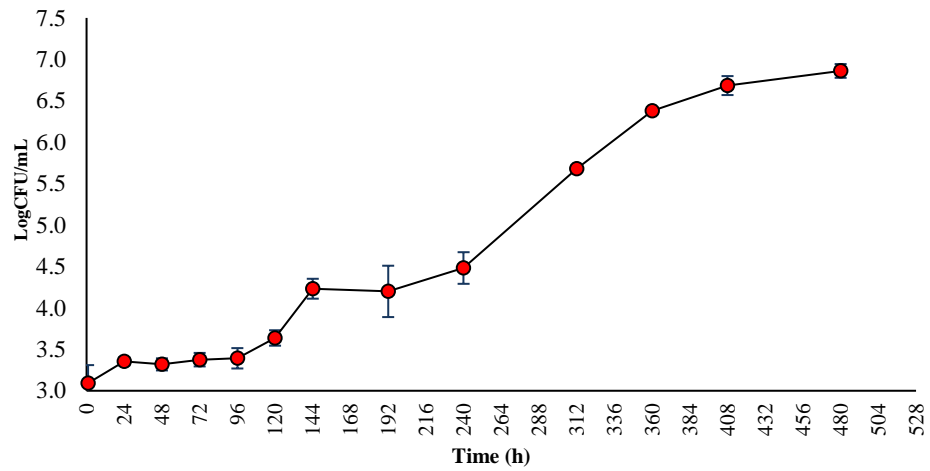


Figure 2.6 Growth curve for *Wickerhamomyces anomalus* SE3 in ASW medium containing B20 biodiesel as sole carbon source. Error bars represent standard deviation of triplicates.

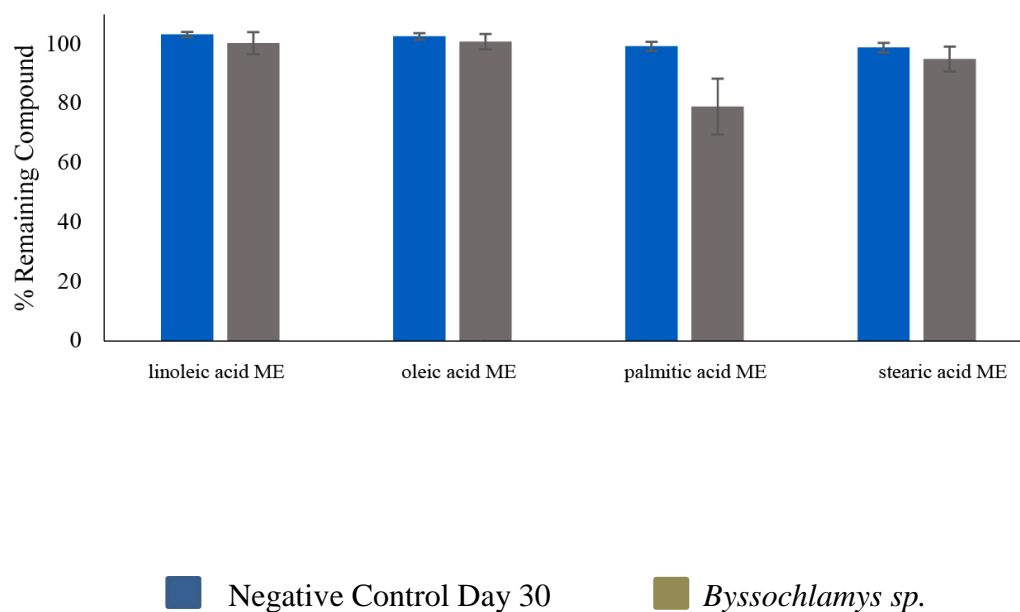


Figure 2.7 Degradation of FAME in B20 biodiesel after 30 days of incubation with isolate *Byssochlamys sp.* SW2. Colored bars represent the remaining percentage of compound after 30 days of incubation compared to negative control at day 0. Error bars represent standard deviation of triplicates.

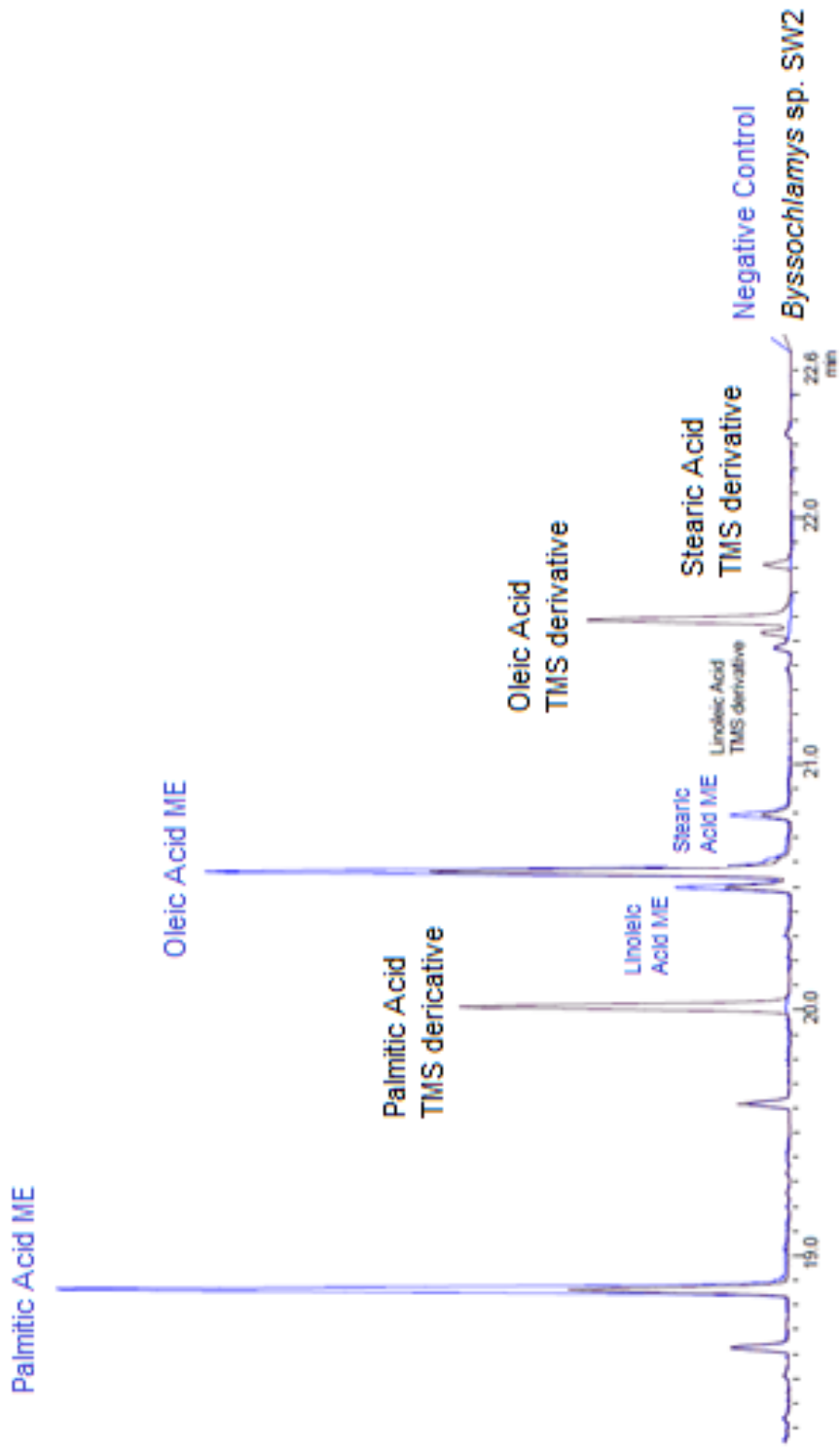


Figure 2.8 Representative total ion chromatograms of aqueous phase after 80 days of biodegradation of B20 with isolate *Byssoschlamys* sp. SW2. Biodegraded sample (black) is compared to negative control (blue).

