AN EVALUATION OF THE FEEDBACK CONTROL OF A STRIPPING

COLUMN MODEL

By

LARRY E. PORTER Bachelor of Science Oklahoma State University Stillwater, Oklahoma

1967

Submitted to the Faculty of the Graduate College of the Oklahoma State University in partial fulfillment of the requirements for the Degree of MASTER OF SCIENCE May, 1969

OKLAHUMA STATE UNIVERSITY LIBRARY

SEP 29 1969

g.Fruz

AN EVALUATION OF THE FEEDBACK CONTROL OF A STRIPPING COLUMN MODEL

Thesis Approved:

iser am

Dean of Graduate College the

PREFACE

A feedback control system has been added to a lumped parameter model, developed to predict the transient behavior of stripping column after an upset in the feed. The effect of column efficiency upon the operation and control of the column was studied. Inconsistencies in the internal stream compositions were also studied. The results indicate that the model can be adequately controlled by conventional feedback control methods.

I wish to express my sincere thanks to Dr. John H. Erbar for the guidance and advice that he gave me while serving as my research adviser. I would also like to thank Dr. R. N. Maddox for his helpful suggestions.

I am indebted to the University Computer Center for the use of its computing facilities. I am also grateful for the NDEA Fellowship which aided in the continuance of my studies.

Finally, I want to especially thank my wife, Linda, who has given me inspiration during my graduate study.

iii

TABLE OF CONTENTS

Chapte	r	Page
I.	INTRODUCTION ,	1
II.	REVIEW OF THE LITERATURE	3
	Development of the Lumped Parameter Model	10 16
III.	DESCRIPTION OF CONTROL MODEL	20
IV.	TESTS OF THE CONTROL MODEL	25
	Proportional Control	26 28 31 33
	Integral, Proportional, and Derivative Control	39 43
	the Internal Vapor Composition	46
ν.	CONCLUSIONS AND RECOMMENDATIONS ,	53
	Conclusions	53 54
NOMENC	LATURE	56
A SELE	CTED BIBLIOGRAPHY	59
APPEND	IX A - FUNCTIONS OF THE FEED DISTURBANCE	61
APPEND	IX B - INCORPORATION OF CONTROL AND HEAT BALANCE EQUATIONS INTO OSBORNE'S COMPUTER PROGRAM	64
APPEND	IX C - SIMULATED COLUMN RESPONSE CURVES	67

. t

LIST OF FIGURES

Figu	re	Page
l.	A Distillation Column According to the Section Concept	12
2.	Stripping Column Control Model	21
3.	Simple Proportional Control of Bottoms Composition	27
4.	Integration of Error Curve by Trapezoidal Rule	29
5.	Simple Integral Control of Bottoms Composition	30
6.	Simple Derivative Control of Bottoms Composition	32
7.	Proportional and Integral Control With Step Disturbance	35
8.	Proportional and Integral Control With Ramp Disturbance of Steep Slope	36
9.	Proportional and Integral Control With Ramp Disturbance of Moderate Slope	37
10.	Proportional and Integral Control With Ramp Disturbance of Gradual Slope	38
11.	Proportional and Integral Control With High Frequency Sinusoidal Disturbance	40
12.	Proportional and Integral Control With Moderate Frequency Sinusoidal Disturbance	41
13.	Proportional and Integral Control With Low Frequency Sinusoidal Disturbance	42
14.	Proportional, Integral, and Derivative Control With Step Disturbance	44

÷,

Figure

15.	A Typical Variation of Tray Efficiency With Vapor Flow Rate	45
16.	Variation of Separation Parameter With Vapor Boilup	47
17.	Integral and Proportional Control With 90% Change in Separation Parameter	48
18,	Experimental and Calculated Data of Burman	49
19.	Comparison Between Experimental and Thermodynamic K Values	51
20.	Low Frequency Sinusoidal Disturbance With No Control	68
21.	Moderate Frequency Sinusoidal Disturbance With No Control	69
22.	High Frequency Sinusoidal Disturbance With No Control	70
23.	Integral and Proportional Control With $K_1 = 1000.0$ and $K_2 = 1000.0$	71
24.	Integral and Proportional Control With $K_1 = 500.0$ and $K_2 = 50.0$	72
25.	Integral and Proportional Control With $K_1 = 10.0$ and $K_2 = 100.0$	73
26.	Integral and Proportional Control With $K_1 = 100.0$ and $K_2 = 1000.0$	74
27.	Integral and Proportional Control With $K_1 = 500.0$ and $K_2 = 100.0$	75
28.	Integral and Proportional Control With $K_1 = 2000.0$ and $K_2 = 100.0$	76
29.	Step Disturbance With $K_1 = 100.0, K_2 = 1000.0,$ and $K_3 = 1000.0$	77
30.	Step Disturbance With $K_1 = 1000.0, K_2 = 100.0,$ and $K_3 = 10000.0$	78
31.	Step Disturbance With $K_1 = 1000.0, K_2 = 100.0,$ and $K_3 = 1000.0$	79
32.	Step Disturbance With $K_1 = 100.0, K_2 = 100.0,$ and $K_3 = 10.0$	80

Figure

33.		Disturbance With $K_1 = 1000.0$, = 100.0, and $K_3 = 100.0$	
34.		Disturbance With $K_1 = 1000.0$, = 1000.0, and $K_3 = 1000.0$	
35.	Step K ₂	Disturbance With $K_1 = 100.0$, = 100.0, and $K_3 = 1000.0$	
36.		Disturbance With $K_1 = 1000.0$, = 1000.0, and $K_3 = 10000.0$	

Page

CHAPTER I

INTRODUCTION

In recent years computer control in the process industries has become increasingly important. With the advent of modern technology, solution of industrial problems necessitates the reliability and rapid service which the computer offers.

The distillation column, an integral part of most processes in the chemical industry, must be accurately controlled. The product compositions and column performance must be maintained at an operating level to insure the desired results of the process.

A lumped parameter model has been developed which has been shown to predict with some degree of accuracy the transient behavior of a stripping column. The model requires only that data which can be obtained from plant column or computer calculations. The model offers simplicity of operation and can be programmed on a computer which can be economically installed in a plant. However, to be suitable for industrial application, the model must be able to control the conditions of the column.

This research project had three objectives. The first objective was to explore the controllability of the

lumped parameter model. A second objective was to apply the control system to a column in which the tray efficiency was variable. By varying the tray efficiency, actual operating conditions of a real-life column could be approximated. The third objective was to resolve inconsistencies in the internal stream compositions between the values predicted by the model and those experimental values taken by Burman (3).

CHAPTER II

REVIEW OF THE LITERATURE

In the last decade many papers have been published in the area of fractionator control. Most of the column and control models involve complex tray-to-tray calculations and matrix manipulations. A description of several of the models will be given in the following paragraphs.

Luyben, et al. (10, 11) did extensive work in the feed-forward control of distillation columns. The transient behavior of a ten-tray pilot-plant column was described by a set of 26 differential equations. Two equations were required for each tray. One equation described the composition-time behavior, while the second showed the liquid rate-time behavior. The remaining equations described the reboiler, reflux drum and sections of the top and feed tray used to preheat cold feed streams to those trays. The 26 equations were then converted by Laplace transforms into the frequency domain. At a particular value of the frequency, 26 algebraic equations resulted with coefficients which were complex numbers. The equations for the top tray or the tenth tray are given by

$$\frac{dI_{no}}{dt} = (1/a)(R - I_{no})$$
(1)

$$\overline{\delta_{10}} \frac{dx_{10}}{dt} = (\overline{x_d} - \overline{x_{10}})R - (\overline{y_{10}} - \overline{y_9})V_{10} + \overline{R} x_d$$
$$- (\overline{L_{10}} + \overline{V_{10}} \overline{m_{s_{10}}}) x_{10} + \overline{V_9} \overline{m_{s_9}} \overline{x_9}, \quad (2)$$

4

where the term, a, is a constant and barred quantities are the steady-state values. In the frequency domain, these equations become

$$j\omega x_{10} = b_3 R - b_3 V_{10} - b_4 x_{10} + b_5 x_9$$
(3)
$$j\omega L_{10} - b_6 R - b_6 L_{10},$$
(4)

where the term, b, is a constant and $j = \sqrt{-1}$. Similar equations for tray 9 can be derived as

$$j\omega x_9 = b_7 I_{10} - b_8 V_9 + b_9 x_{10} - b_{10} x_9 + b_{11} x_8$$
 (5)

$$j\omega Le = b_6 L_{10} - b_6 L_{10}, \qquad (6)$$

Substitution of Equations (3) and (4) into (5) yields

$$x_{B} = (b_{12} + jb_{13})V_{r} - (b_{14} + jb_{15})R.$$
(7)

The stepping procedure was continued on down the column to the feed plate. A similar procedure was followed from the bottom plate to the feed plate. Equations for the feed plate are

$$R = (ba_4 + jb_{a_5})F + (b_{a_5} + jb_{a_7})x_{f}$$
(8)

and

and

$$V_{s} = (b_{a} + jb_{a})F + (b_{a} + jb_{a})x_{f}.$$
(9)

The feedforward controller (FFC) was made up of four separate transfer functions. In matrix notation, these are

$$FFC = \begin{vmatrix} R/F & R/x_{f} \\ V_{s}/F & V_{s}/x_{f} \end{vmatrix}.$$
(10)

Therefore, if the transfer functions are known, the proper values for the corrective inputs are determined by

$$\begin{vmatrix} \mathbf{R} \\ \mathbf{V}_{\mathrm{S}} \end{vmatrix} = \left| \mathbf{FFC} \right| \begin{vmatrix} \mathbf{F} \\ \mathbf{x}_{\mathrm{f}} \end{vmatrix} . \tag{11}$$

Gerster (5) applied the same mathematical model described by Luyben (10, 11) to a ten-tray stripping column. The binary system studied was composed of benzene and acetone. The column and control system were simulated on an analog computer. The liquid temperature on the seventh tray was the controlled variable. The manipulated variable was the vapor rate to the column. Feedback control, as well as feedforward control, was tested on the column.

Brosilow (2) implemented feedback control on a fifteen tray pilot scale rectifying column. The process model consisted of linearized material balance equations for each tray. These equations were very similar in form to those used by Luyben (10, 11). The control law used in the study is described by the following equations:

 $m(t) = a \quad \text{if } \beta' x < 0$ = b \quad \text{if } \beta' x > 0 $\beta' x = \sum_{n=1}^{N} \beta_n x_n \qquad (12)$

 $\beta = PC$ (13)

 $A'P + PA = -Q, \qquad (14)$

where a and b are scalar constraints on the control effort, x_n is the deviation of liquid composition on tray n from the steady-state and β ,P and C are matrix quantities.

Rose, et al. (19, 20) described the control of a five-plate distillation column. The column was represented by a series of finite difference equations, which were very similar to those of Luyben (10, 11). These equations were in differential form and related the rate of change of the composition as a function of the traffics and holdup on each plate. An example of this equation is given for plate 5 by

$$\delta_{5} \frac{dx_{5}}{dt} = V_{4}y_{4} + L_{d}y_{5} - V_{5}y_{5} - L_{5}x_{5}, \qquad (15)$$

where the subscripts refer to the plate number. The terms, V and L, are not constant with time during the transient condition and were represented in transfer notation by

$$V_n = \frac{V_{n-1}}{\tau_1 p+1},$$
 (16)

where $p = \frac{d}{dt}$, τ is a time constant and n is a generalized

plate number.

A generalized model for a distillation column was presented by Holland (8). The component material balances were similar to those presented by Rose (19, 20) and were represented by a set of integral-difference equations. These equations were converted to a set of algebraic equations and put into matrix notation.

The calculational procedure is initiated by assuming a set of temperatures and traffic streams for each plate. The matrix is then solved for the component flow rates from each plate. After convergence has been achieved, a new set of temperatures and flow rates are predicted for the end of the next time period. Holland used this calculational procedure and developed controller equations for both feedback and feedforward control.

