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THE STUDIES OF THE MICROWAVE SPECTRAL LINEWIDTHS OF MOLECULES
AND THE ELECTRONIC ENERGY BAND STRUCTURES OF SOLIDS

A DISSERTATION

SUBMITTED TO THE GRADUATE FACULTY

in partial fulfillment of the requirements for the

degree of

DOCTOR OF PHILOSOPHY

BY

TEH-KAN TUNG

Norman, Oklahoma

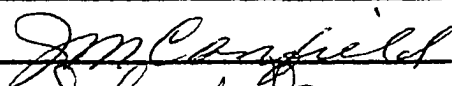
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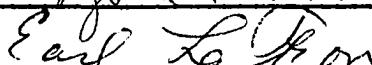
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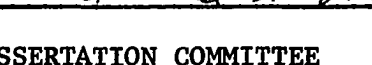
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TABLE OF CONTENTS

| | Page |
|--|------|
| LIST OF TABLES | vi |
| LIST OF ILLUSTRATIONS | vii |
| PART I. LINEWIDTH OF THE ROTATIONAL SPECTRA OF SYMMETRIC-TOP MOLECULES | |
| Chapter | |
| I. INTRODUCTION | 1 |
| II. ANDERSON-TSAO-CURNUTTE THEORY | 3 |
| The Formulation | 3 |
| Tsao-Curnutte Modification | 7 |
| Dipole-dipole Interaction | 9 |
| Computable Form | 12 |
| III. COMPARISON BETWEEN THEORY AND EXPERIMENT | 16 |
| Numerical Calculation | 16 |
| Ammonia Self-broadening | 23 |
| PART II. APPLICATION OF THE GAUSSIAN-TYPE ORBITALS FOR CALCULATING ENERGY BAND STRUCTURES OF SOLIDS BY THE METHOD OF TIGHT BINDING | |
| IV. INTRODUCTION | 26 |

TABLE OF CONTENTS - Continued

| Chapter | Page |
|--|------|
| V. CRYSTAL LATTICE AND CRYSTAL POTENTIAL | 31 |
| Crystal Lattice | 31 |
| Crystal Potential | 33 |
| VI. CRYSTAL SYMMETRY AND GROUP THEORY | 38 |
| VII. BASIS FUNCTIONS | 43 |
| VIII. MULTICENTER INTEGRALS | 46 |
| IX. RESULT AND DISCUSSION | 52 |
| X. THE ENERGY BAND GAP OF MAGNESIUM OXIDE | 58 |
| Crystal Structure | 58 |
| Crystal Potential | 58 |
| Wave Functions | 60 |
| Result and Discussion | 61 |
| APPENDICES | |
| A. DERIVATION OF $S_2(b)$ FOR DIPOLE-DIPOLE | |
| INTERACTION | 68 |
| B. THE FORMULATION OF AHFS CRYSTAL POTENTIAL | 76 |
| C. THE EWALD-TYPE POTENTIAL EXPANSION | 79 |
| D. KINETIC AND POTENTIAL INTEGRALS | 86 |
| REFERENCES | 88 |

LIST OF TABLES

| Table | Page |
|--|------|
| I. The Constants for the Gases | 18 |
| II. Linewidth Parameters of the Symmetric-top Molecules | 19 |
| III. Linewidth Parameters of Self-Broadening of Ammonia | 25 |
| IV. Fourier Coefficients for $V(\vec{r})$ | 37 |
| V. C_i and α_i of Carbon | 44 |
| VI. Comparison of GTO and STO at High Symmetry Points . . . | 53 |
| VII. Comparison of Optical Transitions of Diamond | 54 |
| VIII. Comparison of the Band Structures of STO and GTO | 55 |
| IX. Fourier Coefficients for $V^{\text{Mg}}(\vec{r})$ and $V^0(\vec{r})$ | 62 |
| X. C_i and α_i of Oxygen | 63 |
| XI. β_i and α_i of Oxygen | 64 |
| XII. C_i and α_i of Mg | 65 |
| XIII. β_i and α_i of Mg | 66 |
| XIV. Γ -point Energy Levels of MgO | 67 |
| XV. Comparison of V_{k_y} and V_{2k_y} for Diamond | 84 |
| XVI. Comparison of V_{k_y} and V_{2k_y} for Mg and Oxygen | 85 |

LIST OF ILLUSTRATIONS

| Figure | Page |
|---|------|
| I. Relation Between b and $S(b)$ | 4 |
| II. Diagram of the Coordinates | 10 |
| III. Crystal Structure of Diamond | 32 |
| IV. The Reciprocal Lattice of the fcc Lattice | 32 |

PART I

LINE WIDTH OF THE ROTATIONAL SPECTRA OF
SYMMETRIC-TOP MOLECULES

CHAPTER I

INTRODUCTION

The pressure broadening of spectral lines has been of interest for a long time. Michelson's study¹ of line broadening is perhaps the first one in the field. In that paper, he introduced for the first time the concept of interruption broadening due to collisions. The argument is that a hard-sphere atom (or molecule) will undergo the process of emitting electromagnetic wave of its natural frequency in between collisions. The collision has the effect of terminating the radiations and thus a finite wave train is formed which is our interruption broadened spectral line after a Fourier transform is performed on it. This theory was modified by Lorentz² classically. With the advent of quantum mechanics, new justifications were added by Weisskopf³ and Jablonski.⁴ It was subsequently improved by Foley⁵ and Lindholm.⁶

The Fourier transform treatment gives good results in optical region.⁷ However, it fails in the microwave and infra-red regions. The major difficulty arises from the breakdown of the adiabatic hypothesis. In the optical region, the energy difference is much larger than the average kinetic energy of the colliding molecules. Therefore, the

assumption that the probability of the collision induced transition is small is valid. This assumption is no longer valid in the microwave region since the energy difference is smaller than the average kinetic energy. Precisely for the same reason, the pressure-broadened linewidth of the rotational spectra of molecules may provide a powerful tool to study the general problem of interactions between molecules⁸⁻¹⁰ since the linewidth will yield information about the cross sections for transfer of rotational energy and consequently the intermolecular forces.

The most successful theory for studying microwave linewidth to date was due to Anderson¹¹ and improved by Tsao and Curnutte.¹² This theory which is the central theme of the next chapter takes care of the diabatic effects. A number of other theoretical works on linewidth has appeared in recent years.¹³⁻¹⁶ They vary from using different methods to arrive at the same result (Cooper, Trindle and Illinger) to slightly modified version (Murphy and Boggs). However, Anderson-Tsao-Curnutte theory is still the single most powerful tool in the study of microwave linewidth. In the subsequent chapters, we will use this theory to perform a first principle calculation of the linewidth of the rotational spectra of a number of symmetric-top molecules and compare the results with the experimental values obtained at this laboratory.

CHAPTER II
ANDERSON-TSAO-CURNUTTE THEORY

THE FORMULATION

In 1949, P. W. Anderson¹¹ proposed a generalized theory of collision broadening. This theory has the following basic assumptions:

1. Assumption of a classical path. It is assumed that for all collisions involved, the colliding molecules can be viewed as point dipoles traveling along a classical straight line path. For very close collision, this assumption is not valid. However, we can consider the molecules as wave packets. These wave packets are very small compared to the effective collision radius. This makes the assumption a very good one in almost all cases considered.

2. Zero collision time. The duration of the collision is very small as compared to the interval between collisions. This assumption has the effect that either the lines under consideration are degenerate or they are well-separated.

3. Binary collision. Only the collision involving a pair of molecules -- the emitter and the perturber is considered. This is seen to be valid when the pressure is sufficiently low and the temperature high.

Following Foley's approach, Anderson first wrote down the quantum mechanical expression for intensity of the spectral line due to

dipole radiation as

$$I(\omega) = \text{const} \times \omega^4 \text{Trace} \left[\rho_0 \int_{-\infty}^{+\infty} dt \exp(i\omega t) \mu_z(t) \right. \\ \left. \times \int_{-\infty}^{+\infty} dt' \exp(-i\omega t') \mu_z(t') \right]. \quad (2.1)$$

This equation has since been proved by Margenau and Bloom,¹⁷ where ρ_0 is the density matrix and $\mu(t)$ is the dipole matrix. Using the time-development operator technique

$$\mu(t) = U^{-1} \mu_0 U; \quad T = U_0^{-1} U; \quad U_0 = \exp(-\frac{i}{\hbar} H_0 t) \quad (2.2)$$

where H_0 is the unperturbed Hamiltonian and

$$H = H_0 + H_1(t) \quad (2.3)$$

with $H_1(t)$, the time-dependent interaction Hamiltonian. Then through Correlation Function scheme and a rather formidable mathematical manipulation, this theory gives the cross-section as

$$\sigma = \int_0^{\infty} 2\pi b S(b) db, \quad (2.4)$$

where the function $S(b)$ is a weight factor governing the effectiveness of a particular collision to interrupt a particular radiation and impact parameter or the distance of closest approach of the molecules. The function $S(b)$ increases as b decreases until $b = b_0$ and $S(b_0) = 1$ which means the radiation is completely interrupted.

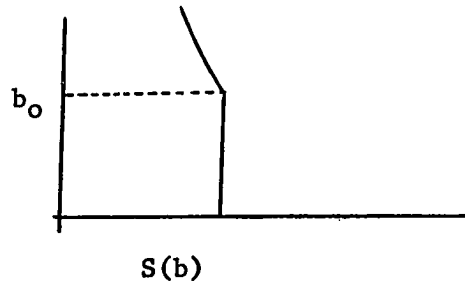


Figure 1. Relation between b and $S(b)$.

$S(b)$ can be expanded as

$$S(b) = S_0(b) + S_1(b) + S_2(b) + \dots \quad (2.5)$$

where

$$S_0(b) = 0$$

$$S_1(b) = i \left[\sum_{m_i, m_2} \frac{(j_i m_i j_2 m_2 | P | j_i m_i j_2 m_2)}{(2j_i + 1)(2j_2 + 1)} - \sum_{m_f, m_2} \frac{(j_f m_f j_2 m_2 | P | j_f m_f j_2 m_2)}{(2j_f + 1)(2j_2 + 1)} \right] \quad (2.6)$$

It is imaginary and contributes only to the line-shift.

$$S_2(b) = S_2(b)_{\text{outer}, i} + S_2(b)_{\text{outer}, f} + S_2(b)_{\text{middle}}$$

with

$$S_2(b)_{\text{outer}, i} = \frac{1}{2} \sum_{m_i, m_2} \frac{(j_i m_i j_2 m_2 | P^2 | j_i m_i j_2 m_2)}{(2j_i + 1)(2j_2 + 1)} \quad (2.7)$$

$$S_2(b)_{\text{outer}, f} = \frac{1}{2} \sum_{m_f, m_2} \frac{(j_f m_f j_2 m_2 | P^2 | j_f m_f j_2 m_2)}{(2j_f + 1)(2j_2 + 1)} \quad (2.8)$$

and

$$S_2(b)_{\text{middle}} = - \sum_{\substack{m_i, m_i' \\ m_f, m_f' \\ m_2, m_2' \\ M}} \sum_{j_2'} \frac{(j_f 1 m_f M | j_i m_i)(j_f 1 m_f' M | j_i m_i')}{(2j_1 + 1)(2j_2 + 1)} \\ \times (j_f m_f j_2 m_2 | P | j_f m_f' j_2' m_2') \\ \times (j_i m_i' j_2' m_2' | P | j_i m_i j_2 m_2). \quad (2.9)$$

Where $j_i m_i$, $j_f m_f$, $j_i' m_i'$, $j_f' m_f'$ are the quantum numbers of the initial and final states of the emitter, $j_2 m_2$, $j_2' m_2'$ that of the perturber and

$$P = \frac{1}{\hbar} \int_{-\infty}^{+\infty} H_1'(t) dt, \quad (2.10)$$

where H_1' is the interaction Hamiltonian with the time-dependence due to

H_1 inserted:

$$(m|H_1|n) = (m|H_1(t)|n) \exp \left[(E_m - E_n)t/i\hbar \right]. \quad (2.11)$$

$(j_f l m_f M | j_i l m_i)$ and $(j_f l m_f' M | j_2 l m_2')$ are the Clebsch-Gordan coefficients.

$S_2(b)$ is real and contributes to the linewidth. The halfwidth $\Delta\nu$ is related to the cross-section by

$$\Delta\nu = \frac{Nv\sigma}{2\pi} \quad (2.12)$$

$Nv\sigma$ gives the number of collisions per unit time. N is the number of molecules per unit volume and v the relative molecular thermal velocity.

Equation (2.4) gives only the collision cross section of a emitter at a particular molecular quantum state J_1, K_1 and a perturber at J_2, K_2 . We may call it the partial cross section and denote it by $\sigma_{J_2 K_2}$. The number of molecules in this particular state is proportional to the statistical weight f_{JK}^{18} ,

$$f_{JK} = \frac{S(I, K) (2J + 1)}{4I^2 + 4I + 1} \sqrt{\frac{B^2 C h^3}{(kT)^3}} \exp \left[BJ(J + 1) + (C - B)K^2 \right] h/kT \quad (2.13)$$

where I is the spin quantum number. B and C are rotational constants; k , the Boltzmann constant; h , Planck constant. T is the temperature. The factor $S(I, K)$ is such that:

for K a multiple of 3, but not 0,

$$S(I, K) = 2(4I^2 + 4I + 3)f(I); \quad (2.14a)$$

for $K = 0$,

$$S(I, K) = (4I^2 + 4I + 3)f(I); \quad (2.14b)$$

for K not a multiple of 3,

$$S(I, K) = 2(4I^2 + 4I)f(I);$$

with

$$f(I) = (2I + 1)/3.$$

The cross section of a given molecule at a state J_1 and K_1 then is given by

$$\sigma_{J_1 K_1} = \sum_{J_2 K_2} f_{J_2 K_2} \sigma_{J_2 K_2}.$$

Where theoretically the summation is taken from $J_2 = 0$ to infinity with K_2 - degeneracy. However, the upper limit in practice is dictated by the statistical weight such that we cut off the series when the upper levels are less populated and give insignificant contributions to the total cross section $\sigma_{J_1 K_1}$.

TSAO-CURNUTTE MODIFICATION¹²

Anderson applied his theory originally to the ammonia inversion spectral line broadening with success. It was Tsao-Curnutte modification which completes the theoretical frame work and takes into consideration the different interactions.

They expand the angular part of the interaction Hamiltonian in spherical harmonics. The general form of the matrix element P is then of the form

$$\begin{aligned} & (j_1 m_1 j_2 m_2 | P | j_1' m_1' j_2' m_2') \\ &= \sum_{\substack{k_1 k_2 \\ \lambda_1 \lambda_2}} a(\lambda k j) (j_1 m_1 j_2 m_2 | Y_{\lambda_1}^{k_1}(1) Y_{\lambda_2}^{k_2}(2) | j_1' m_1' j_2' m_2'). \end{aligned} \quad (2.14)$$

Where $Y_{\lambda}^k(\phi \theta)$ are the apherical harmonics. For symmetric-top molecules the matrix elements are

$$(j'K'm' | Y_{\lambda}^k(\phi \theta) | jKm) = \left[\frac{(2j+1)(2K+1)}{4\pi(2j'+1)} \right]^{\frac{1}{2}} \times (jkKO | j'K') (jkm\lambda | j'm'). \quad (2.15)$$

The quantum numbers are related as

$$j + k = j', K + 0 = K', m + \lambda = m'.$$

The modified expressions of $S_2(b)$ are:

$$S_2(b)_{\text{outer},i} = \frac{1}{32\pi^2} \sum_{\substack{j_1' j_2' \\ k_1 k_2 \\ \lambda_1 \lambda_2}} (j_1 k_1 K_1 0 | j_1' K_1')^2 \times (j_2 k_2 K_2 0 | j_2' K_2')^2 |a(\lambda k j)|^2 \quad (2.16a)$$

$$S_2(b)_{\text{outer},f} = \frac{1}{32\pi^2} \sum_{\substack{j_f' j_2' \\ k_1 k_2 \\ \lambda_1 \lambda_2}} (j_f k_1 K_f 0 | j_f' K_f')^2 \times (j_2 k_2 K_2 0 | j_2' K_2')^2 |a(\lambda k j)|^2 \quad (2.16b)$$

and

$$S_2(b)_{\text{middle}} = (-1)^{j_i + j_f} \frac{\sqrt{(2j_i+1)(2j_f+1)}}{16\pi^2} \times \sum_{k_1 k_2} \sum_{j_2'} (-1)^{k_1 + \lambda_1 + \lambda_2} (j_1 k_1 K_1 0 | j_1' K_1') \times (j_f k_f K_f 0 | j_f' K_f') (j_2 k_2 K_2 0 | j_2' K_2')^2 \times W(j_i j_f j_i j_f, 1 k_1) a(k_1 k_2 \lambda_1 \lambda_2 j) \times a'(k_1 k_2 -\lambda_1 -\lambda_2 j') \quad (2.17)$$

where W is the Racah's coefficient.

DIPOLE-DIPOLE INTERACTION

The classical dipole-dipole interaction between two molecules is given by the expression

$$\vec{P}_1 \cdot \vec{P}_2 - 3(\vec{P}_1 \cdot \vec{R}_0)(\vec{P}_2 \cdot \vec{R}_0)/R^3 ,$$

where \vec{P}_1, \vec{P}_2 are the dipole moments of the molecules, R is the distance between the molecules and \vec{R}_0 is the unit vector along R . For linear or symmetric-top molecules, the dipole moment can be specified by its magnitude and the direction of the symmetry axis.

The interaction Hamiltonian is then

$$(m|p|n) = \frac{P_1 P_2}{\hbar} (G_1 F_1 + G_2 F_2 + G_3 F_3)_{mn}, \quad (2.18)$$

where

$$\begin{aligned} F_1 &= \sin\theta_1 \cos\phi_1 \sin\theta_2 \cos\phi_2 - 2\sin\theta_1 \sin\phi_1 \sin\theta_2 \sin\phi_2 + \cos\theta_1 \cos\theta_2 \\ F_2 &= 3\sin\theta_1 \sin\phi_1 \cos\theta_2 + 3\cos\theta_1 \sin\theta_2 \sin\phi_2 \\ F_3 &= 3\sin\theta_1 \sin\phi_1 \sin\theta_2 \sin\phi_2 - 3\cos\theta_1 \cos\theta_2 \end{aligned} \quad (2.19)$$

$$\begin{aligned} G_1 &= \int_{-\infty}^{+\infty} \frac{\exp(i\omega_m n t)}{R^3} dt \\ G_2 &= \int_{-\infty}^{+\infty} \frac{\exp(i\omega_m n t) \sin\psi \cos\psi}{R^3} dt \\ G_3 &= \int_{-\infty}^{+\infty} \frac{\exp(i\omega_m n t) \cos^2\psi}{R^3} dt , \end{aligned} \quad (2.20)$$

where (ϕ_1, θ_1) and (ϕ_2, θ_2) are the direction of the dipole P_1 and P_2 respectively. The quantities R and ψ depend on time.