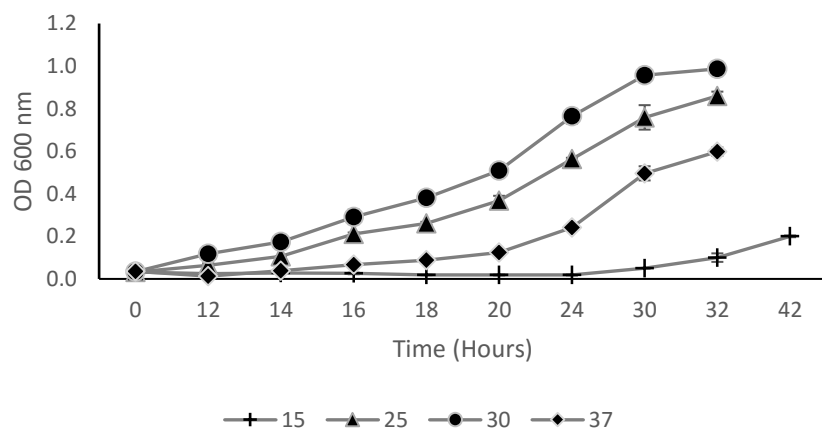


Figure 2.9 Effect of temperature on growth of the yeast *Wickerhamomyces anomalus* SE3. p-value <0.01 between all treatments (Anova-Tukey Test). Error bars represent standard deviation of triplicate samples.

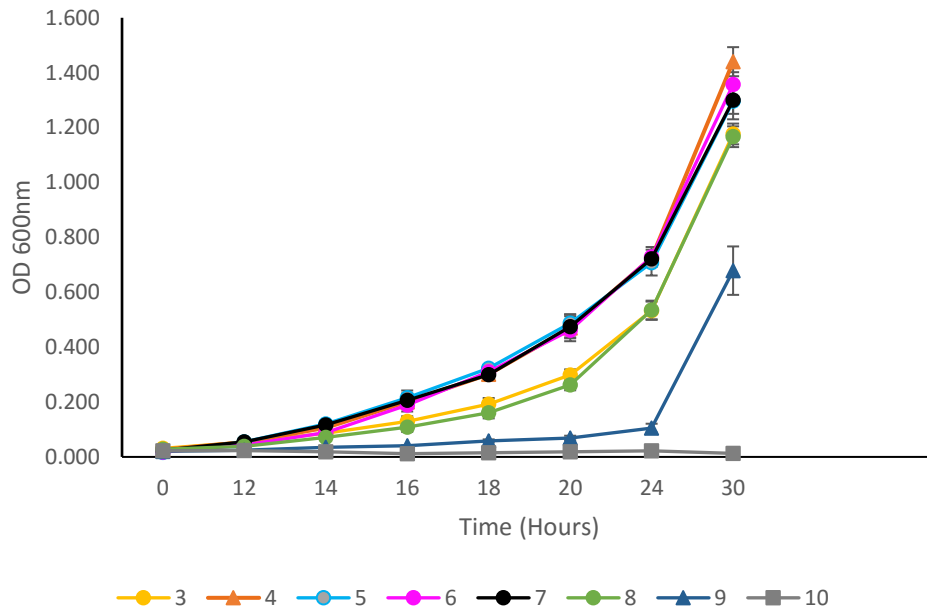


Figure 2.10 Effect of pH on growth of the yeast *Wickerhamomyces anomalus* SE3. Significant differences were found between treatments (p-value = 0.03. According to Tukey's test, pH 4 is the treatments that differs from pH 5 (*p-value<0.05). The other combinations does not differ between each other significantly. Error bars represent standard deviation of triplicate samples.

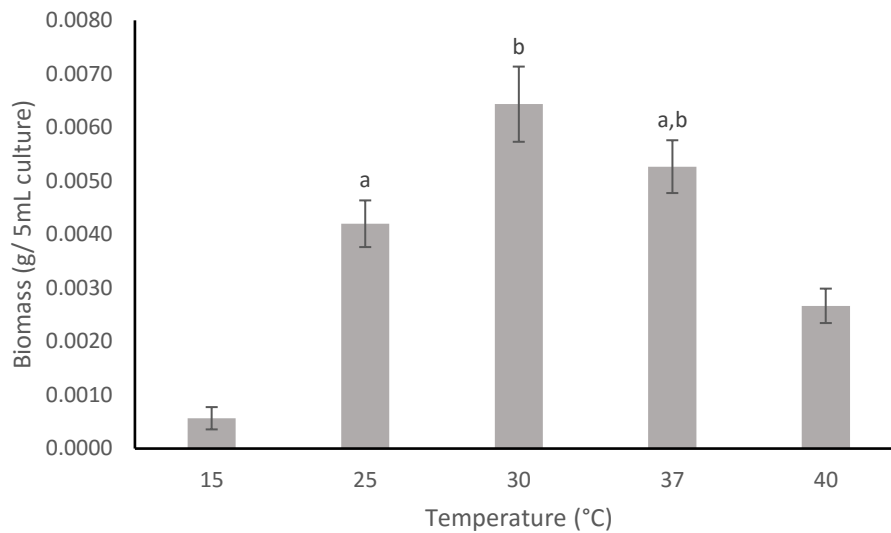


Figure 2.11 Effect of temperature on growth of *Byssochlamys* sp. SW2. According to ANOVA-Tukey's HSD test, treatments sharing letter represent: a, not significantly different; b, p-value<0.05. No shared letters represent p-value<0.01.

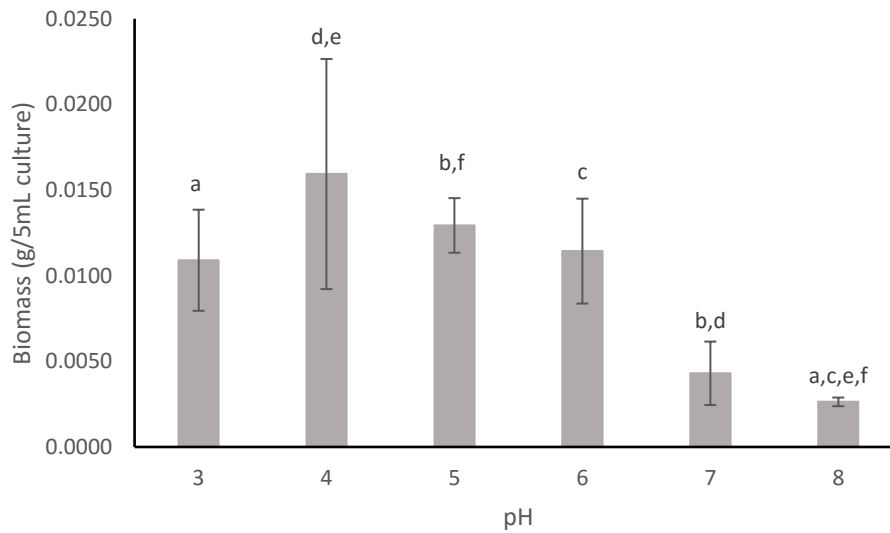


Figure 2.12 Effect of pH on growth of the *Byssochlamys* sp. SW2. According to ANOVA-Tukey's HSD test, shared letters denote significance: a-c, p-value between 0.05-0.1; d-f, p-value <0.05.

