

A STUDY OF THE EFFECT OF RATE OF AMMONIUM NITRATE  
FERTILIZATION, CHEMICAL COMPONENTS, AND TIME OF  
CUTTING ON THE IN VITRO RUMEN DIGESTION OF  
BERMUDAGRASS FORAGE

by

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## INTRODUCTION

The importance of the rumen in the digestion of forages by the ruminant animal has been recognized since the latter part of the 18th century. For many years the accepted method of determining the nutritive value of forages was through the use of approximate chemical analyses and in vivo digestion trials. In recent years, in vitro or artificial rumen techniques have been used considerably in gathering data on the nutritive value of many forages. Although early attempts to correlate the results of in vitro and in vivo digestion trials were not totally successful, recent reports have shown that good correlation is possible.

This investigation was initiated in an attempt to relate the rate of  $\text{NH}_4\text{NO}_3$  fertilization, and time of cutting of bermudagrass forage to selected chemical components and in vitro digestibility, and to determine what, if any, chemical component of bermudagrass forage could be related to dry matter digestibility and hence, nutritive value.

## LITERATURE REVIEW

Barnett and Reid (2), in their review on reactions in the rumen, have divided the various methods of study of rumen digestion in vitro into four categories which are as follows:

- a. Simple incubation of rumen liquor with various substrates.
- b. Impermeable artificial rumens, usually constructed of glass vessels, about which more will be said later.
- c. Semi-permeable sacs which are used in the hope of simulating the part which diffusion plays in the natural rumen.
- d. Artificial rumens of either the impermeable or semi-permeable type which are inoculated with washed suspensions of rumen microorganisms.

Since all the experiments to be considered were conducted with the impermeable type of artificial rumen, it is the only category which will receive further attention.

The impermeable artificial rumen is a system which attempts to approximate many of the environmental conditions found in the rumen of a living animal. In order to accomplish this, certain factors must be held constant. The temperature of the reaction vessel(s) is usually held between 38° and 40° C. through the use of a temperature-controlled water bath. The anaerobic condition which is found in the natural rumen is simulated by constantly passing CO<sub>2</sub> and/or N<sub>2</sub> through the tightly sealed reaction vessel(s). A NaHCO<sub>3</sub> buffer system is used to



assume the role of saliva as a buffer. This buffer system is initially adjusted to a pH of from 6.7 to 7.0, and is held relatively constant through intermittent additions of  $\text{Na}_2\text{CO}_3$ . Trace minerals and a nutrient salt solution necessary for the growth and reproduction of rumen microorganisms are added to the test solution. Rumen liquor samples containing microorganisms natural to the rumen are usually obtained from animals which have an indwelling rumen fistula, or from the rumen of freshly slaughtered animals. The rumen liquor obtained from such animals is subjected to various chemical and/or physical treatments and is then used to inoculate the test solution in the reaction vessel(s). The material which is to be digested is then placed in the reaction vessel(s), and degradation of the substrate is allowed to proceed for a predetermined period of time. At the end of this time, digestion is halted, and the amount of digestion which has occurred is calculated.

There are a wide variety of methods which may be used to measure the amount of digestion occurring in the artificial rumen. While the most common methods are based on the disappearance of one or more chemical components of the substrate, other methods are based on the end products produced by the digestion. Probably the most common method used currently to determine the amount of digestion which has taken place is to measure the disappearance of cellulose. This procedure has been used successfully by Quicke and Bentley (25), Reid et al. (27), Donefer et al. (12), and Kamstra et al. (17, 18), to name only a few. Other common methods currently in use are dry matter disappearance (3, 6, 7, 27, 31), and organic matter disappearance (15), as well as disappearance of other chemical components.

A wide range of techniques has been used in the preparation of

the rumen liquor used to inoculate the test solution. Some methods are very simple and require a minimum of equipment. Walker (34) merely strained the rumen liquor through cheesecloth and obtained good correlation between in vivo and in vitro dry matter digestibility. Quicke et al. (26) reported a procedure which involved extracting pressed rumen contents (pulp) with a phosphate buffer, centrifuging the buffer extract, and resuspending the sediment. Yet a third method was used by Van Dyne (33), in which the rumen contents were extracted with a phosphate buffer, and this extract used as the inoculum.

The effect of the source of the rumen liquor has been studied by many investigators. It has been reported by LeFevre and Kamstra (19) that rumen fluid obtained from sheep and cattle may be used interchangeably if they are fed a similar ration. Quicke et al. (26), in an in vitro fermentation experiment with nine dried forages, showed that feeding different types of forages to a steer used as a rumen liquor source had no significant effect on digestibility of forage cellulose. Barnett (1), reported that rumen liquor obtained from freshly slaughtered animals fed on diets of turnips, grass, silage, and oats, could be used indiscriminately without noticeable effect on results. However, Quicke et al. (26) have reported that rumen liquor obtained from alfalfa-fed animals has a significant effect on in vitro digestion.

While there is considerable variation in the methods used in in vitro fermentation experiments, the results compare very well with those obtained through the use of in vivo procedures (4, 5, 16, 27, 34). Because of the advantages of in vitro procedures over in vivo procedures (29) in the evaluation of forages, they have been and will continue to be used extensively in determining the nutritive value of forage crops.

Numerous investigators (8, 9, 17, 25, 27) have studied the chemical composition and maturity of forages in attempts to relate these properties to in vitro and in vivo digestion trial results. The decrease in digestibility of forages with increasing maturity has been shown to be associated with an increase in lignin content (13, 18, 30, 31), and perhaps with the seasonal decrease in protein content of the forages (2). While lignin is not considered a carbohydrate, it is thought to have a close physical and chemical relationship with the cellulose and other carbohydrate fractions of the plant. Meyer and Lofgreen (21) have shown that crude fiber digestibility may be correlated with lignin content, and Sullivan (31) has shown that there is a correlation between acid-insoluble lignin and digestible dry matter. Quicke and Bentley (25) have shown that with maturing forages, a slight increase in lignin content may cause a considerable decrease in cellulose digestibility. Differences in in vitro digestibility, at all stages of growth, were shown to be more closely related to acid-insoluble than to total lignin, suggesting a change in the nature of the lignin fraction through-out the growing period. Patton and Giesecker (23) reported that increased lignin content in forages caused decreased weight gains in cattle, and suggested the following theories to explain this result:

- a. The lignin may incrust the other digestible constituents, putting them "in a nutshell which the animal can't crack."
- b. Lignin may combine chemically with other constituents, forming unavailable compounds.
- c. Digestion may be retarded through local inhibition of digestive enzymes due to the toxic action of phenolic groups resulting from the partial decomposition of lignin.

More recently, Dehority and Johnson (11) presented evidence obtained through in vitro digestion trials, which tends to verify the theory that deposition of lignin as an "encrusting substance" rather than total concentration of lignin is responsible for decreased digestibility of forages.

The effect of hemicellulose on digestibility has received slight attention. However, Kamstra et al. (17), McBee (20), and Salisbury et al. (28) have shown that hemicellulose neither inhibits nor stimulates cellulose digestion.

