

FIELD PERFORMANCE OF SURTREAT TPS A SURFACE APPLIED, PENETRATING CORROSION INHIBITOR

Construction Report March, 2004

Daniel Q. Humphrey Research Project Manager

Gary Williams, P.E. Engineering Manager



Planning & Research Division
Oklahoma Department of Transportation
200 N.E. 21st Street, Room 3A7
Oklahoma City, Oklahoma 73105
Telephone (405) 521-2671
FAX (405) 521-6948

TG340 .F54 2004 OKDOT Library



TECHNICAL REPORT DOCUMENTATION PAGE

1. REPORT NO. FHW AOK 04(04)	2. GOVERNMENT ACCESSION NO.	3. RECIPIENT'S CATALOG NO.			
4. TITLE AND SUBTITLE		5. REPORT DATE			
Field Performance of Surtreat TPS	A Surface Applied,	March 2004			
Penetrating Corrosion Inhibitor		6. PERFORMING ORGANIZATION CODE			
7. AUTHOR(S)		8. PERFORMING ORGANIZATION REPORT			
Daniel Q. Humphrey					
9. PERFORMING ORGANIZATIONNAME AND ADD	PRESS	10. WORK UNIT NO.			
Oklahoma Department of Transpor	tation				
Planning and Research Division					
200 N.E. 21st Street. Room 2A2		11. CONTRACT OR GRANT NO.			
Oklahoma City, OK 73105		Job Piece Number: 20679 (05)			
12. SPONSORING AGENCY NAME AND ADDRESS		13. TYPE OF REPORT AND PERIOD COVERED			
U.S. Department of Transportation		Construction Report			
Federal Highway Administration		September 2003 - March 2004			
300 N. Meridian Ave., Suite 105-S		14. SPONSORING AGENCY CODE			
Oklahoma City, OK 73107					
15. SUPPLEMENTARY NOTES					
16.ABSTRACT					
		cting with the cement phase of the concrete			
to increase hardness and compressi	ve strength, decrease permeal	rility, and render the goacrete inert to attack			
and penetration by acids, bases and	other aggressive chemical so	betions C. C. C. C.			
17. KEY WORDS	18. DISTRIBUTION STATEM				
Corrosion Inhibitor, Existing Structu		is publication is available from The Planning			
Reinforcing Steel	and Research Divis	ion, Oklahoma DOT.			
19. SECURITY CLASSIF. (OF THIS REPORT) Unclassified	20. SECURITY CLASSIF. (OF THIS PAGE Unclassified) 21. NO. OF PAGES 22. PRICE 42			

ACKNOWLEDGMENTS

The author would like to express his appreciation to Bryan Hurst, Wilson Brewer, David Girdner, Sam Ruble, George Murray, Chuck Donovan, and Jay Herbert from the Planning & Research Division of the Oklahoma Department of Transportation in the role of researching, monitoring, and recording data during construction. Others needing to be acknowledged are Mark Majors and Walt Peters, Project Engineers for the Bridge Division with the procedures and planning of the project: Muskogee Residency for controlling the field operation; Don Olsen of Surtreat the prime contractor. A special thanks goes to Gary Williams, the Engineering Manager from Planning & Research Division that organized and developed the work plan as well as helping oversee the entire project.

The contents of this report reflect the views of the author(s) who is responsible for the facts and the accuracy of the data presented herein. The contents do not necessarily reflect the views of the Oklahoma Department of Transportation or the Federal Highway Administration. This report does not constitute a standard, specification, or regulation. While trade names may be used in this report, it in not intended as an endorsement of any machine, contractor, process, or products.

SI (METRIC) CONVERSION FACTORS

Ap	proximate C	Conversi	ons to SI Un	its	App	roximate Co	nversio	ns from SI U	nits
Symbol	When you know	Multiply by	To Find	Symbol	Symbol	When you know	Multiply by	To Find	Symbol
		LENGTH					LENGTH		
in	inches	25.40	millimeters	min	mm	millimeters	0.0394	inches	in
A	feet	0.3048	meters	m	m	meters	3.281	feet	ft
y.d	yards	0.9144	meters	233	m	meters	1.094	yards	yds
mi	miles	1.609	kilometers	kın	km	kilometers	0.6214	miles	1111
		AREA					AREA		
in^2	square inches	645.2	square millimeters	min ²	mm²	square millimeters	0.00155	square inches	in²
ft ²	square feet	0.0929	square meters	m²	m²	square meters	10.764	square feet	ħ÷
yd ²	square yards	0.8361	square meters	m^2	m²	square meters	1.196	square yards	yd ²
ac	acres	0.4047	hectacres	ha	ha	hectacres	2.471	acres	ac
mi ²	square miles	2.500	square kilometers	km²	km²	square kilometers	0.3861	square miles	mi²
		VOLUME					VOLUME		
fl oz	fluid ounces	29.57	milliliters	mL	mL	milliliters	0.0338	fluid ounces	fl oz
gal	gallon	3.785	liters	L	L	liters	0.2642	gallon	gal
ft ³	cubic feet	0.0283	cubic meters	m³	m³	cubic meters	35.315	cubic feet	tt,
yd³	cubic yards	0.7645	cubic meters	m³	ım²	cubic meters	1.308	cubic yards	y.d³
	51	MASS					MASS		
OZ	ounces	28:35	grams	g	g	grams	0.0353	ounces	oz
lb	pounds	0.4536	kilograms	kg	kg	kilograms	2.205	pounds	lb
Τ	short tons (2000 lb)	0.907	megagrams	Mg	Mg	megagrams	1.1023	short tons (2000 lb)	I.
TEMPERATURE (exact)						TEM	PERATURE	(exact)	
·F	degrees Fahrenheit	(°F-32) 1.8	degrees Celsius	' C	'C	degrees Fahrenheit	9 5(°C)+32	degrees Celsius	°F
	FORCE and	PRESSUR	E or STRESS			FORCE and	PRESSUR	E or STRESS	
lbf	poundforce	4.448	Newtons	N	N	Newtons	0.2248	poundforce	lbf
lbf in ²	poundforce per square inch	6.895	kilopascals	kPa	kPa	kilopascals	0.1450	poundforce per square inch	lbf in ²