Chapter 3: Chemical Analysis of B20 Biodiesel Fuel Exposed to Contaminated Underground Storage Tanks and its Correlation to Fungal Biodegradation

3.1 Introduction

B20 biodiesel is an alternative fuel considered to be a suitable option to replace petroleum diesel as a source of liquid transportation fuel (Hoekman, Broch, Robbins, Cenicerros, & Natarajan, 2012; Knothe, Krahl, & Van Gerpen, 2015). The DoD considers B20 biodiesel to be a means to reduce its carbon footprint and achieve energy security (DOD, 2011; GAO, 2015). During long term; however, storage the fuel can be susceptible to microbial contamination (Bücker et al., 2011; Lee, Ray, & Little, 2010) and oxidative instability (Jakeria et al., 2014). The negative effects associated with these problems include fuel degradation (Zuleta, Baena, Rios, & Calderón, 2012) and damage to the infrastructure due to fouling (Passman, 2013) and corrosion (Lee et al., 2010; Zuleta et al., 2012). The maintenance costs related with these problems can quickly surpass the savings of using biodiesel (Chavez, 2013), which ultimately threaten its broader use.

The fuel B20 is a blend of 20% biodiesel and 80% ultralow sulfur diesel (ASTM-D7467-15, 2015). The biodiesel portion is a mixture of fatty acid methyl esters (FAME) produced by chemical transesterification of raw materials rich in triglycerides (ASTM-D7467-15, 2015; NREL, 2009). The feedstocks for biodiesel

production can be obtained from edible and non-edible sources including vegetable oils, animal fats, waste grease, and even some microalgae species (Hoekman et al., 2012). The triglyceride content of the parental feedstock determines the FAME mixture in the final product (Jakeria, Fazal, & Haseeb, 2014; Knothe, 2005). The most common FAMEs present in biodiesel are composed of 16 to 18 carbons and varying degrees of unsaturation (Table 3.1).

Biodiesel and its blends are more easily degraded by abiotic and biotic mechanisms than ULSD (Bücker et al., 2011; Mariano, Tomasella, De Oliveira, Contiero, & De Angelis, 2008; Zhang et al., 1998), especially during fuel storage. Autoxidation is the abiotic mechanism that is considered to be of primary concern during long term storage (Pullen & Saeed, 2012), however it can be controlled by the addition of antioxidants (Pullen & Saeed, 2012). Fuel biodegradation refers to changes in the fuel composition as the consequence of microbial metabolism, which affects the integrity and properties of the fuel (Pullen & Saeed, 2012). Fuel biodegradation depends on the microbial contaminants and their potential to accelerate degradation processes (Jakeria et al., 2014; Makareviciene & Janulis, 2003), as well as on the fuel chemistry, environmental conditions, presence of fuel additives, storage and handling conditions and the design of the fuel tanks and delivery systems (Passman, 2013).

The evaluation of biodegradation problems in a fuel system is a complex process that requires both the fuel and infrastructure be examined and monitored (Knothe et

al., 2015). An initial examination is usually done by gross observations for signs of microbial contamination such as, changes in fuel color and turbidity, the presence of water, or accumulation of particulates (Passman, 2003). A thorough analysis is necessary to understand the extent of the problem and to define the strategies for an effective control. For further root-cause analysis and monitoring, a correlation between data obtained by physical, chemical and microbiological analyses is required (Hoekman et al., 2012; Passman, 2003).

Recent studies have suggested a correlation between the presence of the fungal genus *Byssochlamys* (family *Trichocomaceae*) and recurrent problems of discolored fuel, presence of flocculent material, and fouling at two USAF facilities (Stamps, 2016). However, the impact of the contamination on the fuel was unknown. We hypothesized that *fungus contamination of USAF B20 fuel tanks changes the chemical composition of the B20 biodiesel*. This hypothesis was tested by measuring the effect of the growth of *Byssochlamys sp. SW2* on the chemical composition of B20 fuel in cultivation experiments. A statistical model was then developed to predict fuel degradation for quality monitoring purposes. To validate our model, we characterized B20 biodiesel samples from contaminated storage tanks at two USAF facilities.

3.2 Materials and Methods

3.2.1 Sampling and GC/MS data collection

Samples of B20 biodiesel were collected from six underground storage tanks at two USAF facilities in the Southwest (SW2, SW3, and SW4) and Southeast (SE3, SE4, and SEE) United States. Samples (1 L) were collected at multiple levels over an 18-month period to measure the temporal variation in fuel composition. A total of sixty samples were collected (17 from SW and 43 from SE) into sterile bottles using a fuel sampler (Koehler Instrument Company, Inc., Holtsville, New York) and transported at room temperature in the dark. The fuel was sterilized by filtration using 0.22 μm polyether sulfone bottle top filters (EMD Millipore, Billerica, MA), prior to chemical analysis. Triplicate 2 mL aliquots of each sample were stored in gas chromatography vials at room temperature in the dark.

The chemical composition of each B20 biodiesel sample was determined by Gas Chromatography/Mass Spectrometry (GC/MS) using a Shimadzu QP 2010 SE (Shimadzu Corporation, USA). Each sample was diluted 1:200 with hexane (Chromasolv[®], for HPLC, >95%, Sigma Aldrich) prior to injection. A volume of 1 μL was injected via autosampler with a split ratio of 1:10. Injection started at 300 $^{\circ}\text{C}$, oven at 40 $^{\circ}\text{C}$ with a 0.5 min hold and increased to 320 $^{\circ}\text{C}$ at a rate of 10 $^{\circ}\text{C min}^{-1}$. Peaks were separated with a Restek Column Rxi 5Sil with dimensions: 30 m, 0.25 mm ID, 0.25 μm . High purity helium was used as carrier gas at a linear

velocity of 36.8 cm s⁻¹. Mass spectra were analyzed in scan mode with the following conditions: interface at 320 °C, ion source 200 °C, solvent cut of 2 min, event time of 0.25 sec and scan speed of 2000.

3.2.2 Chromatographic Data Analysis

Each Total Ion Chromatogram (TIC) was processed using the software LabSolutions version 4.20 (Shimadzu Corporation, USA). Peaks were identified using the mass spectra library NIST version 14. Reference standards for FAME (Supelco® 37 Component FAME Mix, Sigma Aldrich, USA) and B20 (Diesel:Biodiesel (80:20) Blend Standard, RESTEK, USA) were used to confirm major alkanes and FAME peaks identified by the NIST library. Qualitative integration of peak areas was performed using a common detection sensitivity (slope: 1000/min) to distinguish peaks from background/noise. Only well-defined peaks with more than 90% similarity with the NIST library were chosen for further analysis. A similarity index of 100 was used when the spectra were perfectly identical. Individual peaks were quantified using the area normalization method following the equation $i\% = \frac{A_i}{\sum A_i} \times 100$ where A_i is the peak area of a compound and $\sum A_i$ is the sum of peak area of all components (Linskens et al., 1986; Qi et al., 2011).

Data was analyzed in R version 3.2.2, using the Stats (Team, 2014), Caret (Kuhn, 2008), Mass (Venables & Ripley, 2002) and Vegan (Oksanen et al., 2015) packages.

An ANOVA-based variable selection was applied to highlight the compounds with relevant differences among the samples. Each compound was treated as a one-way ANOVA type problem with N measurements represented by the 60 samples and K groups represented by the two facilities (SE and SW). Compounds with p-values <0.05 were selected for multivariate analysis using Permutational Multivariate Analysis of Variance (PERMANOVA) and Principal Component Analysis (PCA).