## EXPERIMENTAL METHODS

Preparation of Bermudagrass Forage Samples: A genetically pure strain of bermudagrass<sup>1</sup> was grown on five series of plots, each series of plots consisting of four replications. Each of the five series of plots received  $\text{NH}_4\text{NO}_3$  fertilizer four times per year in total amounts varying from 0 to 1,400 pounds per acre. The Agronomy Department advised that samples were collected at approximately twenty-one-day intervals in an attempt to simulate the conditions present on normally grazed bermudagrass pasture. Samples were collected over a three year period starting in 1960. The first samples were usually collected during the last week of May, the last samples during the last week of September, with a total of seven cuttings being collected each season.

The samples were dried in a forced draft oven at 105° C. and ground in a Wiley mill to pass through a 40-mesh screen before determining their chemical composition and in vitro dry matter digestibility.

Isolation of Holocellulose: The method used to isolate holocellulose is similar to that of Wise, Murphy, and D'Addieco (37). Ely and Moore (14) reported that this method gave 99 per cent recovery of theoretical holocellulose in ten forages tested.

Initially, four 25-g. samples of bermudagrass forage that had been collected from four replicate plots were combined to form a 100-g. sample. Each of the four 25-g. samples had been collected on the same date from

<sup>1</sup>Midland Variety bermudagrass was used in all experiments.

plots which had received the same amount of  $\text{NH}_4\text{NO}_3$  fertilizer. This 100-g. sample was placed in a 90 x 200 mm. double thimble<sup>1</sup> and was extracted for a minimum of fifteen hours with an ethanol-benzene mixture composed of 1 part ethanol to 2.5 parts benzene by volume. After extraction, the residue was dried, weighed, and thoroughly mixed. Then, 250 ml. of a 0.5 per cent ammonium oxalate solution was added to a fourth of this residue which was refluxed at 85° C. for four hours. After refluxing, the extract was filtered and washed with hot water. A 600-ml. Buchner type funnel with a coarse fritted disk was used for all filtering procedures. A second 250-ml. portion of 0.5 per cent ammonium oxalate was then added to the residue and it was refluxed at 85° C. for fifteen hours. The residue was washed and filtered again. This residue was placed in a 1.5-liter beaker with 625 ml. of water, and 2 ml. of glacial acetic acid and 7.5 g. of reagent grade  $\text{NaClO}_2$  were added in the order indicated. This mixture was placed in a temperature-controlled water bath at 85° C. and stirred occasionally. After fifteen minutes, thirty minutes, and forty-five minutes, glacial acetic acid and  $\text{NaClO}_2$  were added in the amounts indicated above. At the end of one hour, the beaker was cooled in ice water to 10° C. The mixture was filtered, washed six times with ice water, air dried for fifteen hours, and finally dried under an infra-red lamp for approximately four hours.

In Vitro Rumen Digestion: The in vitro method used was a modified version of the technique developed by Walker (34).

Preparation of Inoculum: The rumen fluid used as inoculum was obtained from a steer fitted with a rumen fistula. The steer was maintained on mature, weathered, prairie hay and cottonseed meal, and no

<sup>1</sup>Double extraction thimbles were obtained from W. and R. Balston Ltd., England.

attempt was made to feed a diet consisting solely of the forage being tested in the in vitro rumen apparatus. A sample of rumen contents was withdrawn from the steer, strained through six layers of cheesecloth, and placed in a vacuum bottle preheated to 38° C. for transportation to the laboratory. In the laboratory, the sample was again strained through six layers of cheesecloth and stirred slowly using a magnetic stirrer to prevent sedimentation and to insure that a homogenous sample could be obtained for use as the inoculum.

Preparation of Nutrient Solution: A slightly modified procedure similar to that of Walker (34) was followed in preparing the nutrient solution.

- a. A solution of minerals similar to the solution originally outlined by Warner (35) was prepared and made to a volume of 2 liters. It contained the following:

$(\text{NH}_4)_2\text{SO}_4$	13.20 g.
$\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$	12.30 g.
$\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$	7.38 g.
$\text{K}_2\text{HPO}_4 \cdot 3\text{H}_2\text{O}$	68.50 g.
$\text{CoCl}_2 \cdot 6\text{H}_2\text{O}$	0.04 g.
$\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$	0.04 g.
$\text{FeSO}_4 \cdot 7\text{H}_2\text{O}$	0.20 g.
$\text{MnSO}_4 \cdot \text{H}_2\text{O}$	0.06 g.
$\text{ZnSO}_4 \cdot 7\text{H}_2\text{O}$	0.08 g.
$\text{MoO}_3$	0.02 g.
Glacial acetic acid	12.00 ml.
$\text{H}_3\text{PO}_4$ , 88%, syrupy	16.50 ml.

- b. Two hundred fifty ml. of a 1.0 M  $\text{H}_3\text{PO}_4$  solution was prepared.

- c. An aqueous solution containing the sodium salts listed below was prepared with the final volume being brought to 1 liter.

NaCl 58.50 g.

Na<sub>2</sub>HPO<sub>4</sub> 21.30 g.

NaC<sub>2</sub>H<sub>3</sub>O<sub>2</sub> 82.00 g.

- d. An aqueous solution consisting of the volatile fatty acids listed below was prepared with the final volume being 10 ml.

Propionic acid 2.30 ml.

Isobutyric acid 0.90 ml.

Isovaleric acid 0.80 ml.

Valeric acid 0.80 ml.

- e. Two hundred fifty ml. of a 1.0 M NaHCO<sub>3</sub> solution was prepared.

To prepare the nutrient solution for use in the in vitro rumen, 300 ml. of aqueous mineral solution, 60 ml. of sodium salts solution, 6.0 ml. of volatile fatty acids solution, 6.0 ml. of 1.0 M H<sub>3</sub>PO<sub>4</sub> solution, and 2,508 ml. of water were mixed thoroughly. Immediately before use 120 ml. of a 1.0 M NaHCO<sub>3</sub> solution was added. The total volume of this nutrient solution was 3 liters.

In Vitro Rumen Apparatus: The in vitro rumen apparatus (see figure 1) consisted of a series of twelve wide mouth bottles, each of approximately 500-ml. capacity. The bottles were tightly stoppered with rubber stoppers through which passed two glass tubes. The inlet tube reached to the bottom of the bottle and was drawn out at the end to form a jet. The outlet tube was short and extended approximately





Figure 1. In Vitro Rumen Apparatus

1 cm. below the stopper. The bottles were held in position in a temperature-controlled water bath by a frame constructed of aluminum alloy rods and copper wire. Each bottle was connected in series by a length of 5/16th inch rubber tubing, and the tubing was of such length that it formed an inverted U between adjacent bottles and acted as a condenser to prevent any liquid being carried over from one bottle to another. The water bath was maintained at 38° C. and a stream of CO<sub>2</sub> was bubbled continuously through the bottles. Each of the twelve bottles contained 250 ml. of the nutrient solution, 20 ml. of strained rumen fluid, and 1 g. of substrate (with the exception of three blanks which contained no substrate). Digestion in vitro was continued for seventy-two hours.