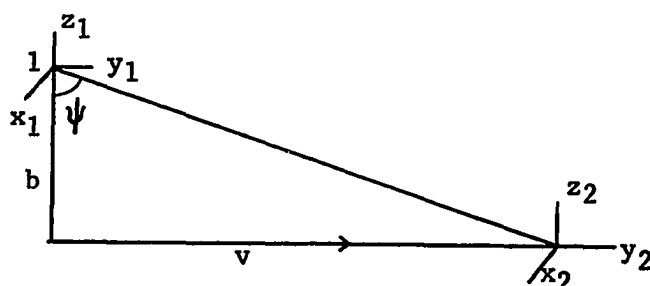


Figure 2. Diagram of the coordinates

If v is the velocity of the collision, then

$$R = \sqrt{(b^2 + v^2 t^2)} \quad (2.21)$$

$$\sin \psi = \frac{vt}{R} \quad \cos \psi = \frac{b}{R} .$$

In Anderson's original application of the theory to ammonia inversion spectral lines, he ignored the G's entirely as he took into consideration only the first order Stark effect and rotational resonance. These approximations are valid since the ammonia molecule is a rather unique case in that the rotational energy levels occur in closely spaced pairs of about 1 cm^{-1} due to inversion and that levels of different rotational quantum numbers J are generally separated by over 20 cm^{-1} . Because of this pattern of energy levels, one needs to consider energy transfer only between the two inversion levels of a given rotational state during the process of collision interruption of radiation. The collision induced transitions of type $J \rightarrow J'$ ($J' \neq J$) are expected to have much smaller cross-sections on account of the large energy differences and are neglected aside from the case of rotational resonance. Furthermore, for the purpose of computing collision induced transition probabilities between the two inversion doublets, the energy spacing between these pairs are set to zero.

For the symmetric-top molecules, the rotational constants

which govern the frequencies of the J transition are less than 14 kMc/sec, hence the approximations are no longer valid. The integrals of G's must be evaluated. It was shown¹² that:

$$\begin{aligned} G_1 &= \frac{1}{vb^2} 2 |k| K_1(|k|) \\ G_2 &= \frac{1}{vb^2} i \frac{2}{3} \frac{k}{|k|} k^2 K_1(|k|) \\ G_3 &= \frac{1}{vb^2} \frac{2}{3} k^2 K_2(|k|) \end{aligned} \quad (2.22)$$

where

$$k = \frac{b}{v} \omega_{mn}, \quad (2.23)$$

and K, the Bessel function of the second kind.

We can also write F's in terms of the spherical harmonics:

$$\begin{aligned} F_1 &= 2\pi[Y_1^1(1)Y_1^1(2) + Y_{-1}^1(1)Y_{-1}^1(2)] \\ &\quad + \frac{2\pi}{3}[Y_1^1(1)Y_{-1}^1(2) + Y_{-1}^1(1)Y_1^1(2)] \\ &\quad + \frac{4\pi}{3}Y_0^1(1)Y_1^1(2), \\ F_2 &= i \cdot 2/\sqrt{2}\pi[Y_1^1(1)Y_0^1(2) + Y_{-1}^1(1)Y_0^1(2) + Y_0^1(1)Y_1^1(2) \\ &\quad + Y_0^1(1)Y_{-1}^1(2)], \\ F &= -2\pi[Y_1^1(1)Y_1^1(2) + Y_{-1}^1(1)Y_{-1}^1(2) + Y_{-1}^1(1)Y_1^1(2) \\ &\quad + Y_1^1(1)Y_{-1}^1(2)] - 4\pi Y_0^1(1)Y_0^1(2). \end{aligned} \quad (2.24)$$

Substituting the expressions of F's and G's into (2.16a),

(2.16b), and (2.17), we obtain

$$\begin{aligned} S_2(b)_{o,i} &= \frac{4}{9} \left(\frac{p_1 p_2}{4\pi v} \right)^2 \frac{1}{b^4} \sum_{j_1 j_2} (j_1 1 K_1 0 | j_1' K_1)^2 \\ &\quad \times (j_2 1 K_2 0 | j_2' K_2)^2 f_1(k), \end{aligned} \quad (2.25)$$

where

$$f_1(k) = \frac{1}{4} k^2 [K_2^2(k) + 4K_1^2(k) + 3K_0^2(k)] \quad (2.26)$$

with similar expression for $S_2(b)_{o,f}$. The function $S_2(b)_m$ is expressed as

$$\begin{aligned} S_2(b)_m = & (-1)^{j_i+j_f+1} \cdot \frac{8}{9} \left(\frac{p_1 p_2}{\hbar v} \right)^2 \frac{1}{b^4} / (2j_i + 1)(2j_f + 1) \\ & \times (j_i 1 K_i 0 | j_i k_i) \times (j_f 1 K_f 0 | j_f k_f) \\ & \times W(j_i j_f j_i j_f, 11) \sum_{j_2'} (j_2 1 K_2 0 | j_2' K_2)^2 \cdot f_1(k) \quad (2.27) \end{aligned}$$

Once the $S_2(b)$ is known, we adopt the "approximation #2" of the interpolation process of Anderson. Let b_0 be the value of b at which

$$S_2(b_0) = 1 \quad (2.28)$$

then

$$\begin{aligned} \sigma_{j_2 k_2} = & \pi b_0^2 + \int_{b_0}^{\infty} (2\pi b) db [S_2(b)_{o,i} + S_2(b)_{o,f} \\ & + S_2(b)_m] \quad (2.29) \end{aligned}$$

The integrand is evaluated to yield

$$\frac{1}{2} b_0^2 [K_3 K_1 + 4K_0 K_2 - K_2^2 - K_1^2 - 3K_0^2]_{k=k_0} \quad (2.30)$$

COMPUTABLE FORM

As previously pointed out, in evaluating the line width of the symmetric-top molecules, the Anderson's approximations applied to ammonia inversion are no longer valid. We must now include all the interaction between the emitter and the perturber during the collision. There are the following transitions which occur in the collision process:

$$\Delta J_1 = \Delta J_2 = 0 \quad (2.31)$$

$$\Delta J_1 = \pm 1, \Delta J_2 = 0 \quad (2.32)$$

$$\Delta J_1 = \pm 1, \Delta J_2 = \mp 1 \quad (2.33)$$

$$\Delta J_1 = 0, \pm 1, \Delta J_2 = \pm 1. \quad (2.34)$$

For the $J_1 = 0 \rightarrow 1$ transition, Birnbaum found that processes (2.31) and (2.32) occur with much higher probability than the others.¹⁹

By neglecting the transitions (2.33) and (2.34) and setting $E(J_1 = 1) - E(J_1 = 0) \simeq 0$, Birnbaum obtained a rather simple expression for the linewidth:

$$\Delta\nu_{J_1 K_1} = \left(\frac{8}{9}\right) \frac{\frac{1}{2} N P_1 P_1}{\pi} \left[1 - \frac{K_1^2}{(J_1+1)^2} \right]^{\frac{1}{2}} \times \left\langle \frac{K_2}{[J_2(J_2+1)]^{\frac{1}{2}}} \right\rangle. \quad (2.35)$$

Where the expression $\langle \rangle$ is the Boltzmann average of the quantity.

However, our calculations in the next chapter will show that for the higher transitions ($J_1 = 1, 2$ or above) this approximation is invalid.

Including all the processes cited above, and the proper values for both Clebsch-Gordan coefficients²⁰ and Racah's coefficients,²¹ we obtain the following equation, the detailed evaluation will be given in Appendix A.

$$\begin{aligned} S_2(b) = & \frac{4}{9} \left(\frac{P_1 P_2}{\pi \nu} \right)^2 \frac{1}{b^4} \left\{ \frac{K_2^2}{J_2(J_2+1)} \left(\frac{K_1^2}{J(J_1+1)} + \frac{K_1^2}{(J_1+1)(J_1+2)} \right. \right. \\ & \left. \left. - \frac{2K_1^2}{(J_1+1)^2} \right) + \frac{J_2^2 - K_2^2}{J_2(2J_1+1)} \left(\frac{(J_1+1)^2 - K_1^2}{(2J_1+1)(J_1+1)} \right. \right. \\ & \left. \left. + \frac{(J_1+2)^2 - K_1^2}{(2J_1+3)(J_1+2)} f_1(x(J_1-J_2+1)) + \frac{(J_1+2)^2 - K_1^2}{(2J_1+3)(J_1+2)} f_1(x(J_1-J_2+2)) \right) \right\} \end{aligned}$$

$$\begin{aligned}
& + \frac{(J_2-1)^2 - K_2^2}{(2J_2+1)(J_2+1)} \left(\frac{J_1^2 - K_1^2}{J_1(2J_1+1)} f_1(x(J_2-J_1+1)) \right. \\
& \quad \left. + \frac{(J_1+1)^2 - K_1^2}{(J_1+1)(2J_1+3)} f_1(x(J_2-J_1)) \right) \\
& + \frac{K_2^2}{J_2(J_2+1)} \left(\frac{(J_1+1)^2 - K_1^2}{(2J_1+1)(J_1+1)} f_1(x(J_1+1)) \right. \\
& \quad + \frac{(J_1+2)^2 - K_1^2}{(2J_1+3)(J_1+2)} f_1(x(J_1+2)) \\
& \quad + \frac{J_1^2 - K_1^2}{J_1(2J_1+1)} f_1(x(J_1)) \\
& \quad \left. + \frac{(J_1+1)^2 - K_1^2}{(J_1+1)(2J_1+3)} f_1(x(J_1+1)) \right) \\
& + \frac{(J_2+1)^2 - K_2^2}{(2J_2+1)(J_2+1)} \left(\frac{K_1^2}{J_1(J_1+1)} + \frac{K_1^2}{(J_1+1)(J_1+2)} \right. \\
& \quad \left. - \frac{2K_1^2}{(J_1+1)^2} f_1(x(J_2+1)) \right) \\
& + \frac{J_2^2 - K_2^2}{J_2(2J_2+1)} \left(\frac{K_1^2}{J_1(J_1+1)} + \frac{K_1^2}{(J_1+1)(J_1+2)} - \frac{2K_1^2}{(J_1+1)^2} f_1(xJ_2) \right) \\
& + \frac{(J_2+1)^2 - K_2^2}{(2J_2+1)(J_2+1)} \left(\frac{(J_1+1)^2 - K_1^2}{(2J_1+1)(J_1+1)} f_1(x(J_1+J_2+2)) \right. \\
& \quad \left. + \frac{(J_1+2)^2 - K_1^2}{(2J_1+3)(J_1+2)} f_1(x(J_1+J_2+3)) \right)
\end{aligned}$$

$$\begin{aligned}
& + \frac{(J_2 - K_2)(J_2 + K_2)}{J_2(2J_2 + 1)} \left(\frac{J_1^2 - K_1^2}{J_1(2J_1 + 1)} f_1(x(J_1 + J_2)) \right. \\
& \quad \left. + \frac{(J_1 + 1)^2 - K_1^2}{(J_1 + 1)(2J_1 + 3)} f_1(x(J_1 + J_2 + 1)) \right) \Big\}
\end{aligned}
\tag{2.36}$$

where $x = 4\pi bB/v$.

CHAPTER III
COMPARISON BETWEEN THEORY AND EXPERIMENT

NUMERICAL CALCULATION

Once the analytical form of $S_2(b)$ is known, we proceed to find b_0 such that $S_2(b_0) = 1$. The method used in this study is the standard Newton's method.²² In the present work, self-broadening of seven different symmetric-top molecules are studied. They are: $\text{CH}_3\text{Br}^{79}$, $\text{CH}_3\text{Br}^{81}$, CH_3I , $\text{CH}_3\text{Cl}^{35}$, $\text{CH}_3\text{Cl}^{37}$, CH_3CN and CH_3CH . Only dipole-dipole interaction is considered since this is the dominating term for collision.

The calculation was carried out on a CDC3400 computer. Except some physical constants such as mass number, Planck's constant, etc., the only input are two rotational constants B and C and the dipole moment. A list of these values and their sources are given in Table I.

The choice of the velocity, which is very crucial to the linewidth, is that of the relative velocity average over the Boltzmann distribution. This choice is justified since at sufficient low pressure and room temperature the gases under consideration do obey the Maxwell-Boltzmann distribution. Thus the velocity of the gas is:

$$\bar{v} = \sqrt{\frac{8KT}{\pi m}} \quad (3.1)$$

For the relative velocity of two colliding molecules, the m is replaced by the reduced mass μ ²³ and

$$\bar{v}_{re} = \sqrt{\frac{8KT}{\pi\mu}} \quad (3.2)$$

The calculated values are listed in Table II. They are compared with the experimental values obtained at this laboratory.²⁵ In the calculation of the linewidths, we have neglected the hyperfine splitting, namely, the width of the $J \rightarrow J+1$ rather than $J, F \rightarrow J+1, F'$ transitions were calculated.

For CH_3Cl , and CH_3Br the broadening effect of different isotopic species, e.g., $\text{CH}_3\text{Cl}^{35} - \text{CH}_3\text{Cl}^{35}$ and $\text{CH}_3\text{Cl}^{35} - \text{CH}_3\text{Cl}^{37}$, etc., must be computed separately. Since the energy differences in (2.31) - (2.34) cannot be neglected in computing $S_2(b)$, one can no longer expect the relative linewidths of a given molecule to be independent of the method of interpolation. For this reason the experimental data will be compared with the theoretical values of the absolute linewidths rather than the relative widths.

Except for the case of the CH_3CCH molecule, the theoretical widths are generally larger than the experimental ones. This is most noticeable in both isotopes of CH_3Cl where one observes a difference of about 30%. Equally larger percentage differences are found for the low- J lines of $\text{CH}_3\text{Br}^{79}$ and $\text{CH}_3\text{Br}^{81}$, but the agreement becomes better at higher J . The CH_3I lines show variations from 2% to 30% between theory and experiment. The same is true also for CH_3CCH except that the measured widths are consistently larger than the theoretical values in this case, because this molecule has a

TABLE I
THE CONSTANTS FOR THE GASES

| Gas | B in Mc/s | C in Mc/s | Dipole-moment in Debye |
|-----------------------------|------------|----------------------|---------------------------|
| $\text{CH}_3\text{Br}^{79}$ | 9568.19* | $148 \times 10^3^\#$ | 1.797* |
| $\text{CH}_3\text{Br}^{81}$ | 9531.84* | $148 \times 10^3^\#$ | 1.797* |
| $\text{CH}_3\text{Cl}^{35}$ | 13292.84* | $150 \times 10^3^+$ | 1.869* |
| $\text{CH}_3\text{Cl}^{37}$ | 13088.137* | $150 \times 10^3^+$ | 1.869* |
| CH_3CH | 8545.85* | $150 \times 10^3^+$ | 0.75* |
| $\text{CH}_3\text{I}^{127}$ | 7501.31* | $150 \times 10^3^+$ | 1.65* |
| CH_3CN | 9198.83* | $150 \times 10^3^+$ | 3.92* |

* reference 14.

+ reference 24.

calculated by the author by standard method.

TABLE II
 LINEWIDTH PARAMETERS (Mc/s - torr)
 OF SYMMETRIC-TOP MOLECULES.

| Gas | ΔJ | K | ΔF | $(\Delta\nu_p)_{\text{Obs}}$ | $(\Delta\nu_p)_{\text{Calc}}$ |
|-----------------------------|------------|---|-----------------------------|------------------------------|-------------------------------|
| $\text{CH}_3\text{Cl}^{35}$ | 0 - 1 | 0 | $\frac{3}{2} - \frac{5}{2}$ | 19.5 | 25.30 |
| | 1 - 2 | 0 | $\frac{5}{2} - \frac{7}{2}$ | 17.6 | 23.72 |
| | | 1 | $\frac{5}{2} - \frac{7}{2}$ | 15.6 | 21.00 |
| $\text{CH}_3\text{Cl}^{37}$ | 0 - 1 | 0 | $\frac{3}{2} - \frac{5}{2}$ | 20.2 | 25.22 |
| | 1 - 2 | 0 | $\frac{5}{2} - \frac{7}{2}$ | 17.9 | 23.60 |
| | 2 - 3 | 0 | $\frac{7}{2} - \frac{7}{2}$ | 12.8 | 21.81 |
| $\text{CH}_3\text{Br}^{79}$ | 0 - 1 | 0 | $\frac{3}{2} - \frac{5}{2}$ | 15.7 | 19.87 |
| | 1 - 2 | 0 | $\frac{1}{2} - \frac{3}{2}$ | 13.6 | 18.37 |
| | | 1 | $\frac{5}{2} - \frac{7}{2}$ | 12.3 | 16.22 |
| | 2 - 3 | 0 | $\frac{7}{2} - \frac{7}{2}$ | 12.8 | 16.72 |
| | | 1 | $\frac{7}{2} - \frac{9}{2}$ | 11.8 | 16.02 |
| | | 2 | $\frac{1}{2} - \frac{3}{2}$ | 11.9 | 13.46 |

TABLE II (Continued)

| Gas | ΔJ | K | ΔF | $(\Delta\nu_p)_{\text{Obs}}$ | $(\Delta\nu_p)_{\text{Calc}}$ |
|-----------------------------|------------|---|------------------------------|------------------------------|-------------------------------|
| $\text{CH}_3\text{Br}^{79}$ | 3 - 4 | 0 | $\frac{5}{2} - \frac{7}{2}$ | 12.8 | 15.50 |
| | | 1 | $\frac{7}{2} - \frac{9}{2}$ | 13.8 | 15.20 |
| | | 2 | $\frac{9}{2} - \frac{11}{2}$ | 12.6 | 14.11 |
| | | 3 | $\frac{7}{2} - \frac{9}{2}$ | 12.0 ^a | 11.85 |
| $\text{CH}_3\text{Br}^{81}$ | 0 - 1 | 0 | $\frac{3}{2} - \frac{5}{2}$ | 15.2 | 19.84 |
| | 1 - 2 | 0 | $\frac{1}{2} - \frac{3}{2}$ | 13.3 | 18.32 |
| | | 1 | $\frac{5}{2} - \frac{7}{2}$ | 12.9 | 16.17 |
| | 2 - 3 | 0 | $\frac{5}{2} - \frac{5}{2}$ | 12.9 | 16.66 |
| | | 2 | $\frac{1}{2} - \frac{3}{2}$ | 12.9 ^a | 13.41 |
| | 3 - 4 | 1 | $\frac{7}{2} - \frac{9}{2}$ | 13.2 | 15.12 |
| | | 2 | $\frac{9}{2} - \frac{11}{2}$ | 12.9 | 14.08 |
| | | 3 | $\frac{3}{2} - \frac{5}{2}$ | 12.7 ^a | 11.81 |
| CH_3I | 0 - 1 | 0 | $\frac{5}{2} - \frac{7}{2}$ | 13.3 | 15.00 |
| | 1 - 2 | 0 | $\frac{7}{2} - \frac{9}{2}$ | 10.8 | 13.97 |
| | | 1 | $\frac{7}{2} - \frac{9}{2}$ | 9.6 | 12.33 |

TABLE II (Continued)

| Gas | ΔJ | K | ΔF | $(\Delta\nu_p)_{\text{Obs}}$ | $(\Delta\nu_p)_{\text{Calc}}$ |
|-------------------------------|------------|---|-------------------------------|------------------------------|-------------------------------|
| CH ₃ I | 2 - 3 | 0 | $\frac{9}{2} - \frac{11}{2}$ | 9.5 ^a | 12.79 |
| | | 2 | $\frac{9}{2} - \frac{9}{2}$ | 9.5 | 10.26 |
| | 3 - 4 | 0 | $\frac{3}{2} - \frac{5}{2}$ | 9.5 | 11.87 |
| | | 1 | $\frac{11}{2} - \frac{11}{2}$ | 9.5 | 11.62 |
| | | 2 | $\frac{11}{2} - \frac{13}{2}$ | 9.6 | 10.81 |
| | | 3 | $\frac{11}{2} - \frac{13}{2}$ | 8.8 | 9.03 |
| | 4 - 5 | 0 | $\frac{11}{2} - \frac{13}{2}$ | 9.8 | 11.37 |
| | | 1 | $\frac{13}{2} - \frac{15}{2}$ | 9.7 | 11.23 |
| | | 2 | $\frac{13}{2} - \frac{15}{2}$ | 9.9 | 10.79 |
| | | 3 | $\frac{13}{2} - \frac{15}{2}$ | 9.7 | 9.94 |
| CH ₃ CN | 0 - 1 | 0 | 1 - 2 | 94.2 | 91.67 |
| C ₃ H ₄ | 0 - 1 | 0 | | 8.7 | 7.04 |
| | 1 - 2 | 0 | | 8.0 ^b | 7.10 |
| | | 1 | | 8.2 ^b | 6.45 |
| | 2 - 3 | 0 | | 7.3 | 7.18 |
| | | 1 | | 8.5 | 6.90 |
| | | 2 | | 8.4 | 5.88 |

TABLE II (Continued)

| Gas | ΔJ | K | ΔF | $(\Delta \nu_p)_{\text{Obs}}$ | $(\Delta \nu_p)_{\text{Calc}}$ |
|-------------------------------|-----------------------|---|------------|-------------------------------|--------------------------------|
| C ₃ H ₄ | 3 \leftrightarrow 4 | 0 | | 8.5 | 7.26 |
| | | 1 | | 8.6 | 7.11 |
| | | 2 | | 7.9 | 6.58 |
| | | 3 | | 7.9 | 5.46 |

^a Poor accuracy because of standing waves.