FIELD PERFORMANCE EVALUATION OF SURTREAT TPS A SURFACE APPLIED, PENETRATING CORROSION INHIBITOR

CONSTRUCTION REPORT

March 2004

Daniel Q. Humphrey
Research Project Manager

Gary Williams, P.E. Engineering Manager

Planning & Research Division

Oklahoma Department of Transportation

200 Northeast 21st Street RM 2A2

Oklahoma City, Oklahoma 73105

TABLE OF CONTENTS

EXECUTIVE SUMMARY	1
BACKGROUND	2
PURPOSE	2
OBJECTIVES	3
PROJECT INFORMATION	4
CONSTRUCTION	5
DATA COLLECTION	6
DISCUSSION	7
RESULTS	8
REFERENCES	9
APPENDIX A, Technical Data Sheet	A
APPENDIX B, Half-Cell Data	В

LIST OF FIGURES

Figure	1.	Bridge during application of Surtreat	4
Figure	2.	Water blasting the deck	5
Figure	3.	Water blasting the deck	5
		Treating the deck	
Figure	5.	Treating the deck	5
Figure	6.	Taking test cores	6
Figure	7.	Gathering chloride samples	6

EXECUTIVE SUMMARY

Corrosion is a major problem for most concrete structures, occurring in the steel bars used to reinforce the concrete. The expense incurred in North America as a result of corrosion-induced repair numbers in the billions of dollars, with more than 50% of the 575,000 bridges in the United States affected by corrosion. Corrosion inhibitors are one of the most cost-effective solutions to this problem, but little independent information is available on their effectiveness in actual use.

Reinforced concrete structures are usually durable in average environments, but their integrity can be compromised when exposed to carbonation or other environments such as chloride salts. [1] These environments reduce the potential service life of reinforced concrete structures and increase maintenance costs. A number of options exist for prolonging the life of concrete structures exposed to aggressive environments, and corrosion inhibitors are among these options. While corrosion inhibitors are available and have been used for increasing the protectiveness of concrete in new construction, [2] the focus of this report is on the reduction of corrosion in existing structures. The costs resulting from damage due to corrosion of existing concrete structures is estimated to be approximately \$8.3 billion per year for highway bridges alone, [3] and there are many other types of concrete structures that are also exposed to corrosion.[4]

Steel corrosion can be mitigated through several approaches. In concrete bridges, the damaged area can be periodically repaired. However, the cost of such repair over the life of a structure can often exceed the cost of total replacement. Another approach is cathodic protection. This can be very effective, but it requires long-term maintenance and a continual supply of electrical power or sacrificial anodes, depending on the selected mode.

BACKGROUND

The Oklahoma Department of Transportation (ODOT) experimentally treated the deck of an existing bridge with *Surtreat TPS*, a surface applied, penetrating corrosion inhibitor. *Surtreat TPS* reportedly can abate corrosion, by creating a passivating ionic film on the reinforcing steel. *Surtreat TPS* claims to permanently modify the cement matrix of existing PCC structures. These modifications result in increased strength and lower permeability. If successful, the use of *Surtreat TPS* could substantially reduce the cost of corrosion maintenance by arresting corrosion - caused spalls, thus reducing the necessity for patching and other repairs.

PURPOSE

The purpose of this study is to evaluate the effectiveness of *Surtreat TPS* corrosion inhibitor in a field performance situation. *Surtreat TPS* is a proprietary product produced by the *Surtreat Corporation* of Pittsburgh, Pennsylvania. According to the manufacturer's literature, *Surtreat TPS* penetrates into the PCC structure being treated, reacting with the cement phase of the concrete to increase hardness and compressive strength, decrease permeability, and render the concrete inert to attack and penetration by acids, bases and other aggressive chemical solutions. This project will focus on *Surtreat's* reported ability to decrease corrosion in an existing structure.

OBJECTIVES

- Conduct half-cell potential tests (ASTM 876-91) on the bridge (Bridge no. 4621 0140 X,
 NBI 17601, McIntosh County), before and after the application of *Surtreat TPS* to evaluate *Surtreat's* ability to reduce corrosion.
- * Conduct Electrical Indication of Concrete's Ability to Resist Chloride Ion Penetration tests (AASHTO 277) to determine if the compound demonstrates the ability to reduce the amount of chloride present in the concrete.
- Conduct pH testing using Germann Instruments Rainbow Dye Test on the existing deck before and after treatment to determine any change in pH.
- Conduct the Standard Test for Abrasion resistance by sandblasting (ASTM 418) before and after *Surtreat TPS* application to document any change in this property.
- Monitor the application process. A. Document application rate, including time required for treatment. B. Document lane closure requirements.
- Conduct surveys of the general condition of the deck at regular intervals after treatment.
- Conduct half-cell potential tests on the bridge (Bridge No. 4621 0140 X, NBI 17601, McIntosh County), one and two years after the application of *Surtreat TPS* to evaluate Surtreat's ability to reduce corrosion on a long-term basis.