Digestion Procedure: A standard, which was a sample of dried and ground bermudagrass forage, was included in each experiment. One gram of this "standard forage" was placed in bottle 8 and this forage was used as an internal check in each experiment. In acceptable experiments its digestibility ranged between 57 and 65 per cent. As a further check, bottles 2 and 12 contained 1 g. each of a purified cellulose sample.<sup>1</sup> Bottles 1, 5, and 9 contained no substrate but had the normal volume of nutrient solution and strained rumen fluid. These bottles provided a value for the quantity of dry matter introduced in the form of rumen fluid. The samples to be tested were added to bottles 3, 4, 6, 7, 10, and 11. Bottles 3 and 7, 4 and 10, and 6 and 11 always contained duplicate samples. The bottles were connected in a series, and an anaerobic environment was maintained throughout the digestion period. The pH of the solution was checked initially and was adjusted to 6.7

<sup>1</sup>Solka Floc BW-200, obtained from Brown Comp., Boston, Mass., is 99.5 per cent pure cellulose when dry.

with  $\text{Na}_2\text{CO}_3$  (106 g./liter). The pH was checked after digestion periods of twenty-four and forty-eight hours, and the samples were corrected to pH 6.7 if they had varied. Alteration of pH was seldom more than 0.2 pH unit per twenty-four hour period. At the end of this digestion period the bottles were disconnected, stirred thoroughly, and the contents were transferred to 250-ml. glass centrifuge bottles. At this point the total volume of each in vitro digestion bottle was approximately 300 ml., making two transfers of 150 ml. each necessary. The samples were centrifuged at 2,000 r.p.m. for twenty minutes in a size 2 International Centrifuge using a no. 267 head. After each 150-ml. sample had been centrifuged, the supernatant was decanted, and the sediments from both fractions were combined. The sediment was washed twice with water, centrifuged at 2,000 r.p.m. for ten minutes after each washing, and the supernatants were discarded. The sediment was then transferred to a tared platinum crucible, dried overnight at 105° C. and weighed. The weight of sediment in bottles 1, 5, and 9 was averaged and then subtracted from the weight of sediment for each in vitro digestion bottle. The figure obtained by these calculations was termed "undigested dry matter." The "undigested dry matter" for each duplicate sample set was then averaged, multiplied by 100, and subtracted from 100 to obtain per cent in vitro dry matter digestibility for each bermudagrass forage or holocellulose sample in question.

Lignin Analysis: The procedure used here combines the salient features of three different methods previously outlined by Thacker (32), Phillips and Smith (24), and Davis (10).

Preparation of Reagents:

- a. A 1 per cent pepsin solution was prepared by adding 1 g.

- of powdered pepsin to 100 ml. of 0.1 N HCl immediately before using.
- b. A 0.25 per cent  $\text{Na}_2\text{CO}_3$  solution was prepared by adding 1 g. of  $\text{Na}_2\text{CO}_3$  to 400 ml. of water.
  - c. A 1.0 N  $\text{H}_2\text{SO}_4$  solution was prepared by placing 55.6 ml. of concentrated  $\text{H}_2\text{SO}_4$  in 2 liters of solution.
  - d. A 72 per cent (by weight)  $\text{H}_2\text{SO}_4$  solution was prepared by adding 405 ml. of concentrated  $\text{H}_2\text{SO}_4$  to 254 ml. of water.

Analytical Procedure: A 0.5 g. sample of the forage was extracted with 50 ml. of an ethanol-benzene mixture (1 part ethanol - 2 parts benzene) for a minimum of twenty-four hours using soil extracting flasks and air condensers. The ethanol-benzene mixture was removed by washing the flask and sample with ethanol followed by water. The sample was then placed in a 250-ml. soil flask, and 40 ml. of a 1 per cent pepsin solution in 0.1 N HCl was added. This mixture was allowed to incubate for twenty-four hours and then the pepsin solution was removed and the sample washed with hot water. At this point, 100 ml. of a 0.25 per cent  $\text{Na}_2\text{CO}_3$  solution was added and the mixture was incubated twenty-four hours at  $37^\circ$  to  $40^\circ$  C. The sample was then filtered and washed with hot water. After transferring the sample to a 600-ml. Berzelius beaker, 100 ml. of 1.0 N  $\text{H}_2\text{SO}_4$  was added and the solution was boiled for one hour using a reflux condenser. After boiling, the sample was filtered and washed with ethanol and ether. The ether was then removed to dry the sample. At this point the sample was placed in a dry Berzelius beaker and 15 ml. of cold 72 per cent  $\text{H}_2\text{SO}_4$  was added. The sample was placed in an ice water bath at  $18^\circ$  C. and stirred frequently for two hours. At the end of two hours, the acid was diluted to approximately 1.0 N by

the addition of 350 ml. of water. This mixture was boiled gently under a reflux condenser for one hour, filtered through gooch crucibles, and washed with hot water, ethanol, and ether. The ether was removed by heating, and the sample was dried in an oven at 105° C. and weighed. As the final step, the sample was ashed at 600° C., cooled in a desiccator and weighed. The loss in weight which occurred through ashing was considered as the amount of lignin originally present in the sample.

Cellulose and Hemicellulose Analysis: The method used was essentially that of Patton (22).

Duplicate 1 g. samples of holocellulose which had been dried, and ground in a Wiley mill to pass a 40-mesh screen, were placed in 6-inch Pyrex centrifuge tubes. Then, 15 ml. of 80 per cent acetic acid and 1.5 ml. of concentrated  $\text{HNO}_3$  were added. Small flask covers were filled with water and placed in the necks of the centrifuge tubes to act as reflux condensers. At this point the tubes were placed on a sand bath and boiled gently for twenty minutes. The centrifuge tubes were then cooled in a water bath and the condensers were removed. Approximately 20 ml. of ethanol was added and the mixture was stirred. The tubes were centrifuged until the sample had sedimented and the supernatant was discarded. The sample was washed, in the centrifuge tube, twice with ethanol, and once each with hot benzene, hot ethanol, and ether. The samples were dried under an infra-red lamp until no ether odor remained and then placed in an oven at 105° C. for one hour. After cooling in a desiccator, the tubes were weighed and ashed at 550° C. overnight. Then the tubes were cooled in a desiccator and reweighed. The loss in weight on ashing was recorded as the amount of cellulose. Since holocellulose is defined as cellulose plus hemicellulose, the weight of the original

holocellulose sample minus the weight of the cellulose was considered to be equal to the amount of hemicellulose present.

## RESULTS AND DISCUSSION

Reproducibility of In Vitro Digestion Method: Tables I and II are sample results, picked at random, which show the reproducibility of this method. Table I indicates that the dry matter digestibility of duplicate bermudagrass forage samples varied only slightly. However, as Table II indicates, the digestibility of duplicate holocellulose samples varied quite a bit. This was probably due to sample variation, and was not caused by the in vitro digestion method. Figure 2 shows that relatively good correlation may be obtained between separate experiments.