^b The K = 0 and K = 1 components of the J = 1 \rightarrow 2 transition of C₃H₄ are separated by only 0.7 Mc/sec. The linewidth data for these two lines are less accurate because of the neighbor-line interference.

rather small dipole moment (0.75D) and because the theoretical widths were calculated using only dipole-dipole force. Inclusion of the quadrupole interaction has been attempted but the result is not impressive. In view of the disagreement of the same magnitude existing in other molecules, the quadrupole contribution to linewidths of CH_3CCH is very difficult to pinpoint. The theoretical width comes very close to experimental value for the $0 \rightarrow 1$ line of CH_3CN . The unusually larger linewidth is due to the large dipole moment (3.92D).

The theoretical linewidths decrease with increasing J for each J , $K \rightarrow J+1$, K series (fixed K), and decrease with increasing K in the sequence $J, K \rightarrow J+1, K$ for a given J . This trend is not always clear in the experimental data; when it is evident, the rate of decrease of linewidths with respect to the quantum numbers is smaller than predicted by theory. If we normalize one theoretical width to its corresponding experimental value for each molecule, the relative theoretical linewidths so obtained naturally show better agreement with experiment. This is especially true of CH_3Cl , but the degree of improvement for the other molecules is not particularly impressive.

AMMONIA SELF-BROADENING

In order to find out the extent of the validity of the approximations employed in the calculation of the linewidth of ammonia inversion spectral lines, we also calculated the line width for ammonia self-broadening using the exact expression of (2.36). The

results are tabulated in Table III where $(\Delta\nu)_{\text{obs}}$ and $(\Delta\nu)_1$ are taken from reference 26. $(\Delta\nu)_2$ is the exact calculation of the absolute linewidth. It is obvious that if $(\Delta\nu)_2$ is normalized, a fairly good agreement between $(\Delta\nu)_1$ and $(\Delta\nu)_2$ is yield which indicates that the approximation is a good one.

TABLE III
 LINEWIDTH PARAMETERS (Mc/s-torr)
 OF SELF-BROADENING OF NH₃

| Lines | $(\Delta\nu)_{\text{obs}}$ | $(\Delta\nu)_1$ | $(\Delta\nu)_2$ | Lines | $(\Delta\nu)_{\text{obs}}$ | $(\Delta\nu)_1$ | $(\Delta\nu)_2$ |
|---------|----------------------------|-----------------|-----------------|--------|----------------------------|-----------------|-----------------|
| (2,2) | 22.3 | 22.6 | 29.7 | (4,3) | 22.1 | 20.1 | 26.1 |
| (3,3) | 24.0 | 23.9 | 31.4 | (4,2) | 16.1 | 14.5 | 19.7 |
| (4,4) | 24.4 | (24.4) | 32.2 | (4,1) | 13.4 | 10.3 | 13.4 |
| (5,5) | 24.6 | 24.7 | 32.6 | (6,5) | 22.0 | 22.0 | 28.5 |
| (6,6) | 24.0 | 24.9 | 32.8 | (6,4) | 19.6 | 18.4 | 24.0 |
| (7,7) | 24.6 | 25.0 | 33.0 | (6,3) | 16.6 | 14.9 | 19.5 |
| (8,8) | 24.7 | 25.1 | 33.2 | (6,2) | 13.8 | 11.5 | 14.9 |
| (9,9) | 24.6 | 25.2 | 33.4 | (8,7) | 21.9 | 22.5 | 29.5 |
| (10,10) | 23.8 | 25.3 | 33.6 | (8,6) | 19.8 | 19.5 | 25.7 |
| (3,2) | 17.6 | 17.6 | 23.0 | (8,5) | 17.5 | 16.7 | 21.9 |
| (3,1) | 12.8 | 11.0 | 14.4 | (10,9) | 21.2 | 23.0 | 30.2 |

PART II

APPLICATION OF THE GAUSSIAN-TYPE ORBITALS FOR
CALCULATING ENERGY BAND STRUCTURES OF
SOLIDS BY THE METHOD OF TIGHT BINDING

CHAPTER IV

INTRODUCTION

The electronic energy band structure of crystalline materials has been of great scientific interest for many years and has been studied very extensively both experimentally and theoretically. There are a number of diverse methods employed by various authors²⁷ in solving this problem. They range from such ab initio calculation of self-consistent-field method of Herman²⁸ to empirically adjusted pseudo-potential approach of Cohen.²⁹ Among them one of the oldest and perhaps the more physical ones is the tight-binding method of Bloch.³⁰

Bloch's method is basically an one-electron model. It consists of making the linear combination of atomic orbitals located at various atomic sites to form the so-called "Bloch sum" and use the Bloch sum as basis functions to solve the eigenvalue problem. Before the advent of high speed computers, the main difficulty of applying this method lay in the fact that, in order to evaluate the necessary matrix elements, a large number of multi-center integrals had to be evaluated, which was an almost impossible task as late as 1954.³¹

The first breakthrough in this respect is due to Lafon and Lin.³² In their calculation of the band structure of lithium, they successfully evaluated all the necessary integrals involved without resorting to any adjustable parameter scheme for tight-binding calculation for the first time.

Atomic orbitals generally assume the form of the linear combinations of exponential functions known as Slater orbitals. The Slater orbitals are the well-known exponential functions of the form $\exp(-\alpha r)$ for s-state, $x\exp(-\alpha r)$ for p-state, etc. For Gaussians, we shall use $\exp(-\alpha r^2)$ or $G^s(\alpha, \vec{r})$ to represent an s-state and $x\exp(-\alpha r^2)$ or $G^{px}(\alpha, \vec{r})$ for a p-state Gaussian; and $G(\alpha, \vec{r})$ for either an s- or a p-state Gaussian. To evaluate the integrals arising from two Slater orbitals situated at two different sites is the major difficulty encountered in the molecular calculations. On the other hand, it is known that the Gaussian function has a certain property which is very desirable in evaluating multi-center integrals, namely, the product of the two Gaussians having different centers A and B is itself a Gaussian with a center C somewhere on the line segment joining the two centers, i.e.,

$$G_i(\alpha_i, \vec{r}_A) G_j(\alpha_j, \vec{r}_B) = K G_k(\alpha_k, \vec{r}_C), \quad (4.1)$$

where K is a constant such that

$$K = \exp\left(-\frac{\alpha_i \alpha_j}{\alpha_i + \alpha_j} \overline{AB}^2\right), \quad (4.2)$$

$$\text{and } \alpha'_k = \alpha_i + \alpha_j; \quad (4.3)$$

$$\begin{aligned}
 c_x &= \frac{\alpha_i^A x + \alpha_j^B x}{\alpha_i + \alpha_j}; & c_y &= \frac{\alpha_i^A y + \alpha_j^B y}{\alpha_i + \alpha_j}; \\
 c_z &= \frac{\alpha_i^A z + \alpha_j^B z}{\alpha_i + \alpha_j}.
 \end{aligned}
 \tag{4.4}$$

Therefore, a Slater orbital of the form $\exp(-\alpha r)$ can be transformed through an integral transformation of the form

$$\exp(-\alpha r) = \left(\frac{\alpha}{2/\pi}\right) \int_0^\infty s^{\frac{3}{2}} \exp\left(-\frac{\alpha^2}{4s}\right) \exp(sr^2) ds.
 \tag{4.5}$$

A multi-center integral can thus be reduced to a single-center integral. The integration over the parameter s is usually handled by numerical integration. In their calculations of lithium and diamond³⁴ band structure, a Gaussian quadrature was employed. They proved that the tight-binding method is not only good for the covalent crystal such as diamond but is also very effective for metal like lithium where the valence electron is not tightly bound. The success of Lafon and Lin points out the attractive features of the tight-binding method applied to solids. In these authors' opinion, it should be properly called LCAO (linear combination of atomic orbitals) method.

As early as 1950, Boys³⁵ proposed that the atomic orbitals be expressed as the linear combination of Gaussian functions. Subsequently a number of investigations were carried out to apply the Gaussian orbitals to the molecular calculations.³⁶ The differences between using Gaussian-type-orbitals (GTO) and Slater-type-orbitals (STO) as atomic functions are:

1. Since the Gaussians fall off more rapidly than the Slater orbitals, it requires more Gaussians with a large range of magnitude of Gaussian exponential parameters to represent a wave function which could be represented by a smaller number of moderate size Slater orbitals. Besides, even with a large number of Gaussians, the GTO still fail to represent the tail part of the wave functions satisfactorily due to the sharp-decay nature of the Gaussian functions.

2. In carrying out the multi-center integrals, the integral transformation (4.5) is no longer necessary. The computational labor involved is greatly reduced.

The present work applies the Gaussian orbitals to the energy band calculation. The purpose is two-fold:

1. To investigate the effect which the longer range wave functions have on the energy band structure, Lafon and Lin have shown that the major cause contributed to the failure of the previous tight-binding calculation was the so-called "nearest neighbor approximation" which only takes care of the interaction between the nearest neighboring atoms or the next-nearest neighboring atoms in the crystal and ignores the ones which were separated at larger distance in order to reduce the number of the troublesome multicenter integrals. On the other hand, it will give us additional insight into the problem if we could find out how important the tail part of the wave function is as far as energy band calculation is concerned. A replacement of STO by GTO amounts to a "cut-off" procedure for the wave functions.

2. A reduction of computer time will make the tight-binding

method a practical and feasible one. It is estimated that for the band structure of lithium,³⁷ the GTO approach requires two minutes on the Univac 1108 at the University of Wisconsin Computer Center. The amount required for diamond is about five minutes. The STO requires hours on the CDC 3400. The results of these two different basis functions give excellent agreement to each other. Hence we speed up the computational time by a fairly large factor without any significant loss of accuracy.

It is believed that the Gaussian-type-orbitals will be a feasible tool to study the one-electron model of the solids. It is likely that electronic energy band structure will be but one of the many calculations GTO is capable of yielding useful information.

CHAPTER V
CRYSTAL LATTICE AND CRYSTAL POTENTIAL

CRYSTAL LATTICE

The space lattice of diamond is face-centered cubic with a basis of two atoms at $(0, 0, 0)$ and $(\frac{1}{4}, \frac{1}{4}, \frac{1}{4})$ associated with each lattice point. The crystal can be considered as composed of two interpenetrating face-centered cubic lattices displaced by a quarter of a lattice constant along the body diagonal.

For the face-centered cubic lattice, the periodicity of the crystal lattice can be represented by a set of vector \vec{R}_n defined such that a translation by any \vec{R}_n will produce the potential at the starting point. \vec{R}_n is represented by the expression

$$\vec{R}_n = n_1 \vec{a}_1 + n_2 \vec{a}_2 + n_3 \vec{a}_3 \quad , \quad (5.1)$$

where the n 's are the shortest possible independent periodic translations in the lattice. For fcc lattice:

$$\begin{aligned} \vec{a}_1 &= \frac{a_0}{2}(1, 1, 0); \\ \vec{a}_2 &= \frac{a_0}{2}(1, 0, 1); \\ \vec{a}_3 &= \frac{a_0}{2}(0, 1, 1), \end{aligned} \quad (5.2)$$

where a_0 is the lattice constant.

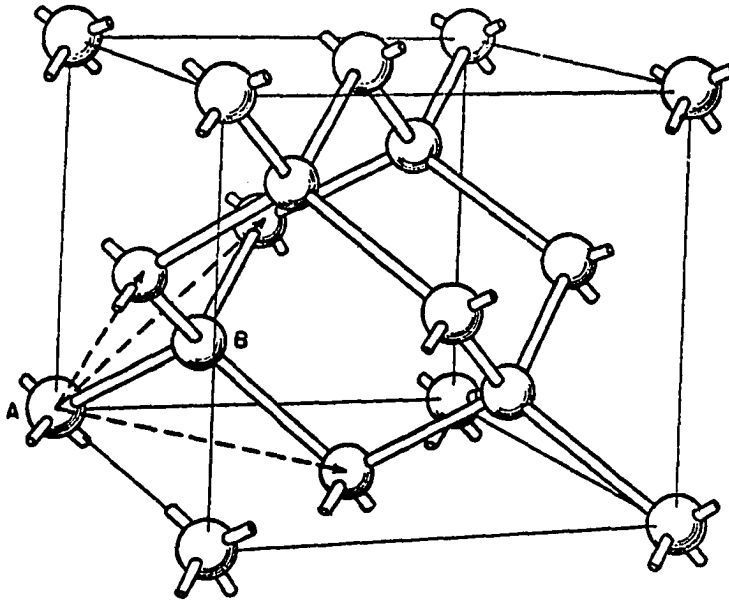


Fig. 3 Crystal Structure of Diamond

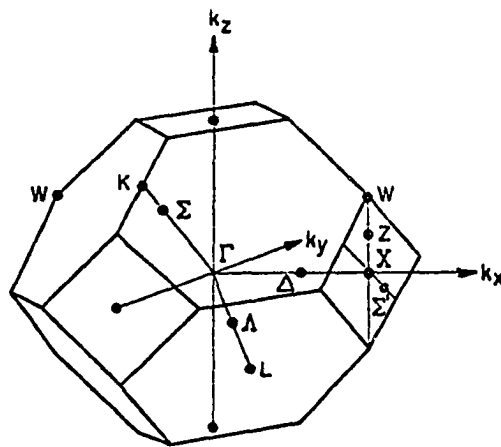


Fig. 4 The Reciprocal Lattice of the fcc Lattice

Now we define another set of vectors which spans the space of the "reciprocal lattice." The significance of this space will be evident in the next chapter. We define another set of vector \vec{K}_m such that

$$\vec{K}_m \cdot \vec{R}_n = 2\pi l_{mn} , \quad (5.3)$$

where l_{mn} are the integers. The set of vector \vec{K}_m form the reciprocal lattice. In terms of the shortest possible independent periodic translations in the reciprocal lattice,

$$\vec{K}_m = m_1 \vec{b}_1 + m_2 \vec{b}_2 + m_3 \vec{b}_3 . \quad (5.4)$$

The reciprocal lattice for fcc lattice is itself a body-centered cubic and

$$\begin{aligned} \vec{b}_1 &= \frac{2\pi}{a_0} (1, 1, -1); \\ \vec{b}_2 &= \frac{2\pi}{a_0} (1, -1, 1) ; \\ \vec{b}_3 &= \frac{2\pi}{a_0} (-1, 1, 1). \end{aligned} \quad (5.5)$$

CRYSTAL POTENTIAL

The Schroedinger equation for an electron moving in a perfectly periodic crystal lattice is

$$\left[-\frac{1}{2} \nabla^2 + V(\vec{r}) \right] \psi(\vec{r}) = E(\vec{r}), \quad (5.6)$$

where atomic units are used.³⁸

unit of mass = the rest-mass of the electron

unit of charge = $|e|$, the magnitude of the charge
on the electron

unit of length = the radius of the first Bohr orbit
of the hydrogen atom

unit of energy = twice the ionization energy of the
normal state of hydrogen atom.

$V(\vec{r})$ is the crystal potential and according to Bloch's theorem, the solution must have the form

$$\psi_{\vec{k}}(\vec{r}) = \exp(i\vec{k} \cdot \vec{r}) U_{\vec{k}}(\vec{r}), \quad (5.7)$$

where \vec{k} is the crystal momentum. The choice of crystal potential is such that it will represent a first approximation of the true crystal potential of the crystal under consideration. Here, in order to compare the result with that of Lafon and Lin using STO, atomic-Hartree-Fock-Slater potential (AHFS) according to the scheme of Woodruff³⁹ is used for diamond. The crystal potential is considered as the superposition of each individual free-atom potential:

$$V(\vec{r}) = \sum_{\nu} V_{\text{atomic}}(\vec{r} - \vec{R}_{\nu}), \quad (5.8)$$

where \vec{R}_{ν} is the primitive translation vector for the atomic sites. The atomic potential is in turn expressed as the sum of the Coulomb and exchange contributions:

$$V_{\text{atomic}}^{\text{Coul}}(\vec{r}) = -\frac{Z}{r} + \left(\frac{4\pi}{r}\right) \int_0^r \rho(r') r'^2 dr' + 4\pi \int_r^{\infty} \rho(r') r' dr', \quad (5.9)$$

$$V_{\text{atomic}}^{\text{exch}}(r) = -\frac{3}{2} [3\rho(r)/\pi]^{\frac{1}{3}}. \quad (5.10)$$

Where Z is the atomic number, $\rho(r)$ is the atomic charge density.

In equation (5.10), Slater exchange term has been used. This term has been subjected to various alternations in recent years.⁴⁰

However, no attempt was made here to adopt any of those alternations of numerical factor. It should be noted, however, the exchange crystal potential should be the cubic root of the sum of the atomic exchange potentials. An approximation is made here to represent the

exchange crystal potential as the sum of the cubic root of each individual free atomic potential. This procedure simplifies the calculation. In his subsequent study of diamond charge density,⁴¹ Lafon made a comparison of these two approaches and found that the simplification is indeed justified.

For diamond, the free atomic charge density $\rho(r)$ is obtained from the Hartree-Fock self-consistent-field calculations of Jucys'.⁴² The expression is

$$4\pi\rho(r) = 2[R_{1s}(r)]^2 + 2[R_{2s}(r)]^2 + 2[R_{2p}(r)]^2, \quad (5.11)$$

where R_{1s} , R_{2s} , and R_{2p} are the radial part of the 1s, 2s, and 2p wave functions respectively. The origin of the potential is located at the mid-point connecting the two face-centered cubic lattice. The Wigner-Seitz cell about the origin has a volume of $\Omega = \frac{1}{4}a_0^3$ and contains two atoms at the locations given by

$$\vec{t}_1 = \frac{1}{8} a_0(1, 1, 1) \quad \text{and} \quad \vec{t}_2 = -\vec{t}_1,$$

for the two face-centered cubic lattices one and two.