PROJECT INFORMATION

Surtreat TPS has the potential to reduce maintenance and life cycle costs and can be used for deck rehabilitation. ODOT will be using this product on S.H. 150 (bridge) over lake Eufaula in McIntosh County as shown in Figure 1. The bridge deck is 100 percent in PONTIS condition state 4 (repaired areas and spalls are more than 10 percent but less than or equal to 25 percent of the total deck area). The project consisted of preparation and treatment of the bridge deck, 2505 square yards, with Surtreat TPS 2, a surface applied inorganic migratory corrosion inhibitor. The goal of the project is to abate corrosion at the level of the reinforcing steel, minimizing the need for future deck patching. ODOT has no prior experience with this product, and considers it an experimental application at this point, the application and the treated bridge will be monitored by the Research Section, Engineering Services Branch of ODOT's Planning & Research Division as described below.

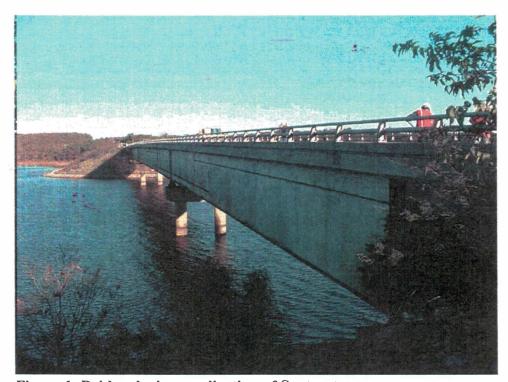


Figure 1. Bridge during application of Surtreat

CONSTRUCTION

Concrete Services Corporation of Tulsa was awarded the contract to prepare and treat the deck. Traffic control was provided by ODOT Division 1 maintenance personnel. ODOT specified treatment of the deck with *Surtreat TPS* 2 at a rate of one gallon per 100 square feet. Preparation of the bridge deck included careful water blasting to remove surface dirt and debris, without causing further delamination or removal of repair material. One lane of traffic was maintained in service at all times during the construction period. The cleaning and treatment of the entire bridge took 6 days. The first 2 days were used to clean the entire bridge deck and the remaining 4 days for application of *Surtreat TPS* 2. *Surtreat* was applied with low pressure sprayers in 3 applications, followed by a clean water misting. Low areas were broomed to minimize puddling of the *Surtreat*. Actual lane closing time was 8 hours or less for each work day. Traffic was resumed on both lanes every day after the work area was rinsed. Traffic was restricted to one lane for less than 48 hours for the entire construction period, minimizing inconvenience to highway users.



Figure 2. Water blasting the deck



Figure 4. Treating the deck



Figure 3. Water blasting the deck



Figure 5. Treating the deck

DATA COLLECTION

Research activity was included in monitoring the application process, including any necessary preparation work such as cleaning, sandblasting, joint repair, etc. Information was gathered on application rate and time required to treat the bridge, with lane closure (time) requirements. A survey of the general condition of the deck was completed before *Surtreat TPS* application. Half-cell testing of the deck was done before and after *Surtreat* application to quantify any reduction in corrosion activity. In order to insure that the *Surtreat* has time to work, the time between application and half-cell testing was 4 months which was recommended by the *Surtreat* representative. Cores were taken before and after *Surtreat* application. These will be tested for abrasion resistance and permeability to determine the amount of improvement after treatment.

A second survey will be done one year after application. This will include half-cell testing, and a survey of the treated area's general condition. A final survey will be done two years after treatment. The same data will be collected for both the one-year and two-year surveys.



Figure 6. Taking test cores



Figure 7. Gathering chloride samples

DISCUSSION

For the *Surtreat TPS* to extend the service life of structures undergoing active chloride induced corrosion they must:

- penetrate the cover concrete to depth of the reinforcing steel
- penetrate accumulated corrosion products in pitting cells to the depth where oxidation of the steel is occurring and or retard or stop the cathode reaction.
- inhibit active chloride induced corrosion cells in the presence of relatively high chloride concentrations

It is difficult to determine if the *Surtreat* actually penetrated the concrete cover. This could be helped by having the manufacturer put some type of dye in the product and conduct some type of color test to determine the presence of the product and to what depth was obtained. A small number of cores were taken for the project. Tests on the cores may show the depth obtained by the product.

A question that needs to be answered is how much of the chemical is getting to the steel surface? Also how much is being absorbed into the concrete and with that how much concentration on the surface is needed to reduce or stop corrosion?

There have been some questions regarding the delamination of the bridge deck and if that would cause the half-cell numbers to be unreliable. The half-cell potential numbers should have decreased if the Surtreat was working as intended. However, our results showed that the majority of the numbers either stayed the same or increased. This could show that the treatment was not as effective as we had expected, 4 months was not long enough to see these improvements in the half-cell numbers, or the delamination in the deck caused some of the numbers to be unreliable.