### Chemical Composition of Bermudagrass Forage as Affected by $\text{NH}_4\text{NO}_3$ Fertilization:

Holocellulose: As shown in Table III, the rate at which  $\text{NH}_4\text{NO}_3$  fertilizer was applied appears to have had an effect on the amount of holocellulose formed in bermudagrass forage. When the total amount of fertilizer applied was increased from 0 through 1,400 pounds per acre, the holocellulose content of the forage samples was decreased. The greatest decrease in holocellulose was shown to occur in the samples collected in late June or early July. In samples which had received no  $\text{NH}_4\text{NO}_3$  fertilizer, the holocellulose fraction comprised an average of 59.80 per cent of the total sample, while in samples which received 1,400 pounds per acre of fertilizer the holocellulose fraction amounted to an average of only 48.40 per cent of the total sample.

TABLE I  
 DIGESTIBILITY IN VITRO OF DRY MATTER IN  
 BERMUDAGRASS FORAGE SAMPLES

Sample	Bottle No.	Weight of	Undigested	Digested
		Sediment g.	Dry Matter g.	Dry Matter %
Blank	1	0.1252	-	-
Blank	5	0.1275	-	-
Blank	9	0.1302	-	-
H-400, <sup>a</sup> 1 g.	3	0.4462	0.3186	68.00
H-400, 1 g.	7	0.4491	0.3215	
I-400, <sup>b</sup> 1 g.	4	0.4642	0.3366	66.00
I-400, 1 g.	10	0.4709	0.3433	
J-400, <sup>c</sup> 1 g.	6	0.5158	0.3882	61.22
J-400, 1 g.	11	0.5149	0.3873	
Solka Floc, 1 g.	2	0.2153	0.0877	91.04
Solka Floc, 1 g.	12	0.2192	0.0916	
Standard Forage, 1 g.	8	0.5096	0.3820	61.80

<sup>a</sup>H-400 - A sample of bermudagrass forage collected from four replicate plots on May 25, 1961.

<sup>b</sup>I-400 - A sample of bermudagrass forage collected from four replicate plots on June 11, 1961.

<sup>c</sup>J-400 - A sample of bermudagrass forage collected from four replicate plots on June 30, 1961.



TABLE II  
 DIGESTIBILITY IN VITRO OF DRY MATTER  
 IN HOLOCELLULOSE SAMPLES

Sample	Bottle No.	Weight of Sediment g.	Undigested Dry Matter g.	Digested Dry Matter %
Blank	1	0.1340	-	-
Blank	5	0.1316	-	-
Blank	9	0.1323	-	-
HC-3, <sup>a</sup> 1 g.	3	0.3209	0.1883	85.23
HC-3, 1 g.	7	0.2397	0.1071	
HC-8, <sup>b</sup> 1 g.	4	0.2950	0.1624	87.11
HC-8, 1 g.	10	0.2280	0.0954	
HC-13, <sup>c</sup> 1 g.	6	0.2719	0.1393	83.02
HC-13, 1 g.	11	0.2612	0.1286	
Solka Floc, 1 g.	2	0.2172	0.0846	92.72
Solka Floc, 1 g.	12	0.1936	0.0610	
Standard Forage, 1 g.	8	0.5579	0.4253	57.47

<sup>a</sup>HC-3 - A holocellulose sample isolated from bermudagrass forage sample H-400.

<sup>b</sup>HC-8 - A holocellulose sample isolated from bermudagrass forage sample I-400.

<sup>c</sup>HC-13 - A holocellulose sample isolated from bermudagrass forage sample J-400.

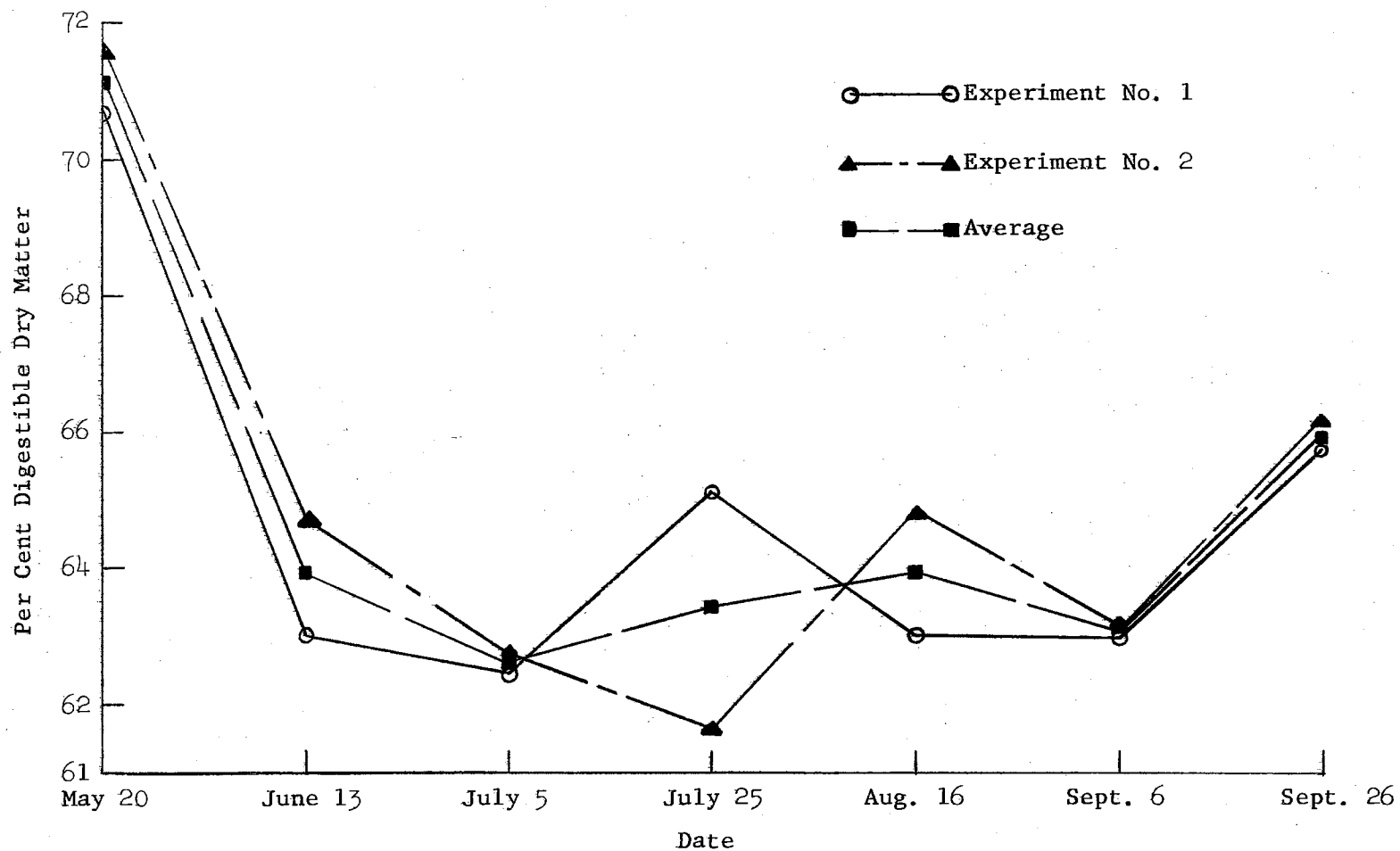


Figure 2. Percentage Digestibility In Vitro of Dry Matter in Bermudagrass Forage as Affected by Time of Cutting (1960).