Then the potential can be expressed as

$$V(\vec{r}) = \sum_{\nu} \sum_{i=1}^2 V_{\text{atomic}} [\vec{r} - (\vec{R}_{\nu} + \vec{t}_i)], \quad (5.12)$$

which is then expanded in the reciprocal lattice

$$V(\vec{r}) = \sum_{\nu} V(\vec{K}_{\nu}) \exp(i\vec{K}_{\nu} \cdot \vec{r}). \quad (5.13)$$

Due to the choice of the origin, the potential is invariant under the inversion operation. Equation (5.13) can be written as

$$V(\vec{r}) = \sum_{\nu} V(\vec{K}_{\nu}) \cos \vec{K}_{\nu} \cdot \vec{r}. \quad (5.14)$$

The Fourier coefficients are given by

$$V(\vec{K}_{\nu}) = \frac{1}{N\Omega} \int V(\vec{r}) \cos \vec{K}_{\nu} \cdot \vec{r} d\tau \quad (5.15)$$

where N is the number of unit cells in the crystal and Ω is the volume of the unit cell. The Fourier coefficients can also be divided into two parts: Coulomb and exchange parts. The detailed derivation of this Fourier transformation is given in appendix B. The result is:

$$V(\vec{K}_y) = -\frac{8\pi}{K^2\Omega} \cos \vec{K}_y \cdot \vec{t} \left\{ Z - \frac{1}{K_y} \int_0^\infty Q(r) \sin K_y r dr + K_y \int_0^\infty E(r) \sin K_y r dr \right\}, \quad (5.16)$$

where $Q(r) = 4\pi r \rho(r)$

$$E(r) = \frac{3}{2} r [3\rho(r)/\pi]^{\frac{1}{3}}.$$

Using $a_0 = 6.728$ a.u., Lafon and Lin generated 4409 Fourier coefficients. The $V(0, 0, 0)$ was set at -1.435 a.u. in order to compare with the OPW calculation of Bassani and Yoshimine⁴³ using the same potential. The first fifteen coefficients are listed in Table IV.

TABLE IV
FOURIER COEFFICIENTS FOR $V(r)$

| $\frac{a_0 \vec{k}}{2\pi} = \vec{l}$ | | | Fourier Coefficients |
|--------------------------------------|-------|-------|----------------------|
| l_x | l_y | l_z | V_K |
| 0 | 0 | 0 | -1.435 |
| 1 | 1 | 1 | -0.4550 |
| 2 | 2 | 0 | -0.2080 |
| 3 | 1 | 1 | -0.1586 |
| 4 | 0 | 0 | -0.1148 |
| 3 | 3 | 1 | -0.0990 |
| 4 | 2 | 2 | -0.08094 |
| 3 | 3 | 3 | -0.07312 |
| 5 | 1 | 1 | -0.07312 |
| 4 | 4 | 0 | -0.06312 |
| 5 | 3 | 1 | -0.05838 |
| 6 | 2 | 0 | -0.05192 |
| 5 | 3 | 3 | -0.04870 |
| 4 | 4 | 4 | -0.04414 |
| 5 | 5 | 1 | -0.04179 |

CHAPTER VI

CRYSTAL SYMMETRY AND GROUP THEORY

The full symmetry group of a crystalline solid is the space group which consists of translational-symmetry operators and rotational-symmetry operators. The former form the pure translational subgroup and the latter form the point group. A well-known group-theoretical principle in quantum mechanics states that the wave functions of a quantum system must form bases for irreducible representations of the group of operators which commute with the Hamiltonian of the system.

A general space-group element may be written as $\{\vec{R}|\vec{t}\}$ where \vec{R} is the rotational operator of the point group and \vec{t} is the translational operator of the translation group,

$$\{\vec{R}|\vec{t}\}\vec{x} = \vec{R}\vec{x} + \vec{t}. \quad (6.1)$$

When $\vec{t} = 0$, the aggregate of the operators form the point-group. When \vec{R} is the identity operator \vec{E} , the aggregate of the pure translational operators form the invariant and Abelian subgroup. This translational subgroup contains all the translations which can be written as

$$\vec{t}_n = n_1\vec{a}_1 + n_2\vec{a}_2 + n_3\vec{a}_3 \quad (6.2)$$

Bloch Theorem states that if we have an operator $\vec{T}_{\vec{x}}$ such

that

$$\vec{T}_{\vec{a}}(\vec{r}) = \vec{r} + \vec{a},$$

then for a wave function $\psi_{\vec{k}}$

$$\vec{T}_{\vec{a}}(\psi_{\vec{k}}) = \exp(i\vec{k} \cdot \vec{a})\psi_{\vec{k}}. \quad (6.3)$$

The Hamiltonian H belongs to the translational group with

element $\vec{T}_{\vec{t}_n}$. Since for any arbitration translation group (6.3) holds, we can consider $\exp(i\vec{k} \cdot \vec{t}_n)$ as the representation of $\vec{T}_{\vec{t}_n}$, then

$$\vec{T}_{\vec{t}_n}\psi(\vec{r}) = \exp(i\vec{k} \cdot \vec{t}_n)\psi(\vec{r}). \quad (6.4)$$

We now label $\psi(\vec{r})$ as $\psi_{\vec{k}}(\vec{r})$, then write

$$\psi_{\vec{k}}(\vec{r}) = \exp(i\vec{k} \cdot \vec{r})U_{\vec{k}}(\vec{r}), \quad (6.5)$$

$$\vec{T}_{\vec{t}_n}\psi_{\vec{k}}(\vec{r}) = \exp(i\vec{k}(\vec{r} + \vec{t}_n))U_{\vec{k}}(\vec{r} + \vec{t}_n). \quad (6.6)$$

Therefore,

$$U_{\vec{k}}(\vec{r} + \vec{t}_n) = U_{\vec{k}}(\vec{r}), \quad (6.7)$$

and $U_{\vec{k}}$ is periodic. This satisfies the periodic property of the crystal.

In chapter V, we have seen that a typical vector in the reciprocal space is defined as

$$\vec{K}_m = 2\pi(m_1\vec{b}_1 + m_2\vec{b}_2 + m_3\vec{b}_3),$$

and

$$\begin{aligned} \vec{K}_m \cdot \vec{t}_n &= 2\pi(m_1\vec{n}_1 + m_2\vec{n}_2 + m_3\vec{n}_3) \\ &= 2\pi \times \text{integer}. \end{aligned}$$

Then

$$\exp(i(\vec{k} + \vec{K}_m) \cdot \vec{t}_n) = \exp(i\vec{k} \cdot \vec{t}_n). \quad (6.8)$$

If we apply the translational operator $\vec{T}_{\vec{t}_n}$ on the wave

function $\psi_{\vec{k} + \vec{K}_m}$,

$$\begin{aligned} \vec{T}_{\vec{t}_n} \psi_{\vec{k} + \vec{K}_m} &= \exp(i(\vec{k} + \vec{K}_m) \cdot \vec{t}_n) \psi_{\vec{k} + \vec{K}_m} \\ &= \exp(i\vec{k} \cdot \vec{t}_n) \psi_{\vec{k} + \vec{K}_m}(\vec{r}) \end{aligned} \quad (6.9)$$

This shows that $\psi_{\vec{k}}$ and $\psi_{\vec{k} + \vec{K}_m}$ have the same symmetry property. Thus \vec{k} and \vec{K} are equivalent vectors in the sense as far as the relation (6.4) is concerned, \vec{k} and $\vec{k} + \vec{K}_m$ are not distinguishable. Therefore we can consider them to have essentially the same \vec{k} value. It is justified now that we confine all our attention to \vec{k} vectors lying inside a finite zone of the k -space. This zone is known as the first Brillouin zone which is constructed by setting up perpendicular bisecting planes on lines connecting the origin to all reciprocal-lattice points and then taking the volume about the origin enclosed by these planes. The set of points $\vec{k} + \vec{K}_m$ all designates the same irreducible representations of the translation group. All of k -space can be filled by fitting together a multitude of such zones, each centered on a lattice point, in which case the energy is a periodic function k throughout the k -space. Conventionally, one chooses the wave vector \vec{k} so that it always lies within the first Brillouin zone. This procedure is known as reduced zone scheme.

If \vec{R} is the operator of the point group, then when we apply this operator to a wave function of Bloch form

$$\begin{aligned} \vec{R} \psi_{\vec{k}} &= \vec{R} \exp(i\vec{k} \cdot \vec{r}) U_{\vec{k}}(\vec{r}) \\ &= U_{\vec{k}}(\vec{R}^{-1} \vec{r}) \exp(i\vec{k} \cdot \vec{R}^{-1} \vec{r}) \end{aligned} \quad (6.10)$$

~

It is seen that

$$\begin{aligned}\vec{k} \cdot \vec{R}^{-1}\vec{r} &= \vec{R} \cdot \vec{k} \cdot \vec{R}\vec{R}^{-1}\vec{r} \\ &= \vec{R} \cdot \vec{k} \cdot \vec{r},\end{aligned}\quad (6.11)$$

and the result of operation is

$$\vec{R}\psi_{\vec{k}} = \exp(i\vec{R}\vec{k} \cdot \vec{r})U_{\vec{k}}(\vec{R}^{-1}\vec{r}). \quad (6.12)$$

$U_{\vec{k}}(\vec{R}^{-1}\vec{r})$ is periodic if $U_{\vec{k}}(\vec{r})$ is since if \vec{r} is a primitive translation, $\vec{R}^{-1}\vec{r}$ must also be a primitive translation. Periodicity is all that is required for $U_{\vec{k}}(\vec{r})$. Hence we may write $U_{\vec{k}}(\vec{R}^{-1}\vec{r})$ as $U_{\vec{R}\vec{k},\vec{k}}(\vec{r})$. $U_{\vec{R}\vec{k},\vec{k}}(\vec{r})$ and $U_{\vec{k}}(\vec{r})$ only differ by a phase factor. The result is that by applying an operator to the wave function, we are only rotating the \vec{k} -vector. The eigenfunction is still in Bloch form. This conclusion enables us to inspect the \vec{k} -vector and find out all the symmetry-related \vec{k} vectors, all of them must have associated eigenfunction of the same energy. In other words, if ψ is an acceptable wave function with eigenvalue E , $\vec{R}\psi$ is also an acceptable wave function with the same energy.

If we start with an arbitrary \vec{k} -vector and apply all the rotational operations and generate all the orientations of the \vec{k} -vector, such aggregate of orientations is called a "star",⁴⁴ and all the \vec{k} 's are different. However, if we place the \vec{k} -vector along a pre-chosen symmetry position, then some of the $\vec{R}\vec{k}$'s are the same. The subgroup of \vec{R} 's which leave \vec{k} unchanged is called the group of the wave vector. The irreducible representations of the group are called small representations. The small group is the subgroup of the point group.

To summarize: the translational symmetry properties of the crystal makes it possible for us to confine our discussion of the energy band structure within the first Brillouin zone. The rotational symmetry properties give us the information about the energy levels and degeneracies in the crystal.

CHAPTER VII

BASIS FUNCTIONS

The atomic wave functions have the usual form

$$\psi = \sum_i c_i \phi_{g,i} \quad , \quad (7.1)$$

where

$$\phi_{g,i} = R_n(r) Y_{lm}(\theta, \phi) \quad . \quad (7.2)$$

The radial-part of the wave functions is taken from Huzinaga's self-consistent-Hartree-Fock calculation of carbon,⁴⁵

$$R_n(r) = N_i r^{n-1} \exp(-\alpha_i r^2) \quad , \quad (7.3)$$

where N_i , the normalization constant is given by

$$N_i = \left[\frac{2^{2n+\frac{3}{2}}}{(2n-1)!!/\pi} \right]^{\frac{1}{2}} (\alpha_i)^{\frac{(2n+1)}{4}} \quad .$$

The coefficients C_i and the Gaussian exponential parameter α_i for 1s, 2s, and 2p orbitals of carbon are listed in Table V.

Equation (7.1) could be written more explicitly as the following expressions,

$$\psi_{1s, 2s} = \left[8 \left(\frac{2}{\pi} \right)^{\frac{3}{2}} \right]^{\frac{1}{2}} \left\{ \sum_j C_j \alpha_j^{\frac{3}{4}} \exp(-\alpha_j r^2) \right\} Y_{0,0}(\theta, \phi), \quad (7.4a)$$

$$\psi_{2p} = \left[\frac{32 \left(\frac{2}{\pi} \right)^{\frac{3}{2}}}{3} \right]^{\frac{1}{2}} r \left\{ \sum_j C_j \alpha_j^{\frac{5}{4}} \exp(-\alpha_j r^2) \right\} Y_{1,m}(\theta, \phi). \quad (7.4b)$$

TABLE V
C_i AND α_i OF CARBON

| C _i (1s) | C _i (2s) | C _i (2p) | α_i |
|---------------------|---------------------|---------------------|------------|
| 0.43809 | -0.17699 | | 5.14773 |
| 0.15459 | -0.03606 | | 42.4974 |
| 0.04534 | -0.00974 | | 146.097 |
| 0.00934 | -0.00202 | | 634.882 |
| 0.14581 | -0.05267 | | 1.96655 |
| 0.00199 | 0.57408 | | 0.49624 |
| 0.35867 | -0.08938 | | 14.1892 |
| 0.00122 | -0.00026 | | 4232.61 |
| 0.00041 | 0.54768 | | 0.15331 |
| | | 0.50734 | 0.35945 |
| | | 0.30611 | 1.14293 |
| | | 0.09150 | 3.98640 |
| | | 0.01469 | 18.1557 |
| | | 0.31735 | 0.11460 |

We thus have all the free atomic wave functions at our disposal to construct the basis functions for the crystal Hamiltonian, the Bloch sum.

The Bloch sum is formed from these atomic functions for each of the two sub-lattices,

$$b_{\alpha}^j(\vec{k}, \vec{r}) = \sum_{\vec{y}} \exp(i\vec{k} \cdot \vec{R}_{\vec{y}}) \psi_{\alpha}[\vec{r} - (\vec{R}_{\vec{y}} + \vec{e}_j)] , \quad (7.5)$$

where $\alpha = 1s, 2s, 2p_x, 2p_y, 2p_z$ and the index j indicates the sub lattice, thus $b_{1s}^1(\vec{k}, \vec{r})$ designates a $1s$ -Bloch sum associated with the first sub-lattice, etc. In order to avoid the complex factor in the energy matrix, it is a common practice to choose a phase factor for the Bloch sum so that the matrix elements are all real. In the case of diamond, at $(0, 0, 0)$ in k -space which in BSW notation⁴⁴ is known as Γ -point, the calculation will be further simplified by forming "bonding" and anti-bonding" combinations,

$$b_{\alpha}^{+}(\vec{k}, \vec{r}) = I_{\alpha}^{+}(\alpha) [N \Omega_{\alpha}^{+}(k)]^{-\frac{1}{2}} \sum_{\nu} \exp(i\vec{k} \cdot \vec{R}_{\nu}) \times \{ \psi_{\alpha}[\vec{r} - (\vec{R}_{\nu} + \vec{t}_1)] \pm \psi_{\alpha}[\vec{r} - (\vec{R}_{\nu} + \vec{t}_2)] \} , \quad (7.6)$$

the phase factor $I_{\alpha}^{+}(\alpha)$ are defined as

$$I_{\alpha}^{+}(\alpha) = -i I_{\alpha}^{-}(\alpha) = \begin{cases} 1 & \text{for } \alpha = ns \\ i & \text{for } \alpha = np_x, y, z \end{cases} ,$$

and $\Omega_{\alpha}^{+}(k)$ are the normalization constants. At Γ -point, the Γ group of diamond lattice divides into the irreducible representations:

$$\Gamma_1 + \Gamma_{15} + \Gamma_{25}' + \Gamma_2'$$

where Γ_1 and Γ_2' are both singlet and Γ_{15} and Γ_{25}' are both triplet. From the transformation properties of various wave functions, it is obvious that the bonding of the s -functions forms the representation of Γ_1 ; anti-bonding Γ_2' . The bonding of the p -functions forms the representation of Γ_{15} and anti-bonding Γ_{25}' .⁴⁶

CHAPTER VIII

MULTICENTER INTEGRALS

In order to find the energy band, the ten Bloch functions are used as basis for the secular equation,

$$| H_{nlm, n'l'm'}(\vec{k}) - E S_{nlm, n'l'm'}(\vec{k}) | = 0, \quad (8.1)$$

where H represents the one-electron Hamiltonian

$$H = -\frac{1}{2}\nabla^2 + V(\vec{r}). \quad (8.2)$$

The matrix elements consist of the overlap, kinetic, and potential integrals.

$$\begin{aligned} S_{ij}(\vec{k}) &= \int b_i^*(\vec{k}, \vec{r}) b_j(\vec{k}, \vec{r}) d\tau \\ &= [\Omega_i(\vec{k}) \Omega_j(\vec{k})]^{-\frac{1}{2}} \sum_{\nu} \exp(i\vec{k} \cdot \vec{R}_{\nu}) \int \psi_i^*(\vec{r}) \psi_j(\vec{r} - \vec{R}_{\nu}) d\tau, \\ T_{ij}(\vec{k}) &= \int b_i^*(\vec{k}, \vec{r}) (-\frac{1}{2}\nabla^2) b_j(\vec{k}, \vec{r}) d\tau \\ &= [\Omega_i(\vec{k}) \Omega_j(\vec{k})]^{-\frac{1}{2}} \sum_{\nu} \exp(i\vec{k} \cdot \vec{R}_{\nu}) \\ &\quad \times \int \psi_i^*(\vec{r}) (-\frac{1}{2}\nabla^2) \psi_j(\vec{r} - \vec{R}_{\nu}) d\tau, \\ V_{ij}(\vec{k}) &= \int b_i^*(\vec{k}, \vec{r}) V(\vec{r}) b_j(\vec{k}, \vec{r}) d\tau \\ &= [\Omega_i(\vec{k}) \Omega_j(\vec{k})]^{-\frac{1}{2}} \sum_{\nu} \exp(i\vec{k} \cdot \vec{R}_{\nu}) \\ &\quad \times \int \psi_i^*(\vec{r}) V(\vec{r}) \psi_j(\vec{r} - \vec{R}_{\nu}) d\tau. \end{aligned} \quad (8.3)$$

The expressions $\int \psi_i^*(\vec{r}) \psi_j(\vec{r} - \vec{R}_{\nu}) d\tau$, etc., are known as the multicenter integrals. Since the atomic orbitals ψ 's are expressed as linear combinations of Gaussians, provided that the individual integrals involving Gaussians are evaluated, we have all the

integrals at our disposal.

Let a 1s Gaussian be placed at point A, and another 1s Gaussian at B, then the overlap integral for these two Gaussians is

$$\begin{aligned} & \int \exp(-\alpha_1 r_A^2) \exp(-\alpha_2 r_B^2) d\tau \\ &= \left(\frac{\pi}{\alpha_1 + \alpha_2} \right)^{\frac{3}{2}} \exp\left(-\frac{\alpha_1 \alpha_2}{\alpha_1 + \alpha_2} \overline{AB}^2\right), \end{aligned} \quad (8.4)$$

where $\vec{r}_A = \vec{r} - \vec{A}$ and $\vec{r}_B = \vec{r} - \vec{B}$.