RESULTS

Electrical Indication of Concrete's Ability to Resist Chloride Ion Penetration (AASHTO 277):

Cha	rge Passed (Coulombs)	Relative Chloride Permeability
Prior to application	4712	High
4 months after application	4176	High

Abrasion Resistance of concrete by sandblasting (ASTM C418):

SSD Weight, g

	Before	After	Loss, g	% Loss
Prior to application	229.47	228.07	1.40	0.61
4 months after application	168.28	167.30	0.98	0.58

^{*} The reason for the difference in weight before and after is due to size of the cores that were tested.

Half-cell results can be found in Appendix B:

pH testing using Germann Instruments Rainbow Dye Test:

	pH level
prior to application	11
4 months after application	11

REFERENCES

- Bjegovic and B. Miksic (CORTEC). "Migrating Corrosion Inhibitor Protection of Concrete".
 Materials Performance Vol. 38, No. 11, pp. 52-56 (1999) November.
- 2. D. Bjegovic and B. Miksic, "MCI Protection of Concrete," Cortec Supplement to Materials Performance, January 2001, pages 10-13.
- 3. Cost of Corrosion Study, http://nace.org/nace/content/nacenews/news_win/html/costcorr.htm, July 2002.
- 4. Corrosion in Concrete, http://corrosion.ksc.nasa.gov/html/corrincon.htm, July 2002.

APPENDIX A

Technical Surtreat Data



SURTREAT TPS 2 Material Safety Data Sheet

SECTION I SUPPLIER INFORMATION

Common Name:

SURTREAT-TPS 2

Chemical Name Formula:

Supplier

SURTREAT

437 Grant Street, 1210 Frick Building

Pittsburgh, PA 15219

Phone:

(412) 281 - 1202

Emergency Phone:

Chem-Tel 1-800-255-3924 (24 hours)

Date Prepared

Monday, June 09, 2003

SECTION II HAZARDOUS INGREDIENT INFORMATION

CFR 29 Part 1910.1000 Table Z-1 (July 1, 1996 issue)

Ingredient

CAS Number

PEL-OSHA (ppm)

TWA-OSHA (mg/m³) TLV-ACGIH (ppm)

STEL-ACGIH (ppm)

This product contains no hazardous

materials.

SECTION III PHYSICAL/CHEMICAL CHARACTERISTICS

Boiling Point:

212 1.094

Specific Gravity: Melting Point:

N/A

12

N/A

Vapor Pressure (mm Hg):

Vapor Density (Air=1):

N/A

Evaporation Rate (Butyl Acetate = 1):

N/A

Solubility in Water:

100%

Appearance and Odor:

A clear liquid with a sweet odor.

SECTION IV FIRE AND EXPLOSION HAZARD DATA

Flash Point:

none

Auto-Ignition Temperature:

°F N/A

LEL:

% N/A

UEL:

N/A

Extinguisher Media:

This material is not expected to burn. None known.

Special Fire Fighting Procedures: Unusual Fire and Explosion Hazards:

None known

SECTION V REACTIVITY DATA

Stability

Conditions and Materials to Avoid

Flammable hydrogen gas may be produced on prolonged contact with metals such as aluminum, tin, lead, and zinc. Avoid contact with glass and reactive metals.

Hazardous Decomposition or By-Products Polymerization

Hydrogen gas. Will not occur.

Conditions to Avoid

None known

SECTION VI HEALTH HAZARD DATA

Inhalation

Acute

Causes irritation to the respiratory tract.

Chronic

No data available.

SURTREAT CORP. INFORMATION

INTRODUCTION

Technological advances made by Surtreat Corp. are based on the fact that the deterioration of Portland cement concrete is largely a chemical process which can be chemically inhibited and reversed.

The internationally accepted SURTREAT® method is the most effective and inexpensive way to restore the original properties of deteriorating concrete and to offer long-term protection to all surfaces exposed to hostile environments.

BENEFITS

The unique SURTREAT system increases the ability of concrete to resist physical and chemical breakdown by:

- Increasing compression strength, hardness
- Inhibiting reinforcing steel corrosion
- Resisting attack and penetration by acids, salts and other aggressive chemical solutions
- Extracting chlorides and pacifying chloride activity
- Pacifying alkali-silica reaction
- Elevating the pH

APPLICATION

The benefits are achieved by continuous application of a combination of SURTREAT proprietary chemical formulations which penetrate into the concrete microstructure in liquid and vapor state to combine with the cement phase of the concrete and deposit on steel components. Application is made by Surtreat Corp. or a qualified contractor.

TESTING

Performance of **SURTREAT-GPHP** has been verified on numerous projects as well as through extensive laboratory testing performed by Professional Service Industries, Inc., Pittsburgh, PA. Results of the tests reflect the following:

Compression Strength Increase

ASTM C-42 increased by 300 to 2000 psi depending on original strength and number of applications.

Water Penetration Reduction

100% resistance after 14 days exposure to 6 inch column of water.
75% resistance after 28 days exposure to 6 inch column of water.
90% resistance after 24 hours @ 100 psi.

Freeze Thaw Resistance

ASTM C-672 – 50 cycles no surface loss. AASHTO T-161 – 146 cycles no surface loss.

