Eu-doped CaF₂ grown on Si(100) substrates by molecular beam epitaxy

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(Received 22 May 1995; accepted for publication 23 July 1995)

Eu is incorporated into CaF₂ films grown on Si(100) by molecular beam epitaxy using elemental Eu evaporation. Eu doping as high as 4.05 at. % does not significantly degrade the surface morphology, indicating a relatively high solubility of Eu in CaF₂. Photoluminescence spectra from Eu-doped CaF₂ show strong blue emissions from Eu²⁺ ions in cubic sites. The inhomogeneous broadening of the zero-phonon line near 24 190 cm⁻¹ is reduced by ~20% upon *in situ* annealing at 1100 °C. © 1995 American Institute of Physics.

The increasing interest in the incorporation of rare-earth (RE) elements in epitaxial thin films of the alkaline-earth fluorides is motivated by potential applications in solid state microfilm or microcavity lasers. Growth of these materials on Si substrates offers potential compatibility with Si-based technology. Nd- and Er-doped CaF₂ films grown on Si substrates by molecular beam epitaxy (MBE) have exhibited strong photoluminescence at 1.04 μ m and 1.54 μ m, respectively.¹⁻³ Eu-doped CaF₂ bulk crystals have been extensively investigated⁴⁻⁹ for use as blue emission devices since a strong fluorescence at 420 nm is observed upon UV irradiation. Although laser action in Eu-doped CaF₂ bulk crystals is hindered by excited state absorption,¹⁰ such films can be used as light emitters in multicolor thin-film electroluminescence devices.¹¹ Recently, Eu²⁺ incorporated in CaF₂ films grown on Si(111) substrates by MBE has been used as a sensitive probe of elastic strain in the heteroepitaxial layers.^{12,13} As reported, the growth of the Eu-doped CaF₂ films was carried out by sublimation of a CaF₂ single crystal containing 0.05% divalent Eu.

In this letter, we present the MBE growth of Eu-doped CaF_2 on Si(100) substrates using elemental Eu evaporation. The incorporation of Eu in epitaxial CaF_2 films was investigated by x-ray photoelectron spectroscopy (XPS), reflection high-energy electron diffraction (RHEED), and photoluminescence (PL) measurements. We demonstrate that Eu concentrations as high as 4.05 at. % do not significantly degrade the surface morphology, indicating a relatively high solubility of Eu in the CaF₂ matrix. PL spectra from Eu-doped CaF₂ show a strong blue to violet emission expected from Eu²⁺ ions in cubic sites. *In situ* post-growth annealing improves the morphology of the films and reduces the inhomogeneous broadening of the zero-phonon line of Eu²⁺ interconfiguration transitions.

Growth of Eu-doped CaF_2 on Si(100) was carried out in

an Intevac Modular GEN II MBE system with a background pressure of ~10⁻¹⁰ Torr throughout the deposition. The 3-inch diameter *p*-type Si(100) substrates (Silicon Sense, Inc.) were cleaned using the Shiraki method.¹⁴ The passivating oxide formed during the *ex situ* cleaning procedure was thermally desorbed in the growth chamber after several minutes at ~1100 °C. A surface prepared in this manner exhibits no discernible oxygen or carbon traces as examined by Auger electron spectroscopy. High purity polycrystalline CaF₂ was evaporated from a graphite-coated PBN crucible at a typical growth rate of 20 Å /min for a cell temperature of 1300 °C. An elemental source of Eu was evaporated from a separate low-temperature effusion cell heated to 300–400 °C to give beam equivalent pressures of 3.5×10^{-10} – 1.9×10^{-8} Torr.

CaF₂ films grown on Si(100) are known to always exhibit a {111} faceted morphology because the surface energy of the CaF₂(100) face is much larger than for Si(100) and CaF₂(111) faces.¹⁵ However, growth morphology and crystalline quality can be improved by *in situ* annealing.^{16,17} In our experiments, a 400 Å CaF₂ buffer layer, a 3600 Å Eudoped CaF₂ layer, and a 200 Å CaF₂ top layer were deposited on a Si(100) substrate at 580 °C. Most of the samples in this work were annealed *in situ* at 1100 °C for a few minutes at the end of growth.