Note: Experiments no. 1 and 2 were run on the same samples at different times to check reproducibility of the in vitro digestion method.

TABLE III  
 THE PERCENTAGE OF HOLOCELLULOSE IN BERMUDAGRASS  
 FORAGE AS AFFECTED BY  $\text{NH}_4\text{NO}_3$   
 FERTILIZATION

Sample No. <sup>a</sup>	0 lb./acre $\text{NH}_4\text{NO}_3$			1,400 lb./acre $\text{NH}_4\text{NO}_3$		
	1960	1961	average	1960	1961	average
	%	%	%	%	%	%
1	49.72	-	49.72	45.88	44.44	45.16
2	57.20	62.76	59.58	53.72	48.32	51.02
3	55.84	63.76	59.80	48.08	48.72	48.40
4	55.52	58.20	56.86	51.44	53.88	52.66
5	55.20	59.60	57.40	45.72	50.92	48.32
6	55.84	-	55.84	52.44	48.28	50.36
7	52.40	58.48	55.44	44.10	48.72	46.41

<sup>a</sup>All samples numbered from 1 through 7 were collected on the following dates: (1) May 20, 25, and 24; (2) June 13, 11, and 13; (3) July 5, June 30, and July 5; (4) July 25, 20, and 26; (5) August 16, 10, and 16; (6) September 6, August 31, and September 6, and (7) September 26, 28, and October 1 for the years 1960, 1961, and 1962 respectively.

Cellulose: The rate of fertilization seems to have had little or no effect on the amount of cellulose laid down in bermudagrass forage. As Table IV indicates, various samples from forage which was not receiving  $\text{NH}_4\text{NO}_3$  fertilizer had a greater cellulose content than samples from forage which had received a maximum of 1,400 pounds per acre of  $\text{NH}_4\text{NO}_3$  fertilizer. However, in other samples the reverse of the above was shown to occur, in that forage which had received 1,400 pounds per acre of fertilizer laid down a greater amount of cellulose than forage which

received no fertilizer.

TABLE IV

THE PERCENTAGE OF CELLULOSE IN BERMUDAGRASS  
FORAGE AS AFFECTED BY  $\text{NH}_4\text{NO}_3$   
FERTILIZATION

Sample No. <sup>a</sup>	0 lb./acre $\text{NH}_4\text{NO}_3$			1,400 lb./acre $\text{NH}_4\text{NO}_3$		
	1960	1961	average	1960	1961	average
	%	%	%	%	%	%
1	21.97	-	21.97	25.64	20.89	23.26
2	28.07	31.86	29.96	30.61	24.89	27.75
3	25.28	31.76	28.52	27.18	22.14	24.66
4	27.06	26.52	26.79	28.10	25.92	27.01
5	24.92	26.10	25.51	23.62	24.86	24.24
6	26.34	-	26.34	27.62	23.57	25.60
7	25.04	25.69	25.36	25.28	22.52	23.90

<sup>a</sup>See footnote a to Table III.

Hemicellulose: Hemicellulose formation in bermudagrass forage was decreased by increased amounts of  $\text{NH}_4\text{NO}_3$  fertilizer. Table V indicates that forage which received 1,400 pounds per acre of  $\text{NH}_4\text{NO}_3$  fertilizer formed less hemicellulose than forage which had received no  $\text{NH}_4\text{NO}_3$  fertilizer. The greatest reduction in hemicellulose content was shown to occur in late June or early July. At that time the average percentage of hemicellulose in samples which had received no fertilizer was 31.28 per cent, while samples which had received 1,400 pounds per

acre of  $\text{NH}_4\text{NO}_3$  fertilizer contained an average of 23.29 per cent hemicellulose.

TABLE V  
THE PERCENTAGE OF HEMICELLULOSE IN BERMUDAGRASS  
FORAGE AS AFFECTED BY  $\text{NH}_4\text{NO}_3$   
FERTILIZATION

Sample No. <sup>a</sup>	0 lb./acre $\text{NH}_4\text{NO}_3$			1,400 lb./acre $\text{NH}_4\text{NO}_3$		
	1960	1961	average	1960	1961	average
	%	%	%	%	%	%
1	27.75	-	27.75	20.24	23.55	21.90
2	29.13	30.90	30.02	23.11	23.43	23.27
3	30.56	32.00	31.28	20.90	26.58	23.29
4	28.46	31.68	30.07	23.34	27.96	25.65
5	30.28	33.50	31.89	22.09	26.06	24.08
6	29.50	-	29.50	24.82	24.71	24.76
7	27.36	32.79	30.08	18.82	26.17	22.50

<sup>a</sup>See footnote a to Table III.

Unpublished data (36) have shown that increasing the amount of  $\text{NH}_4\text{NO}_3$  fertilizer applied to bermudagrass forage caused an increase in the protein content of the forage. This suggests that as the protein content of bermudagrass forage increased, there was a subsequent decrease in the amount of holocellulose laid down by the forage. Since holocellulose is defined as the cellulose plus hemicellulose fractions of the forage, the decrease in holocellulose could have been caused by a

decrease in hemicellulose or cellulose, or both. However, since it was shown that the greatest decrease in holocellulose and hemicellulose occurs simultaneously, it would seem that the decreased holocellulose content can be accounted for in the hemicellulose fraction alone.

Per Cent Digestibility In Vitro of Dry Matter in Bermudagrass Forage as Affected by  $\text{NH}_4\text{NO}_3$  Fertilization: Figure 3 and Table VI indicate that the rate at which  $\text{NH}_4\text{NO}_3$  fertilizer was applied had negligible effect on the per cent digestibility in vitro of dry matter in bermudagrass forage. Comparison of Figure 3 with Table III shows that the holocellulose fraction of bermudagrass forage had little effect on its in vitro digestibility. Earlier reports (17, 20, and 28) have shown that hemicellulose has no effect on in vitro cellulose digestion by rumen microorganisms. When the results given in Table V are compared with Figure 3, a similar conclusion may be drawn as to the effect of hemicellulose on in vitro dry matter digestibility of bermudagrass forage.