Chloride Penetration Resistance

AASHTO T-250 – 30% decrease at 1 inch depth. ASTM C-672 – 42% decrease at 2 inch depth.

Water-soluble Chloride Reduction

Gel pour extraction method. Reduced by 58% at 1-inch depth. Reduced by 67% at 2-inch depth.

Increase in Hardness

ASTM C-418 – 14% increase in hardness of new concrete.
ASTM C-414 – 64% increase in hardness of deteriorated concrete.
ASTM C-501 – 1000 cycles – 38% increase in wear index.

Reduced Rebar Corrosion Potential

ASTM C-876 – Half-cell potential – reduced voltage by 50% in 14 days (400 to 200mV).

Increased Surface Adhesion

No failure of epoxy bonded to treated surface.

Reduced Chemical Reactivity

Resistant to attack and pentration by concentrated hydrochloric acid.

Flexural Strength

Increased from 423 to 543 psi.

The majority of projects that involve installation of SURTREAT products include comprehensive testing and analysis before and after the treatment. Tests performed on site often include: Compression Strength (CAPO Pull Out), Water Permeability (Pressure Cell), pH (Indicator Dye), Corrosion Potential (Silver Chloride Half-Cell), Chloride Ion and Acid Reactivity. Customer representatives or engineers are invited to observe testing procedures and verify the results. Comprehensive project reports are submitted on request.

Surtreat Corp., 437 Grant Street, 1210 Frick Building, Pittsburgh, PA 15219 phone 412-281-1202, fax 412-281-1282, e-mail srtrt@nauticom.net

SURTREAT® CORROSION INHIBITION PROCESS AND FORMULATIONS

The Process of Corrosion

Reinforcing steel corrosion is the most common cause of failure of concrete structures. Steel corrosion is an oxidation process requiring the presence of air (oxygen) and moisture. Corrosion is promoted by anions, like chloride, and an acid environment. Corrosion is inhibited by an alkaline environment, cations and barrier coatings.

Freshly poured low water-cement ratio concrete provides excellent protection for imbedded reinforcement. However, as time passes, under the influence of salts and atmospheric acids cement bonds in concrete begin to deteriorate, concrete becomes increasingly permeable and susceptible to intrusion of contaminants and moisture. As the pH of concrete falls, electrochemical activity increases. The process perpetuates itself as oxidized rebar expands forcing concrete to crack and delaminate, thus encouraging air and water penetration. Once started, rebar corrosion can not be stopped by simply waterproofing the surface of the concrete, in fact application of a waterproofing membrane or coating will make the situation worse by trapping the chlorides and moisture inside the concrete microstructure which in combination with a low pH is a perfect environment for corrosion.

SURTREAT® Process

The corrosion process is influenced and promoted by penetration of air, water, chloride, and an acid environment.

The SURTREAT process for corrosion inhibition includes application of SURTREAT Volatile Corrosion Inhibitor (VCI), which permeates throughout the structure as a vapor and deposit on rebar forming a passive film on the surface of the white steel. Application of SURTREAT VCI is usually followed by an application of SURTREAT General Purpose Restoration and Protection Formulation (GPHP).

SURTREAT GPHP penetrates the concrete surface reacting with Portland cement and solidifies inside becoming a permanent part of the structure. SURTREAT GPHP inhibits corrosion by:

- Complexing water soluble chlorides
- Purging and inhibiting penetration of chlorides
- Raising the pH
- Restricting access of air and water to the rebar

SURTREAT GPHP also reacts with iron oxide on the surface of imbedded reinforcement to convert it into an inert mass resistant to water and air penetration.

The effects of SURTREAT® formulations on the corrosion process have been verified through field evaluation as well as lab analysis. The following is a brief summary of the test results performed in our laboratory as well as in the field. All tests are done using a silver chloride half cell, the threshold point for corrosion is -190mV.

LABORATORY EVALUATION

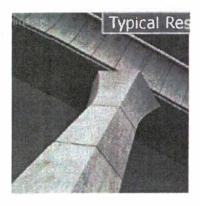
ID	CONDITION/EXPOSURE	mV	Kohm
<u>A</u>	Base Sample	+45 ~370	8.0 0.5
	Exposure to HCl/CaCl 5 applications of GPHP and	~3 /0	0.3
	1 application of VCI	-115	2.2
<u>B</u>	Base Sample 3 applications of GPHP and	-300	0.5
	1 application of VCI	-200	1.0
	After one week	-125	1.5
<u>C*</u>	Base Sample	-325	0.4
	3 applications of GPHP	-300	0.4
	1 application of VCI	-275	0.5
	2 applications of GPHP	-235	0.6
	After one week	-200	0.7

^{*}Sample labeled C came from a balcony of the Ocean Pines Complex in Cape Canaveral FL.

SUMMARY OF SILVER CHLORIDE HALF CELL MEASUREMENTS ON A BALCONY AT WINDJAMMER CONDOMINIUM, COCOA BEACH, FL

AREA	BEFORE mV	AFTER mV	NUMBER OF MEASUREMENTS			
A R	-326 -350	-153 -155	6			
<u>B</u> <u>C</u>	-229	-64	3			
D	-285	-179	3			
Surtreat a	pplication rate		50 sf/gal			
Number o	f applications		2			
Curing tin	ne		24 hours			
Half cell c	corrosion point		-190 mV			
Date perfe	ormed		6/16 - 6/17/95.			