3.2.3 Fungal Biodegradation Classification Model

The results from biodegradation experiments of B20 with *Byssochlamys* sp. SW2 were used to construct a Linear Discriminant Analysis (LDA) classification model with prediction properties. Biodegraded and non-biodegraded datasets were defined as classes for discrimination. For the biodegraded group, four samples of B20 fuel that were retained prior to addition to tanks, that were never exposed to the biology in the tanks, were individually exposed during 30 days of incubation to fungal biodegradation using *Byssochlamys* sp. SW2. The non-biodegraded group consisted of negative controls with no inoculum for each fuel. Samples from tanks SE3, SE4 and SEE were used as test sets to validate the model (Table 3.4). In the test set we included two samples representing a pre- and post-cleaning procedure applied to the tanks after a contamination event (Table 3.4).

Experimental conditions included destructive sampling at days 0, 5, 14 and 30 of triplicates for each experimental tube and control, to simulate a fuel biodegradation

progression. Each experiment was set up in 16 x 100 mm glass tubes filled with 1 mL of ASW and 10 μ L of filter-sterilized B20 as the sole carbon and energy source, sealed with vials with a cap that had a PTFE liner to avoid evaporation. An inoculum resulting in 1×10^6 spores mL^{-1} of *Byssochlamys* was added to each tube and incubated at 25°C, shaking at 250 rpm. Un-amended controls containing no electron donor were included to evaluate carbon source carryover. At the end of each incubation period, the fuel phase was extracted and a chromatographic analysis was performed (see above, Section 3.2.2).

3.3 Results

3.3.1 Characterization of the Composition of B20 Fuel Samples

B20 fuel samples exposed to bio-contaminated tanks were analyzed by GC/MS. The TIC of each sample showed a complex mixture of compounds (Fig. 3.1). Alkanes and FAMES were detected in all B20 fuel samples. Well-defined peaks with a similarity index of >90% to the NIST library were identified (Table 3.2). The identified peaks were well spread over the retention time range in the chromatogram (Table 3.2) in accordance with the reference standards. Minor peaks that were not qualified by the selection criteria were discarded from further analysis and included various branched alkanes, naphthalenes and aromatics. Sixteen major fuel components including alkanes of various chain lengths and FAMES were retained.

Patterns of variation in the dataset were identified using a combined approach of variable selection and multivariate analysis. Nine peaks were highlighted with significant differences among samples (Table 3.2), using a one-way ANOVA approach. Fuel samples between SE and SW (GC/MS data 1-factor PERMANOVA, $F=25.32$, $R^2=0.30$, $p=0.001$) showed significant differences in their composition of the selected variables. Geographic location appeared as a discrimination factor (Fig. 3.2). Linoleic and palmitic acid methyl esters (loadings = 0.4445 and -0.4388 respectively) accounted for most of the discrimination between locations SE and SW, with an indirectly proportional trend in their concentrations (Table 3.3).

Variations within location SE were also investigated (Fig. 3.2). Fuel composition between tanks at SE3, SE4 and SEE varied significantly (GC/MS data 1-factor PERMANOVA, $F=3.8017$, $R^2=0.21$, $p=0.013$). The discrimination was based on the first three principal components (Table 3.3), capable of explaining 75% of the variability in the dataset.

Four samples of B20 obtained at SE and never exposed to the biology in the tanks showed differences in their proportion of FAMES (Fig. 3.3). Unexposed fuels A and C showed similar relative abundances of linoleic and oleic acid methyl esters, while B and D had similar abundances of palmitic and oleic acid methyl ester (Fig. 3.3). Unexposed fuels were discriminated according to their patterns and clustered in two separated groups (Fig. 3.2).

3.3.2 Fungal Biodegradation Model

An LDA model constructed with biodegraded and non-biodegraded datasets allowed us to predict the severity of biodegradation in B20 fuel samples. The variables used to construct the model were identical to those used for the characterization of *in situ* B20 fuel samples (Table 3.2). The model consisted of three clusters that showed separation between each other (Figure 3.4), corresponding to the length of incubation and a progression from non-degraded (days 0, 5), degraded at day 14, and the most degraded group at day 30. The linear discriminant LD1 accounted for 97.1% of the total discrimination, responsible for

the observed separation. The largest linear discriminant coefficients were Linoleic and Palmitic Acid ME (-16.969, -12.964 respectively).

Samples of B20 fuel obtained from SE (Table 3.4) were objectively classified by the apparent severity of their fungal biodegradation. Nine samples including an unexposed fuel control were classified in the non-degraded group (Figure 3.5). These samples corresponded to fuels obtained from different locations (nozzle, middle and bottom) inside tanks SE4 and SEE, and differences between them were not detectable. Four samples obtained from the bottom of tank SE3 (sample numbers 27, 30, 16, 36; Table 3.4) were classified as degraded. Samples 27 and 30 were similar to fuels exposed to *Byssochlamys sp.* SW2 for 14 and 30 days. Sample 30 was obtained in March 2015 and sample 27 in the month of May 2015. Both represent the pre- cleaning stage of tank 3 where contamination was highest. Sample 16 represents the post-cleaning state of tank 3 in the Month of May 2015 and it was classified as degraded but in the group of Day 14. Sample 36 was classified as the most degraded since it most closely related to the degraded sample at day 30. These results represented a demonstration of the predictive properties of the LDA model.

3.4 Discussion

We hypothesized that *fungus contaminants of USAF B20 biodiesel fuel tanks metabolize components of the fuel, causing changes in its chemical composition.*

The stability and properties of B20 are influenced by its composition (Ghazali, Mamat, Masjuki, & Najafi, 2015), which can be affected during its storage. We detected variation in fuel components in B20 fuel samples obtained from underground storage tanks with recurrent problems of fungal contamination. Our results suggested that *the observed differences are the result of different FAME feedstocks that are susceptible to fungal biodegradation.* These results are important to assessing the risk of fungal growth on particular feedstocks used to produce B20 blends.

Biodiesel blends can contain over 2000 compounds (Marchal, Penet, Solano-Serena, & Vandecasteele, 2003) that can be characterized by GC/MS (Pauls, 2011). Pattern recognition methods (Brereton, 2009; Hochkirchen, 2010; Johnson & Synovec, 2002; Sutro, 1971; Wongravee et al., 2009) have been successfully used in the discrimination of fuel types (Flood, Goding, O'Connor, Ragon, & Hupp, 2014), identification of fuel adulteration (Skrobot, Castro, Pereira, Pasa, & Fortes, 2007) and monitoring of fuel degradation (Johnson, Rose-Pehrsson, & Morris, 2004). We found 9 compounds that change significantly among samples and used them to evaluate how fuel changes can be altered by fungal biodegradation using

multivariate techniques. This allowed us to identify differences in fuel composition *in situ* and monitor degradation *in vivo*.

Fuel components varied among facilities and tanks within facilities (Fig. 3.2). The unexposed fuels varied in their content of FAMEs, suggesting the use of different feedstocks in biodiesel production (Jakeria et al., 2014). Feedstock availability and production capacity directly affect the supply chain (DOD, 2007). It is reasonable to expect different FAME profiles between or within different storage facilities. The provenance of the feedstock oil was not known or available, but the source of feedstock could potentially explain the differences observed between facilities. Common biodiesel feedstocks in the U.S are soybean and canola (EIA, 2016b), and have high concentrations of mono- and poly-unsaturated fatty acids (Knothe, 2008). Other well-known feedstocks are palm oil and tallow (EIA, 2016b), which contain abundant saturated fatty acids (Knothe, 2008).