XPS experiments were conducted at room temperature in an adjoining analysis chamber to eliminate complications caused by exposure to the atmosphere. This chamber is equipped with a VG100AX hemispherical analyzer and a twin-anode x-ray source used in conjunction with a VGX900 data acquisition system (Fisons Instruments). XPS spectra were taken using the Al $K\alpha_{1,2}$ source ($h\nu$ =1486.6 eV) at a potential of 15 kV and with 10 mA emission current. The binding energies were referenced to F 1s at 684.5 eV. RHEED patterns were produced using a Varian electron gun operated at 9.5 keV and at an angle of incidence of 1° to 3° and digitized using a CCD camera and data acquisition software developed by *k*-space Associates, Inc. Photolumines-

Appl. Phys. Lett. 67 (13), 25 September 1995

0003-6951/95/67(13)/1891/3/\$6.00

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FIG. 1. Dependence of Eu concentration (solid circles) and integrated intensity of Eu $3d_{5/2}$ XPS peak (open circles) on inverse Eu cell temperature.

cence (PL) measurements were performed at 10 K using the 325 nm line of a HeCd laser with an excitation power of 5-7 mW. PL signals were collected by a 1 m Spex double-grating monochromator and detected by a S20 photomultiplier connected to a photon counting data acquisition system.

Figure 1 shows Eu concentrations in epitaxial CaF_2 films grown at 580 °C as a function of inverse Eu cell temperature. The atomic percentage of Eu is calculated using the relations

at.% =
$$\frac{J_{\text{Eu}}}{J_{\text{CaF}_2} + J_{\text{Eu}}} = \frac{1}{(1 + J_{\text{CaF}_2}/J_{\text{Eu}})}$$

and

$$\frac{J_{\text{CaF}_2}}{J_{\text{Eu}}} = \frac{P_{\text{CaF}_2}}{P_{\text{Eu}}} \cdot \frac{\eta_{\text{Eu}}}{\eta_{\text{CaF}_2}} \sqrt{\frac{T_{\text{CaF}_2} \cdot M_{\text{Eu}}}{T_{\text{Eu}} \cdot M_{\text{CaF}_2}}}$$

where *J* is the beam flux, *P* is the measured beam equivalent pressure, η is the ionization efficiency relative to nitrogen (2.4 and 1.7 for Eu and CaF₂, respectively), *T* is the absolute temperature of the effusion cell, and *M* is the molecular weight. Arrhenius behavior with an activation energy of 1.40 eV is observed up to ~7.48 at. %.

The XPS spectra of Fig. 2 show Eu $3d_{3/2}$ and Eu $3d_{5/2}$ core levels from three Eu-doped CaF₂ samples grown at different Eu cell temperatures. For comparison, the XPS spectrum of an undoped CaF₂ film is also shown. The heights of the Eu 3d peaks increase as the Eu cell temperature increases and the integrated intensities of these peaks depend exponentially on the inverse of the Eu cell temperature. As an example, the temperature dependence of the integrated intensity of the Eu $3d_{5/2}$ peak is plotted in Fig. 1. The fit of the XPS data to an Arrhenius plot with an activation energy of 1.39 eV indicates nearly the same temperature dependence as the Eu concentration.

Figure 3 shows RHEED patterns recorded with the electron beam along the [011] and [001] directions during growth of a CaF₂ film with a Eu concentration of ~4.05 at. %. A well defined two-domain (2×1) RHEED pattern of the Si(100) substrate appears after oxide desorption [Fig.



FIG. 2. XPS spectra showing Eu $3d_{3/2}$ and Eu $3d_{5/2}$ core levels and the x-ray induced Ca LMM Auger electron transition for samples grown using various Eu cell temperatures: (a) 400 °C; (b) 380 °C; (c) 360 °C; (d) undoped.

3(a)]. The spotty RHEED patterns shown in Fig. 3(b), indicating three-dimensional island growth, were recorded after 400 Å of CaF₂ growth. The incorporation of Eu does not cause any apparent change in the surface structure of the CaF₂ matrix as evidenced by comparing the RHEED patterns of the subsequent Eu-doped CaF₂ layer [Fig. 3(c)] with those of the undoped CaF₂ buffer layer [Fig. 3(b)]. However, surface degradation becomes visible under optical microscopy as the doping level reaches 7.48 at. % even though the RHEED patterns remain unchanged. Precipitation of Eu at high doping levels may be responsible for the surface degradation. The effect of *in situ* annealing is evident in the RHEED patterns shown in Fig. 3(d). Sharp diffraction streaks with visible Kikuchi lines emerge immediately after the substrate temperature is raised to 1100 °C, which indi-



FIG. 3. RHEED patterns along the $[0\overline{1}1]$ and [001] azimuths recorded: (a) after desorption of oxide from Si substrate; (b) after deposition of a 400-Å-thick CaF₂ layer; (c) after deposition of a 3600-Å-thick Eu-doped CaF₂ layer; (d) after *in situ* annealing at 1100 °C for two minutes.