Integrals involving higher orbitals can be obtained by successive differentiations of the 1s-1s integral, i.e.,

$$\begin{aligned} & \int \exp(-\alpha_1 r_A^2) x_B \exp(-\alpha_2 r_B^2) d\tau \\ &= \frac{1}{-2\alpha_2} \frac{\partial}{\partial B_x} \exp(-\alpha_1 r_A^2) \exp(-\alpha_2 r_B^2) d\tau \\ &= -\left(\frac{\pi}{\alpha_1 + \alpha_2} \right)^{\frac{3}{2}} \left(\frac{\alpha_1}{\alpha_1 + \alpha_2} \right) \overline{AB}_x \exp\left(-\frac{\alpha_1 \alpha_2}{\alpha_1 + \alpha_2} \overline{AB}^2\right). \end{aligned} \quad (8.5)$$

Various integrals are listed in the appendix D.

The Hamiltonian for the potential integral has a general form of

$$V(\vec{r}_y) \cos(\vec{k}_y \cdot \vec{r}_c). \quad (8.6)$$

The potential integral then assumes the form

$$\exp(-\alpha_1 r_A^2) \cos(\vec{k}_y \cdot \vec{r}_c) \exp(-\alpha_2 r_B^2) d\tau, \quad (8.7)$$

from chapter IV, we know that

$$\begin{aligned} & \exp(-\alpha_1 r_A^2) \exp(-\alpha_2 r_B^2) \\ &= \exp\left(-\frac{\alpha_1 \alpha_2}{\alpha_1 + \alpha_2} \overline{AB}^2\right) \exp(-(\alpha_1 + \alpha_2) r_D^2) \\ D_i &= (\alpha_1 A_i + \alpha_2 B_i) / (\alpha_1 + \alpha_2); i = x, y, z. \end{aligned}$$

Thus the Hamiltonian $\cos \vec{K}_\nu \cdot \vec{r}_C$ can be written

as

$$\begin{aligned} \cos \vec{K}_\nu \cdot \vec{r}_C &= \cos[\vec{K}_\nu \cdot (\vec{r}_{CD} + \vec{r}_D)] \\ &= \cos \vec{K}_\nu \cdot \vec{r}_{CD} \cos \vec{K}_\nu \cdot \vec{r}_D - \sin \vec{K}_\nu \cdot \vec{r}_{CD} \\ &\quad \times \sin \vec{K}_\nu \cdot \vec{r}_D . \end{aligned} \quad (8.8)$$

The integral becomes

$$\begin{aligned} &\int \exp\left(-\frac{\alpha_1 \alpha_2}{\alpha_1 + \alpha_2} \overline{AB}^2\right) \exp(-(\alpha_1 + \alpha_2) r_D^2) \\ &\quad \times [\cos \vec{K}_\nu \cdot \vec{r}_{CD} \cos \vec{K}_\nu \cdot \vec{r}_D - \sin \vec{K}_\nu \cdot \vec{r}_{CD} \sin \vec{K}_\nu \cdot \vec{r}_D] d\tau . \end{aligned} \quad (8.9)$$

Let I_1 be the integral

$$\int \exp(-(\alpha_1 + \alpha_2) r_D^2) \cos \vec{K}_\nu \cdot \vec{r}_D d\tau ,$$

then I_1 can be evaluated as

$$\begin{aligned} I_1 &= \int_0^\infty r_D^2 \exp(-(\alpha_1 + \alpha_2) r_D^2) dr_D \\ &\quad \times \int \cos(K_\nu r_D \cos \theta) \sin \theta d\theta d\varphi \\ &= -\frac{2\pi}{K_\nu} \int_0^\infty r_D \exp(-(\alpha_1 + \alpha_2) r_D^2) dr_D \int_{K_\nu r_D}^{-K_\nu r_D} \cos u du \\ &= \frac{4\pi}{K_\nu} \int_0^\infty r_D \exp(-(\alpha_1 + \alpha_2) r_D^2) \sin K_\nu r_D dr_D \\ &= \left(\frac{\pi}{\alpha_1 + \alpha_2}\right)^{\frac{3}{2}} \exp\left(-\frac{K_\nu^2}{4(\alpha_1 + \alpha_2)}\right) \end{aligned} \quad (8.10)$$

and

$$I_2 = \int \exp(-(\alpha_1 + \alpha_2) r_D^2) \sin \vec{K}_\nu \cdot \vec{r}_D d\tau = 0 .$$

Therefore:

$$\begin{aligned} &\int \exp(-\alpha_1 r_A^2) V(\vec{r}) \exp(-\alpha_2 r_B^2) d\tau \\ &= \sum_\nu V(\vec{K}_\nu) \cdot \exp\left(-\frac{\alpha_1 \alpha_2}{\alpha_1 + \alpha_2} \overline{AB}^2\right) \left(\frac{\pi}{\alpha_1 + \alpha_2}\right)^{\frac{3}{2}} \\ &\quad \exp\left(-\frac{K_\nu^2}{4(\alpha_1 + \alpha_2)}\right) \cos \vec{K}_\nu \cdot \vec{r} . \end{aligned} \quad (8.11)$$

At this point, one major difficulty arises.

This is the convergence of K_y which is dictated primarily by the factor

$$V(K_y) \exp\left(-\frac{K_y^2}{4(\alpha_1 + \alpha_2)}\right) ;$$

for cases where $(\alpha_1 + \alpha_2)$ is large, convergence is reached only after a large number of K_y is employed which makes it almost impossible to carry out. This slow convergence arises mainly from the singularity of the crystal potential which varies like $-Z/|\vec{r} - \vec{R}_\nu|$ about each nucleus. In order to remove this difficulty we divide the crystal potential into two parts,

$$V(\vec{r}) = V_1(\vec{r}) + V_2(\vec{r}) , \quad (8.12)$$

where $V_1(\vec{r})$ and $V_2(\vec{r})$ have the following properties.

(1) $V_1(\vec{r})$ has the full symmetry of the crystal and constructed as a superposition of "localized" contributions from each atom. Its Fourier expansion is identical to that of $V(\vec{r})$ for large K_y and each atomic contribution is readily integrated with the Gaussian at the same center.

(2) $V_2(\vec{r})$ also has the full symmetry of the crystal but represents a relatively smooth function which converges rapidly in K_y . Once $V_1(\vec{r})$ is known, $V_2(\vec{r})$ could be gotten by

$$V_2(\vec{r}) = V(\vec{r}) - V_1(\vec{r}) .$$

With the properties of $V_1(\vec{r})$ and $V_2(\vec{r})$ in mind, we can see that $V_1(\vec{r})$ behaves like $-Z/|\vec{r} - \vec{R}_\nu|$ about each nucleus. $V_2(\vec{r})$, which is a relatively smooth function of r , can be expanded in a rapidly convergent series,

$$\begin{aligned}
& \langle G(\alpha_1, \vec{r}_A) | V_2(\vec{r}) | G(\alpha_2, \vec{r}_B) \rangle \\
&= \sum_{\vec{r}} a_{\vec{r}} \langle G(\alpha_1, \vec{r}_A) | \cos \vec{K}_{\vec{r}} \cdot \vec{r} | G(\alpha_2, \vec{r}_B) \rangle .
\end{aligned} \tag{8.13}$$

The details about the convergence will be given in appendix C.

On the other hand, $V_1(\vec{r})$ is now expanded in the direct space as a superposition of function \mathcal{V} centered at each site,

$$\langle G(\alpha_1, \vec{r}_A) | V_1(\vec{r}) | G(\alpha_2, \vec{r}_B) \rangle . \tag{8.14}$$

$$= \sum_{\vec{r}_\mu} \langle G(\alpha_1, \vec{r}_A) | \mathcal{V}(\vec{r} - \vec{r}_\mu) | G(\alpha_2, \vec{r}_B) \rangle . \tag{8.14}$$

The only restriction upon $\mathcal{V}(\vec{r})$ is that it reproduces $-Z/|\vec{r} - \vec{r}_\mu|$ near the origin. We are free to choose the form of \mathcal{V} for the region away from the origin in such a way to facilitate the calculation. By making $\mathcal{V}(\vec{r})$ negligibly small before it reaches the next neighbor, one can improve the summation. The particular form of $\mathcal{V}(\vec{r})$ employed is

$$\mathcal{V}(\vec{r}) = -(Z/r) (1 + \frac{1}{2} r^2) \exp(-\frac{1}{2} r^2), \tag{8.15}$$

with $\frac{1}{2} = 2.5$. For the case where both Gaussians centered at the same center and $\alpha_1 + \alpha_2 > 40$,

$$\begin{aligned}
& \langle G(\alpha_1, \vec{r}_A) | V_1(\vec{r}) | G(\alpha_2, \vec{r}_A) \rangle \quad \text{may be approximated by} \\
& \simeq \langle G(\alpha_1, \vec{r}_A) | \mathcal{V}(\vec{r}_A) | G(\alpha_2, \vec{r}_A) \rangle .
\end{aligned} \tag{8.16}$$

where A and B refer to two different sites, the only non-negligible integrals of V_1 for $\alpha_1 + \alpha_2 > 40$ occur when $\alpha_1 \gg \alpha_2$ or $\alpha_2 \gg \alpha_1$. The approximation used are

$$\begin{aligned}
& \langle G(\alpha_1, \vec{r}_A) | V_1(\vec{r}) | G(\alpha_2, \vec{r}_B) \rangle \\
& \simeq \langle G(\alpha_1, \vec{r}_A) | \mathcal{V}(\vec{r}_A) | G(\alpha_2, \vec{r}_B) \rangle \quad \text{for } \alpha_1 \gg \alpha_2 \\
& \simeq \langle G(\alpha_1, \vec{r}_A) | \mathcal{V}(\vec{r}_B) | G(\alpha_2, \vec{r}_B) \rangle , \quad \text{for } \alpha_2 \gg \alpha_1 .
\end{aligned} \tag{8.17}$$

In actual calculation, for the case $\alpha_1 \gg \alpha_2$ we expand $G(\alpha_2, \vec{r}_B)$ about point A in Taylor series and equation (8.17) can be readily integrated out. The validity of this approximation is borne out by the fact that for the single-center integral, $\alpha_1 = 4232.61$ and $\alpha_2 = 4232.61$, this cut-off procedure gives the $(1s|V|1s)$ integral as -0.4392×10^{-2} while a rigorous summation of 1000 K_V 's gives -0.4391×10^{-2} and a summation of 3000 K_V 's gives -0.4386×10^{-2} . For the high-low combinations, we tested the pair with $\alpha_1 = 42.4974$ and $\alpha_2 = 0.15331$, a rigorous summation of K_V 's takes 944 K_V 's before it reaches convergence and gives the $(1s|V|1s)$ integral value as $-0.22862507 \times 10^{-1}$ and the cut-off procedures gives $-0.22869800 \times 10^{-1}$. From these figures, it is established that the approximation used here is indeed justified. The detail of the expansion scheme is also given in appendix C.

CHAPTER IX

RESULT AND DISCUSSION

In this calculation, all the necessary integrals are evaluated by the method described in the previous chapter. It is found that a big reduction of time is obtained by going from STO to GTO even though the latter has a much larger number of basis functions, i.e., in the STO calculation of diamond band structure, six STO exponential parameters are used which results in a total of 21 integrals. In the GTO calculation, 14 GTO Gaussian parameters are used and 105 integrals are evaluated. Even though the number of integrals to be evaluated in the GTO formulation is five times the number of integrals required in the STO, the total computational time of the GTO approach is still far less than the STO method. The results and their comparison with the STO calculation are listed in tables VI - VIII. It is seen that they agree with each other very well.

TABLE VI

COMPARISON OF GTO AND STO AT SYMMETRY POINTS
(ENERGY IN a.u.)

| | STO | GTO |
|----------------|--------|--------|
| Γ_1 | -0.743 | -0.734 |
| Γ_{25}' | 0.0 | 0.0 |
| Γ_{15}' | +0.230 | +0.232 |
| Γ_2' | +0.494 | +0.494 |
| X_1 | -0.446 | -0.437 |
| X_4 | -0.217 | -0.205 |
| X_1 | +0.296 | +0.321 |
| X_3 | +0.597 | +0.593 |
| L_2' | -0.553 | -0.553 |
| L_1 | -0.452 | -0.419 |
| L_3 | -0.100 | -0.096 |
| L_1 | +0.368 | +0.371 |
| L_3 | +0.369 | +0.371 |
| L_2' | +0.653 | +0.657 |

TABLE VII
COMPARISON OF OPTICAL TRANSITIONS OF DIAMOND
(ENERGY IN eV)

| Transition | Experiment ⁵⁴ | STO ³⁴ | GTO |
|--------------------------------------|--------------------------|-------------------|------|
| $\Gamma_{25}' - \Gamma_{15}$ | 7.3 | 6.3 | 6.3 |
| $\Sigma_2 - \Sigma_3$ $X_4 - X_1$ | 12.2 | 13.9 | 14.1 |
| $X_1 - X_1$ | 16 | 20.1 | 20.6 |
| $\Gamma_{25}' - \Gamma_1$ | 23 | 20.2 | 20.2 |

TABLE VIII
COMPARISON OF THE BAND STRUCTURE OF STO AND GTO
(ENERGY IN a.u.)

| K_x | Basis | 0.0 | 0.1 | 0.2 | 0.4 | 0.5 | 0.6 | 0.75 | 0.9 | 1.0 |
|------------------|-------|--------|--------|--------|--------|--------|--------|--------|--------|--------|
| $\Delta_1^{(1)}$ | STO | -1.243 | -1.240 | -1.230 | -1.189 | -1.160 | -1.124 | -1.062 | -0.990 | -0.938 |
| | GTO | -1.238 | -1.235 | -1.225 | -1.185 | -1.156 | -1.122 | -1.061 | -0.991 | -0.941 |
| $\Delta_5^{(1)}$ | STO | -0.503 | -0.514 | -0.542 | -0.608 | -0.637 | -0.662 | -0.690 | -0.704 | -0.707 |
| | GTO | -0.504 | -0.516 | -0.543 | -0.609 | -0.639 | -0.664 | -0.691 | -0.707 | -0.709 |
| $\Delta_2^{(1)}$ | STO | -0.503 | -0.511 | -0.534 | -0.612 | -0.661 | -0.714 | -0.799 | -0.883 | -0.938 |
| | GTO | -0.504 | -0.513 | -0.536 | -0.615 | -0.665 | -0.719 | -0.804 | -0.887 | -0.941 |
| $\Delta_1^{(2)}$ | STO | -0.273 | -0.272 | -0.271 | -0.278 | -0.281 | -0.279 | -0.261 | -0.225 | -0.192 |
| | GTO | -0.272 | -0.272 | -0.274 | -0.280 | -0.280 | -0.275 | -0.256 | -0.219 | -0.189 |
| $\Delta_5^{(2)}$ | STO | -0.273 | -0.257 | -0.219 | -0.115 | -0.062 | -0.012 | +0.049 | +0.085 | +0.092 |
| | GTO | -0.272 | -0.257 | -0.220 | -0.119 | -0.066 | -0.017 | +0.045 | +0.082 | +0.089 |
| $\Delta_2^{(2)}$ | STO | -0.009 | -0.005 | +0.003 | +0.009 | -0.002 | -0.026 | -0.084 | -0.151 | -0.192 |
| | GTO | -0.010 | -0.007 | +0.001 | +0.009 | -0.001 | -0.022 | -0.076 | -0.142 | -0.183 |

TABLE VIII (Continued)

| K_x | Basis | 0.0 | 0.1 | 0.2 | 0.4 | 0.5 | 0.6 | 0.75 | 0.9 | 1.0 |
|---------------|-------|--------|--------|--------|--------|--------|--------|--------|-----|-----|
| $\Sigma_1(1)$ | STO | -1.243 | -1.237 | -1.216 | -1.138 | -1.084 | -1.026 | -0.955 | | |
| | GTO | -1.238 | -1.231 | -1.211 | -1.136 | -1.085 | -1.029 | -0.961 | | |
| $\Sigma_3(1)$ | STO | -0.503 | -0.535 | -0.601 | -0.739 | -0.798 | -0.847 | -0.902 | | |
| | GTO | -0.504 | -0.536 | -0.603 | -0.742 | -0.802 | -0.851 | -0.906 | | |
| $\Sigma_1(2)$ | STO | -0.503 | -0.525 | -0.575 | -0.691 | -0.741 | -0.780 | -0.791 | | |
| | GTO | -0.504 | -0.527 | -0.577 | -0.694 | -0.743 | -0.780 | -0.791 | | |
| Σ_2 | STO | -0.503 | -0.507 | -0.521 | -0.571 | -0.604 | -0.637 | -0.678 | | |
| | GTO | -0.504 | -0.509 | -0.522 | -0.572 | -0.605 | -0.639 | -0.680 | | |
| Σ_4 | STO | -0.273 | -0.265 | -0.242 | -0.163 | -0.114 | -0.052 | +0.029 | | |
| | GTO | -0.272 | -0.265 | -0.244 | -0.166 | -0.113 | -0.056 | +0.024 | | |
| $\Sigma_1(3)$ | STO | -0.273 | -0.249 | -0.200 | -0.076 | -0.014 | +0.023 | -0.023 | | |
| | GTO | -0.272 | -0.251 | -0.204 | -0.077 | -0.014 | +0.022 | -0.022 | | |
| $\Sigma_3(2)$ | STO | -0.273 | -0.245 | -0.196 | -0.139 | -0.136 | -0.144 | -0.168 | | |
| | GTO | -0.272 | -0.247 | -0.199 | -0.142 | -0.136 | -0.142 | -0.163 | | |

TABLE VIII (Continued)

| K_x | Basis | 0.0 | 0.1 | 0.2 | 0.4 | 0.5 | 0.6 | 0.75 | 0.9 | 1.0 |
|----------------|-------|--------|---------|--------|--------|--------|--------|--------|-----|-----|
| $\Sigma_3(3)$ | STO | -0.009 | -0.001 | +0.017 | +0.057 | +0.068 | +0.076 | +0.086 | | |
| | GTO | -0.010 | -0.003 | +0.014 | +0.054 | +0.065 | +0.072 | +0.081 | | |
| $\Lambda_1(1)$ | STO | -1.243 | -1.233 | -1.203 | -1.093 | -1.052 | | | | |
| | GTO | -1.238 | -1.228 | -1.200 | -1.095 | -1.057 | | | | |
| $\Lambda_1(2)$ | STO | -0.503 | -0.557 | -0.661 | -0.872 | -0.927 | | | | |
| | GTO | -0.504 | -0.558 | -0.664 | -0.872 | -0.923 | | | | |
| $\Lambda_3(1)$ | STO | -0.503 | -0.518 | -0.547 | -0.592 | -0.598 | | | | |
| | GTO | -0.504 | -0.519 | -0.549 | -0.594 | -0.600 | | | | |
| $\Lambda_3(2)$ | STO | -0.273 | -0.251 | -0.206 | -0.139 | -0.132 | | | | |
| | GTO | -0.272 | -0.251 | -0.209 | -0.141 | -0.133 | | | | |
| $\Lambda_1(3)$ | STO | -0.273 | -0.234 | -0.183 | -0.137 | -0.132 | | | | |
| | GTO | -0.272 | -0.237 | -0.186 | -0.142 | -0.133 | | | | |
| $\Lambda_1(4)$ | STO | -0.009 | +0.003 | +0.028 | +0.125 | +0.148 | | | | |
| | GTO | -0.01 | +0.0001 | +0.032 | +0.132 | +0.153 | | | | |

CHAPTER X

THE ENERGY BAND GAP OF MAGNESIUM OXIDE

CRYSTAL STRUCTURE

The crystal structure of magnesium oxide (MgO) is same as that of rock-salt. It is composed of two interpenetrating face-centered cubic lattices. One fcc lattice consists of oxygen atoms while the other consists of magnesium atoms. These two fcc lattices are displaced by half lattice constant a_0 along the body diagonal. The a_0 for MgO is 4.21Å.