Rosenbrook (21) had a different approach to the control of a column. Rather than calculate transient data, he was concerned only with initial and final steady-state date. He defined a "disturbance function", D_s, and a "variation of product", G. For a binary system, D_s is given by

$$D_{s} = \Sigma_{r} \left| \frac{d}{dt} \delta_{r} x_{r} \right|, \qquad (18)$$

where the sum is taken over all plates. δ_r is the liquid holdup. If the column were operating at steady-state and the reflux rate were increased by ΔL and if all liquid flows were increased by the same amount, the material

balance around the condenser, the reboiler, and each plate would yield

$$\frac{\mathrm{d}}{\mathrm{d}t} \delta_{\mathbf{n}+1} \mathbf{x}_{\mathbf{n}+1} = 0 \tag{19}$$

$$\frac{\mathrm{d}}{\mathrm{d}t} \delta_0 x_0 = \Delta L(x_1 - x_0) \qquad (20)$$

$$\frac{\mathrm{d}}{\mathrm{dt}} \delta_{\mathbf{r}} \mathbf{x}_{\mathbf{r}} + \Delta \mathbf{L}(\mathbf{x}_{\mathbf{r+1}} - \mathbf{x}_{\mathbf{r}}).$$
(21)

The subscripts refer to the plate number. The values of the composition, x, are those in the previous steady-state. On summing, the amount of disturbance in the column after the change in reflux is found to be

$$D_{g} = \Delta L(x_{n+1} - x_{0}). \qquad (22)$$

A similar equation can be written for an increase in vapor flow rate, ΔV , as

$$D_{s} = \Delta V(y_{n} - x_{o}). \qquad (23)$$

Optimum control would have the property of making the rate of decrease of D_s a maximum. Therefore, if a disturbance should enter the column through the feed, control would be applied by a subsequent disturbance in the liquid or vapor flow rate which would annul part of the disturbance previously introduced.

Lupfer, et al. (12, 13, 14) have also done extensive work in the area of feedforward control. However, their work has been confined to actual column operation. A rather complicated control scheme was suggested to optimize column operation. General equations for a debutanizer were developed which stated that the product purities are functions of 16 variables. By applying internal reflux control, three of the variables were removed from consideration. The addition of feed enthalpy control reduced the number of variables to 11. The regulation of the reboiler heat input reduced the number of independent variables to 10. The general equation for the controlled variables are given by

 $B = f_{9}(C_{3}, iC_{4}, nC_{4}, iC_{5}, nC_{5}, F, F_{t}, F_{e}, iC_{5d}, nC_{4b})$ $R_{i} = f_{10}(C_{3}, iC_{4}, nC_{4}, iC_{5}, nC_{5}, F, F_{t}, F_{e}, iC_{5d}, nC_{4b}),$ (24)

where F_t is the feed tray location, F_e is the feed enthalpy, and R, is the internal reflux rate. The equations show that the bottoms flow rate and the internal reflux rate required to maintain the desired product purities are a function of feed composition, feed flow rate, feed tray location, feed enthalpy, and product specification. The values of the feed enthalpy and feed tray location were optimized to minimize operating costs. The development of an equation for predictive control of the internal reflux rate involved a regression analysis of data obtained by tray-to-tray calculations. The data were "curve-fitted" to a forty-five term second-order equation. A multi-order time lag was used on the bottom and reflux

flow rates. Feedback control was then used to compensate for the predictive system errors which result because of measurement errors, imperfect equations, and computing equipment errors. A composition analyzer was placed on the product stream, and a controller trimmed the computed bottom product signal to insure specified overhead product purity.

Williams, et al. (23, 24) and Williams in conjunction with Rose (19, 20) concerned himself with determining the optimum sampling point and the range of stability for a given sampling period. The mathematical model, presented by Rose, was used. The maximum sampling period was found to be less than one-half the system time constant. The sampling point was found to vary according to the amount of dead space error.

Other articles have been written which, although they do not discuss a particular mathematical model, contain useful material. Talbot (22) discusses the various methods which are presently being used in fractionator control. The advantages and disadvantages of each control scheme are discussed. Bertrand (1) also discusses the different control schemes for fractionators. Composition, pressure, and inventory controls are illustrated.

> Development of the Lumped Parameter Model (16, 17)

In 1962 Marr (15) suggested a new concept for

predicting the transient behavior of a distillation column. He suggested that a model should be developed in which some parameter could be used to describe the degree of separation that was occurring in a distillation column. All model development until that time had been directed toward the complicated plate-to-plate model.

Reynolds (18) developed a model which used the concept of the separation parameter. He envisioned a distillation column as being composed of several sections in which there could be any number of trays. According to the section concept, as shown in Figure 1, a section of a distillation column is that part of the column which lies between the points at which the feed streams enter or product streams leave the column. According to Reynolds, the rate at which mass is transferred from the vapor phase to the liquid phase can be expressed by the equation

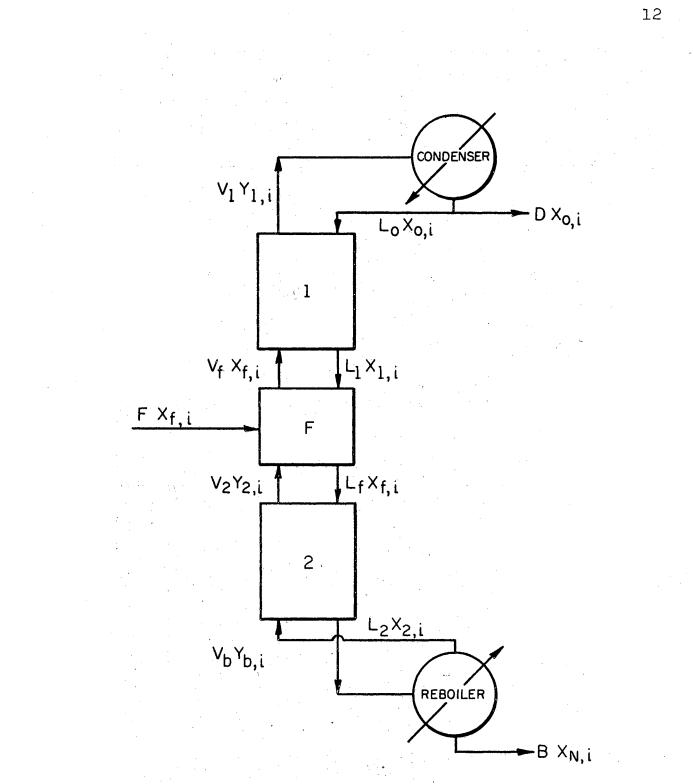
$$N_{n,i}^{V} = -J_{n,i}(y^{*} - y)_{n,i},$$
 (25)

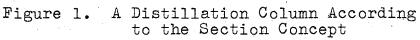
where $J_{n,i}$ is the parameter which describes the degree of separation occurring in a section and $(y^* - y)_{n,i}$ is the driving force for mass transfer in the section.

Although Equation (25) appears to be identical in form to the equation normally used for mass transfer

$$N = K_{OG}(y^* - y) ,$$

there are fundamental differences. The coefficient, K_{OG} , is related to the diffusivity of the component being





transferred. The coefficient, $J_{n,i}$, however, is not directly related to the physical properties of the component being transferred. $J_{n,i}$ is a parameter that describes the degree of separation that occurs in a section and is an empirically determined factor.

Reynolds developed a set of differential equations to predict the transient behavior of the liquid and vapor streams leaving the section. Two major assumptions were made by Reynolds. The first assumption was that $J_{n,i}$ remains constant for small changes in column conditions. The second was constant molal overflow throughout the section. Reynolds was unable to get good agreement between values predicted by his model and experimental data.

Osborne (16) felt that Reynold's concept was basically sound. Rather than assume constant molal overflow, he extended Reynold's model to relate the net rate at which mass is transferred from the liquid phase to the net rate at which mass is transferred from the vapor phase. His development of the model follows.

From the law of conservation of matter,

$$input = V_{n+1} y_{n+1,i}$$
(27)

output =
$$V_{n+1} y_{n+1,i} + \frac{\partial (V_n y_{n,i})}{\partial z} dz + N_{n,i}^V$$
 (28)

accumulation =
$$\frac{\partial(\delta_n^{\vee} y_{n,i})}{\partial t}$$
. (29)

The output term is the sum of all of the material leaving the vapor phase, either by flow or mass transfer. The flow rate of a component from the section is equal to the flow rate in $V_{n+1} y_{n+1,i}$ plus the increase in the stream flow rate that occurs in the section $\frac{\partial(V_n y_{n,i})dz}{\partial z}$. The vapor phase is represented by the term $N_{n,i}^V$. The material balance equation becomes

$$\frac{\partial(\delta_{n}^{V} y_{n,i})}{\partial t} = -\frac{\partial(V_{n} y_{n,i})}{\partial z} dz - N_{n,i}^{V}.$$
(30)

A similar material balance equation

$$\frac{\partial(\delta_{n}^{L} \mathbf{x}_{n,i})}{\partial t} = -\frac{\partial(L_{n} \mathbf{x}_{n,i})}{\partial z} dz - N_{n,i}^{L}$$
(31)

can be derived for the liquid stream passing through a section of a distillation column.

If the change in height Δz is small, the term, $\frac{\partial(V_n y_{n,i})}{\partial z}$ can be approximated by

$$\frac{\partial(V_n y_{n,i})}{\partial z} = \frac{V_n y_{n,i} - V_{n+1,i}}{\Delta z} .$$
 (32)

Since $\delta_n^V y_{n,i}$ is now a function of time only, the partial derivative with respect to time can be replaced with a total derivative. Osborne then assumed constant molal holdup and a constant rate of interphase mass transfer in a section. According to Reynolds,

$$N_{n,i}^{V} = -J_{n,i}(y^{*} - y)_{n,i}.$$
 (33)

Equation (30) can then be written in the form

$$\delta_{n}^{V} \frac{d(y_{n,i})}{dt} = -(V_{n} y_{n,i} - V_{n+1} y_{n+1,i}) + J_{n,i}(y^{*} - y)_{n,i}.$$
(34)

If a section is considered to be subdivided into an infinite number of subsections, the driving force for one of these infinitesimal subsections can be represented by the equation

$$(y^* - y)_{k,i} = (Kx)_{k-1,i} - y_{k+1,i}$$
 (35)

By summing the values of the subintervals, the driving force for the section can be approximated by the equation

$$(y^* - y)_{n,i} = (Kx)_{n-1,i} - y_{n+1,i}$$
 (36)

Using Equation (36), Equation (34) can be rewritten in the form

$$\delta_{n}^{V} \frac{d(y_{n,i})}{dt} = -(V_{n} y_{n,i} - V_{n+1} y_{n+1,i}) + J_{n,i}[(Kx)_{n-1,i} - y_{n+1,i}].$$
(37)

A similar derivation for the liquid phase gives the equation

$$\delta_{n}^{L} \frac{d(x_{n,i})}{dt} = -(L_{n} x_{n,i} - L_{n-1} x_{n-1,i}) - J_{n,i}[(Kx)_{n-1,i} - y_{n+1,i}].$$
(38)

By assuming that J_{n,i} is constant for small changes

in column conditions, a solution to the transient behavior of the composition of the vapor and liquid streams leaving the section can be obtained by integrating Equations (37) and (38).

At steady-state $\frac{d(y_{n,i})}{dt} = 0$. Therefore, the value of $J_{n,i}$ can be calculated by the equations

$$J_{n,i} = \frac{(v_n y_{n,i} - v_{n+1} y_{n+1,i})}{(Kx)_{n-1,i} - y_{n+1,i}}$$
(39)

or

$$J_{n,i} = - \frac{(L_n x_{n,i} - L_{n-1} x_{n-1,i})}{(Kx)_{n-1,i} - y_{n+1,i}}$$
(40)

and with steady-state data.

Introduction to Feedback Control (6, 25)

A feedback control system is defined as one in which the controlled variable is compared with the reference variable, and any difference between the two is used to reduce the difference. The control instrument continuously measures the output variable of the controlled process and compares this output with a pre-established desired value. The instrument then uses any resulting error to compute the required correction to the setting of a basic element of the piece of equipment being controlled.

A control system must fulfill three basic requirements:

1. The controlled system must be stable. The

response to a disturbance must reach a steady value within a reasonable period of time. Instability in the system produces oscillations of the output variable and could drive the output to some limiting value.

- 2. The second requirement, accuracy, insures that the control system will reduce any error to some tolerable value.
- 3. Speed of response, the third requirement, is essential in maintaining the desired product specifications.

The three modes of control normally used in control systems are proportional, integral, and derivative. Stability, accuracy, and speed of response of the control system are usually obtained by some combination of the three modes with appropriate proportionality constants. Gardner (4) and Hoffman (7) discuss the common combinations of the control modes. A discussion of each control mode follows:

Proportional control applies a correction to the process which is directly proportional to the error. The equation which describes this action is

$$\Delta m = K_1 E_r. \tag{41}$$

For any particular column or apparatus being controlled, there is a small range of values of K_1 which gives optimum

correction. Too small a value of K₁ results in a continuous steady-state error. Too large a value results in instability and overcorrection. An error must be present for a correction to be made; therefore, proportional control can never completely correct an upset in the column.