We identified the FAMEs Palmitic (C16:0) and Linoleic Acid (C18:2) ME as the compounds that accounted for most of the variation among clusters. Interestingly, Palmitic and Linoleic Acid ME explained most of the variation in the fungal community structure (Stamps, 2016), suggesting a selective biodegradation. The genus *Byssochlamys* was an abundant OTU in the fungal community (Stamps, 2016). Because *Byssochlamys sp.* can use B20 as sole source of carbon and energy (Andrade, Chapter 2), we used it to investigate selective biodegradation.

An LDA (Hochkirchen, 2010) was constructed using GC/MS chemical profile data of the fuel component of the B20 biodegradation experiments with *Byssochlamys sp.* SW2 (Fig. 3.4). Palmitic and linoleic acid ME corresponded to the largest and negative linear coefficients (Brereton, 2009), which indicate a preferential degradation of palmitic and linoleic acid ME by this organism. This is consistent with other studies that show that *Byssochlamys sp.* and a close relative *Paecilomyces variotti*, can degrade palmitic acid ME in storage tanks (Andrade, Chapter 2; Curvelo, Almeida, Nunes, & Feitosa, 2011). New feedstocks rich in palmitic acid ME (Chen et al., 2015; Fazaeli & Aliyan, 2015) are considered a promising alternative to satisfy future biodiesel demands. In this study; however, we quite clearly show the susceptibility of palmitic acid ME in B20 biodiesel to microbial contamination during storage, especially by fungi of the genus *Byssochlamys*.

The LDA model was used to objectively classify samples of B20 fuel from storage tanks in service into groups and identify patterns of fungal biodegradation. Samples of B20 obtained from the bottom of these tanks were classified as biodegraded (Fig. 3.5). This prediction was reasonable considering that most of the microbial contaminants will thrive in the bottom of tanks due to water accumulation (Bento & Gaylarde, 2001) and only very limited mixing occurs in these tanks. The water-fuel interface is often the location of dense fungal biofilms and thus biodegradation of the fuel (Passman, 2003).




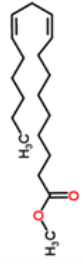
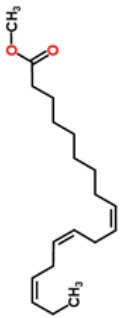
The LDA prediction of patterns of biodegradation appeared to be an effective tool to evaluate fungal contamination during fuel storage. In the test set, we detected a biodegraded sample from the bottom of tank SE3 (#36, Fig. 3.5) that contained only low amounts of adenosine triphosphate (ATP; 3.2×10^2 RLU, (Stamps, 2016)) measured *in situ*. Quantification of ATP is a standard method used to detect microbial contamination in fuel systems (ASTM:D7687, 2011), but the results can be deceiving if the values are low. For instance, fungal spores have $\ll 1$ fg ATP/spore (Passman, 2013), and their higher hydrophobicity can pull them to the fuel phase compared with mycelium that lies in the fuel-water interface (Linder, Szilvay, Nakari-Setälä, & Penttilä, 2005). Thus, when sampling fuel contaminated with fungi, it is possible that only fungal spores might be present (ASTM:D7687, 2011), generating a false negative result with low ATP measurements below the detection limit (Passman, 2013; Rakotonirainy, Heraud, & Lavédrine, 2003).

The LDA model was also useful in assessing the efficacy of a procedure used to remove water from the bottom of a storage tank (Figure 3.5). Samples obtained before the removal of water from SE3 tank were categorized as the most degraded (Table 3.4). Samples taken after removal of water (and some fuel); however, still indicated that biodegradation had occurred, but noticeably less. This data suggests that the biodegraded fuel is most closely associated with water and/or the bottom of the tank. Fuel higher up in the tank does not carry the same chemical profile that would be recognized as biodegradation. A storage tank with fungal contamination that affects the fuel composition, fouling, and potentially microbially influenced

corrosion may be asymptomatic according to fuel samples taken from the dispenser or top of the tank.

Chemical characterization of B20 biodiesel samples from contaminated underground storage tanks is an effective method for detection of past or current fungal contamination. This signal is largely due to the fungi altering the content of the FAMES palmitic and linoleic acid methyl ester. Biodiesel containing a significant proportion of these FAMES is more likely to support the growth of the fungi studied here and; therefore, more susceptible to fungal biodegradation. The research presented here not only provides a methodology that should be considered for monitoring B20 biodiesel for fungal contamination and proliferation, but also suggests that the feedstocks used in production of biodiesel should be reconsidered for their susceptibility to biodegradation.

Table 3.1 Common fatty acid methyl esters found in biodiesel ^a.

Common Name	Formal Name	CAS. Number	Abbreviation	Molecular Weight	Molecular structure ^b
Palmitic acid methyl ester	Hexadecanoic acid methyl ester	112-39-0	16:0	270.45	
Stearic acid methyl ester	Octadecanoic acid methyl ester	112-61-8	18:0	298.50	
Oleic acid methyl ester	(9Z)-9-Octadecenoic acid methyl ester	112-62-9	18:1	296.49	
Linoleic acid methyl ester	9,12-Octadecadienoic acid (Z,Z)-, methyl ester	112-63-0	18:2	294.47	
Linolenic acid methyl ester	(Z,Z,Z)-9,12,15-Octadecatrienoic acid, methyl ester	301-00-8	18:3	292.46	

a. Table adapted from Hoekman et al., 2012

b. Figures for the molecular structure were taken from <https://pubchem.ncbi.nlm.nih.gov>

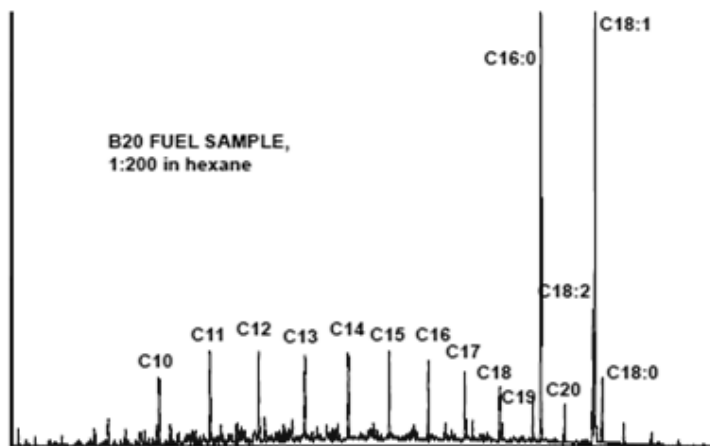


Figure 3.1 Representative total ion chromatogram (TIC) obtained from a B20 fuel sample. Major peaks for n-alkanes and FAMES are labeled.

Table 3.2 Retention times of major peaks identified in B20 fuel samples.

Variable	Compound	Retention Time (min)
n-C9 ^a	<i>n</i> -nonane	5.316 - 5.337
n-C11	<i>n</i> -undecane	8.502 - 8.525
n-C12	<i>n</i> -dodecane	10.030 - 10.053
n-C13 ^a	<i>n</i> -tridecane	11.473 - 11.497
n-C14 ^a	<i>n</i> -tetradecane	12.834 - 12.859
n-C16	<i>n</i> -hexadecane	15.335 - 15.361
n-C17	<i>n</i> -heptadecane	16.485 - 16.512
n-C18	<i>n</i> -octadecane	17.578 - 17.605
n-C19 ^a	<i>n</i> -nonadecane	18.617 - 18.645
n-C20 ^a	<i>n</i> -eicosane	19.609 - 19.637
C16:0 ^a	<i>n</i> -palmitic acid methyl ester	18.854 - 18.895
C18:2 ^a	<i>n</i> -linoleic acid methyl ester	20.496 - 20.594
C18:1 ^a	<i>n</i> -oleic acid methyl ester	20.564 - 20.594
C18:0 ^a	<i>n</i> -stearic acid methyl ester	20.738 - 20.811

^a Compounds selected for multivariate analysis.