CRYSTAL POTENTIAL

The crystal potential adopted for this calculation is again the AHFS potential. The MgO lattice belongs to the point group O_h^5 . The positions of Mg and O atoms in the unit cell are given by

$$\vec{r}_1 = a_0(0, 0, 0) , \quad \vec{r}_2 = \frac{a_0}{2}(1, 1, 1) .$$

The crystal potential $V(\vec{r})$ again can be expanded in the reciprocal lattice with reciprocal lattice vector \vec{K}_ν ,

$$V(\vec{r}) = \sum_{\nu} V(\vec{K}_\nu) \exp(-i\vec{K}_\nu \cdot \vec{r}) .$$

Now we center the potential at either one of the atoms. The choice of the origin gives us a real energy matrix since the inversion symmetry is preserved. $V(\vec{K}_\nu)$ can be evaluated as,

$$\begin{aligned}
V(\vec{K}_\nu) &= \frac{1}{N\Omega} \sum_{\mu} \sum_{i=1}^2 \int V_i(\vec{r} - (\vec{R}_\mu + \vec{t}_i)) \exp(-i\vec{K}_\nu \cdot \vec{r}) d\tau \\
&= \frac{1}{N} \sum_{i=1}^2 \int V_i(\vec{r} - \vec{t}_i) \exp(-i\vec{K}_\nu \cdot \vec{r}) d\tau \\
&= \frac{1}{N} \sum_{i=1}^2 \int (V_i(\vec{r} - \vec{t}_i) \exp(-i(\vec{K}_\nu \cdot (\vec{r} - \vec{t}_i))) \exp(-i\vec{K}_\nu \cdot \vec{t}_i) d\tau \\
&= \frac{1}{N} \sum_{i=1}^2 \int V_i(\vec{r}) \exp(-i\vec{K}_\nu \cdot \vec{r}) d\tau \exp(-i\vec{K}_\nu \cdot \vec{t}_i) \\
&= \frac{1}{N} \left\{ \int V_1(\vec{r}) \cos \vec{K}_\nu \cdot \vec{r} d\tau + \cos \vec{K}_\nu \cdot \vec{t}_2 \int V_2(\vec{r}) \cos \vec{K}_\nu \cdot \vec{r} d\tau \right\}. \quad (10.1)
\end{aligned}$$

The potential now is divided into two kinds: the symmetric and anti-symmetric parts. The Fourier coefficients are:

$$\begin{aligned}
V^S(\vec{K}_\nu) &= \frac{1}{N} \left[\int V_1(\vec{r}) \cos \vec{K}_\nu \cdot \vec{r} d\tau + \int V_2(\vec{r}) \cos \vec{K}_\nu \cdot \vec{r} d\tau \right]; \\
&\text{for } K_x, K_y \text{ and } K_z \text{ all even} \quad (10.2)
\end{aligned}$$

$$\begin{aligned}
V^A(\vec{K}_\nu) &= \frac{1}{N} \left[\int V_1(\vec{r}) \cos \vec{K}_\nu \cdot \vec{r} d\tau - \int V_2(\vec{r}) \cos \vec{K}_\nu \cdot \vec{r} d\tau \right]; \\
&\text{for } K_x, K_y, \text{ and } K_z \text{ all odd.} \quad (10.3)
\end{aligned}$$

In the calculation, we take Mg as 1 and 0 as 2,

therefore

$$\begin{aligned}
V^S(\vec{K}_\nu) &= V^{Mg}(\vec{K}_\nu) + V^0(\vec{K}_\nu) \\
V^A(\vec{K}_\nu) &= V^{Mg}(\vec{K}_\nu) - V^0(\vec{K}_\nu),
\end{aligned}$$

where

$$V^{Mg}(\vec{K}_\nu) = \frac{1}{N} \int V_1(\vec{r}) \cos \vec{K}_\nu \cdot \vec{r} d\tau \quad (10.4)$$

$$V^0(\vec{K}_\nu) = \frac{1}{N} \int V_2(\vec{r}) \cos \vec{K}_\nu \cdot \vec{r} d\tau. \quad (10.5)$$

The evaluation of these coefficients are the same as the case of diamond we tabulate the first fifteen $V^{Mg}(\vec{K}_\nu)$ and $V^0(\vec{K}_\nu)$ in table IX. The atomic wave functions used are given by Clementi.⁴⁷

60
WAVE FUNCTIONS

For the free oxygen atomic wave function, Stewart's small Gaussian expansion⁴⁸ is much preferable since it contains less number of Gaussians and each Gaussian is of moderate size as compared to the ones in Huzinaga's paper. The C_i 's and α_i 's are listed in Table X. For 3s and 3p states, we use Hosoya's⁴⁹ HF calculation of STO and fit them with the least square fitting.⁵⁰ The results are,

$$\psi_{3s} = \left(\frac{1}{4\pi}\right)^{\frac{1}{2}} \sum_i \beta_i \exp(-\alpha_i r^2) \quad (10.6)$$

$$\psi_{3p} = \left(\frac{3}{4\pi}\right)^{\frac{1}{2}} \sum_i \beta_i \exp(-\alpha_i r^2) . \quad (10.7)$$

The various β_i 's and α_i 's are tabulated in Table XI.

For magnesium, the 1s, 2s, 3s, 2p wave functions are taken from Veillard's calculations.³⁶ The C_i 's and α_i 's are tabulated in Table XII. The 3p state are taken from the tabulated values by Trefftz and Bierman⁵¹ who used the HFSCF scheme to calculate the wave functions. The β_i 's and α_i 's are tabulated in Table XIII.

We therefore have all the free atomic wave functions at our disposal to construct the basis functions for the crystal Hamiltonian — the Bloch sum.

The Bloch functions are constructed in the usual manner as

$$b_{nlm}(\vec{k}, \vec{r}) = [N\Omega(\vec{k})]^{-\frac{1}{2}} \sum_j \exp(-i\vec{k} \cdot \vec{R}_j) \times \psi_{nlm}(\vec{r} - \vec{R}_j) . \quad (10.8)$$

The energy matrix is set up by 1s, 2s, 3s, 2p, 3p functions of both magnesium and oxygen and all the multicenter integrals are evaluated in the same manner as before. The same "cut-off" procedure with a

little revision is employed. The details are given in Appendix C. The 18x18 energy matrix is diagonalized at the Γ -point.

RESULT AND DISCUSSION

At the Γ -point, the energy levels splits into ten levels four of them are triplets. They are tabulated in Table XIV. We can identify Γ_A , Γ_B , Γ_C and Γ_D as the Mg 1s, oxygen 1s, Mg 2s, Mg 2p core states respectively. Γ_F corresponds to experimental value²⁹ at Γ_{15} and is the top of the valence band. Γ_G corresponds to their Γ_1 which is the bottom of conduction band. The band gap is 10.7 eV vs. the experimental value of 7.8 eV. The existence of Γ_E and Γ_H at their respective positions have not been confirmed by experimental data. Since there are no other first principle calculations that we know of at present time, no conclusion can be drawn about this point. It is suggested that a SCF approach be applied to this problem and further ascertain the exact positions of these levels. In view of the computer time involved, a complete band analysis should be followed only after the problem at Γ -point is resolved. Despite the difficulty cited here, the LCAO method does give fairly good agreement between the theoretical and experimental band gap values. This points out the fact that with an improved crystal potential, the LCAO method is able to yield better result.

TABLE IX
FOURIER COEFFICIENTS FOR $V_{\nu}^{Mg}(\vec{r})$ AND $V_{\nu}^0(\vec{r})$

| $\frac{a_0 \vec{k}}{2\pi} = \vec{l}$ | | | $V_{K\nu}^{Mg}$ | $V_{K\nu}^0$ |
|--------------------------------------|-------|-------|-----------------|--------------|
| l_x | l_y | l_z | | |
| 0 | 0 | 0 | -1.0000 | -1.0000 |
| 1 | 1 | 1 | -0.22965 | -0.19667 |
| 2 | 0 | 0 | -0.19500 | -0.16307 |
| 2 | 2 | 0 | -0.12893 | -0.10146 |
| 3 | 1 | 1 | -0.10421 | -0.080168 |
| 2 | 2 | 2 | -0.97981 | -0.074982 |
| 4 | 0 | 0 | -0.079098 | -0.059642 |
| 3 | 3 | 1 | -0.069162 | -0.051742 |
| 4 | 2 | 0 | -0.066400 | -0.049559 |
| 4 | 2 | 2 | -0.057348 | -0.042417 |
| 3 | 3 | 3 | -0.052114 | -0.038295 |
| 5 | 1 | 1 | -0.052114 | -0.038295 |
| 4 | 4 | 0 | -0.045385 | -0.032988 |
| 5 | 3 | 1 | -0.042205 | -0.030473 |
| 4 | 4 | 2 | -0.041252 | -0.029720 |

TABLE X
 C_i AND α_i OF OXYGEN

| $C_i(1s)$ | $C_i(2s)$ | $C_i(2p)$ | α_i |
|-------------|------------|-----------|------------|
| 0.137884 | -0.0297337 | | 149.926 |
| 0.481402 | -0.145386 | | 27.1311 |
| 0.502395 | -0.194778 | | 7.18802 |
| 0.0351062 | 0.618588 | | 0.91247 |
| -0.00978845 | 0.499792 | | 0.272406 |
| | | 0.0138204 | 41.6051 |
| | | 0.122175 | 7.40821 |
| | | 0.382355 | 1.87952 |
| | | 0.487390 | 0.56275 |
| | | 0.253299 | 0.177164 |

TABLE XI
 β_i AND α_i OF OXYGEN

| $\beta_i(3s)$ | $\beta_i(3p)$ | α_i |
|---------------|---------------|------------|
| -0.395933 | -0.00448594 | 0.177164 |
| -0.234483 | | 1.87952 |
| -0.686843 | 0.121193 | 0.56275 |
| 0.42302 | | 0.272406 |
| 0.386842 | | 7.18802 |
| 0.728111 | | 27.1311 |
| 0.051912 | -0.00706031 | 0.01506 |
| 0.295691 | | 0.91247 |
| 0.723929 | | 149.926 |
| 0.169199 | -0.00322285 | 0.004045 |
| | 0.401516 | 1.87952 |
| | 0.432280 | 7.40821 |
| | 0.295612 | 41.16051 |

TABLE XII
 C_i AND α_i OF Mg

| $C_i(1s)$ | $C_i(2s)$ | $C_i(3s)$ | $C_i(2p)$ | α_i |
|-----------|-----------|-----------|-----------|------------|
| 0.00038 | -0.00009 | 0.00001 | | 43643.6 |
| 0.00293 | -0.00073 | 0.00014 | | 6585.89 |
| 0.01496 | -0.00384 | 0.00074 | | 1510.32 |
| 0.05863 | -0.01514 | 0.00291 | | 431.833 |
| 0.17661 | -0.04979 | 0.00971 | | 142.071 |
| 0.37113 | -0.11832 | 0.002298 | | 51.4089 |
| 0.40079 | -0.20111 | 0.04123 | | 19.9361 |
| 0.12384 | -0.02310 | 0.00306 | | 8.04158 |
| 0.00375 | 0.57420 | -0.13501 | | 2.50948 |
| 0.00052 | 0.53517 | -0.25490 | | 0.871939 |
| -0.00003 | 0.02118 | 0.59531 | | 0.108819 |
| 0.00001 | 0.00801 | 0.52058 | | 0.040455 |
| | | | 0.00475 | 192.644 |
| | | | 0.03462 | 45.7493 |
| | | | 0.14372 | 14.2558 |
| | | | 0.34665 | 4.99324 |
| | | | 0.46489 | 1.78234 |
| | | | 0.24109 | 0.61510 |

TABLE XIII

 β_i AND α_i OF Mg

| $\beta_i(3p)$ | α_i |
|---------------|------------|
| -0.0039006 | 0.01506 |
| -0.0361927 | 0.040455 |
| -0.146854 | 0.177164 |
| 0.233841 | 0.272406 |
| -0.642646 | 0.6151 |
| 0.931061 | 0.871939 |
| -1.11535 | 1.78234 |
| 1.44632 | 2.50948 |
| -0.0480317 | 4.99324 |
| 1.67100 | 1.42558 |

TABLE XIV
 Γ -POINT ENERGY LEVELS OF MgO

| Γ | E in a.u. |
|----------------------|-----------|
| Γ_A (singlet) | -48.758 |
| Γ_B (singlet) | -20.865 |
| Γ_C (singlet) | - 4.20029 |
| Γ_D (triplet) | - 2.97507 |
| Γ_E (singlet) | - 2.12606 |
| Γ_F (triplet) | - 2.05207 |
| Γ_G (singlet) | - 1.64585 |
| Γ_H (triplet) | - 1.54222 |
| Γ_I (triplet) | - 5.37088 |
| Γ_J (singlet) | 2.67848 |

APPENDIX A

DERIVATION OF $S_2(b)$ FOR DIPOLE-DIPOLE INTERACTION

For dipole-dipole interaction, the $S_2(b)_{o,i}$ term is given by

$$S_2(b)_{o,i} = \frac{4}{9} \left(\frac{p_1 p_2}{\hbar v} \right)^2 \frac{1}{b^4} \sum_{j_1, j_2} (j_1 1 K_1 0 | j_1, K_1)^2 \times (j_2 1 K_2 0 | j_2, K_2)^2 f_1(k) \quad (A.1)$$

according to the selection rule

$$\begin{array}{ccc} j_1 + 1 & & j_2 + 1 \\ j_1' = j_1 & \text{and} & j_2' = j_2 \\ j_1 - 1 & & j_2 - 1 \end{array} \quad (A.2)$$

Therefore we have nine different matrix elements; namely,

$$\begin{array}{cc} (j_1 1 K_1 0 | j_1 + 1 K_1)^2 & (j_2 1 K_2 0 | j_2 + 1 K_2)^2 \\ \times & \\ (j_1 1 K_1 0 | j_1 K_1)^2 & (j_2 1 K_2 0 | j_2 K_2)^2 \\ (j_1 1 K_1 0 | j_1 - 1 K_1)^2 & (j_2 1 K_2 0 | j_2 - 1 K_2)^2 \end{array}$$

using the table for C - G coefficients,²⁰ these nine different coefficients can be evaluated. $f_1(k)$ is defined in (2.26) with

$$k = \frac{2b}{v} B [j_1(j_1' + 1) - j_1(j_1 + 1) + j_2' + 1 - j_2(j_2 + 1)].$$

Define $x = \frac{4\pi b B}{v}$ and $k = xD$,

the transitions are classified as:

for $\Delta j_1 = 0$, $\Delta j_2 = 0$, $D = 0$,

$$\begin{aligned}
 S_2(b)_{o,i} &= (j_1 1 K_1 0 | j_1 K_1)^2 (j_2 1 K_2 0 | j_2 K_2)^2 \\
 &= \frac{K_1^2}{j_1(j_1+1)} \frac{K_2^2}{j_2(j_2+1)} \\
 S_2(b)_{o,f} &= \frac{K_1^2}{(j_1+1)(j_1+2)} \frac{K_2^2}{j_2(j_2+1)} \\
 S_2(b)_m &= (-1)^{j_1+j_f+1} \frac{8}{9} \left(\frac{P_1 P_2}{\hbar v} \right)^2 \frac{1}{b^4} \\
 &\quad \times \sqrt{(2j_1+1)(2j_f+1)} (j_1 1 K_1 0 | j_1 K_1) \\
 &\quad \times (j_f 1 K_f 0 | j_f K_f) W(j_1 j_f j_1 j_f, 11) \\
 &\quad \sum_{j_2} (j_2 1 K_2 0 | j_2 K_2)^2 f_k(k)
 \end{aligned}$$

where $W(j_1 j_f j_1 j_f, 11)$ is the Racah coefficients the value of it can be found in reference 21. In our case

$$W(abcd; if) \quad \text{with } f = 1 \text{ and } a = b - 1, c = d - 1:$$

$$\begin{aligned}
 W(abcd; 11) &= (-1)^{b+d-f} \\
 &\quad \times \left[\frac{(f+b+d+1)(f+b+d)(-f+b+d)(-f+b+d-1)}{4(ab+1)b(2b-1)(2d+1)d(ad-1)} \right]^{\frac{1}{2}}.
 \end{aligned}$$

We see that $S_2(b)_m$ exists only when $\Delta j_1 = 0$, i.e., only diagonal matrix elements exist. The $S_2(b)_m$ in this case is given by:

$$S_2(b)_m = -\frac{4}{9} \left(\frac{P_1 P_2}{\hbar v} \right)^2 \frac{1}{b^4} \frac{K_1 K_f}{(j_1+1)^2}.$$