Integral control applies a correction to the process which is directly proportional to the time integral of the error. Expressed mathematically,

$$\Delta m = K_2 \int_0^t E_r dt. \qquad (42)$$

The main advantage of integral control is the elimination of the steady-state error. The integral controller continues to correct until the error is zero. If, however, the error develops a change in sign, the accumulated correction must be integrated out before the proper correction can be made. For example, if there had been a large positive error, the integral of the error would be represented by a positive area under the error curve. If the error should then change sign, the negative area under the error curve would have to equal the positive area before the total area under the error curve could change sign permitting the proper correction to be made. Therefore, integral control tends to be sluggish and can lead to overcorrections and instability.

Derivative control applies a correction to the process which is directly proportional to the time derivative of the error. Expressed mathematically,

$$\Delta m = K_3 \frac{dE_r}{dt}.$$
 (43)

Derivative control is desirable, because it responds to the rate of change of error and can apply a large correction while the error is forming. However, derivative control cannot be used alone, because it will not respond to a steady-state error.

Combination of the control modes can utilize the good qualities of each and eliminate the undesirable characteristics of each individual mode. Proportional plus integral control is often used to obtain the stability of proportional control and the error elimination of integral control. The addition of derivative control provides the early detection of the error and adds damping to the system, permitting a higher proportional gain.

CHAPTER III

DESCRIPTION OF CONTROL MODEL

Osborne (16, 17) developed a mathematical model which described the transient behavior of a distillation column. The model offers a more simple solution than the conventional tray-to-tray model. A single parameter, $J_{n,i}$, describes the separation occurring in each section of the column. The model's simplicity makes it possible to use smaller computers for transient condition calculations.

A feedback control system was developed as an extension of the lumped parameter model, in order to test the controllability of a column described by the model. A stripping column, shown in Figure 2, was selected for the simulated control study. The composition of the bottoms product was selected as the controlled variable. The reboiler heat duty was adjusted by a controller at each time interval to maintain the desired bottoms composition. Simple proportional, integral, or derivative control modes or combinations of these modes were selected for the simulated controller.

Osborne assumed that all flow rates changed to their final steady-state values instantaneously after the feed upset occurred. Rather than make this assumption, the

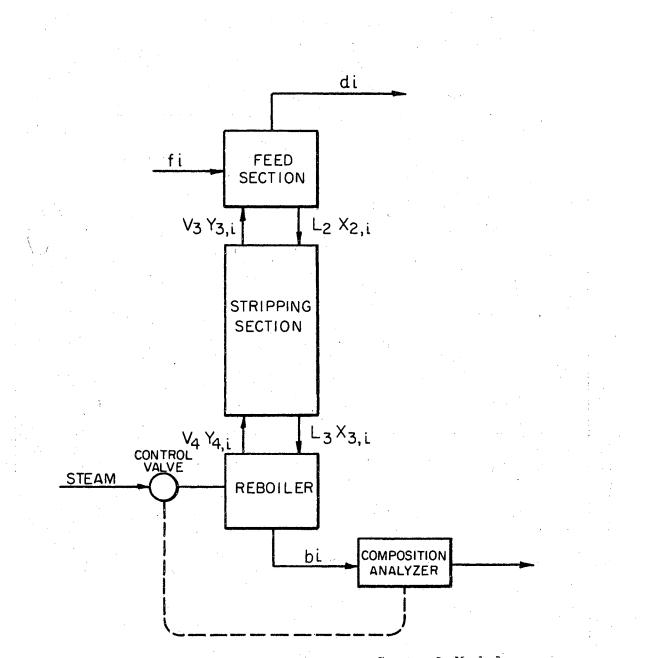


Figure 2. Stripping Column Control Model

flow rates were calculated by heat and material balances at each time interval. An over-all heat and material balance can be written as

$$\mathbf{F} = \mathbf{D} + \mathbf{B} \tag{44}$$

$$F h_{f} = D H_{d} + B h_{b} - Q_{r}.$$
(45)

The molar enthalpies, h_f , H_d , and h_b , were assumed to remain constant for small changes in the respective flow rates. This assumption is valid at small time increments, because the temperature and composition of the components change only slightly. Equations (44) and (45) may then be written as

$$\Delta \mathbf{F} = \Delta \mathbf{D} + \Delta \mathbf{B} \tag{46}$$

$$h_{f} \Delta F = H_{d} \Delta D + h_{b} \Delta B - \Delta Q_{r}.$$
(47)

Substitution of (46) into (47) yields

$$\Delta D = \Delta F \frac{(h_b - h_f) - \Delta Q_r}{(h_b - H_d)}$$
(48)

$$\Delta B = \Delta F - \Delta D. \tag{49}$$

Similar equations used for the reboiler are

$$\Delta B = \Delta L_3 - \Delta V_4 \tag{50}$$

$$\Delta Q_{r} + h_{3} \Delta L_{3} = h_{3} \Delta B + H_{4} \Delta V_{4}$$
(51)

$$\Delta V_{4} = \frac{\Delta B(h_{b} - h_{3}) - \Delta Q_{r}}{(h_{3} - H_{4})}$$
(52)

$$\Delta \mathbf{L}_3 = \Delta \mathbf{B} + \Delta \mathbf{V}_4 \,. \tag{53}$$

Assuming that the column is at thermal steady-state at all times, the following equations can be written for the stripping section

$$\Delta \mathbf{L}_{2} + \Delta \mathbf{V}_{4} = \Delta \mathbf{L}_{3} + \Delta \mathbf{V}_{3} \tag{54}$$

$$h_{2} \Delta L_{2} + H_{4} \Delta V_{4} = h_{3} \Delta L_{3} + H_{3} \Delta V_{3}.$$
 (55)

Simultaneous solution of Equations (54) and (55) yields

$$\Delta V_{3} = \frac{h_{2} (\Delta L_{3} - \Delta V_{4}) + H_{4} \Delta V_{4} - h_{3} \Delta L_{3}}{(H_{3} - h_{2})}$$
(56)

$$\Delta L_2 = \Delta L_3 + \Delta V_3 - \Delta V_4.$$
 (57)

The new flow rates are calculated as

$$X = X_{\text{initial}} + \Delta X, \qquad (58)$$

where X equals the respective flow rates.

The control procedure can now be described. The values of the separation parameter are determined from the steady-state operating conditions. Once the separation parameters have been determined, the feed rate and composition are changed to new values. The reboiler duty is then adjusted by the simulated controller to maintain a constant composition of one of the components in the bottoms product. Heat and material balances are solved to determine the new flow rates.

The new values of the flow rates are then used in the lumped parameter model to determine the new composition of

the bottom product. The reboiler duty is again adjusted. This procedure is continued until the column reaches steady-state at the desired composition level or the computer time allotted for the run is exceeded.

CHAPTER IV

TESTS OF THE CONTROL MODEL

A binary system was selected to test the control model. The hexane-heptane system was selected, because data for this system was readily available. Pertinent data were selected from Osborne's work. Due to the generality of the model, any binary or multicomponent system could reasonably be expected to exhibit similar results to those shown by this system, provided reasonably good thermodynamic data are available and the system is not a complex one.

In order to test the controllability of the model, disturbances in the feed composition and feed flow rate were induced on the model of the stripping column. The three types of feed disturbances considered were step, ramp, and sinusoidal. Combinations of proportional, reset or integral, and derivative control modes were used to maintain the composition of the bottoms product at the desired level. Another test of the model was to vary the column efficiency and observe the action taken by the controller. The tests and results are discussed in the remainder of the chapter.

The working equation for proportional control, as previously discussed, is

$$\Delta m = K_1 E_{p}. \tag{41}$$

The error is defined by the equation

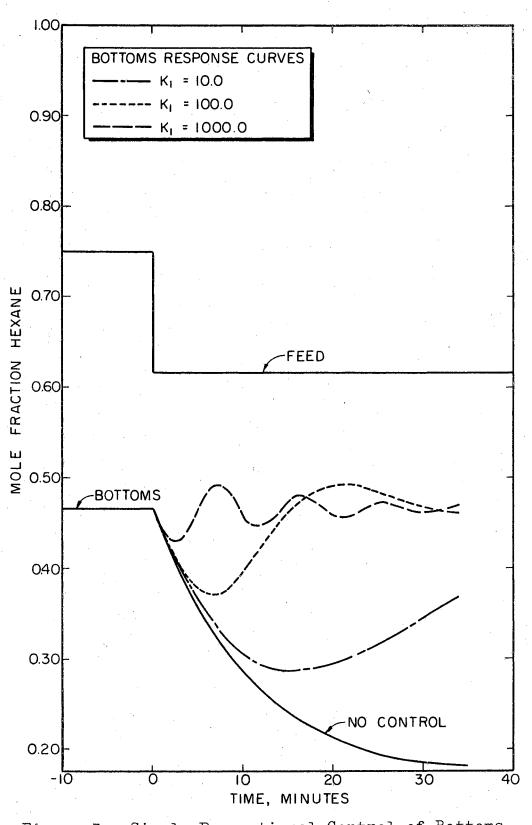
$$\mathbf{E}_{\mathbf{r}} = \mathbf{x}_{\mathrm{bd}} - \mathbf{x}_{\mathrm{bm}}.$$
 (59)

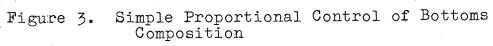
Substitution of the appropriate variables into Equation (41) gives

$$\Delta Q_r = K_1 (x_{bd} - x_{bm}).$$
 (60)

The measured value of the composition was compared with the specified composition after each time interval. The reboiler duty was then adjusted to its new value.

The value of the proportionality constant was varied to obtain the best control for a step disturbance in composition and flow rate to the feed. Selected results are shown in Figure 3. By observing these results, certain conclusions can be made. With a value of the proportional constant, K_1 , equal to 10.0, little correction to the error can be observed. Increasing the value of the constant by an order of magnitude, correction is observed; however, control is still sluggish, and a large amount of off-quality bottoms product would result. When the value of the constant was increased to 1000.0 to obtain an early correction, an overcorrection resulted. Values of the





constant larger than 1000.0 resulted in calculational instability of the model.

Integral Control

The equation for integral control is given by

$$\Delta m = K_2 \int_0^t E_r dt \qquad (42)$$

and

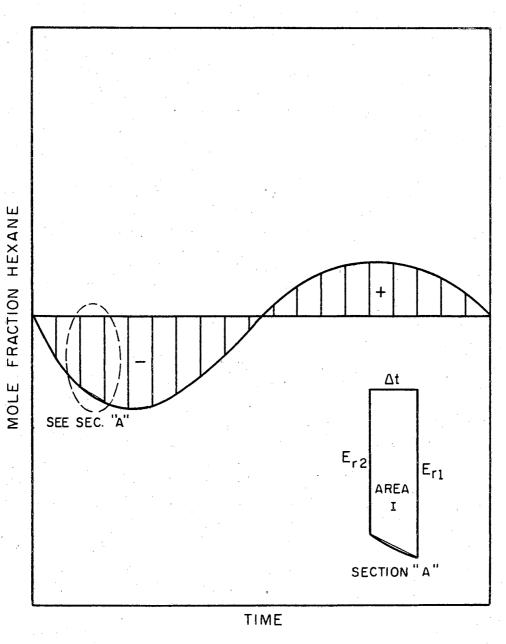
$$\Delta Q_{r} = K_{2} \int_{0}^{t'} (x_{bd} - x_{bm}) dt'. \qquad (61)$$

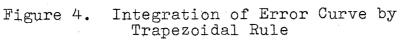
The trapezoidal rule was used to approximate the value of the integral. The trapezoidal rule can be expressed as

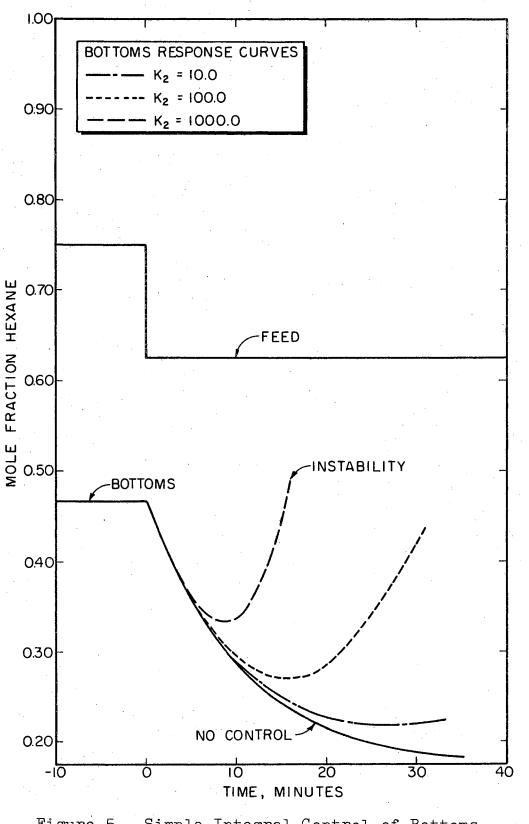
$$I = \frac{\Delta t'}{2} (E_{r2} + E_{r1}).$$
 (62)

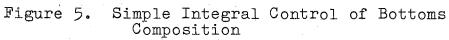
The areas, calculated for each time increment, were summed to obtain the total area under the error curve at time, t'. The method is illustrated in Figure 4.