Table 3.3 Principal component analyses (PCAs) of global B20 fuel dataset.

Order of importance	PC1 39%^a	PC2 20%^a	PC3 16%^a
1 st	Linoleic Acid ME (0.44)	n-C20 (-0.50)	n-C9 (0.51)
2 nd	Palmitic Acid ME (-0.43)	n-C19 (-0.46)	n-C14 (-0.50)
3 rd	Stearic Acid ME (0.38)	n-C13 (0.43)	Stearic Acid ME (0.42)
4 th	n-C19 (0.36)	n-C14 (0.32)	n-C13 (-0.29)
5 th	n-C13 (0.35)	Oleic Acid ME (0.28)	n-C20 (-0.27)
6 th	n-C14 (0.29)	Palmitic Acid ME (-0.24)	Palmitic Acid ME (-0.24)
7 th	Oleic Acid ME (-0.25)	n-C9 (0.23)	Oleic Acid ME (0.19)
8 th	n-C20 (0.22)	Stearic Acid ME (-0.15)	Linoleic Acid ME (0.16)
9 th	n-C9 (-0.023)	Linoleic Acid ME (0.13)	n-C19 (0.08)

^a Percentage correspond to the variation explained by the Principal Component.

Variable loadings of the three principal components (PC) are shown in parenthesis.

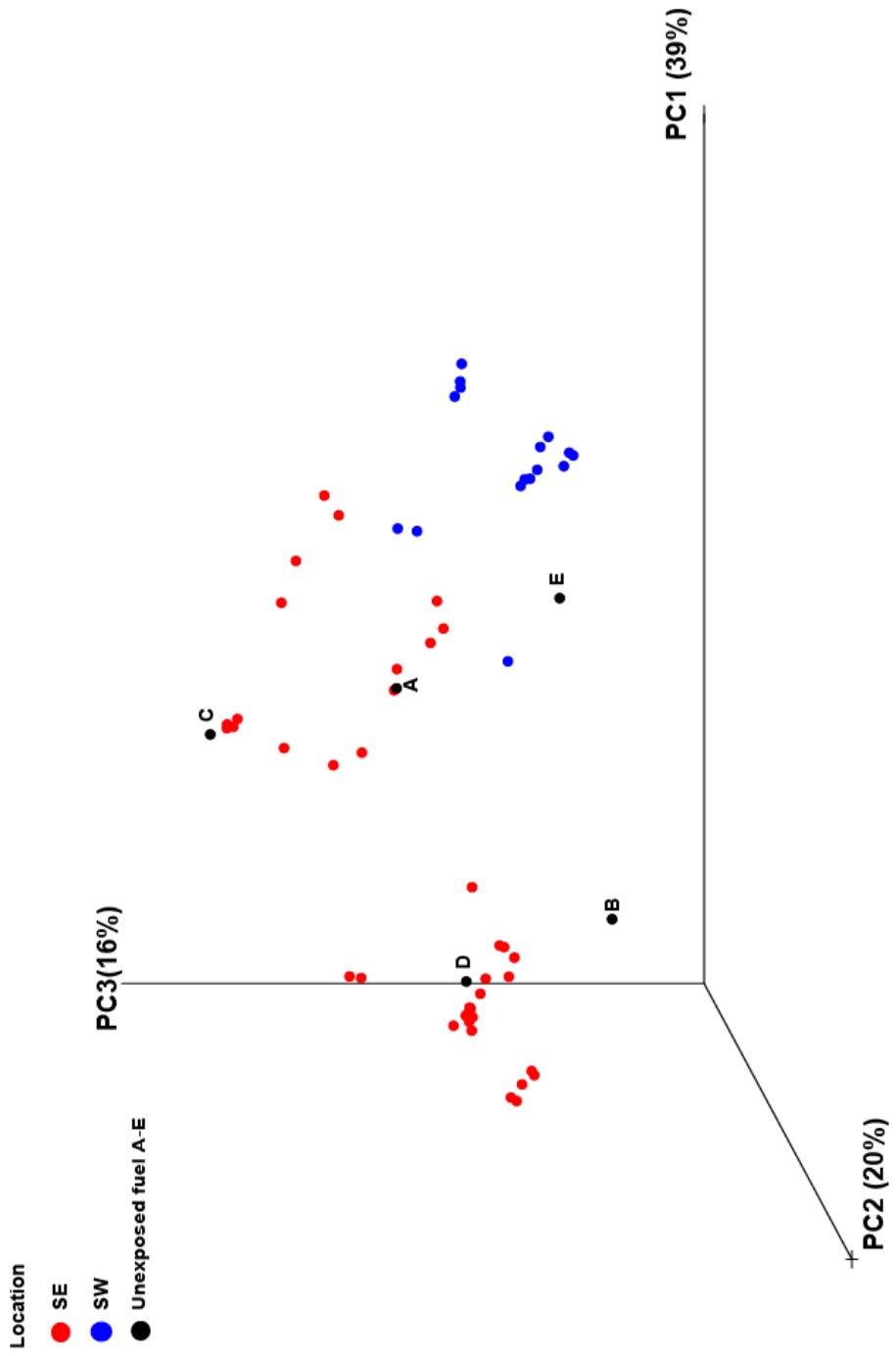


Figure 3.2 3-dimensional ordination of B20 fuel samples by principal components (PC) 1, 2 and 3. Each fuel sample from SE (red), SW (blue), and unexposed fuel (black, A-D) is represented by a circle.