Therefore for $\Delta J_1 = 0$, $\Delta J_2 = 0$,

$$\begin{aligned}
 S(b) &= \frac{4}{9} \left(\frac{P_1 P_2}{\hbar v} \right)^2 \frac{1}{b^4} \frac{K_2^2}{j_2(j_2+1)} \left[\frac{K_1^2}{j_1(j_1+1)} \right. \\
 &\quad \left. + \frac{K_1^2}{(j_1+1)(j_1+2)} \frac{K_1^2}{(j_1+1)^2} \right].
 \end{aligned}$$

$$\Delta J_1 = 1, \Delta J_2 = -1, D = J_1 - J_2 + 1,$$

$$\begin{aligned} S_{inel_1}(b)_{o,i} &= \frac{4}{9} \left(\frac{P_1 P_2}{\hbar v} \right)^2 \frac{1}{b^4} (j_1 1 K_1 0 | j_1 + 1 K_1)^2 \\ &\quad \times (j_2 1 K_2 0 | j_2 - 1 K_2)^2 f_1 \{ x(J_1 - J_2 + 1) \} \\ &= \frac{4}{9} \left(\frac{P_1 P_2}{\hbar v} \right)^2 \frac{1}{b^4} \left\{ \frac{J_2^2 - K_2^2}{J_2(2J_2 + 1)} \right. \\ &\quad \left. \frac{(J_1 + 1)^2 - K_1^2}{(2J_1 + 1)(J_1 + 1)} f_1 \{ x(J_1 - J_2 + 1) \} \right\} \\ S_{inel_1}(b)_{o,f} &= \frac{4}{9} \left(\frac{P_1 P_2}{\hbar v} \right)^2 \frac{1}{b^4} \left\{ \frac{K_2^2 - K_2^2}{J_2(2J_2 + 1)} \right. \\ &\quad \left. \frac{(J_1 + 2)^2 - K_1^2}{(2J_1 + 1)(J_1 + 1)} f_1 \{ x(J_1 - J_2 + 2) \} \right\} \end{aligned}$$

$$\Delta J_1 = -1, \Delta J_2 = 1, D = J_2 - J_1 + 1,$$

$$\begin{aligned} S_{inel_2}(b)_{o,i} &= \frac{4}{9} \left(\frac{P_1 P_2}{\hbar v} \right)^2 \frac{1}{b^4} (j_1 1 K_1 0 | j_1 - 1 K_1)^2 \\ &\quad \times (j_2 1 K_2 0 | j_2 + 1 K_2)^2 f_1 \{ x(J_2 - J_1 + 1) \} \\ &= \frac{4}{9} \left(\frac{P_1 P_2}{\hbar v} \right)^2 \frac{1}{b^4} \left\{ \frac{(J_1 - K_1)(J_1 + K_1)}{J_1(2J_1 + 1)} \right. \\ &\quad \left. \times \frac{(J_2 + 1)^2 - K_1^2}{(2J_1 + 1)(J_1 + 1)} \times f_1 \{ x(J_2 - J_1 + 1) \} \right\} \\ S_{inel_2}(b)_{o,f} &= \frac{4}{9} \left(\frac{P_1 P_2}{\hbar v} \right)^2 \frac{1}{b^4} \left\{ \frac{(J_1 + 1)^2 - K_1^2}{(J_1 + 1)(2J_1 + 3)} \right. \\ &\quad \left. \times \frac{(J_2 + 1)^2 - K_1^2}{(2J_1 + 1)(J_1 + 1)} f_1 \{ x(J_2 - J_1) \} \right\} \end{aligned}$$

$$\begin{aligned}
S_{inel} = & \frac{4}{9} \left(\frac{p_1 p_2}{\hbar v} \right)^2 \frac{1}{b^4} \left\{ \frac{(J_2 - K_2)(J_2 + K_2)}{J_2(2J_2 + 1)} \right. \\
& \left[\frac{(J_1 + 1)^2 - K_1^2}{(2J_1 + 1)(J_1 + 1)} f_1 \{x(J_1 - J_2 + 1)\} \right. \\
& + \left. \frac{(J_1 + 2)^2 - K_1^2}{(2J_1 + 3)(J_1 + 2)} f_1 \{x(J_1 - J_2 + 2)\} \right] \\
& + \frac{(J_2 + 1)^2 - K_2^2}{(2J_2 + 1)(J_2 + 1)} \left[\frac{J_1^2 - K_1^2}{J_1(2J_1 + 1)} f_1 \{x(J_2 - J_1 + 1)\} \right. \\
& + \left. \left. \frac{(J_1 + 1)^2 - K_1^2}{(J_1 + 1)(2J_1 + 3)} f_1 \{x(J_2 - J_1 + 1)\} \right] \right\}
\end{aligned}$$

$$\Delta J_1 = 1 \quad \Delta J_2 = 0 \quad D = J_1 + 1$$

$$\begin{aligned}
S'_{inel_1}(b)_{0,1} = & \frac{4}{9} \left(\frac{p_1 p_2}{\hbar v} \right)^2 \frac{1}{b^4} (j_1 1 K_1 0 | j_1 + 1 \quad K_1)^2 \\
& (j_2 1 K_2 0 | j_2 K_2)^2 f_1 \{x(J_1 + 1)\} \\
= & \frac{4}{9} \left(\frac{p_1 p_2}{\hbar v} \right)^2 \frac{1}{b^4} \frac{K_2^2}{J(J + 1)} \\
& \frac{(J_1 + 2)^2 - K_1^2}{(2J_1 + 3)(J_1 + 2)} f_1 \{x(J_1 + 2)\}
\end{aligned}$$

$$\Delta J_1 = -1, \quad \Delta J_2 = 0, \quad D = J_1,$$

$$\begin{aligned}
S'_{inel_2}(b)_{0,1} = & \frac{4}{9} \left(\frac{p_1 p_2}{\hbar v} \right)^2 \frac{1}{b^4} (j_1 1 K_1 0 | j_1 - 1 K_1)^2 \\
& (j_2 1 K_2 0 | j_2 K_2)^2 f_1 \{xJ_1\} \\
= & \frac{4}{9} \left(\frac{p_1 p_2}{\hbar v} \right)^2 \frac{1}{b^4} \frac{K_2^2}{J_2(J_2 + 1)} \frac{J_1^2 - K_1^2}{J_1(2J_1 + 1)} \\
& f_1(xJ_1)
\end{aligned}$$

$$S'_{inel_2}(b)_{o,f} = \frac{4}{9} \left(\frac{p_1 p_2}{\hbar v} \right)^2 \frac{1}{b^4} \frac{K_2^2}{J_2(J_2+1)}$$

$$\frac{(J_1+1)^2 - K_1^2}{(J_1+1)(2J_1+3)} f_1(x(J_1+1))$$

$$\Delta J_1 = 0, \Delta J_2 = 1, D = J_2 + 1$$

$$S'_{inel_3}(b)_{o,i} = \frac{4}{9} \left(\frac{p_1 p_2}{\hbar v} \right)^2 \frac{1}{b^4} (j_1 1 K_1 0 | j_1 K_1)^2$$

$$(j_2 1 K_2 0 | j_2 + 1 K_2) f_1(x(J_2+1))$$

$$S'_{inel_3}(b)_m = (-1)^{j_i+j_f+1} \frac{8}{9} \left(\frac{p_1 p_2}{\hbar v} \right)^2 \frac{1}{b^4}$$

$$\times \sqrt{(2J_1+1)(2J_f+1)} (j_1 1 K_1 0 | j_1 K_1)$$

$$\times (j_f 1 K_f 0 | j_f K_f) W(j_1 j_f j_1 j_f, 11)$$

$$\times (j_2 1 K_2 0 | j_2 + 1 K_2)^2 f_1(x(J_2+1))$$

$$S'_{inel_3}(b)_{o,i} = \frac{4}{9} \left(\frac{p_1 p_2}{\hbar v} \right)^2 \frac{1}{b^4} \frac{(J_2+1)^2 - K_2^2}{(2J_2+1)(J_2+1)}$$

$$\frac{K_1^2}{J_1(J_1+1)} f_1(x(J_2+1))$$

$$S'_{inel_3}(b)_{o,f} = \frac{4}{9} \left(\frac{p_1 p_2}{\hbar v} \right)^2 \frac{1}{b^4} \frac{(J_2+1)^2 - K_2^2}{(2J_2+1)(J_2+1)}$$

$$\frac{K_1^2}{(J_1+1)(J_1+2)} f_1(x(J_2+1))$$

$$S'_{inel_3}(b)_m = \frac{4}{9} \left(\frac{p_1 p_2}{\hbar v} \right)^2 \frac{1}{b^4} (-1) \frac{2K_1^2}{(J_1+1)^2}$$

$$\frac{(J_2+1)^2 - K_2^2}{(2J_2+1)(J_2+1)} f(x(J_2+1))$$

$$S'_{inel_3}(b) = \frac{4}{9} \left(\frac{p_1 p_2}{\hbar v} \right)^2 \frac{1}{b^4} \frac{(J_2+1)^2 - K_2^2}{(2J_2+1)(J_2+1)} \left(\frac{K_1^2}{J_1(J_1+1)} \right. \\ \left. + \frac{K_1^2}{(J_1+1)(J_1+2)} - \frac{2K_1^2}{(J_1+1)^2} \right) f_1(x(J_2+1))$$

$$\Delta J_1 = 0, \quad \Delta J_2 = -1, \quad D = J_2,$$

$$S'_{inel_4}(b)_{o,i} = \frac{4}{9} \left(\frac{p_1 p_2}{\hbar v} \right)^2 \frac{1}{b^4} (j_1 1 K_2 0 | j_1 K_1)^2 \\ \times (j_2 1 K_2 0 | j_2 -1 K_2)^2 f(x J_2)$$

$$S'_{inel_4}(b)_m = (-1)^{j_1+j_f+1} \frac{8}{9} \left(\frac{p_1 p_2}{\hbar v} \right)^2 \frac{1}{b^4} \\ \times \sqrt{(2J_1+1)(2J_f+1)} (j_1 1 K_1 0 | j_1 K_1) \\ \times (j_f 1 K_f 0 | j_f K_f) W(j_1 j_f j_1 j_f, 11) \\ \times (j_2 1 K_2 0 | j_2 -1 K_2)^2 f(x J_2)$$

$$S'_{inel_4}(b)_{o,i} = \frac{4}{9} \left(\frac{p_1 p_2}{\hbar v} \right)^2 \frac{1}{b^4} \frac{J_2^2 - K_2^2}{J_2(2J_2+1)} \\ \times \frac{K_1^2}{J_1(J_1+1)} f_1(x J_2)$$

$$S'_{inel_4}(b)_{o,f} = \frac{4}{9} \left(\frac{p_1 p_2}{\hbar v} \right)^2 \frac{1}{b^4} \frac{J_2^2 - K_2^2}{J_2(2J_2+1)} \\ \times \frac{K_1^2}{(J_1+1)(J_1+2)} f(x J_2)$$

$$S'_{inel_4}(b)_m = \frac{4}{9} \left(\frac{p_1 p_2}{\hbar v} \right)^2 \frac{1}{b^4} \frac{J_2^2 - K_2^2}{J_2(2J_2+1)} \\ (-1) \frac{2K_1^2}{(J_1+1)^2} f_1(x J_2)$$

$$\begin{aligned}
s'_{inel_4}(b) &= \frac{4}{9} \left(\frac{p_1 p_2}{\hbar v} \right)^2 \frac{1}{b^4} \frac{J_2^2 - K_2^2}{J_2(2J_2+1)} \frac{K_1^2}{J_1(J_1+1)} \\
&\quad + \frac{K_1^2}{(J_1+1)(J_1+2)} - \frac{2K_1^2}{(J_1+1)^2} f_1(xJ_2) \\
s'_{inel} &= \frac{4}{9} \left(\frac{p_1 p_2}{\hbar v} \right)^2 \frac{1}{b^4} \left[\frac{k_2}{J_2(J_2+1)} \left\{ \frac{(J_1+1)^2 - K_1^2}{2J_1+1} \frac{f_1(x(J_1+1))}{(J_1+1)} \right. \right. \\
&\quad \left. \left. + \frac{(J_1+2)^2 - K_1^2}{(2J_1+3)(J_1+2)} f_1(x(J_1+2)) \right. \right. \\
&\quad \left. \left. + \frac{J_1^2 - K_1^2}{J_1(2J_1+1)} f_1(xJ_1) + \frac{(J_1+1)^2 - K_1^2}{(J_1+1)(2J_1+3)} f_1(x(J_1+1)) \right\} \right. \\
&\quad \left. + \frac{(J_2+1)^2 - K_2^2}{(2J_2+1)(J_2+1)} \left\{ \frac{K_1^2}{J_1(J_1+1)} + \frac{K_1^2}{(J_1+1)(J_1+2)} \right. \right. \\
&\quad \left. \left. - \frac{2K_1^2}{(J_1+1)^2} f_1(x(J_2+1)) + \frac{J_2^2 - K_2^2}{J_2(2J_2+1)} \frac{K_1^2}{J_1(J_1+1)} \right. \right. \\
&\quad \left. \left. + \frac{K_1^2}{(J_1+1)(J_1+2)} - \frac{2K_1^2}{(J_1+1)^2} \right\} f_1(xJ_2) \right]
\end{aligned}$$

$$\Delta J_1 = 1, \Delta J_2 = 1, \quad D = J_1 + J_2 + 2,$$

$$\begin{aligned}
s''_{inel_1}(b)_{o,1} &= \frac{4}{9} \left(\frac{p_1 p_2}{\hbar v} \right)^2 \frac{1}{b^4} (j_1 1 K_1 0 | j_1 + 1 K_1)^2 \\
&\quad \times (j_2 1 K_2 0 | j_2 + 1 K_2)^2 f_1(x(J_1+J_2+2)) \\
&= \frac{4}{9} \left(\frac{p_1 p_2}{\hbar v} \right)^2 \frac{(J_1+1)^2 - K_1^2}{(2J_1+1)(J_1+1)} \\
&\quad \times \frac{(J_2+1)^2 - K_2^2}{(2J_2+1)(J_2+1)} f_1(x(J_1+J_2+2))
\end{aligned}$$

$$S''_{inel_2(b)_{o,f}} = \frac{4}{9} \left(\frac{p_1 p_2}{\hbar v} \right)^2 \frac{1}{b^4} \frac{(J_1+2)^2 - K_1^2}{(2J_1+3)(J_1+2)} \\ \times \frac{(J_2+1)^2 - K_2^2}{(2J_2+1)(J_2+1)} f_1(x(J_1+J_2+3))$$

$$\Delta J_1 = -1, \Delta J_2 = -1, \quad D = J_1 + J_2$$

$$S''_{inel_2(b)_{o,i}} = \frac{4}{9} \left(\frac{p_1 p_2}{\hbar v} \right)^2 \frac{1}{b^4} (j_1 1 K_1 0 | j_1 - 1 K_1)^2 \\ (j_2 1 K_2 0 | j_2 - 1 K_2)^2 f_1(x(J_1+J_2)) \\ = \frac{4}{9} \left(\frac{p_1 p_2}{\hbar v} \right)^2 \frac{1}{b^4} \frac{J_1^2 - K_1^2}{J_1(2J_1+1)}$$

$$\frac{J_2^2 - K_2^2}{J_2(2J_2+1)} f_1(x(J_1+J_2))$$

$$S''_{inel_2(b)_{o,i}} = \frac{4}{9} \left(\frac{p_1 p_2}{\hbar v} \right)^2 \frac{1}{b^4} \frac{(J_1+1)^2 - K_1^2}{(J_1+1)(2J_1+3)}$$

$$\frac{J_2^2 - K_2^2}{J_2(2J_2+1)} f_1(x(J_1+J_2+1))$$

$$S''_{inel(b)} = \frac{4}{9} \left(\frac{p_1 p_2}{\hbar v} \right)^2 \frac{1}{b^4} \left[\frac{(J_2+1)^2 - K_2^2}{(2J_2+1)(J_2+1)} \right.$$

$$\left\{ \frac{(J_1+1)^2 - K_1^2}{(2J_1+1)(J_1+1)} f_1(x(J_1+J_2+2)) \right.$$

$$\left. + \frac{(J_1+2)^2 - K_1^2}{(2J_1+3)(J_1+2)} f_1(x(J_1+J_2+3)) \right\}$$

$$+ \frac{J_2^2 - K_2^2}{J_2(2J_2+1)} \left\{ \frac{J_1^2 - K_1^2}{J_1(2J_1+1)} f_1(x(J_1+J_2)) \right.$$

$$\left. + \frac{(J_1+1)^2 - K_1^2}{(J_1+1)(2J_1+3)} f_1(x(J_1+J_2+1)) \right\} \Big] .$$

APPENDIX B

THE FORMULATION OF AHFS CRYSTAL POTENTIAL

The crystal potential can be considered as a superposition of atomic potentials. It can be divided into two parts: the Coulomb part and the exchange part. For the Coulomb part

$$v_{\text{cry}}^{\text{Coul}}(\vec{r}) = \sum_{\vec{R}_\nu} \sum_{i=1}^2 v_{\text{atomic}}^{\text{Coul}}[\vec{r} - (\vec{R}_\nu + \vec{t}_i)], \quad (\text{B.1})$$

where

$$v_{\text{atomic}}^{\text{Coul}}(\vec{r}) = -\frac{Z}{r} + \frac{4\pi}{r} \int_0^r \rho(r') r'^2 dr' + 4\pi \int_r^\infty (r') r' dr'. \quad (\text{B.2})$$

$v_{\text{cry}}^{\text{Coul}}(\vec{r})$ can be expanded in a Fourier series

$$v_{\text{cry}}^{\text{Coul}}(\vec{r}) = \sum_{\vec{K}_\nu} v_{\text{cry}}^{\text{Coul}}(\vec{K}_\nu) \exp(i\vec{K}_\nu \cdot \vec{r}) \quad (\text{B.3})$$

and $v_{\text{cry}}^{\text{Coul}}(\vec{K}_\nu)$ is defined as

$$v_{\text{cry}}^{\text{Coul}}(\vec{K}_\nu) = \frac{1}{N\Omega} \int v_{\text{cry}}^{\text{Coul}}(\vec{r}) \exp(-i\vec{K}_\nu \cdot \vec{r}) d\tau. \quad (\text{B.4})$$

Upon substituting (B.3) into (B.4)

$$\begin{aligned} v_{\text{cry}}^{\text{Coul}}(\vec{K}_\nu) &= \frac{1}{N\Omega} \sum_{\vec{R}_\nu} \sum_{i=1}^2 \int v_{\text{atomic}}^{\text{Coul}}[\vec{r} - (\vec{R}_\nu + \vec{t}_i)] \exp(-i\vec{K}_\nu \cdot \vec{r}) d\tau \\ &= \frac{2\cos\vec{K} \cdot \vec{t}}{2} \int v_{\text{atomic}}^{\text{Coul}}(\vec{r}) \exp(-i\vec{K}_\nu \cdot \vec{r}) d\tau \end{aligned} \quad (\text{B.5})$$

where $\vec{t} = \vec{t}_2 = -\vec{t}_1$.