FIG. 4. PL spectra of (a) annealed at ~1100 °C and (b) as-grown Eu-doped CaF₂ films grown under the same conditions and with the same Eu concentration of ~0.77 at. %. The full widths at half-maximum of the zero-phonon lines are 38 cm⁻¹ and 46 cm⁻¹, respectively. Both films are ~4200 Å thick. Both spectra were taken at 10 K.

cates a smooth surface.¹⁷ While *in situ* annealing improves the growth morphology by smoothing out the $\{111\}$ facets on the surface,¹⁷ it does not alleviate the morphological features attributed to Eu precipitation.

Figure 4(a) shows the PL spectrum of a CaF₂ film with a Eu concentration of ~ 0.77 at. %. A sharp zero-phonon line near 24 190 cm⁻¹ accompanies a broad vibronic sideband peaking at about 23 500 cm⁻¹. This spectrum is very similar to the fluorescence spectra of Eu²⁺ ions in CaF₂ bulk crystals reported previously.^{5,8} The zero-phonon line shown in Fig. 4(a) arises from the electronic transition from the lowest lying $\Gamma_8(4f^{65}d)$ level of Eu²⁺ to the ${}^8S_{7/2}(4f^{7})$ ground state. The peaks between the vibronic sideband maximum and the zero-phonon line are associated with the density of states maxima of various optical and acoustic phonons. All our samples, grown with various Eu concentrations, exhibit the spectral features shown in Fig. 4(a). No emission indicative of a ${}^5D_0 \rightarrow {}^7F_J$ Eu³⁺ transition was observed in the entire scanned range of 15 000–25 000 cm⁻¹.

Our PL data demonstrate the predominance of Eu²⁺ ions in cubic symmetry sites of CaF₂ single crystal films prepared by MBE. This differs from previous reports on the preparation of Eu-doped CaF₂ bulk crystals^{6,9} using the Bridgman-Stockbarger technique, where a small amount of Eu mixed with CaF₂ in an excess of fluorine resulted in predominately Eu³⁺ ions in cubic or other lower symmetry sites. Chemical or irradiation reduction processes are required to convert the Eu ions in the bulk crystals from trivalent states to divalent states. Our results suggest that interstitial F⁻ ions, substitutional O^{2-} ions, or other charged species which are necessary to compensate the additional positive charge of Eu^{3+} in CaF₂ bulk crystals are absent in the MBE-grown Eu-doped CaF₂ films. No excess F⁻ ions are available because our experiments employ molecular CaF2 and elemental Eu beams provided by sublimation of CaF₂ and Eu sources in a low background pressure environment. In Nd-doped CaF₂ samples prepared by MBE,¹ NdF₃ was intentionally selected to provide excess F^- ions to enable substitution of Nd^{3+} ions for divalent cations in the CaF₂ matrix. These results imply that the valence states of RE elements in fluorides can be controlled by using different RE dopant species.

To illustrate the effect of in situ annealing on the PL efficiency of Eu²⁺ ions in epitaxial CaF₂ films, the PL spectra of annealed and as-grown films with the same growth conditions and Eu concentration are shown in Figs. 4(a) and 4(b), respectively. The peak intensity of the zero-phonon line remains nearly the same while the line width is narrowed by \sim 20% and the integral intensity of the vibronic sideband is reduced by $\sim 13\%$ upon annealing. It is evident that in situ annealing, which significantly improves the growth morphology, causes only minor changes in the PL efficiency of Eu^{2+} . The insensitivity of the optical activity of Eu^{2+} to in situ annealing at temperatures as high as 1100 °C suggests that most of the Eu ions entering the CaF₂ matrix are already located substitutionally in divalent cation sites during growth at 580 °C. The reduced broadening of the zero-phonon line may result from embedded Eu ions moving closer to their equivalent positions upon annealing.

In summary, the growth of Eu-doped CaF₂ thin films on Si(100) has been realized by MBE using an elemental Eu source and a substrate temperature of 580 °C. Good surface morphology is observed in samples with Eu concentration as high as 4.05 at. %. Photoluminescence spectra from these samples show strong blue emissions attributed to Eu²⁺ ions in cubic sites. *In situ* annealing at ~1100 °C reduces the inhomogeneous broadening of the zero-phonon line and significantly improves the growth morphology without dramatically changing the PL efficiency of Eu²⁺ ions.

We thank Joel Young for technical support with the MBE laboratory. This work was partially supported by the National Science Foundation through grant number OSR-9108771.

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