The performance of Surtreat's chemistry has been verified by numerous independent engineering-consulting firms in the field as well as through extensive independent laboratory testing. The results of the testing reflect the following typical result characteristics:

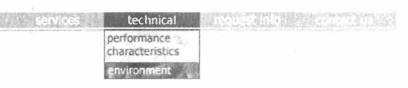
- Reduced Rebar Corrosion Potential ASTM (American Society for Testing Materials) C-876 - Half-cell potential showed a reduced voltage by 70% in 14 days (0.8 to 0.3)
- Water Penetration Reduction 100% resistance after 14 days exposure to 6-inch column of water. 75% resistance after 28 days exposure to 6-inch column of water. 90% resistance after 24 hours when 100 psi is applied.
- Freeze Thaw Resistance ASTM C-672 -50 cycles with no surface loss. AASHTO (American Association of State Highway and Transportation Officials) T-161 - 146 cycles with no surface loss.
- Chloride Penetration Resistance AASHTO T-250 -30% decrease at 1-inch depth.
- Water-soluble Chloride Reduction Reduced by 58% and 67% at 1 and 2 inches, respectively.
- Increase in Hardness ASTM C-418 14% increase in hardness of new concrete. ASTM C-414 64% increase in hardness of deteriorated concrete. ASTM C-501 1,000 cycles with a 38% increase in wear index.
- Reduced Rebar Corrosion Potential ASTM (American Society for Testing Materials) C-876 - Half-cell potential showed a reduced voltage by 70% in 14 days (0.8 to 0.3)
- Increased Surface Adhesion No failure of epoxy bonded to treated surface.
- Reduced Chemical Reactivity Resists reaction with concentrated hydrochloric acid
- Flexural Strength Increased from 423 to 543 psi.
- Increased pH Level Edges of concrete increased from 5 to 9, center of concrete increased from 9 to 12.

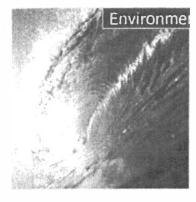
*Comprehensive project reports for select projects are available upon request. To request further information click here.

Intranet Login

Lega! Notice / Copyright @ 2002 Surtreat®. All rights reserved







Surtreat operates according to the principle that responsibility for the environment is an integral part of every project that we execute.

Surtreat Project Managers operate to comply with all applicable laws and regulations, with emphasis on prevention and to minimize adverse environmental impacts by identifying and controlling potential environmental risks in advance.

Furthermore, Surtreat uses water-soluble organic and inorganic materials that are environmentally safe. Our systems contain no VOC's (volatile organic compounds) and have been tested by leading environmental firms showing no effect on the turbidity, pH, or dissolved oxygen content levels in water

Surtreat's formulations bond inorganic compounds to the structure where they become part of the steel and concrete matrix indefinitely. Leaving no residues, coatings or materials that could potentially harm humans, animals, fish and the environment. Surtreat has been granted waivers and exemptions from the Environmental Quality Act, California Costal Commission, and other state and county agencies for use of its chemical treatment formulas, due to their non-toxic nature.

Intrarret Login

Legal Notice / Copyright @ 2002 Surtreat® All rights reserved.

APPENDIX B

Half-cell results

Mun to the heren 9/29/03 DATE: WEATHER: BRIDGE NO: 4621 0140 X NBI NO: 17601 TESTED BY: D. HUMPHIZEY HALF-CELL TEST PAGE / OF #5 PAGES a P6

MV

					· · · · · · · · · · · · · · · · · · ·	PAGE 2	OF #78	PAGES		-
	413	S48	545	450	ಬೀ	40\$	પલા	472	267	19
25			1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1				•	* * * * * * * * * * * * * * * * * * *		
		1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	1						9	
	319	331	પ ક્ષ્મ	524	248	324	402	451	201	23
20		8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8				1		7 7 7 8		
		1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	7 7 8 8 8 8			1 1 1 1 1			0 0 0 0 0	
-	415	9 90	388	49 5	228	420	274	પ _{ર્મિ}	81	19
15				9 9 9 8				# 9 9 9 9 9 9 9 9 9 9 9 9 9 9 9 9 9 9 9		*****
		50 50 60 60 60 60 60 60 60 60 60 60 60 60 60	6 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0		1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1					
	444	567	346	482	445	425	182	249	189	84
10					9 8 8 8					
			1 7 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1		· · · · · · · · · · · · · · · · · · ·	0 0 1 1 0 0 0				
	430	481	293	3 21	394	418	94	(3)	138	25
5				24						
					8 9 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8			7 9 9 6 8 8		
	518	541	584	449	405	576	490	628	234	22
50	55	60	65	70	75	80	85	90	95	

	541						PAGE 3	OF	F8 PAGES	,	
25	,106		147	391	135	195	276	464	190	232	ЧЗ
	24.0		216	श्टा	775	251	20	436	<i>2</i> 15	184	B
20	3 %						30		<i>C</i> 5	(96	
15	138		74	83	6 3	368	194	442	77	Gγ	72
				1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1				* * * * * * * * * * * * * * * * * * *			
10	U7	•4•	289	331,	(34	ાવર	262	126	251	406	16
5	206		246	116	62	99	382	121	129	124	29
	IC.	,				UD	20-		201	10000000000000000000000000000000000000	1 1 1 1 1 1 1 1 1 1 1 1
100	IS ₀	105	4 28 1	10 523	50o 115	120	125	130	135	140	145