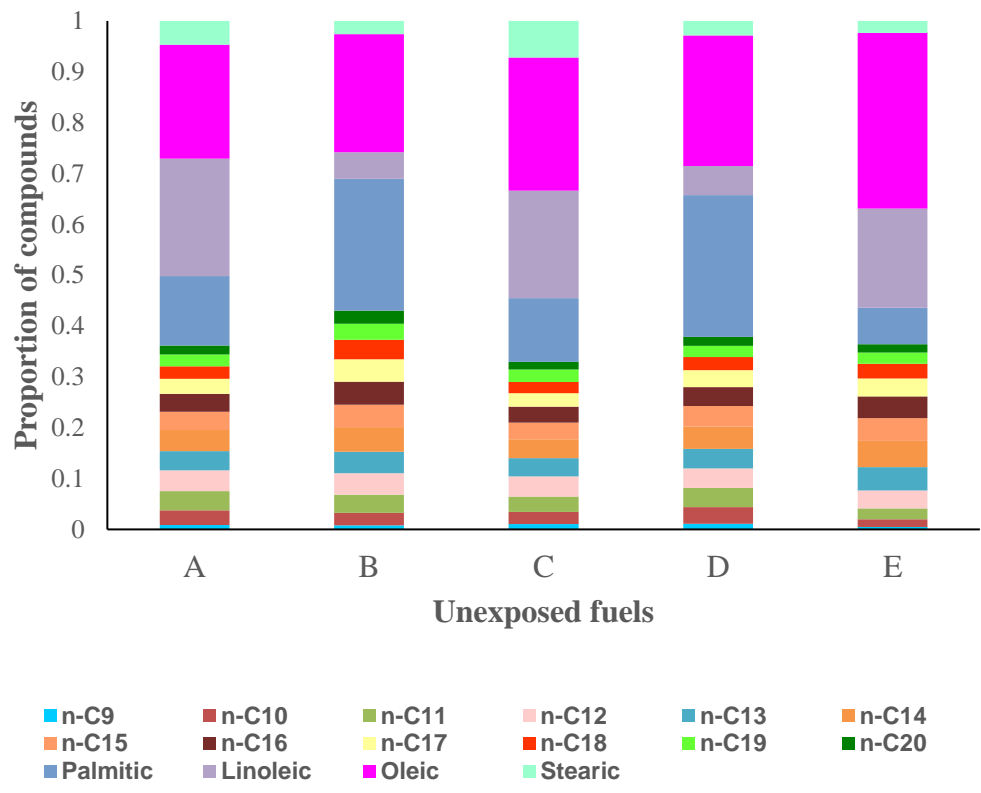


Figure 3.3 Proportion of 16 major compounds of B20 found in unexposed fuel samples. Fuels 'A to D' were obtained at SE location, while fuel 'E' at SW location.

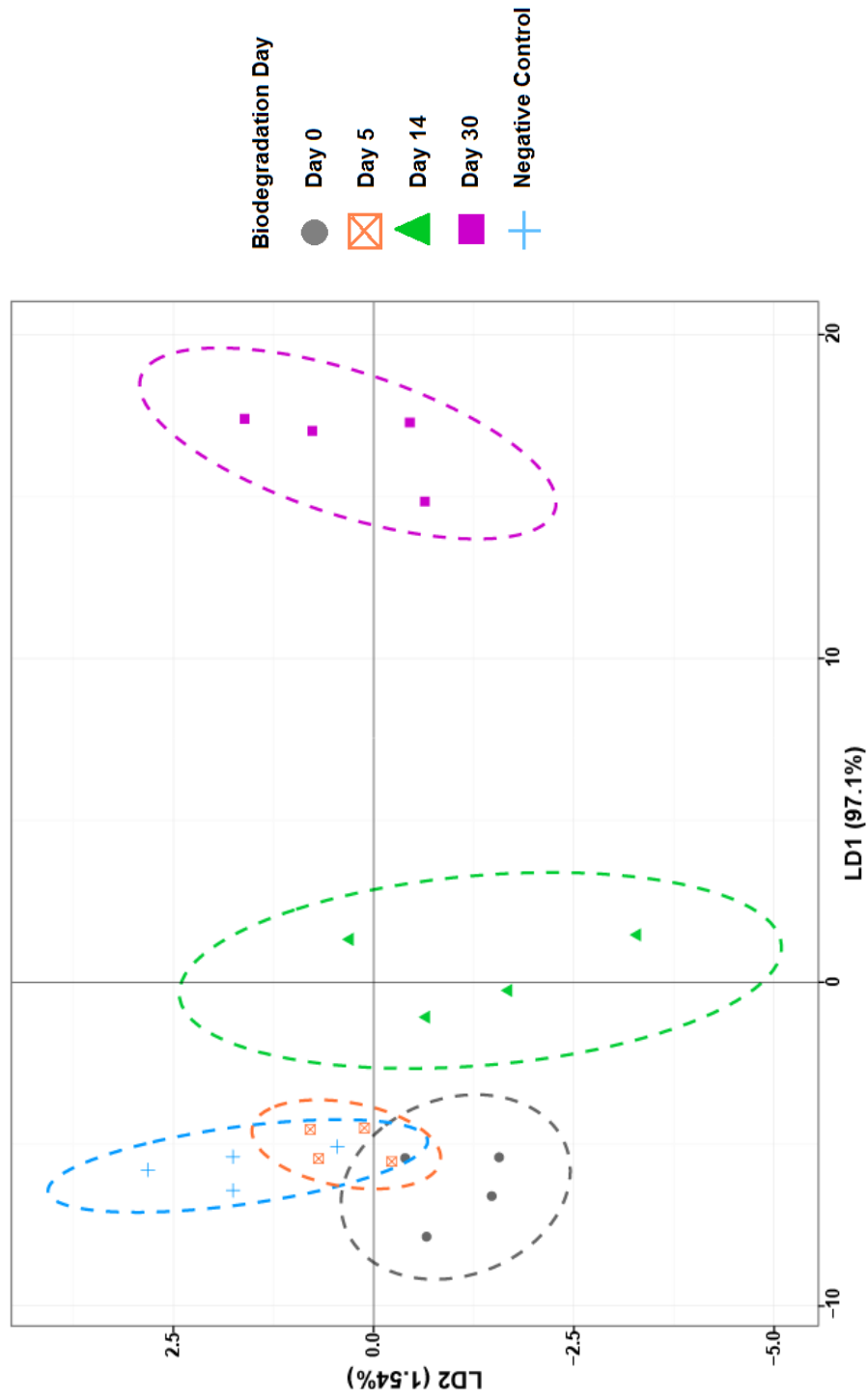


Figure 3.4 Score plot for the first two linear discriminant factors of biodegradation patterns obtained after incubation of *Byssochlamys* sp. SW2 in B20 fuels. The scores were calculated by LDA of nine major peaks found in B20 (see table 3.2). The 95% confidence ellipses for the individual groups are also shown.

Table 3.4 Description of the test set of B20 samples obtained from SE facility and used to validate the LDA model.

Sample #	Sampling Location Description ^a	Sampling Date	LDA prediction ^b
36	SE,3,B	10/28/2014	D
42	SE,E,B	3/12/2015	ND
30	SE,3,B	3/12/2015	D
14	SE,-,N	5/7/2015	ND
48	SE,E,B	5/7/2015	ND
52	SE,E,M	5/7/2015	ND
53	SE,E,N	5/7/2015	ND
4	SE,3,N	5/7/2015	ND
13	SE,3,R	5/7/2015	ND
16	SE,3,B (post-cleaning)	5/7/2015	D
27	SE,3,B (pre-cleaning)	5/7/2015	D
44	SE,4,B	5/7/2015	ND
47	SE,4,M	5/7/2015	ND

^a Sample descriptors include the facility Southeast (SE), tank (3, 4 or E), location in the tank (B, bottom; M, middle; N, nozzle; -, unknown location), fuel unexposed to the tanks (R).

^b Sample predicted to be not degraded (ND). Sample predicted to be degraded (D).