Per Cent Digestibility In Vitro of Dry Matter in Bermudagrass Forage as Affected by Time of Cutting: Table VII and Figures 2, 4, and 5 show that there was considerable seasonal variation in the in vitro dry matter digestibility of bermudagrass forage. Samples collected in May of 1960 were 71.15 per cent digested, while similar samples collected in July were only 62.62 per cent digestible. Similar results were noted in 1961, when May samples were 68.00 per cent digestible, while samples collected on June 30 were only 61.22 per cent digestible. Samples collected during the summer of 1962 show a decrease in digestibility of similar magnitude occurring during the period from June 13 to July 5. During the period from early July to late September, digest-

TABLE VI

PERCENTAGE DIGESTIBILITY IN VITRO OF DRY MATTER IN BERMUDAGRASS  
FORAGE AS AFFECTED BY  $\text{NH}_4\text{NO}_3$  FERTILIZATION

Date Collected	Rate of $\text{NH}_4\text{NO}_3$ Fertilization		
	0 lb./acre	400 lb./acre	1,400 lb./acre
	%	%	%
May 20, 1960	72.75	72.64	71.16
July 5, 1960	62.54	64.66	65.70
August 16, 1960	64.42	64.89	62.10
September 26, 1960	59.88	61.77	61.04

TABLE VII

PERCENTAGE DIGESTIBILITY IN VITRO OF DRY MATTER IN BERMUDAGRASS  
FORAGE AND HOLOCELLULOSE AS AFFECTED BY TIME OF CUTTING

Sample No. <sup>a</sup>	Digestibility of Dry Matter in Bermudagrass Forage			Digestibility of Holocel- lulose Isolated from 1961 Bermudagrass Forage
	1960 <sup>b</sup>	1961	1962	
	%	%	%	%
1	71.15	68.00	62.02	85.23
2	63.87	65.60	67.69	87.11
3	62.62	61.22	58.26	86.60
4	63.41	62.30	62.46	-
5	63.93	61.09	63.07	89.85
6	63.12	62.43	59.88	88.92
7	66.00	61.61	63.57	88.38

<sup>a</sup> See footnote a to Table III.

<sup>b</sup> Each value for 1960 is the average result of two experiments.

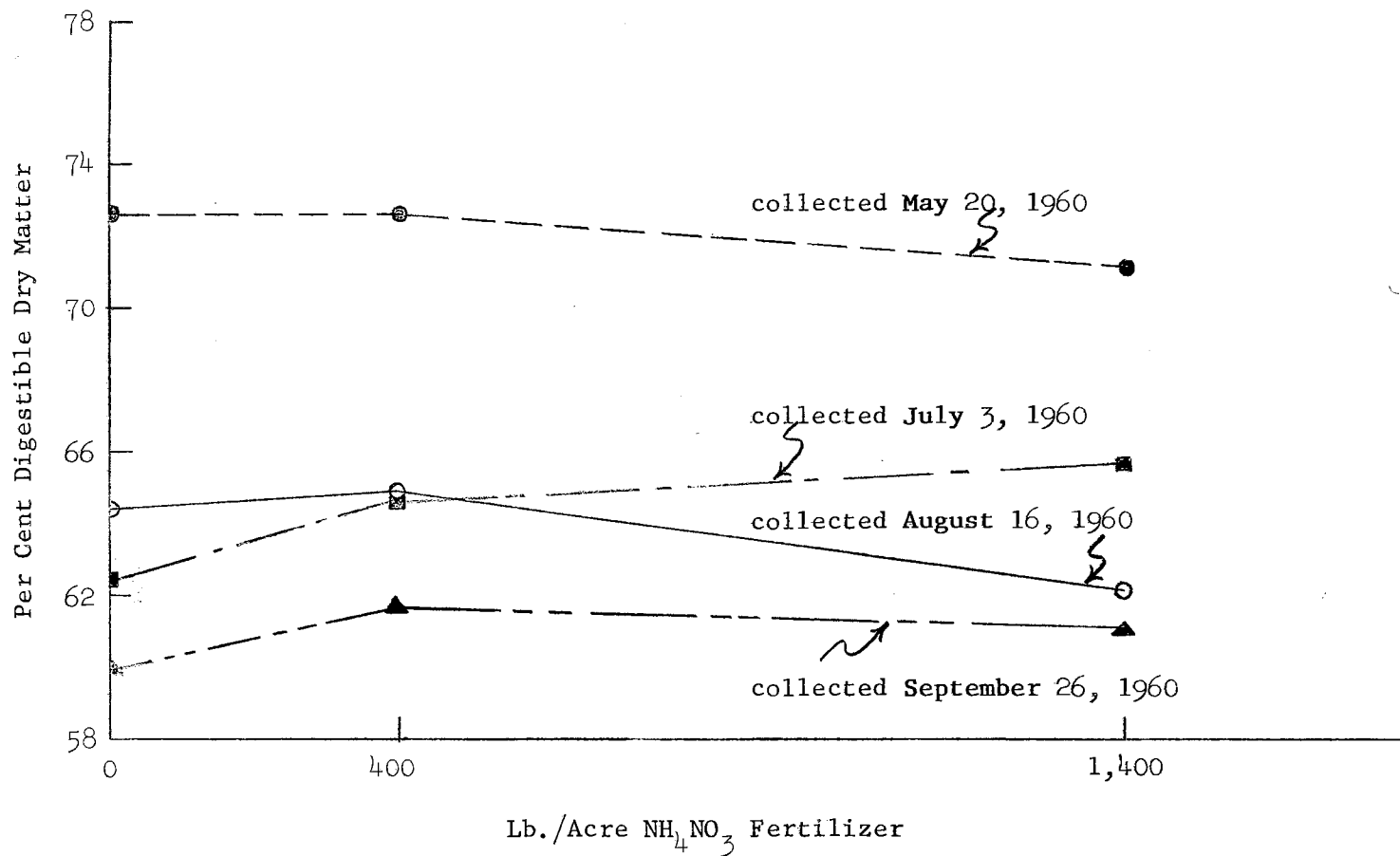


Figure 3. Percentage Digestibility In Vitro of Dry Matter in Bermudagrass Forage as Affected by  $\text{NH}_4\text{NO}_3$  Fertilization.