DATE:	WEATHER:
BRIDGE NO:	
TESTED BY:	



				-		PAGE 5	OF #78	PAGES		_
0.5	478	576	448	216	334	252	446	393	195	332
25					6 6 6 6 5 1	1		۸	1 1 1 2 4 5 5 7	
	416	પુરુક	<i>3</i> 45	uu .	2 61 4	264	196	420	ul	299
20		450	343					(10		-40
					6 8 8	* * * * * * * * * * * * * * * * * * *		4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4		
15	465	455	418	435	338	466	247	254	110	349
	1,0					1 1 1 1 1 1 1 1 1 1 1			6 2 1 1 1 1 2 4 4	
	(Jo-	413	402	યા વ	Yes	434	28 1	341	273	263
10		11:2				•				
	4,		6 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4							
5	401	496	281	380	<i>ዓ</i> ን5	2de	35	295	92	157
				./*						
	428	505	५७ ३ 215	5TL	474	ઽ ૫૩	421	524	21A	407
200	205	210	215`	220`	225	230	235	240	245	

11.	1	Ť	(1)	Ī	1	317	1	4)		T.		(#)	1/	1	1
	DATE:		19/03						V	VEATHE	R:				
v.	BRIDG	ED BY:							-		-				

		-				PAGE 6	OF #78	PAGES		-
)He	10.				
25	405	473	250	346	212	496	253	325	45B	Sk
20					• • • • • • • • • • • • • • • • • • •	*))) (
				9 3 3 5 0 0		6 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0		•		
	44.	449	203	34	376	212	405	298	प्रा ।	35
20]					1					
				4 7 8 8		8		1		*
								1 0 1 1		
15	425	182	145	192	388	lbi	342	201	223	218
				0 0 0 0 0	1 1 1 1	6 3 3 5 7 7	* * * * * * * * * * * * * * * * * * *	0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0		İ
				0 0 1 0 2 3 8		• • • • • • • • • • • • • • • • • • •				
	248	148	175	364	348	418	3 0	138	252	33
10								9 b. F 6		
								5 6 6 8 8		***************************************
	264	24	218	333	2.5	196	236		261	1,
5		40	48		312	146	235	(15	256	Hol
									9	
	132	360	(৪ ৩ 265	340	383	200	360	140	278	304
50	255	260	265	270	275	280	285	290	295	

						PAGE 7	OF #78	PAGES	-
	285	243	422	S23	526	3T)	291	367	240
25				*	,				
- 0	580	410	555	SW	609	<i>475</i>	403	467	198
20		9 9 9 8 8 8		4 A A A A A A A A A A A A A A A A A A A					
- 4	732	313	392	332	142	82	25/4	424	40(
15									
-	39	थाउ	73	181	149	189	229	222	329
10									
will be all the same	173	3-2	²⁸ 2	315	134	(3) 24 (c)	8	86	89
5									
2	73	776	136	241	721	157	755	la	162
00	305	310	13 ∕√ 315	241 320	325	330	335	340	1 /3 345

DATE:	9	/29	103	1
BRIDGE	No):		

BRIDGE NO:	
TESTED BY:	

						PAGE 8	OF 18	PAGES	
	214	234	512	72N	218	207			
25		1 1 1 1 1 1 1 1 1 1 1 1		1					
	445	295	zen	332	204	<i>2</i> 71			
20		8 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4		0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0		A			
	u	365	417	247	206	277			
15									
	182	231	<i>3</i> 7%	460	192	120			
10						1 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0			
	378	าร	lo]	3 €0	90	144			e.
5									
	2-0	21/	4 -	UZI	175	o =			
350	355	360	365	370	375	380	385	390	395

DATE: WEATHER: BRIDGE NO: 4621 0140 X NBI NO: 17601 TESTED BY: D. HUMPLEY HAUF - CELL TEST N PAGE / OF - 18 **PAGES**

-- 12

- 1 -

MV

DATE: 2	/18/04
BRIDGE N	O:
TESTED B	V·

			,		PAGE 2	OF 🚛	PAGES		
529	643	631	S\$5	5%	475	56z	\$22	3 1(
25									1 1 1 1 1 1 1
431	318	<i>5</i> 86	547	300	a fo	430	424	199	٥
20				y					
452	470	442	S24	<i>Q</i> ₀ 3	4716	201	471	144	2
5		YG 5	11	<i>∞6</i> 3	Tio				
	71		1106	*					
0 542	SYL	392	489	437	538	ISb	Bi	191	9
5	571	Er	S60	<i>3</i> 71	536	412	116	119	2
		1 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0			0		90 10 10 10 10		6 6 7 7 8
50	653	60	65	63 €	643 75	62 80	5 <i>A</i>	90	ح 95
50	55	00	00	70	75	80	00	90	95

V.