Several runs were made with the integral controller in which K_2 was varied to obtain the best control for a step disturbance in composition and flow rate to the feed. The results are shown in Figure 5. The results reveal that simple integral control is unacceptable for good control. A value of the constant, K_2 , equal to 1000.0 provided a reasonable degree of control but caused instabilities in the calculational procedure of the lumped parameter model. Smaller values of the constant, K_2 ,









allowed stable operation, but little control. These results are verified in the theory presented earlier. In practice, pure integral control tends to be sluggish and can lead to instability.

Derivative Control

The working equation for the derivative controller is written as

$$\Delta m = K_3 \frac{dE_r}{dt}$$
(43)

or

$$\Delta Q_{r} = K_{3} \frac{E_{r} |_{t'=t'} - E_{r} |_{t'=t'_{0}}}{t' - t'_{0}}.$$
 (63)

The feed disturbance was assumed to occur at $t'_0 = 0$. Because the system was assumed to be at the desired steadystate before the upset, the initial error was assumed to be zero. Equation (63) reduces to

$$\Delta Q_r = \frac{K_3 E_r | t' = t'}{t'}, \qquad (64)$$

where t' equals the time elapsed since the column upset.

The results of the simple derivative controller are presented in Figure 6 for the control of the column which underwent a step disturbance in feed composition and flow rate. Small values of the constant K_3 provided rather slow control. With K_3 equal to 500.0, the system came close to a condition of being critically damped. When the value of the constant was increased by two orders of

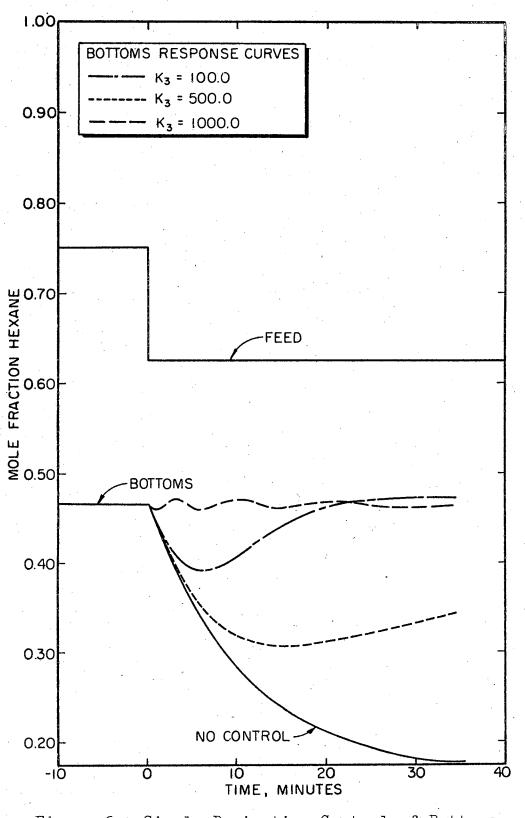


Figure 6. Simple Derivative Control of Bottoms Composition

32

magnitude, good control was achieved, despite oscillation about the desired value.

The derivative controller functioned well and provided better control than the pure proportional or pure integral controller. The derivative controller detected the error more quickly than the other modes and applied a large correction at small values of time, t'. However, at larger values of time, t', the amount of control diminished. In practice, derivative control is not used alone. Because of its mathematical nature, derivative control will not respond to a steady-state error. This is shown when K_3 is equal to 500.0. Although the composition curve levels out at a steady value, this value is above the desired product composition. For this reason, derivative control must be used in conjunction with the other control modes.

Proportional and Integral Control

The proportional and integral control modes can be combined to give

$$\Delta Q_{r} = K_{1} E_{r} + K_{2} \int_{0}^{t'} E_{r} dt'. \qquad (65)$$

This combination of control modes is frequently used in industry. Because of its relative importance, this control scheme was tested on three types of feed disturbances.

The first type of disturbance considered was the step disturbance. The feed composition and flow rate were stepped by the same amount as the case of the pure control

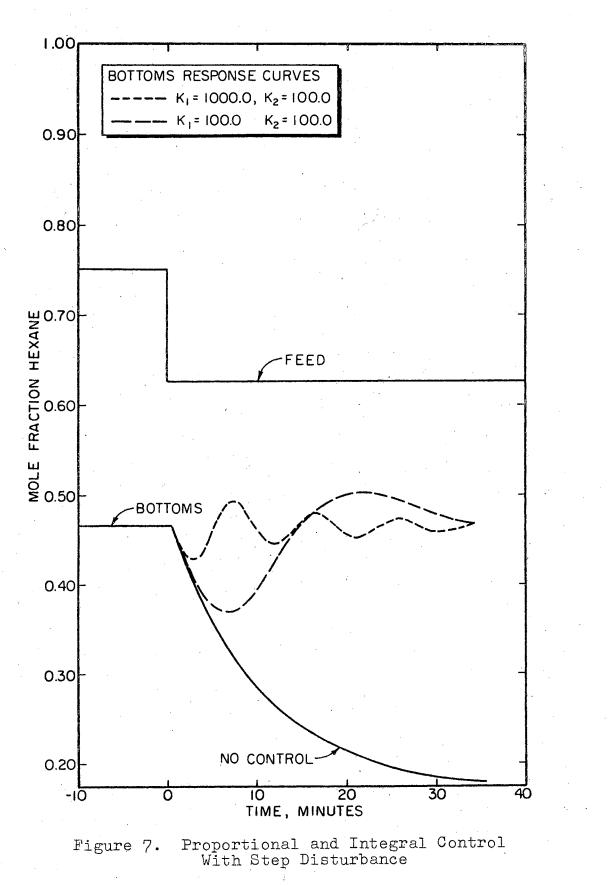
modes discussed earlier. The values of the constants K_1 and K_2 were varied to obtain optimum control. Selected results for the step disturbance are presented in Figure 7.

The control curves reveal that the main control effort results from the proportional controller. The effect of integral control appears to be negligible. The importance of the integral controller, however, lies in its ability to eliminate all the error.

A ramp disturbance in the feed composition and flow rate was the second type of disturbance considered in the tests. The slope of the ramp function was varied while the constants K_1 and K_2 remained at a fixed value. The values of K_1 and K_2 were selected as those which gave good results during the step disturbance.

In Figure 8 the control curve is similar to the curve obtained during a step change. The slope of the ramp function was decreased in Figure 9, and the control was much better. A further decrease in the slope allowed almost perfect control, as shown in Figure 10. The results are much as one should expect. A steep slope of the ramp function closely approximates a step change. As the slope of the ramp decreases, the severity of the disturbance also decreases and can be approximated by a series of small step changes.

The third type of disturbance studied was the sinusoidal variation of the feed composition and flow rate. The frequency of the sine wave was varied with the



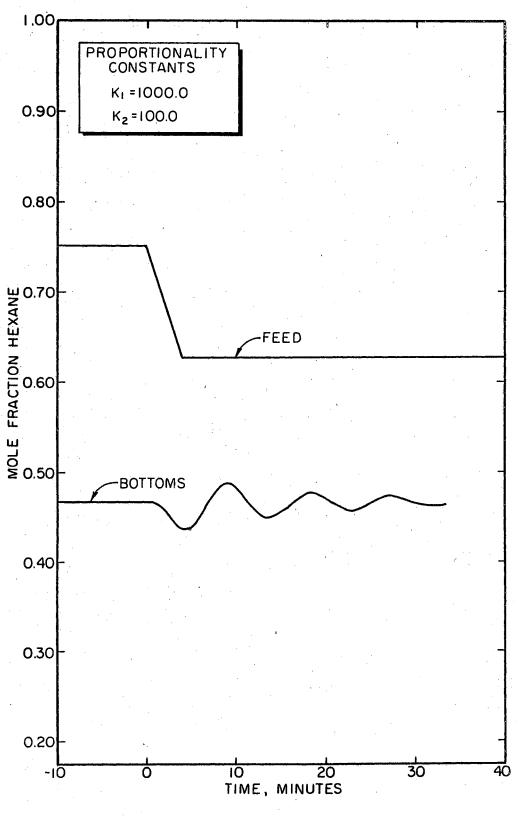
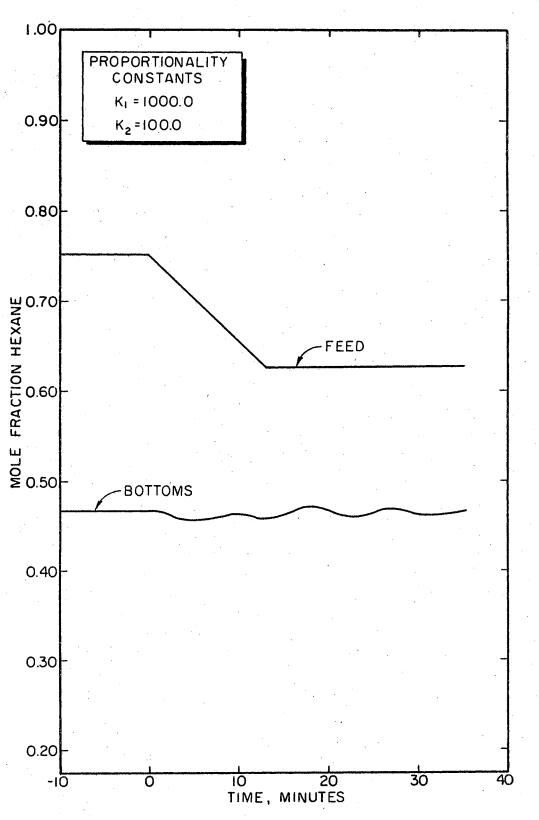
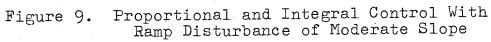
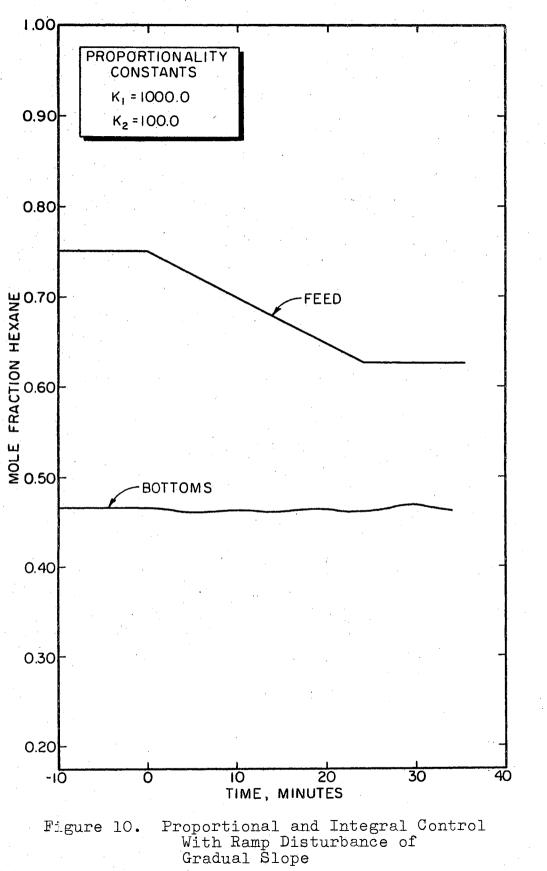


Figure 8. Proportional and Integral Control With Ramp Disturbance of Steep Slope







constants K_1 and K_2 set at the same values used to control the ramp disturbance. The amplitude of the sine function was set at one-half the magnitude of the step change.

At sufficiently high or low frequencies, the desired bottoms composition was maintained quite well. At intermediate values of frequency, the amplitude of the oscillation about the desired bottoms composition increased. At high frequencies, the oscillation of the feed composition closely approximates a steady-state value. At low frequencies the severity of the disturbance decreases in much the same manner described for the ramp function. The results are presented in Figures 11, 12, and 13.