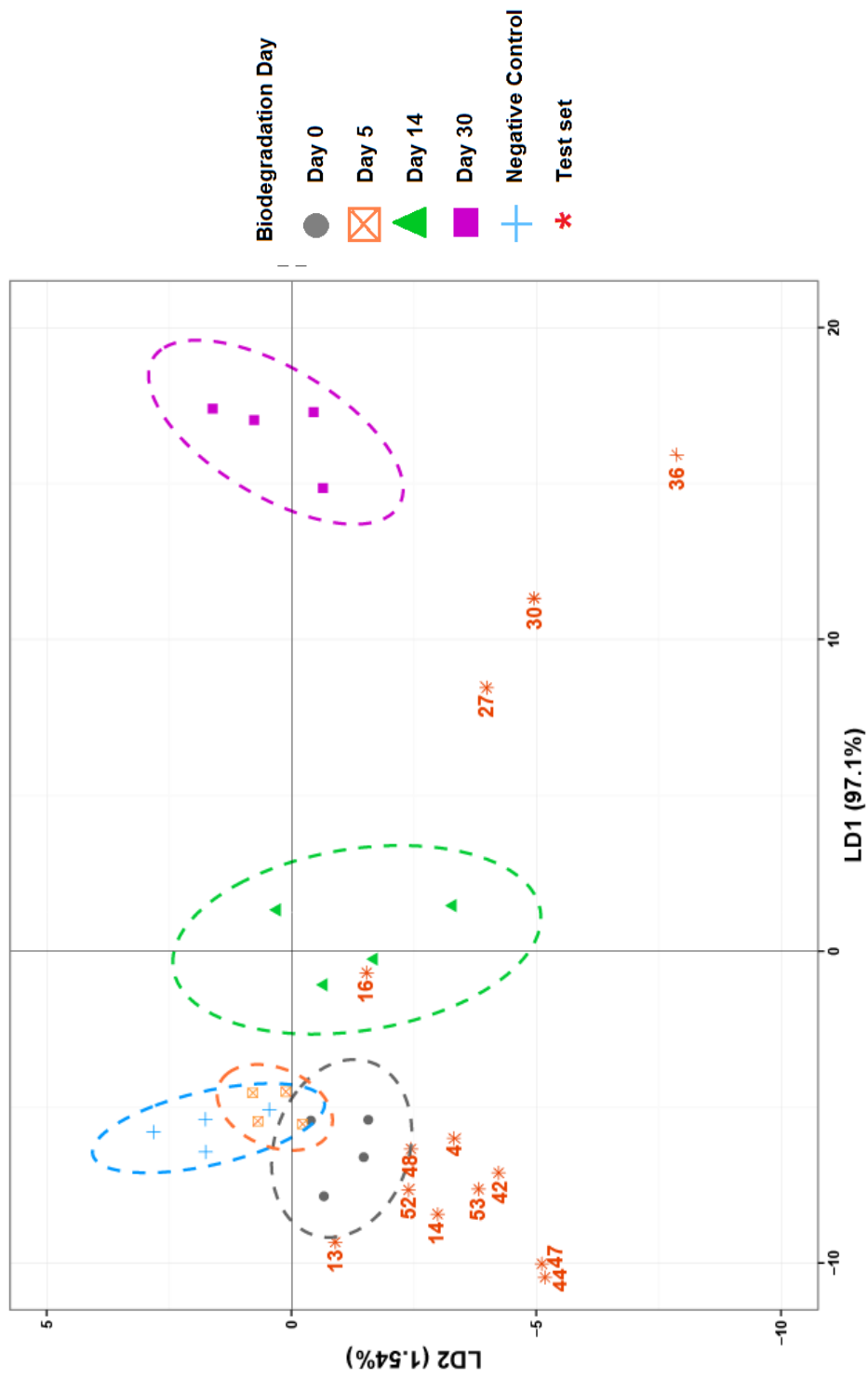


Figure 3.5 LDA prediction plot containing data from the test set samples. The 95% confidence ellipses for the individual groups are also shown.

Chapter 4: Summary and Future Directions

The Department of Defense (DoD) has been increasing the use of alternative fuels in ground vehicles and equipment as part of its Strategic Energy Plan (DOD, 2011). As a result, many military bases have infrastructure dedicated to the storage and dispensing of B20 biodiesel. Biodiesel is composed of single chain fatty acid methyl esters (FAME) derived from plant or animal fats, and B20 is an 80:20 blend of petroleum-derived ultra-low sulfur diesel (ULSD) and biodiesel (ASTM-D7467-15, 2015). Biodiesel contains more oxygen, is more hygroscopic, and is more oxidatively unstable compared to ULSD (Jakeria et al., 2014). This potentially increases the susceptibility of biodiesel to microbial contamination and degradation (Bücker et al., 2011; Mariano et al., 2008; Prince et al., 2008).

We have studied B20 biodiesel from storage tanks at several Air Force Bases, both with and without reported issues with fuel quality (color, clarity, particulates). Fuels of compromised quality from two different AFBs had substantial microbial contamination, which was believed to be the root cause of reported issues. Molecular characterization of the microbial assemblages showed that these fuels harbored a high concentration of the fungal genera *Byssochlamys* and *Wickerhamomyces*. We isolated ten different genera of fungi from B20 storage tanks at two USAF facilities. Two of these isolates were representatives of the most abundant genera in the B20 storage tanks (Stamps, 2016). *Byssochlamys* sp. SW2 and *Wickerhamomyces anomalus* SE3 are able to grow in B20 as the sole source of

carbon and energy. These fungi were capable of aerobic degradation of many carbon substrates in low water environments over a broad pH range.

In this research we hypothesized that *fungal contamination of USAF B20 fuel tanks cause changes in the chemical composition of the fuel*. Fuel components varied among facilities and tanks within facilities (Figure 3.2). The unexposed fuels varied in their content of FAMES, suggesting the use of different feedstocks in biodiesel production (Jakeria et al., 2014). *Byssochlamys sp. SW2* and *Wickerhamomyces anomalus SE3* preferentially degraded palmitic and linoleic acid methyl esters, and our *in situ* model supports the hypothesis that palmitic and linoleic acid methyl esters are the most susceptible components to biodegradation.

In our data, we lack the information regarding the specific feedstocks used to produce the B20 fuel stored in the USAF tanks, which was a limitation. Based on our findings, we suggest that *biodiesel from different feedstocks will be differentially susceptible to fungal contamination*. This hypothesis remains to be tested by biodegradation experiments comparing the susceptibility of different feedstocks to fungal attack. Feedstocks commonly used with high content of palmitic acid methyl ester include palm oil and cottonseed (Knothe, 2008), as well as novel feedstocks from marine microalgae (Chen et al., 2015). Feedstocks with high concentrations of linoleic acid include soybean, sunflower and cottonseed (Jakeria, Fazal, & Haseeb, 2014). Current investigation about the design of optimal biodiesel blends (Knothe, 2008) is focused only in the improvement of the oxidative stability, cold flow, and

increased NO_x exhaust. We suggest that considering the effect of microbial degradation in the search of new feedstocks and mixtures for biodiesel production is also important.

We show that the presence of *Byssochlamys sp. SW2* can alter the composition of B20 biodiesel in storage tanks, and we offer a model for predicting severity of biodegradation. We used *Byssochlamys sp. SW2* in our research, because it represented an abundant organism in the tanks that were studied (Stamps, 2016). However, we are aware that microorganisms that contribute to alterations of the fuel quality during storage are part of microbial communities and consortia (Bücker et al., 2014; Lee, Ray, & Little, 2010; Passman, 2003; Stamps, 2016). In consortia, microorganism can accomplish complex tasks that are not possible individually (Brune & Bayer, 2012; Passman, 2003). Our understanding of the most abundant organism represents an essential baseline for future studies of root-cause analysis. Based on our results, our hypothesis is that *Byssochlamys sp. and Wickerhamomyces anomalus* have an active role in biodegradation of B20 inside the storage tanks. Modeling and monitoring of mesocosm experiments where *Byssochlamys sp.* and *Wickerhamomyces anomalus* interact with other bacteria and fungi, will be a first approach to test this hypothesis.

In conclusion, we show that the presence of *Byssochlamys sp.* alters the composition of B20 biodiesel in storage tanks, and we offer a model for predicting severity of biodegradation. *Byssochlamys sp. SW2* and *Wickerhamomyces anomalus SE3*

preferentially degraded palmitic and linoleic acid methyl esters, and our *in situ* model supports the hypothesis that palmitic and linoleic acid methyl esters are the most susceptible components to biodegradable. We suggest the use of alternative feedstocks containing less palmitic and linoleic acid for B20 biodiesel production to increase fuel stability in storage tanks.

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