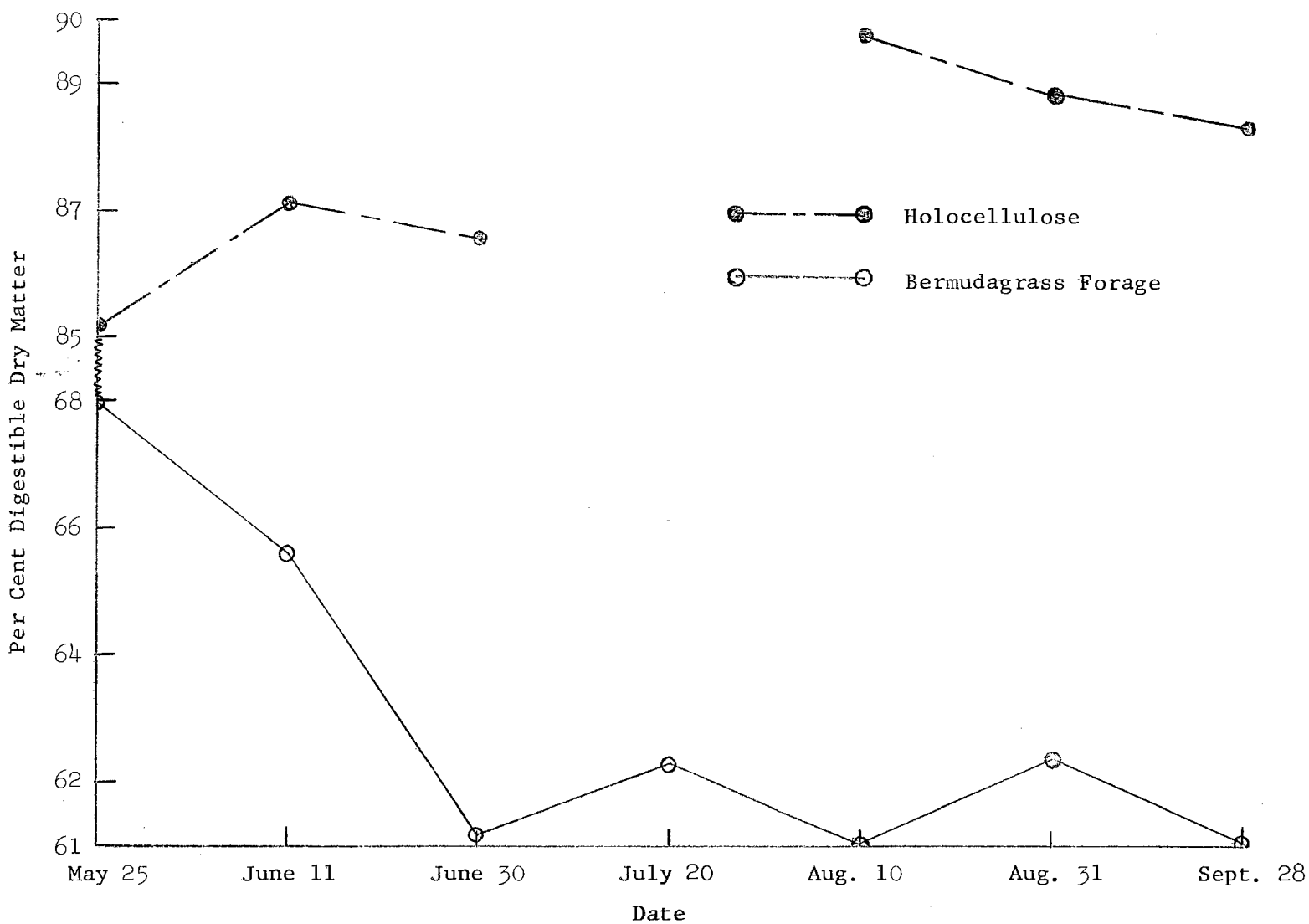


Figure 4. Percentage Digestibility In Vitro of Dry Matter in Bermudagrass Forage and Holocellulose as Affected by Time of Cutting (1961).

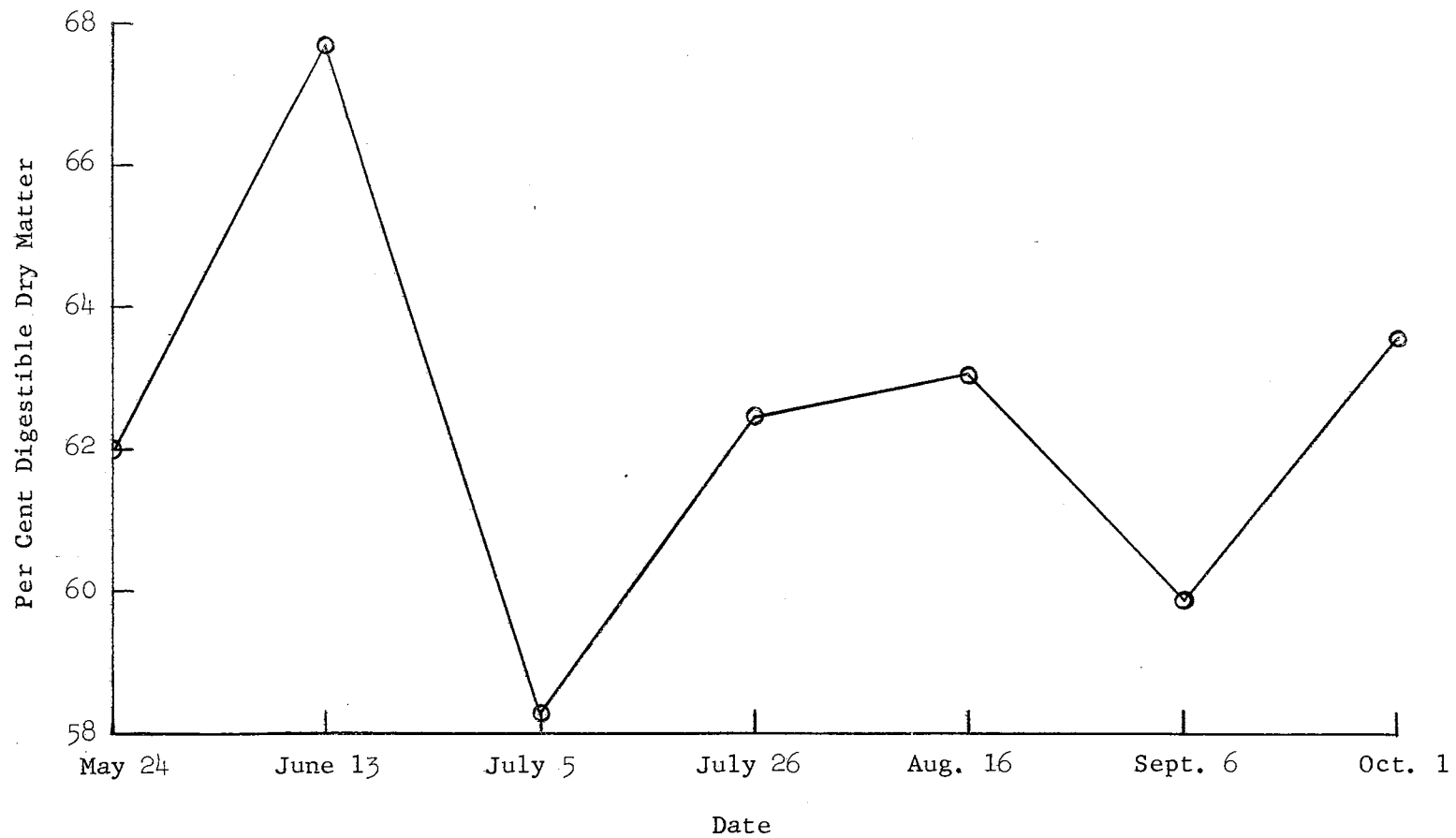


Figure 5. Percentage Digestibility In Vitro of Dry Matter in Bermudagrass Forage as Affected by Time of Cutting (1962).

ibility leveled off, with a few minor fluctuations, at an average figure of approximately 62 per cent.