Since the choice of the origin of the crystal potential is such that it is invariant under inversion equation (B.5) can be

written as

$$V_{\text{cry}}^{\text{Coul}}(\vec{K}_\nu) = \frac{2\cos\vec{K}_\nu \cdot \vec{t}}{\Omega} \int V_{\text{atomic}}^{\text{Coul}}(\vec{r}) \cos\vec{K}_\nu \cdot \vec{r} d\tau, \quad (\text{B.6})$$

$$\begin{aligned} \int V_{\text{atomic}}^{\text{Coul}}(\vec{r}) \cos\vec{K}_\nu \cdot \vec{r} d\tau &= \int_0^\infty V_{\text{atomic}}^{\text{Coul}}(r) r^2 dr \\ &\quad \times \int_0^\pi \cos(K_\nu r \cos\theta) \sin\theta d\theta \int_0^{2\pi} d\phi \\ &= \frac{4\pi}{K_\nu} \int_0^\infty r V_{\text{atomic}}^{\text{Coul}}(r) \sin(K_\nu r) dr. \end{aligned}$$

Therefore

$$V_{\text{cry}}^{\text{Coul}}(\vec{K}_\nu) = \frac{8\pi\cos\vec{K}_\nu \cdot \vec{t}}{K_\nu \Omega} \int_0^\infty r V_{\text{atomic}}^{\text{Coul}}(r) \sin(K_\nu r) dr. \quad (\text{B.7})$$

Substituting (B.2) into (B.7) and effecting the integration yields

$$V_{\text{cry}}^{\text{Coul}}(\vec{K}_\nu) = \frac{-8\pi\cos\vec{K}_\nu \cdot \vec{t}}{K_\nu^2 \Omega} \left\{ Z - 4\pi \int_0^\infty r^2 \rho(r) \frac{\sin K_\nu r}{K_\nu r} dr \right\}. \quad (\text{B.8})$$

For the exchange part, we use the standard Slater exchange approximation

$$V_{\text{atomic}}^{\text{exch}}(r) = -3 \left[\frac{3}{8\pi} \rho(r) \right]^{\frac{1}{3}}. \quad (\text{B.9})$$

$$V_{\text{cry}}^{\text{exch}}(\vec{r}) = \sum_{\vec{R}} \sum_{i=1}^2 V_{\text{atomic}}^{\text{exch}}[\vec{r} - (\vec{R}_i + \vec{t}_i)]$$

$$\begin{aligned} V_{\text{cry}}^{\text{exch}}(\vec{K}_\nu) &= \frac{1}{N\Omega} \int V_{\text{cry}}^{\text{exch}}(\vec{r}) \exp(-i\vec{K}_\nu \cdot \vec{r}) d\tau \\ &= \frac{8\pi\cos\vec{K}_\nu \cdot \vec{t}}{K_\nu} \left(\frac{3}{8\pi} \right)^{\frac{1}{3}} \int_0^\infty r [\rho(r)]^{\frac{1}{3}} \\ &\quad \times \sin(K_\nu r) dr. \end{aligned} \quad (\text{B.10})$$

The crystal potential is a sum of the Coulomb and exchange terms,

$$\begin{aligned}
 v_{\text{cry}}(\vec{r}) &= v_{\text{cry}}^{\text{exch}}(\vec{r}) + v_{\text{cry}}^{\text{Coul}}(\vec{r}) \\
 &= \sum_{\vec{K}_y} v_{\text{cry}}(\vec{K}_y) \cos \vec{K}_y \cdot \vec{r} .
 \end{aligned}
 \tag{B.11}$$

and we obtain

$$\begin{aligned}
 v_{\text{cry}}(\vec{K}_y) &= v_{\text{cry}}^{\text{Coul}}(\vec{K}_y) + v_{\text{cry}}^{\text{exch}}(\vec{K}_y) \\
 &= -\frac{8\pi}{K_y^2 \Omega} \cos \vec{K}_y \cdot \vec{t} \left\{ Z - \frac{1}{K_y} \int_0^\infty Q(r) \sin K_y r dr \right. \\
 &\quad \left. + K_y \int_0^\infty E(r) \sin K_y r dr \right\} ,
 \end{aligned}
 \tag{B.12}$$

where

$$\begin{aligned}
 Q(r) &= 4\pi r \rho(r) , \quad \frac{1}{3} \\
 E(r) &= \frac{3}{2} [3\rho(r)/\pi]^{1/3} .
 \end{aligned}$$

APPENDIX C

THE EWALD-TYPE POTENTIAL EXPANSION

In chapter IX, we divide the crystal potential into two parts,

$$V(\vec{r}) = V_1(\vec{r}) + [V(\vec{r}) - V_1(\vec{r})] ,$$

$$V_1(\vec{r}) = \sum_{\nu} \mathcal{V}(\vec{r} - \vec{R}_{\nu}) ,$$

where the requirement for $\mathcal{V}(r)$ is such that it behaves like Z/r near the origin. The part $V_2(\vec{r}) = V(\vec{r}) - V_1(\vec{r})$ is in effect a "cut-off" potential with the singularity cut off. Therefore $V_2(\vec{r})$ is a relatively smooth function of r and consequently requires less number of Fourier coefficients to represent it. As shown in the calculation of lithium band structure by Gaussian basis functions³⁷

we can approximate

$$\begin{aligned} & \langle G(\alpha_1, \vec{r} - \vec{A}) | V_1(\vec{r}) | G(\alpha_2, \vec{r} - \vec{B}) \rangle \\ & \approx \langle G(\alpha_1, \vec{r} - \vec{A}) | \mathcal{V}(\vec{r} - \vec{A}) | G(\alpha_2, \vec{r} - \vec{A}) \rangle . \end{aligned}$$

In the process of subtracting $V_1(\vec{r})$ and form $V_2(\vec{r})$, we have

$$\begin{aligned} V(\vec{K}_{\nu}) &= \frac{1}{N\Omega} \int V_1(\vec{r}) \exp(-i\vec{K}_{\nu} \cdot \vec{r}) d\tau \\ &= \frac{8\pi \cos \vec{K}_{\nu} \cdot \vec{t}}{K_{\nu} \Omega} \int_0^{\infty} r \mathcal{V}(r) \sin K_{\nu} r dr , \end{aligned} \quad (C.1)$$

assuming that $\mathcal{V}(r)$ can be written as

$$\begin{aligned} \mathcal{V}(r) &= \frac{1}{r} (Z + a_1 r + a_2 r^2 + \dots) \exp(-br^2) \\ &= \frac{1}{r} (Z + \sum_{i=1}^n a_i r^i) \exp(-br^2) \end{aligned}$$

where b is chosen with an appropriate magnitude such that $\mathcal{V}(r)$ remains well localized.

$$V_1(\vec{K}_\nu) = \frac{8\pi \cos \vec{K}_\nu \cdot \vec{t}}{K_\nu r} \left\{ Z \int_0^\infty \exp(-br^2) \sin K_\nu r dr \right. \\ \left. + \sum_{i=1}^n a_i \int_0^\infty r^i \exp(-br^2) \sin K_\nu r dr \right\} . \quad (C.2)$$

The integral of the type $\int_0^\infty x^{2\nu-2} e^{-dx^2} \sin(yx) dx$ is evaluated by using Hypergeometric function and Gamma function⁵² such that

$$\int_0^\infty x^{2\nu-2} \exp(-dx^2) \sin(yx) dx \\ = \frac{1}{2} d^{-\frac{\nu}{2}} \Gamma(\nu) y F_1\left(\nu; \frac{3}{2}; -\frac{1}{4} y^2 / d\right) . \quad (C.3)$$

In the case of diamond, we use

$$\mathcal{V}(r) = -\left(\frac{Z}{r}\right) (1 + k r^2) \exp(-k r^2)$$

with $k = 2.5$.

In the case of MgO, we obtain the "cut-off" potentials for both magnesium and oxygen and then form the symmetric and antisymmetric parts of the crystal potential. In MgO, instead of fixing the values of b and a_i 's, we let the high Fourier coefficients be the $V_1(\vec{K}_\nu)$ and using

$$V(r) = \frac{1}{r} \left\{ \alpha \exp(-\beta_1 r^2) - (Z + d) \exp(-\beta_2 r^2) \right\} . \quad (C.4)$$

The α and β 's are not pre-determined but were fitted by a least-square curve-fitting such that $V_1(\vec{K}_\nu)$ will be the same as the tail part of $V(\vec{K}_\nu)$. The values of α , β_1 , and β_2 for magnesium are

$$\alpha = -14.7853, \quad \beta_1 = 19.3866, \quad \beta_2 = 60.3715,$$

and for oxygen are

$$\alpha = -9.94075, \quad \beta_1 = 16.8809, \quad \beta_2 = 66.8269 .$$

We list in table XV and table XVI the comparison of $V_2(\vec{K}_\nu)$ and $V(\vec{K}_\nu)$ which we call V_{K_ν} and V_{2K_ν} .

To evaluate the integral

$$\langle G^S(\alpha_1, \vec{r}_A) | \mathcal{V}(\vec{r}_A) | G^S(\alpha_2, \vec{r}_B) \rangle \quad \text{for } \alpha_1 \gg \alpha_2$$

we use the scheme of expanding $\exp(-\alpha_2 r_B^2)$ about the point A in Taylor series and then integrate the series until it reaches convergence. For the $-Z/r_A$ term, the integral has the form

$$I = - \int \exp(-\alpha_1 r_A^2) (Z/r_A) \exp(-\alpha_2 r_B^2) d\tau.$$

Since

$$\vec{r}_B = \vec{r}_A - \vec{AB},$$

$$\begin{aligned} \text{so } r_B^2 &= r_A^2 + \overline{AB}^2 - 2\vec{r}_A \cdot \vec{AB} \\ &= r_A^2 + \overline{AB}^2 - 2r_A \overline{AB} \cos \theta, \end{aligned}$$

where θ is the angle between \vec{r}_A and \vec{r}_B . We can therefore write

$$\exp(-\alpha_2 r_B^2) = \exp(-\alpha_2 (r_A^2 + \overline{AB}^2)) \exp(2 r_A \overline{AB} \cos \theta)$$

and expand the second exponent as

$$\exp(2\alpha_2 r_A \overline{AB} \cos \theta) \simeq 1 + \sum_n \frac{(2\alpha_2 r_A \overline{AB} \cos \theta)^n}{n!}.$$

The integral now becomes

$$\begin{aligned} I &= -Z \exp(-\alpha_2 \overline{AB}^2) \int r_A \exp(-(\alpha_1 + \alpha_2) r_A^2) \\ &\quad \times \sum_{n=0}^{\infty} \frac{(2\alpha_2 r_A \overline{AB} \cos \theta)^n}{n!} \sin \theta d\theta d\phi dr \\ &= -2\pi Z \exp(-\alpha_2 \overline{AB}^2) \sum_{n=0}^{\infty} \left(\frac{2\alpha_2 \overline{AB}}{n!} \right)^n \\ &\quad \times \left[\int_0^{\infty} r_A^{n+1} \exp(-(\alpha_1 + \alpha_2) r_A^2) dr \right] \\ &\quad \times \left[\int_0^{\pi} \cos^n \theta \sin \theta d\theta \right] \\ &= -4\pi Z \exp(-\alpha_2 \overline{AB}^2) \sum_{n=0}^{\infty} \frac{(2\alpha_2 \overline{AB})^{2n}}{(2n+1)!} \int_0^{\infty} r_A^{2n+1} \exp(-(\alpha_1 + \alpha_2) r_A^2) dr_A. \end{aligned}$$

The radial part of the integral is of the form

$$\int_0^{\infty} r^{2n+1} \exp(-\lambda r^2) dr = \frac{n!}{2\lambda^{n+1}}.$$

If the purpose is to evaluate $(-Z/r)\psi r^2$ then the only modification will be

$$\int_0^{\infty} r^{2(n+1)+1} \exp(-\lambda r^2) dr = \frac{(n+1)!}{2\lambda^{n+2}}.$$

Therefore, the result for $-Z/r$ can also be used for $(-Z/r)\psi r^2$ provided that we multiply the result of the former by a factor of $\psi(n+1)/\lambda$. The integral I is evaluated as

The integral I is evaluated as

$$I = -2 Z \exp(-\alpha_{2AB}^2) \sum_{n=0}^{\infty} \frac{(2\alpha_{2AB})^{2n}}{(2n+1)!} \frac{n!}{(\alpha_1 + \alpha_2)^{n+1}}$$

The convergence of the summation is dictated by the ratio of the nth term with respect to the sum of the (n-1) terms. If the ratio is less than 10^{-6} , we stop the summation.

The s-p, and p-p integral again could be obtained by partial differentiation with respect to B_x , B_y or B_z :

$$\begin{aligned} & \langle G^s(\alpha_1, \vec{r}_A) | -\frac{Z}{r_A} | G^{p_x}(\alpha_2, \vec{r}_B) \rangle \\ &= \frac{1}{2\alpha_2} \frac{\partial}{\partial B_x} \langle G^s(\alpha_1, \vec{r}_A) | -\frac{Z}{r_A} | G^s(\alpha_2, \vec{r}_B) \rangle \\ &= 2\pi Z \exp(-\alpha_{2AB}^2) \vec{AB} \left[\sum_{n=0}^{\infty} \frac{(2\alpha_{2AB})^{2n}}{(2n+1)!} \frac{n!}{(\alpha_1 + \alpha_2)^{n+1}} \right. \\ & \quad \left. - 4\alpha_2 \sum_{n=1}^{\infty} \frac{n(2\alpha_{2AB})^{2n-2}}{(2n+1)!} \frac{n!}{(\alpha_1 + \alpha_2)^{n+1}} \right]. \end{aligned}$$

$$\begin{aligned}
& \langle G^{Px}(\alpha_1, \vec{r}_A) | -\frac{Z}{r_A} | G^S(\alpha_2, \vec{r}_B) \rangle \\
&= -\frac{2\pi Z}{\alpha_2^2} \exp(-\alpha_2^2 \overline{AB}^2) \overline{AB}_x \sum_{n=1}^{\infty} \frac{n! (2\alpha_2)^{2n}}{(2n+1)! (\alpha_1 + \alpha_2)^{n+1}} n (\overline{AB}^2)^{n-1}
\end{aligned}$$

$$\begin{aligned}
& \langle G^{Px}(\alpha_1, \vec{r}_A) | -\frac{Z}{r_A} | G^{Py}(\alpha_2, \vec{r}_B) \rangle \\
&= -\frac{\pi Z}{\alpha_2^2} \exp(-\alpha_2^2 \overline{AB}^2) \overline{AB}_x \overline{AB}_y \sum_{n=1}^{\infty} \frac{(2\alpha_2)^{2n} n!}{(2n+1)! (\alpha_1 + \alpha_2)^{n+1}} \\
&\quad \times \left\{ n(n-1) (\overline{AB}^2)^{n-2} - 2\alpha_2 n (\overline{AB}^2)^{n-1} \right\}^{53}
\end{aligned}$$

TABLE XV
COMPARISON OF $V_{K_{\mu}}$ AND $V_{2K_{\mu}}$ FOR DIAMOND

| K_x | K_y | K_z | $V_{K_{\mu}}$ | $V_{2K_{\mu}}$ |
|-------|-------|-------|---------------|----------------|
| 0 | 0 | 0 | -1.435 | -0.64276 |
| 2 | 2 | 0 | -0.20803 | 0.19417 |
| 4 | 0 | 0 | -0.11484 | 0.10115 |
| 4 | 2 | 2 | -0.080940 | 0.045115 |
| 5 | 1 | 1 | -0.073118 | 0.032615 |
| 5 | 3 | 1 | -0.58377 | 0.012904 |
| 5 | 3 | 3 | -0.048701 | 0.0019987 |
| 5 | 5 | 1 | -0.041788 | 0.0011086 |
| 6 | 4 | 2 | -0.038372 | 0.00023358 |
| 7 | 3 | 1 | -0.036572 | -0.000049802 |

TABLE XVI

COMPARISON OF V_{K_y} AND V_{2K_y} FOR Mg and O

| K_x | K_y | K_z | $V_{K_y}(O)$ | $V_{2K_y}(O)$ | $V_{K_y}(Mg)$ | $V_{2K_y}(Mg)$ |
|-------|-------|-------|--------------------------|-------------------------|--------------------------|-------------------------|
| 9 | 5 | 1 | -0.1115×10^{-1} | 0.2426×10^{-2} | -0.1607×10^{-1} | 0.5833×10^{-2} |
| 23 | 11 | 7 | -0.1817×10^{-2} | 0.1999×10^{-3} | -0.2717×10^{-2} | 0.2440×10^{-3} |
| 30 | 8 | 2 | -0.1319×10^{-2} | 0.5143×10^{-4} | -0.1968×10^{-2} | 0.5562×10^{-4} |
| 31 | 11 | 1 | -0.1182×10^{-2} | 0.2940×10^{-4} | -0.1761×10^{-2} | 0.2923×10^{-4} |
| 30 | 16 | 14 | -0.9487×10^{-3} | 0.8585×10^{-5} | -0.1412×10^{-2} | 0.5213×10^{-5} |
| 39 | 7 | 7 | -0.7753×10^{-3} | 0.3368×10^{-5} | -0.1181×10^{-2} | 0.4302×10^{-7} |

APPENDIX D

KINETIC AND POTENTIAL INTEGRALS

The integrals of kinetic and potential energies are as follows:

$$\begin{aligned}
 \langle G^S(\alpha_1, \vec{r}-\vec{A}) | -\frac{1}{2}\nabla^2 | G^S(\alpha_2, \vec{r}-\vec{B}) \rangle &= \lambda \zeta (3 - 2\lambda \overline{AB}^2), \\
 \langle G^{PX}(\alpha_1, \vec{r}-\vec{A}) | -\frac{1}{2}\nabla^2 | G^S(\alpha_2, \vec{r}-\vec{B}) \rangle &= \lambda^2 \Delta \zeta \overline{AB}_x (5 - 2\lambda \overline{AB}^2) / \alpha_1, \\
 \langle G^{PX}(\alpha_1, \vec{r}-\vec{A}) | -\frac{1}{2}\nabla^2 | G^{PX}(\alpha_2, \vec{r}-\vec{B}) \rangle &= \lambda^2 \Delta \zeta (\frac{5}{2} - 7\lambda \overline{AB}_x^2 - \lambda \overline{AB}^2 + 2\lambda^2 \overline{AB}^2 \overline{AB}_x^2) / \alpha_1 \alpha_2, \\
 \langle G^{PX}(\alpha_1, \vec{r}-\vec{A}) | -\frac{1}{2}\nabla^2 | G^{PY}(\alpha_2, \vec{r}-\vec{B}) \rangle &= \lambda^3 \Delta \zeta \overline{AB}_x \overline{AB}_y (2\lambda \overline{AB}^2 - 7) / \alpha_1 \alpha_2, \\
 \langle G^S(\alpha_1, \vec{r}-\vec{A}) | \cos(\vec{K}_\nu \cdot \vec{r}_C) | G^S(\alpha_2, \vec{r}-\vec{B}) \rangle &= \Delta \delta \zeta \cos(\vec{K}_\nu \cdot \vec{r}_{CD}), \\
 \langle G^{PX}(\alpha_1, \vec{r}-\vec{A}) | \cos(\vec{K}_\nu \cdot \vec{r}_C) | G^S(\alpha_2, \vec{r}-\vec{B}) \rangle &= \\
 &= \Delta \delta \zeta [(\lambda \overline{AB}_x / \alpha_1) \cos(\vec{K}_\nu \cdot \vec{r}_{CD}) \\
 &\quad - (\vec{K}_\nu)_x \sin(\vec{K}_\nu \cdot \vec{r}_{CD}) / 2(\alpha_1 + \alpha_2)], \\
 \langle G^{PX}(\alpha_1, \vec{r}-\vec{A}) | \cos(\vec{K}_\nu \cdot \vec{r}_C) | G^{PX}(\alpha_2, \vec{r}-\vec{B}) \rangle &= \Delta \delta \zeta \{ (\lambda / \alpha_1 \alpha_2) [\frac{1}{2} - \lambda \overline{AB}_x^2 \\
 &\quad - (\lambda / 4 \alpha_1 \alpha_2) (\vec{K}_\nu)_x^2] \cos(\vec{K}_\nu \cdot \vec{r}_{CD}) \\
 &\quad + (\lambda / 2 \alpha_1 \alpha_2) \overline{AB}_x (\vec{K}_\nu)_x (2u-1) \sin(\vec{K}_\nu \cdot \vec{r}_{CD}) \} , \\
 \langle G^{PX}(\alpha_1, \vec{r}-\vec{A}) | \cos(\vec{K}_\nu \cdot \vec{r}_C) | G^{PY}(\alpha_2, \vec{r}-\vec{B}) \rangle &= \\
 &= (\lambda / \alpha_1 \alpha_2) \Delta \delta \zeta \{ \frac{1}{2} [u (\vec{K}_\nu)_x \overline{AB}_y \\
 &\quad - (1-u) (\vec{K}_\nu)_y \overline{AB}_x] \sin(\vec{K}_\nu \cdot \vec{r}_{CD}) \\
 &\quad - \lambda [\overline{AB}_x \overline{AB}_y + (\vec{K}_\nu)_x (\vec{K}_\nu)_y / 4 \alpha_1 \alpha_2] \cos(\vec{K}_\nu \cdot \vec{r}_{CD}) \} ,
 \end{aligned}$$

where

$$\lambda = \alpha_1 \alpha_2 / (\alpha_1 + \alpha_2), \Delta = [\pi / (\alpha_1 + \alpha_2)]^{3/2}, \quad \mathcal{J} = \exp(-\lambda \overline{AB}^2),$$

$$\delta = \exp[-K_y^2 / 4(\alpha_1 + \alpha_2)], \quad u = \alpha_1 / (\alpha_1 + \alpha_2),$$

and \overline{AB}_x refers to the x component of the line joining the points A and

B.

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