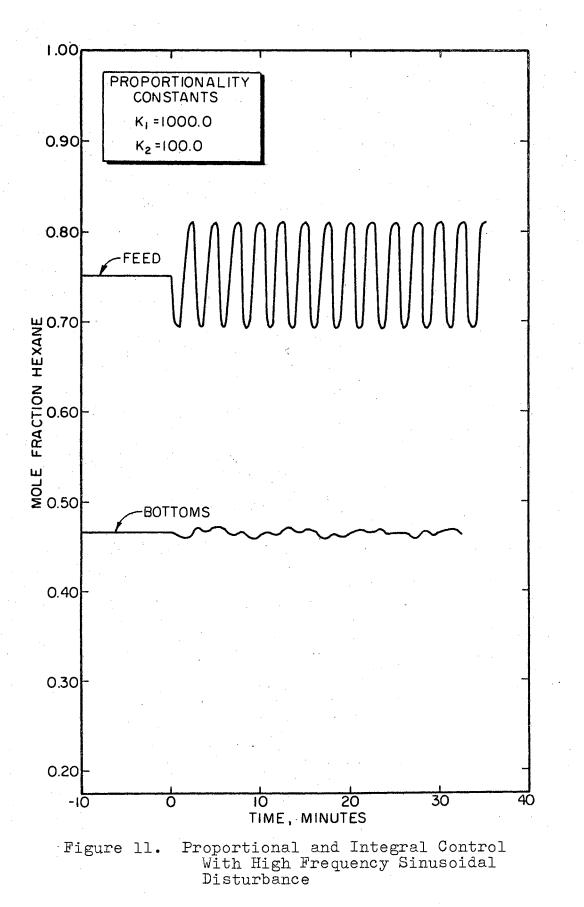
An intermediate frequency was selected and the constants K_1 and K_2 were varied. Typical results are shown in Appendix C.

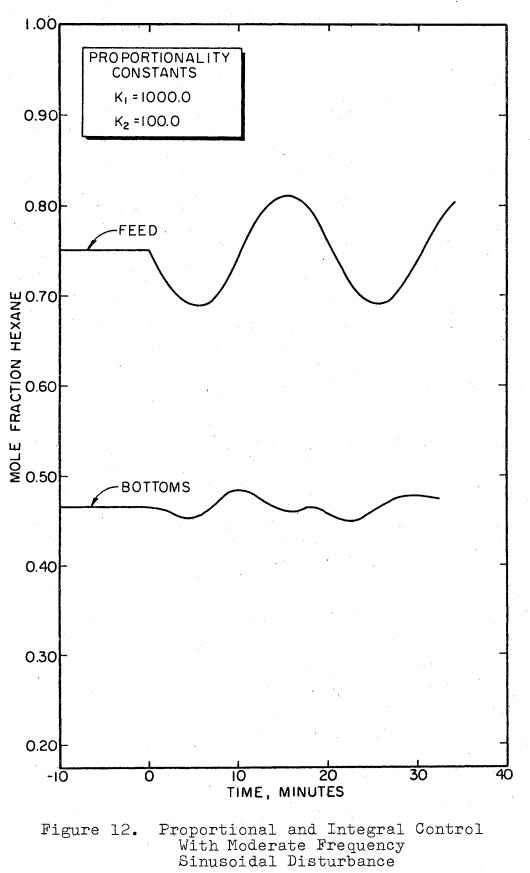
Integral, Proportional, and Derivative Control

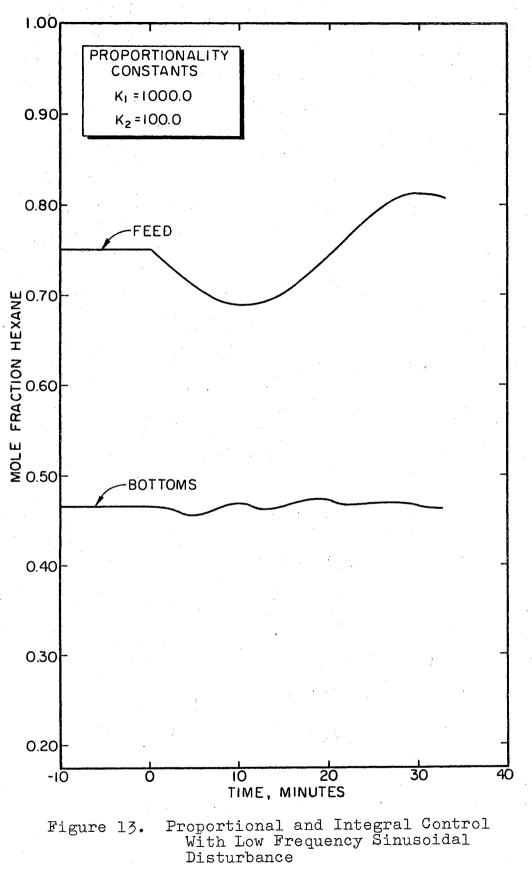
Although the combination of integral and proportional modes provided satisfactory control, the success of the pure derivative control prompted its inclusion in the control scheme. The working equation for the combination of the three control modes becomes

$$\Delta Q_{\mathbf{r}} = K_1 E_{\mathbf{r}} + K_2 \int_0^t E_{\mathbf{r}} dt' + K_3 \frac{dE_{\mathbf{r}}}{dt'}. \quad (66)$$

The control scheme was tested on the step disturbance only. The results obtained with the proportional-integral







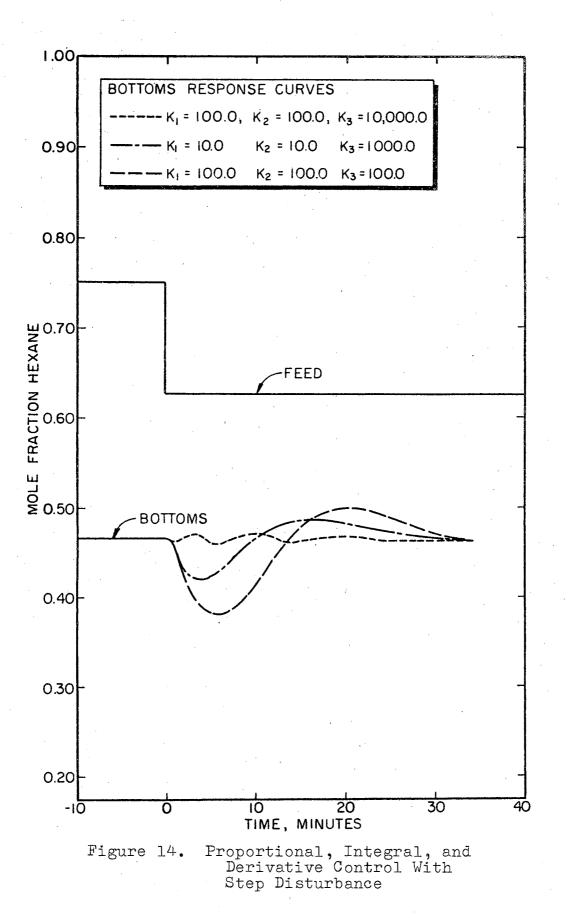
controller have shown that good control of the step disturbance also provides good control of the ramp and sinusoidal disturbances. The best control was achieved with large values of K_3 and moderate values of K_1 and K_2 . Typical results are presented in Figure 14 and Appendix C.

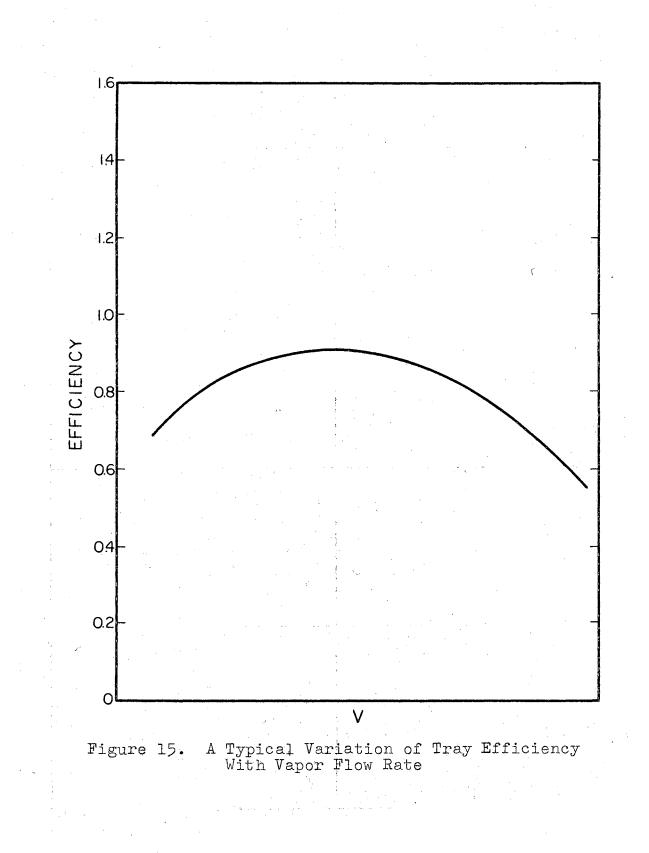
Variation of the Separation Parameter

Osborn (16, 17), when developing the lumped parameter model, made an assumption that the value of the separation parameter remained constant. The efficiency of the plates in a "real" column will vary as a function of vapor and liquid loadings (9). A typical curve is presented in Figure 15 with efficiency as a function of the vapor flow rate. The Murphree equation for efficiency is

$$E = \frac{y_n - y_{n+1}}{y^* - y_{n+1}},$$
 (67)

where the subscript, n, is the tray number. This equation is very similar in form to Equation (39) derived for the separation parameter, $J_{n,i}$. Since there appears to be a relationship between tray efficiency and the separation parameter, the separation parameter in a "real" column could be assumed to be a function of the liquid and vapor traffics. A quadratic equation was selected as the form of the function. The maximum efficiency was assumed to occur at the initial steady-state operation of the column. The separation parameter was then decreased as a function





of the vapor boilup rate after the feed upset. Typical variations of $J_{n,i}$ with vapor flow rate are presented in Figure 16.

Several tests were conducted with various lower limits on the separation parameter. The only effect noted in the system was a displacement of the liquid and vapor rates. As the value of the separation parameter decreased, the boilup vapor rate increased to compensate for the decrease in the amount of separation occurring in the column. However, the limiting vapor velocity within the fractionator was not reached. Subsequent work in this area would be to describe what would happen if flooding occurred. The control of the bottoms composition was not seriously affected as a result of the decrease in efficiency. The control curve output was quite similar to those curves which were obtained with a constant separation parameter, as shown in Figure 17.

Tests to Resolve an Inconsistency in the Internal Vapor Composition

Burman (3) described an inconsistency in the internal stream traffics predicted by the mathematical model and data measured in a twelve-inch diameter column. The model consistently predicted a vapor composition leaving the reboiler higher in the more volatile component than was actually observed in the experimental column. His results are shown in Figure 18.