	T	1		-	,	PAGE 3	OF 🚍 8	PAGES		-
	128	To	349	طما	269	Zþ	550	243	2678	464
25						1				
			4	, , , , ,	4 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8	#	1 1 1 1 1 1			
20	346	251	391	263	ZN	34	471 -	198	26	136
		4 (m)		4	9 # # # # # # # # # # # # # # # # # # #	0 4 4 5 7 7 8 8	9 Y A A A A A A A A A A A A A A A A A A		10	
	162	98	98	(DZ	<u>3</u> ∤3	258	448	la-l	149	NS
15		4		10 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0				8 8 9 9 9 9 9 9		
				8 6 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8		4 4 5 6	6 6 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7	1 1 1 1 1 2 3 3 1 1 1 1		
10	130	328	324	120	136	201	201	265	429	M
10				1 4 7 7 8			i i i i i	100 100 100 100 100 100 100 100 100 100		
		-		8 1 8 1 8 1 8 1 8 1 8 1 8 1 8 1 8 1 8 1						
5	24(SN.	121	120	92	768	103	152	<u> তি</u>	271
					2 2 3 1 1	, , , , , ,	0 0 0 0 0 0 0			
	152.	102	ωχ.	yen .	ШО	277	42/	₹.0	3/0	426
100	152 105	110	ዛ ን 115	120	125	<i>3</i> 77 130	135	140	145	146
										1

DATE:	2/	13/	a	6	
BRIDGE	NO:	l	-		
TESTED	BY:				

5	

				· ·	PAGE 4	of 🚛	PAGES		
363 25	3 92	248	265	367	3 E0	213	Y69	329	⁴ વ્હ
		1 1 1 1 1 1 1 1 1 1 1 1 1							
20	353	169	461	119	258	428	205	240	375
15	136	143	259	167	145	334	192	164	495
					#				1 1 1 0 0 0 0
35 /	ß	98	287)	74	115	304	3 47	ħ9	464
					0 0 0 0 0 0 0 0 0 0		-		, , , , , , , , , , , , , , , , , , ,
<u>43</u>	153	124	716	222	348	46	328	209	405
		1 1 1 1 1 1					0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0		
440	57 <i>6</i> 155	333	310	214	<i>S</i> \$7	353	407	430	Jey.
150	155	160	165	170	1/5	180	165	190	195

DATE:	2/	18	104	6	
BRIDGE	NO:	-			
TESTED	BY:				

						PAGE 5	OF 278	PAGES		-
	tn	57A	519	220	344	2 33	467	486	K ₁ 0	43
25								# # # # # # # # # # # # # # # # # # #	77 14 14 15 16 16 16 16 16 16 16 16 16 16 16 16 16	0 0 0 0 0
	1 52	1100	262	40	m)	2.7		ch >	134	
20	36	4.7.7	297	7 0 ι	612	201	137	763)SC	
								* * * * * * * * * * * * * * * * * * *	1 1 1 1 5 6 6	
15	40	474	48	448	3 53	<i>4</i> 63	219	316	96	50/
					• • • • • • • • • • • • • • • • • • •				2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2	
)6 4	453	396	526	40z	442	557	3 57	296	27
10									1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	
2	ક રા	M6	3/3	//ac	468	177	343	2 -	1(9	7/0
5			<u> </u>	425			343	302	<i>X.1</i>	
									1.00 0.00 0.00 0.00 0.00 0.00 0.00 0.00	
200	529	584 210	866	484	53.	32	477	50/	04	50

BRIDGE NO:		
TESTED BY:		

						PAGE 6	OF 178	PAGES		
25	148	438	Do	291	262	576	535	51.8	478	5:
20	34	440	27.6	335	3 04	264	42	241	3 .1	38
	hi	166	160	205	3°A	<i>2</i> 63	Z4r	/BZ	2 28	
15										
10	34	200	189	395	855	/13 3	354	135	28 7	3
5	218	36	2 5	313	241	218	249	.3g	33 2	47
250	167	4n 260	In	310	378	241	445	11%	335	37

DATE:	2	18	/af	
BRIDGE	NO:	1		
TESTED	BY:			

						PAGE 7	OF 578	PAGES		_
25	298	264	420	550	555	<i>3</i> 88	281	406	201	19
25					**************************************	6 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4		6 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0		
	93	SIZ	584	619	617	SH	438	463	201	24
20				7 4 5 5 7 6 7 7 8 8 8 8 8 8 8 8 8 8 8 8 9 9	# d d d d d d d d d d d d d d d d d d d	# # # # # # # # # # # # # # # # # # #	100 mm	# 1		***********
15	236	Lby	232	30	43	54	20 8	403	44	23
			* * * * * * * * * * * * * * * * * * *	1 1 1 1 1 1 1 2 2 3 3 3 3 3 3 3 3 3 3 3					1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	
10	87	30	84	FL	253	170	215	232	315	17
	19	8 (1) 8 (1) 8 (1) 8 (1) 8 (1) 8 (1) 8 (1)								
5	139	34	2n	237	210	31	щ	Мо	124	411
				1						
	240	293	160	214	206	148	257	135	136	401
00	240 305	310	<i>Ц∕</i> 0 315	320	325	330	335	340	345	0

						PAGE 8	OF #8	PAGES	
	224	306	389	3(2	193	268			
25				1 1 1 1 1 1 1 1 1					
	420	æ3	356	32	208	255			
20									
	115	303	47	3 1 -	162	2144		2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2	
15									
10	197	253	316	465	ЛЗ	143			11
						i i			
5	3 %	85	109	435	92	147			
	193	250	206	460	167	14 380	4		
350	355	360	365	370	375	380	385`	390	395