Per Cent Digestibility In Vitro of Dry Matter in Bermudagrass

Forage as Affected by Lignin Content: Comparison of percentage lignification with in vitro digestibility data indicates that the lignin in bermudagrass forage affected its digestibility. Table VIII and Figure 6 show that, in general, as lignification increases, in vitro digestibility decreases. Analysis of these data proves that they are statistically significant at the 5 per cent level of probability. Figure 4 indicates that when bermudagrass forage was subjected to the acid chlorite treatment for removal of lignin, the digestibility was increased by as much as 30 per cent. Dehority and Johnson (11) have presented evidence showing that the manner in which lignin was laid down may affect digestibility as much as the total concentration of lignin present in the forage, and this may be true for bermudagrass forage.

TABLE VIII

PERCENTAGE DIGESTIBILITY IN VITRO OF DRY MATTER  
 IN BERMUDAGRASS FORAGE AS AFFECTED BY  
 LIGNIN CONTENT

Sample No. <sup>a</sup>	Digestibility			Lignin Content		
	1960	1961	average	1960	1961	average
	%	%	%	%	%	%
1	70.70	68.00	69.35	6.63	6.11	6.37
2	63.03	65.60	64.32	7.07	6.48	6.78
3	62.45	61.22	61.84	6.62	7.46	7.04
4	63.48	60.41	61.94	8.08	6.60	7.34
5	61.36	59.20	60.28	8.87	7.18	8.02
6	66.15	60.12	63.14	7.65	6.72	7.18
7	68.87	59.30	64.08	7.18	7.40	7.29

<sup>a</sup>See footnote a to Table III.

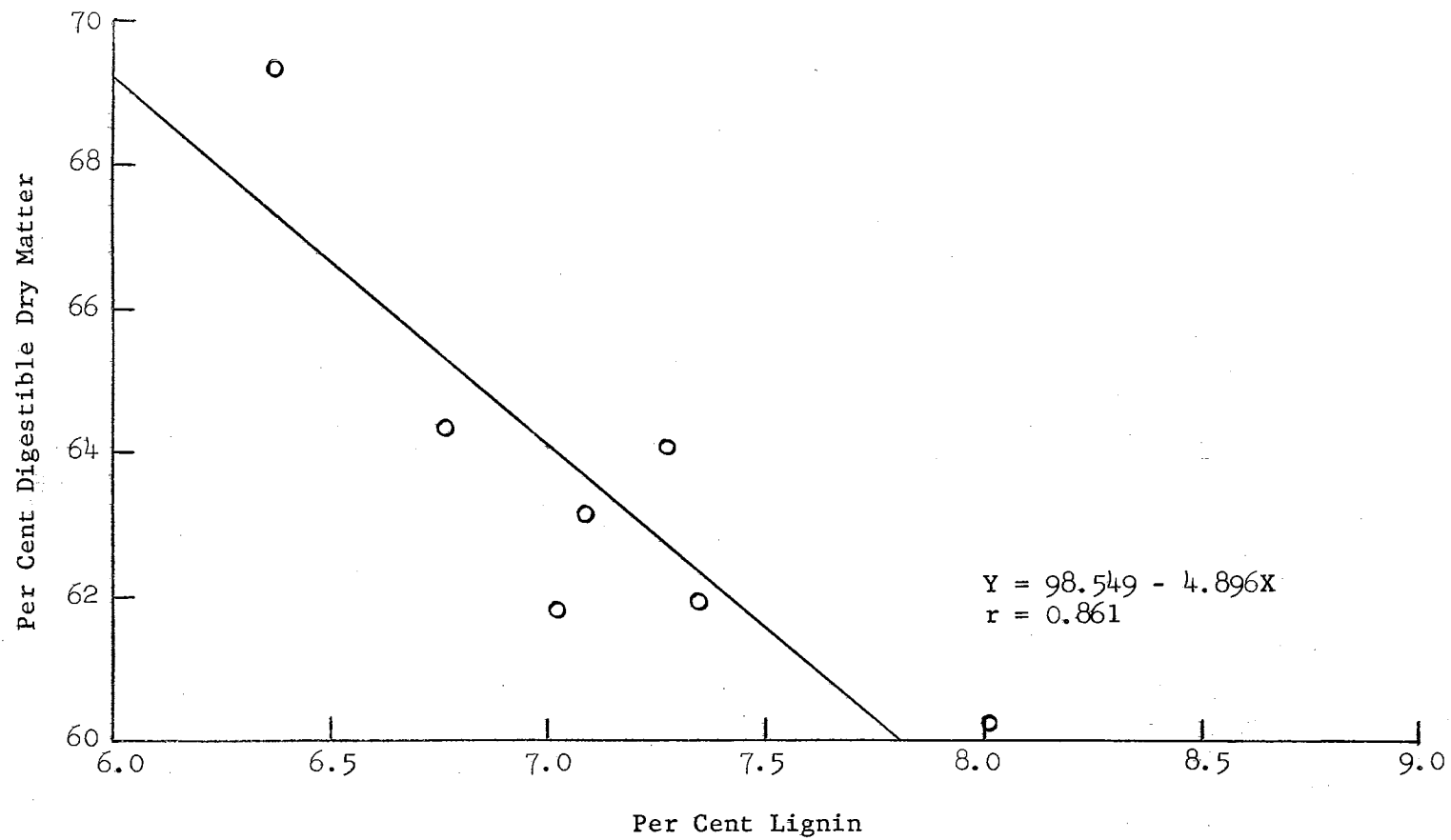


Figure 6. Percentage Digestibility In Vitro of Dry Matter in Bermudagrass Forage as Affected by Lignin Content.

## SUMMARY

Samples of bermudagrass forage collected over a three year period were studied to determine the effect of rate of  $\text{NH}_4\text{NO}_3$  fertilization on selected chemical components, and the effect of rate of fertilization, chemical composition, and time of cutting on in vitro dry matter digestibility.

The forage samples were analyzed for holocellulose, hemicellulose, cellulose, lignin, and in vitro dry matter digestibility.

The following general trends were noted:

- a. Increased  $\text{NH}_4\text{NO}_3$  fertilization caused a decrease in the percentage of holocellulose laid down by the forage.
- b. The rate of  $\text{NH}_4\text{NO}_3$  fertilization had little effect on the percentage of cellulose laid down by the forage.
- c. Increased  $\text{NH}_4\text{NO}_3$  fertilization caused a decrease in the percentage of hemicellulose laid down by the forage.
- d. In vitro dry matter digestibility appeared to be little affected by the rate of  $\text{NH}_4\text{NO}_3$  fertilization, and holocellulose, cellulose, and hemicellulose content of the forage.
- e. Forage samples collected during May and early June were more digestible than similar samples collected during the remainder of the growing season.
- f. Increased lignification caused a decrease in in vitro digestibility.

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VITA

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Master of Science

Thesis: A STUDY OF THE EFFECT OF RATE OF AMMONIUM NITRATE FERTILIZATION, CHEMICAL COMPONENTS, AND TIME OF CUTTING ON THE IN VITRO RUMEN DIGESTION OF BERMUDAGRASS FORAGE

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