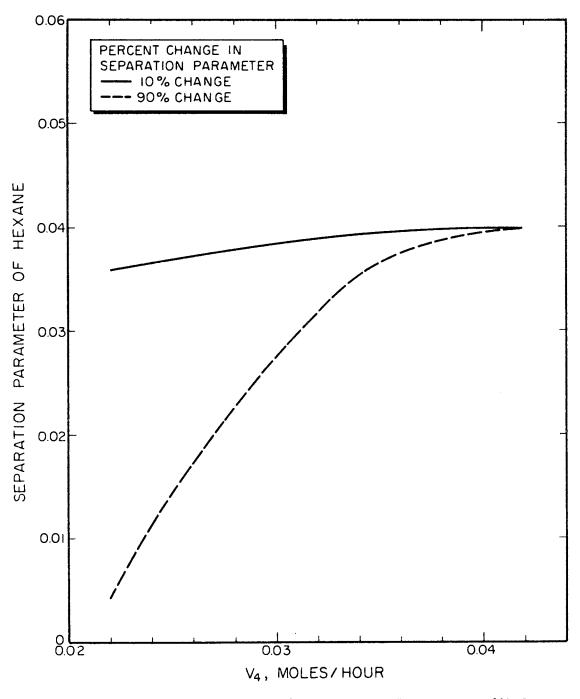
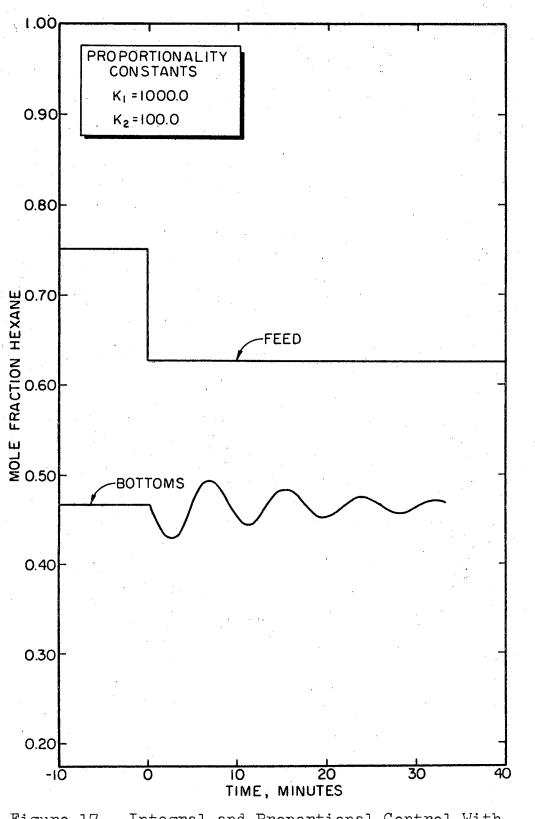
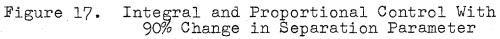
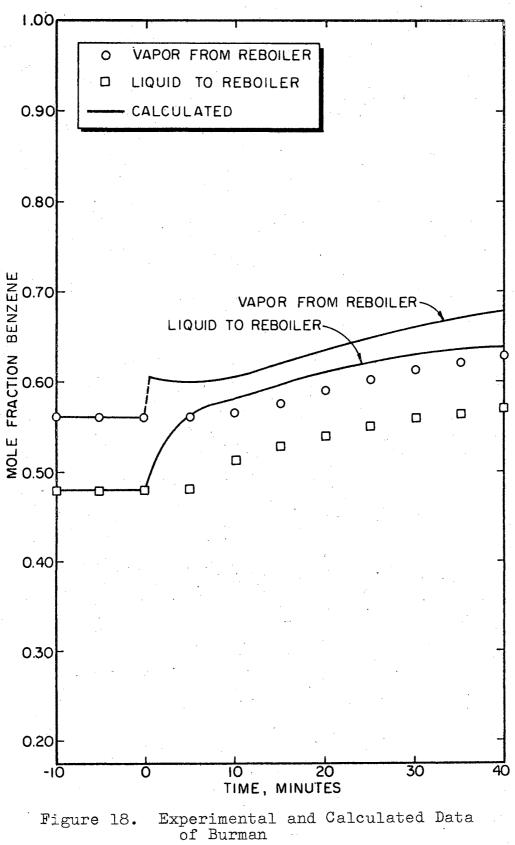


Figure 16. Variation of Separation Parameter With Vapor Boilup







Osborne's model, which Burman used in his research, assumes that the reboiler is an equilibrium stage. By taking the experimental compositions of the vapor leaving the reboiler and the bottoms product, one can calculate a set of "K values". Differences were found when comparing these "K values" with thermodynamic K values. The sets of K values are plotted in Figure 19. The difference had a maximum value near the time of the feed disturbance and decreased as the column approached the final steady-state condition. This leads to the conclusion that the reboiler is not operating as a theoretical stage during transient operation.

An efficiency term which forced the thermodynamic K values to equal those calculated from experimental data was placed in the reboiler calculations in the model. The bottom composition of the model was controlled to follow the compositions measured on the experimental column. A combination of proportional, integral, and derivative modes was used to control the reboiler heat input. The agreement with the experimental data improved; however, the difference was not completely resolved.

When an upset occurs in the feed section of a column, a time lag occurs before the effect is noted in the bottoms product stream. By adjusting the reboiler holdup term or the column holdup term in the column model, the time lag in the experimental column can be more closely approximated. Doubling the reboiler holdup in the calculations gave the

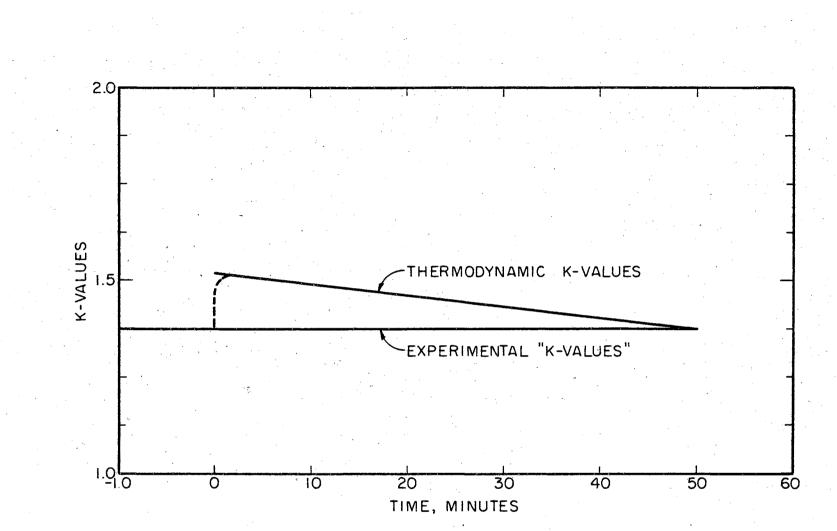


Figure 19. Comparison Between Experimental and Thermodynamic K Values

best results. The vapor boilup concentration was more closely approximated. However, there was still a discrepancy between the calculated and experimental values of the concentration of the liquid stream to the reboiler.

A third possible factor which could explain the inconsistency is the external and internal material balances. Burman assumed that there was constant molal overflow during the initial and final steady-state. However, Osborne's model does not make this assumption. Therefore, the initial steady-state values of the internal stream flow rates were varied to observe the effect upon the internal compo-The inconsistency in the liquid stream composisition. tion was decreased; however, the value of the separation parameter, which is calculated from the initial steadystate data, became negative when the best results were achieved. A negative separation parameter of the more volatile component violates the principles of distillation, because there must be some movement of the more volatile component up the column. However, it is felt that accurate measurement of the steady-state internal flow rates will give better correlation than was exhibited by assuming constant molal overflow.

1

CHAPTER V

CONCLUSIONS AND RECOMMENDATIONS

Conclusions

The objectives of this research project were threefold. The first objective was to apply feedback control to the lumped parameter model. The second objective was to vary the column efficiency to observe the results upon the model and the control of the model. The third objective was to attempt to resolve inconsistencies in the internal vapor composition from the reboiler.

Step, ramp, and sinusoidal disturbances were induced in the feed rate and composition. The bottoms composition was controlled by adjusting the reboiler heat duty. Combinations of the proportional, integral, and derivative modes were used to achieve the desired control.

The results indicate that the lumped parameter model can be readily controlled within a feedback control loop. The control curves appear to be consistent with normal control modes and exhibit characteristics of the control of a first order differential equation. The derivative controller appears to provide the bulk of the control. However, the combination of the three modes should be used to allow for the possibility of a steady-state error.

A decrease in the column efficiency resulted in an increase of the vapor boilup rate. The control function of the model was not seriously affected by the variable separation parameter.

The inconsistency in the internal vapor composition of the model and Burman's data can be partially resolved by placing an efficiency factor on the reboiler, by considering a time lag between the feed upset and the corresponding change in the bottoms composition and by measuring accurately the steady-state internal flow rates.

Recommendations

The dynamic model and control scheme should be extended to describe the operation and control of a complete column.

A combination of feedforward and feedback controllers should be developed. This control scheme would provide much faster and more reliable control of the column products.

A separation parameter should be determined for the reboiler. This would probably eliminate the error in the vapor boilup composition.

To obtain an actual working control system, a small computer should be tied into an experimental distillation column. Putting the theory into practice would more thoroughly test the practicality of the control model.

The effect of a variable tray efficiency upon the control function should also be tested experimentally.

, »

NOMENCLATURE

°'4

. .

Major Symbols

English Letters

В	= bottoms product flow rate, moles/hour.
D	= distillate flow rate, moles/hour.
Ds	= disturbance function.
Έ	= efficiency.
Er	= error in the controlled variable.
F	= feed flow rate, moles/hour.
₽e	= feed enthalpy, BTU/lbmole.
Ft	= feed tray location.
h	= enthalpy of liquid, BTU/lbmole.
H	= enthalpy of vapor, BTU/lbmole.
J	= parameter which describes the degree of
	separation occurring in a column section,
	moles/hour-mole fraction.
K	= vapor liquid equilibrium coefficient.
K1	= constant for proportional control.
Ka	= constant for integral control.
K ₃	= constant for derivative control.
L	= liquid flow rate, moles/hour.
m	= manipulated variable which is adjusted to
	correct error in controlled variable.

^m s	slope of equilibrium curve.
m(t)	deviation of reflux flow rate about the
	steady-state.
N	= net rate of mass transfer between phases,
	moles/hour.
Q _r	= reboiler heat duty, BTU/hour.
R	= reflux flow rate, moles/hour.
t	time, hour.
ť'	time, minutes.
V	= vapor flow rate, moles/hour.
x	= liquid composition, mole fraction.
У	vapor composition, mole fraction.
Z	= height of the column section, ft.

Greek Letters

δ	Ξ	holdup in a section or on a tray, moles.
τ	, ff	time constant, hour.
ω	=	frequency, cycles/hour.

Subscripts

b = bot	toms.
---------	-------

bd = desired bottoms composition.

bm = measured bottoms composition.

d = distillate.

f = feed.

k = subsection number.

I = internal.

i = component number.

n = section number.

o = initial condition.

r = rectifying section.

s = stripping section.

Superscripts

* = equilibrium value.

Groups

$$\frac{d}{dt} = \text{total derivative with respect to time.}$$

$$\frac{\partial}{\partial z} = \text{partial derivative with respect to distance.}$$

$$\frac{\partial}{\partial t} = \text{partial derivative with respect to time.}$$

A SELECTED BIBLIOGRAPHY

- 1. Bertrand, L., and J. B. Jones. Chemical Engineering, 68 (1961), 139.
- 2. Brosilow, C. B., and K. R. Handley. <u>A I Ch E Journal</u>, Number 3, 14 (1968), 467.
- 3. Burman, L. D. Ph. D. Thesis, Oklahoma State University, Stillwater, Oklahoma (1968).
- 4. Gardner, W. C. <u>Hydrocarbon Processing and Petroleum</u> <u>Refiner</u>, Number 8, 40 (1961), 95.
- 5. Gerster, J. A., and J. R. Haden, Jr. <u>Instrumentation</u> <u>in the Chemical and Petroleum Industries</u>, 3 (1966), 41.
- 6. Harrison, H. L., and J. S. Ballinger. <u>Introduction</u> <u>to Automatic Controls</u>. Scranton, Pennsylvania: International Textbook Company, 1963.
- 7. Hoffman, Harold. <u>Hydrocarbon Processing and</u> <u>Petroleum Refiner</u>, Number 2, 42 (1963), 108.
- 8. Holland, Charles D. <u>Unsteady State</u> <u>Processes With</u> <u>Applications in Multicomponent</u> <u>Distillation</u>. Englewood Cliffs, New Jersey: Prentice-Hall, Inc., 1966.
- 9. Kastanek, F., and G. Standart. <u>Separation Science</u>, Number 2, 4 (1967), 439-486.
- 10. Luyben, W. L. <u>Chemical Engineering Progress</u>, Number 8, 61 (August, 1965), 74.
- 11. Luyben, W. L., and J. A. Gerster. <u>Industrial and</u> <u>Engineering Process Design and Development</u> <u>Quarterly</u>, Number 4, 3 (October, 1964), 374.
- 12. Lupfer, D. E. <u>Industrial and Engineering Chemistry</u>, Number 12, 53 (December, 1961), 963.
- 13. Lupfer, D. E., and M. L. Johnson. <u>I</u> <u>S</u> <u>A</u> <u>Transactions</u>, Number 2, 3 (1964), 165.

- 14. Lupfer, D. E., and J. R. Parsons. <u>Chemical Engineer-</u> <u>ing Progress</u>, Number 9, 58 (September, 1962), 37.
- 15. Marr, G. R. "Distillation Column Dynamics: A Suggested Mathematical Model." Paper presented at the A. I. Ch. E. Meeting, Baltimore, Maryland, May, 1962.
- 16. Osborne, W. G., Jr. M. S. Thesis, Oklahoma State University, Stillwater, Oklahoma (1964).
- 17. Osborne, W. G., Jr. Ph. D. Thesis, Oklahoma State University, Stillwater, Oklahoma (1967).
- 18. Reynolds, D. L. M. S. Thesis, Oklahoma State University, Stillwater, Oklahoma (1964).
- 19. Rose, A. <u>Industrial and Engineering Chemistry</u>, Number 6, 48 (June, 1956), 1008.
- 20. Rose, A., and T. J. Williams. <u>Industrial and Engineer</u>ing <u>Chemistry</u>, 47 (November, 1955), 2284.
- 21. Rosenbrock, H. G. <u>Transactions of the Institution of</u> <u>Chemical Engineers</u>, 40 (1962), 35.
- 22. Talbot, F. D. <u>Instrumentation in the Chemical and</u> <u>Petroleum</u> <u>Industries</u>, 2 (1965), 69.
- 23. Williams, T. J. <u>Chemical</u> Engineering Progress, 53 (May, 1957), 220.
- 24. Williams, T. J. <u>Industrial and Engineering Chemistry</u>, Number 9, 50 (1958), 1214.
- 25. Williams, T. J., and V. A. Lauher. <u>Automatic Control</u> of <u>Chemical and Petroleum Processes</u>. Houston, Texas: Gulf Publishing Company, 1961.

APPENDIX A

FUNCTIONS OF THE FEED DISTURBANCE

FUNCTIONS OF THE FEED DISTURBANCE

Three different types of feed disturbances were used in the study. The disturbances were represented by step, ramp, and sinusoidal functions.

A step change in the feed flow rate and composition can be represented by the equations

$$\Delta \mathbf{F} = \text{constant}$$
 (68)

and

$$\Delta \mathbf{x}_{\mathbf{f}} = \text{constant.} \tag{69}$$

The values of the final steady-state feed flow rate and composition were substituted for the initial values at the time of the upset.

The equations for the ramp functions are given by

$$\mathbf{F} = \mathbf{F}_0 + \mathbf{a}_1 \mathbf{t} \tag{70}$$

$$\mathbf{x}_{\mathbf{f}} = \mathbf{x}_{\mathbf{f}_0} + \mathbf{a}_2 \mathbf{t}, \tag{71}$$

where a_1 and a_2 are arbitrarily determined constants which determine the slopes of the respective ramps. The values of the constants were varied to change the slope or severity of the disturbance. The values of a_1 were 1.05 x 10⁻³, 2.1 x 10⁻³, 4.2 x 10⁻³, 8.4 x 10⁻³, and

16.8 x 10⁻³ moles/(hour)². The values of a_2 were 0.15, 0.31, 0.62, 1.24, and 2.48 hour⁻¹.

The sinusoidal functions are represented by the equations

$$\mathbf{F} = \mathbf{A}_1 \operatorname{sin\omegat} \tag{72}$$

$$\mathbf{x}_{\mathbf{f}} = \mathbf{A}_{\mathbf{2}} \operatorname{sin\omegat}. \tag{73}$$

The amplitudes of the sine functions were set at one-half the value of the respective step changes. The frequency was arbitrarily varied to obtain a range of good operation or control. The frequency, ω , had values of 1.5, 3, 6, 12, and 24 cycles/hour.

APPENDIX B

INCORPORATION OF CONTROL AND HEAT BALANCE EQUATIONS INTO OSBORNE'S COMPUTER PROGRAM

INCORPORATION OF CONTROL AND HEAT BALANCE EQUATIONS INTO OSBORNE'S COMPUTER PROGRAM

The calculation of the transient column conditions was made by a computer program written by Osborne (16, 17). However, some changes were made in the original program to incorporate a feedback control scheme.

The first major change involved the calculations of the flow rates after the upset. Osborne's program assumed that the flow rates changed to their final steady-state values at the time of the upset. These steady-state values were read into the computer on data cards. In the revised program, the flow rates were calculated through the use of heat and material balances.

A second change was the addition of a subroutine to the program. This subroutine contained the heat and material balance equations and the controller equations. The error in bottoms composition is given by

$$\mathbf{E}_{\mathbf{r}} = \mathbf{x}_{\mathrm{bd}} - \mathbf{x}_{\mathrm{bm}}.$$
 (59)

The controller corrects the error by the equation

$$\Delta Q_{r} = K_{1}E_{r} + K_{2}\int_{0}^{t'}E_{r} dt' + K_{3} \frac{dE_{r}}{dt'}.$$
 (66)

The trapezoidal rule was the method of integration. The differentiation was made by assuming for small intervals that

$$\frac{\mathrm{dE}_{\mathbf{r}}}{\mathrm{dt}} = \frac{\mathrm{E}_{\mathbf{r}}|_{\mathrm{t=t}}}{\mathrm{t}}.$$
 (74)

The new value of Q_r is then used in the heat and material balance equations to calculate the new flow rates. The disturbance functions were also included in the subroutine. The subroutine was called by the main program at the desired time intervals. For this study, the time interval was set at one-half minute.

APPENDIX C

SIMULATED COLUMN RESPONSE CURVES

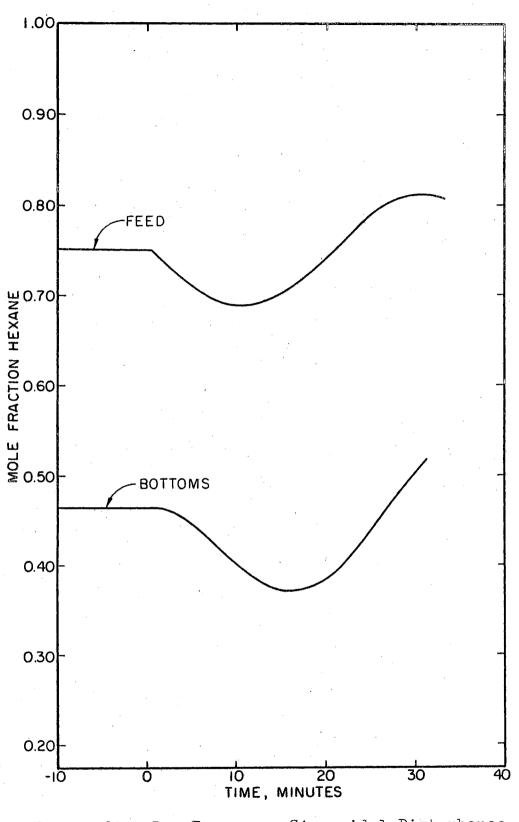


Figure 20. Low Frequency Sinusoidal Disturbance With No Control

68

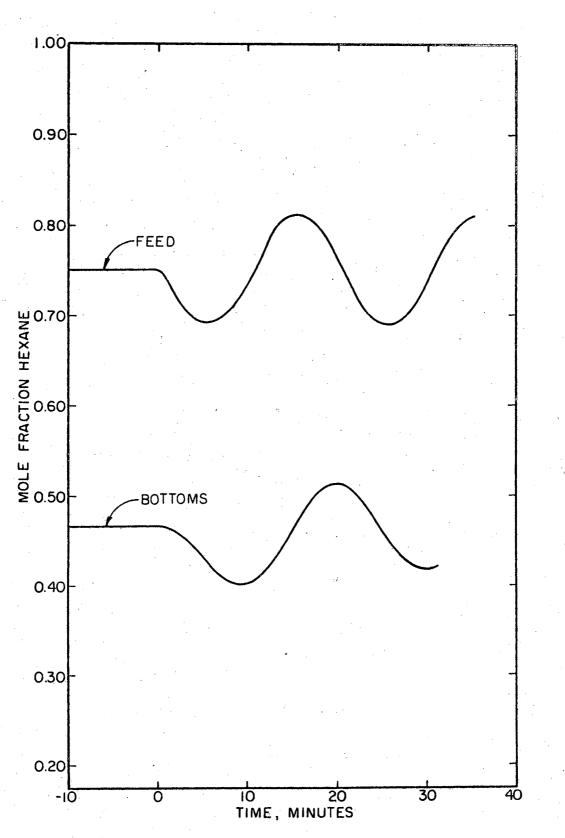
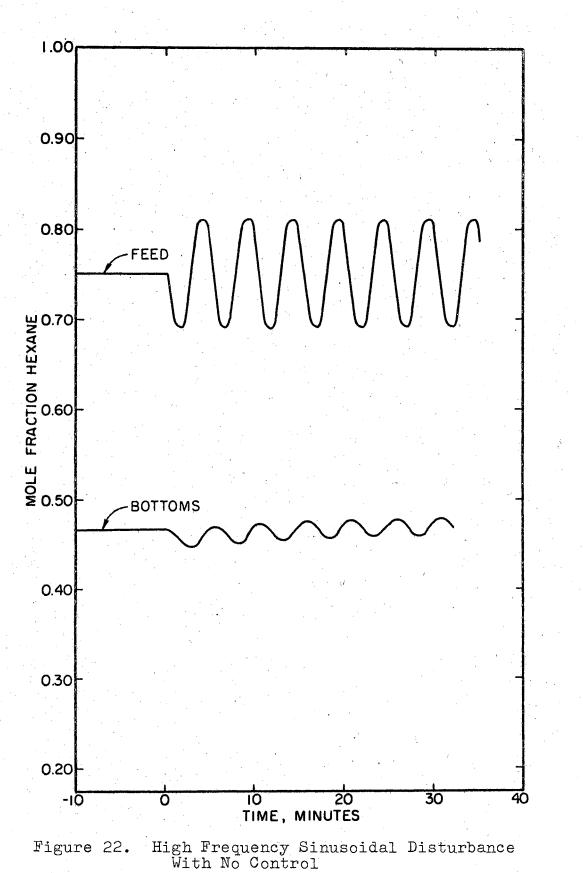
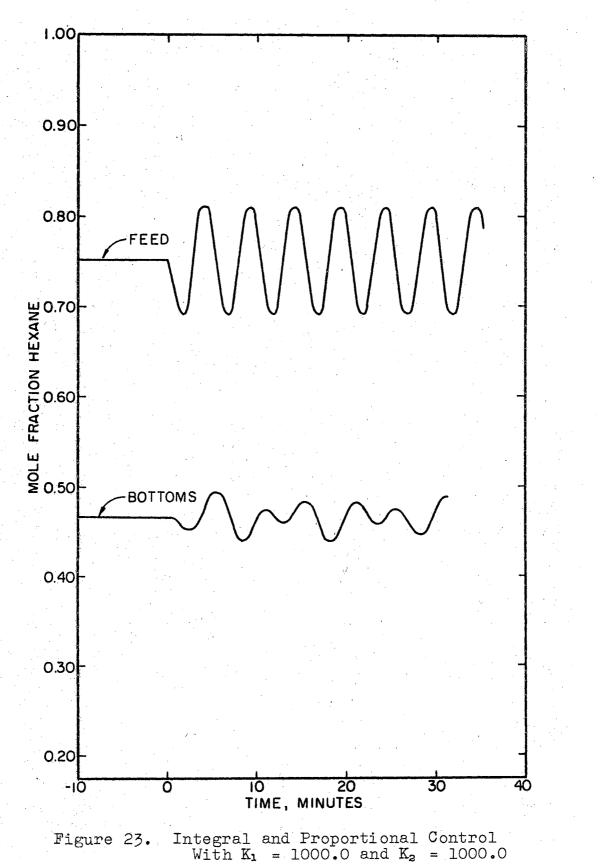
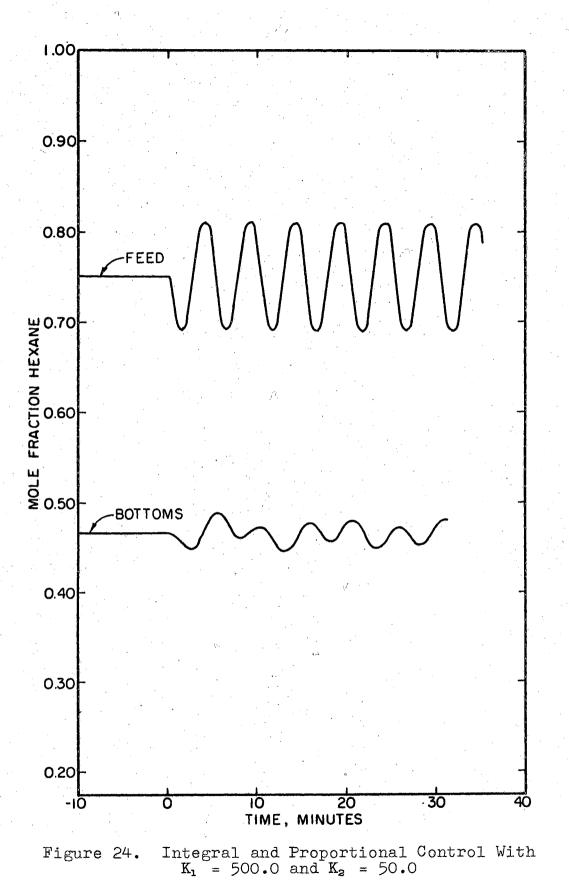


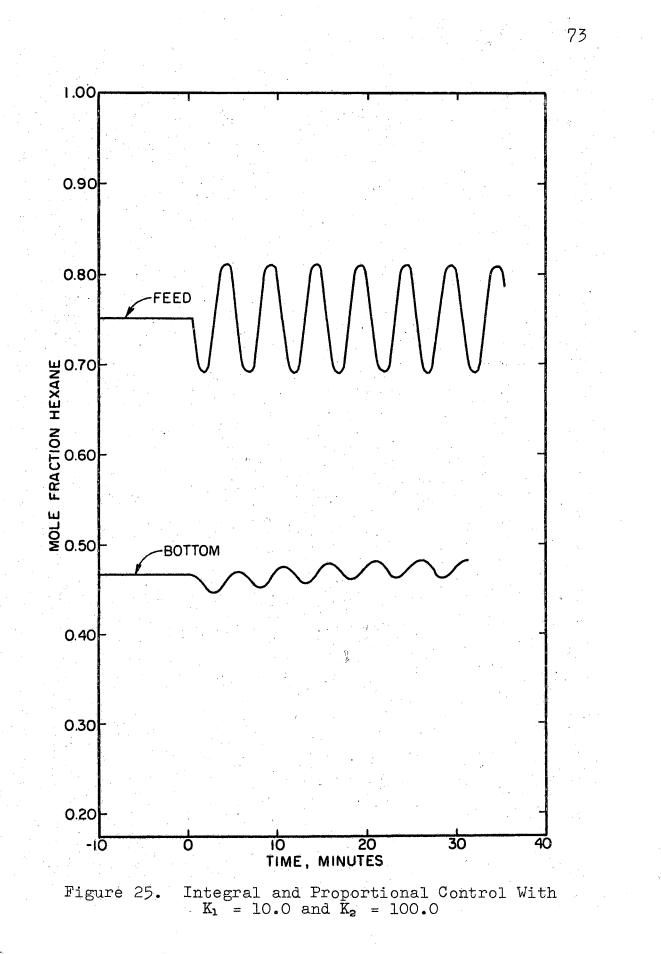
Figure 21. Moderate Frequency Sinusoidal Disturbance With No Control

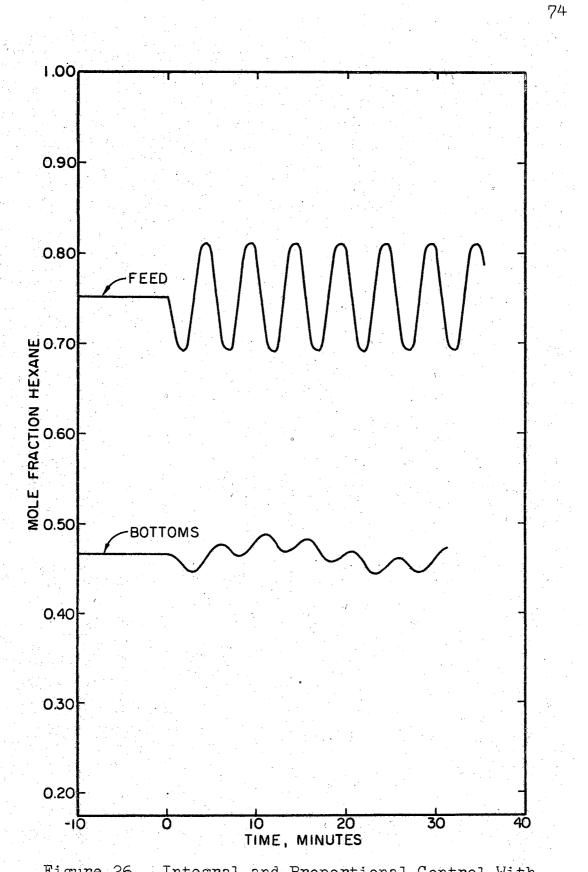
69

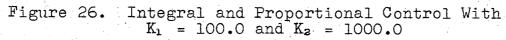


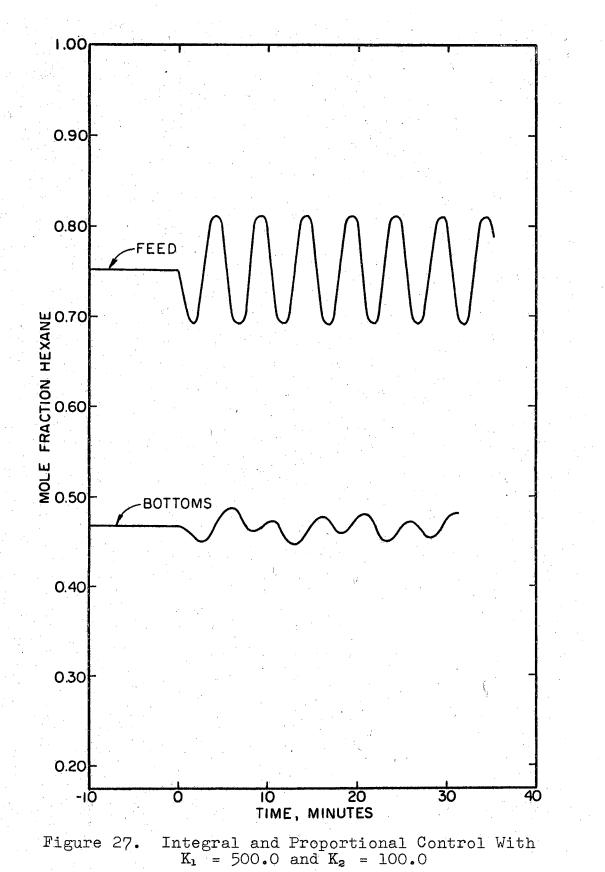


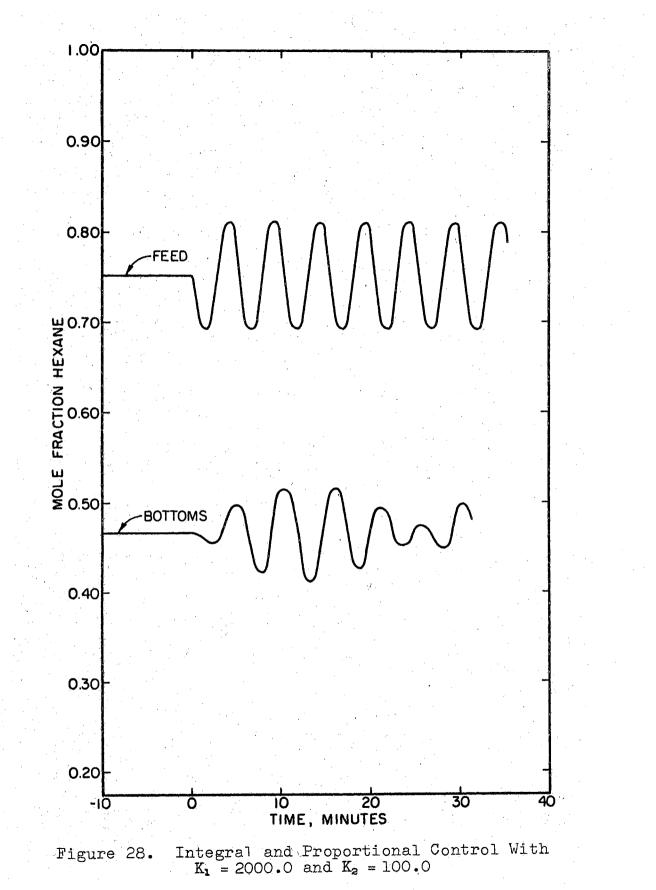


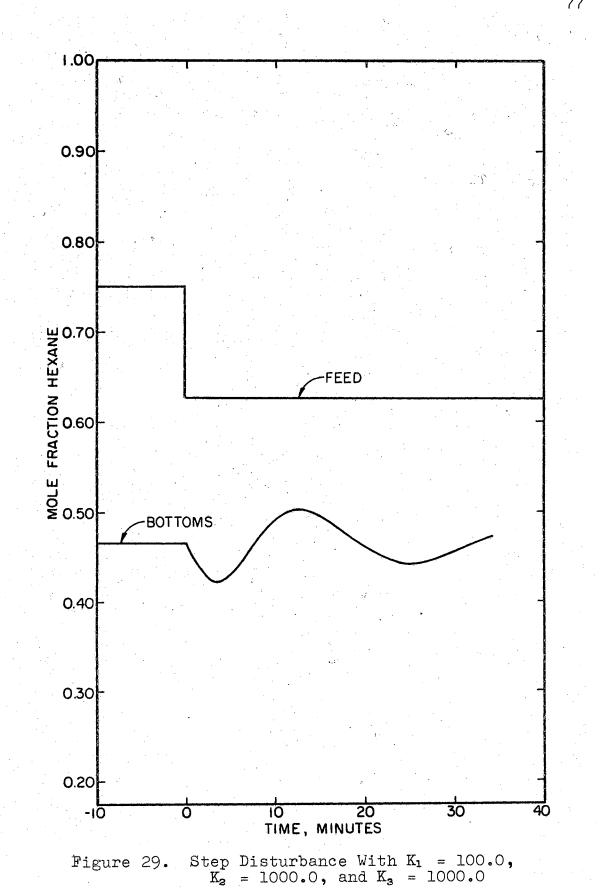


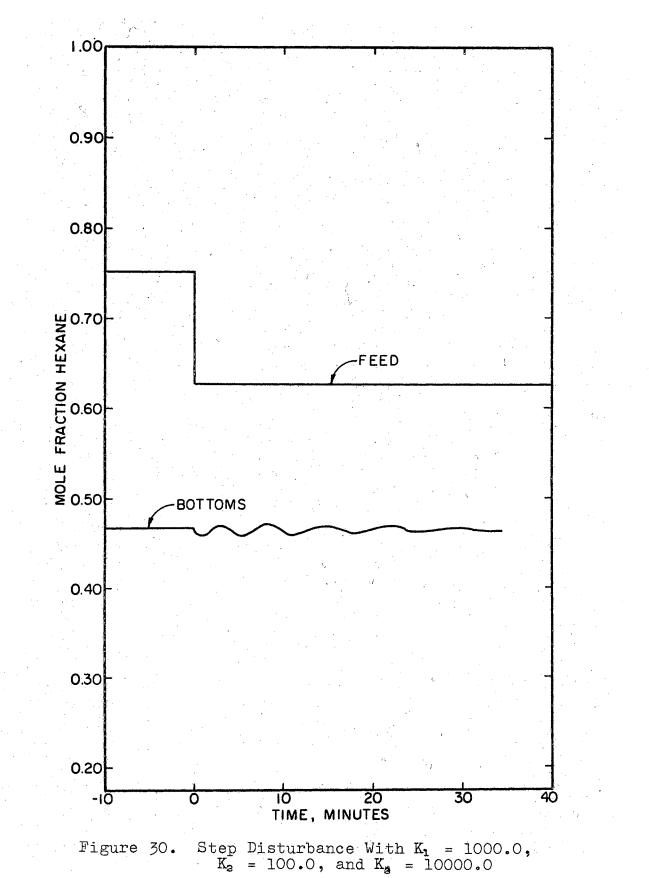


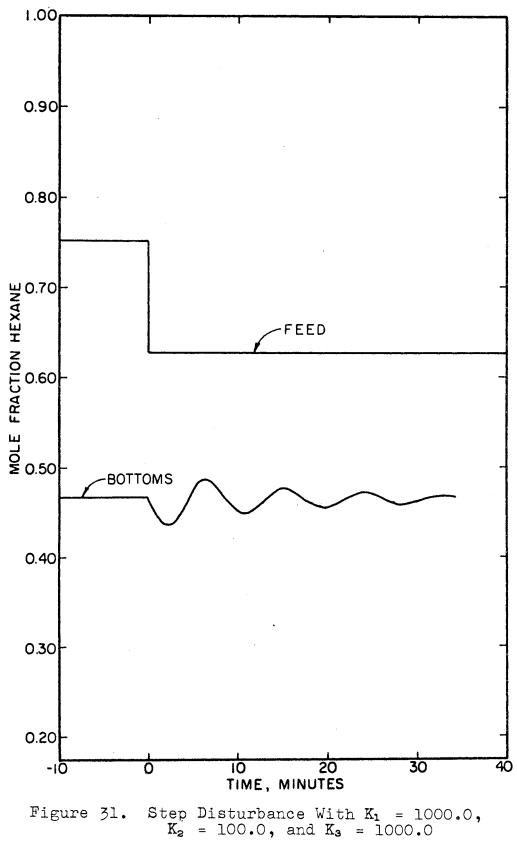


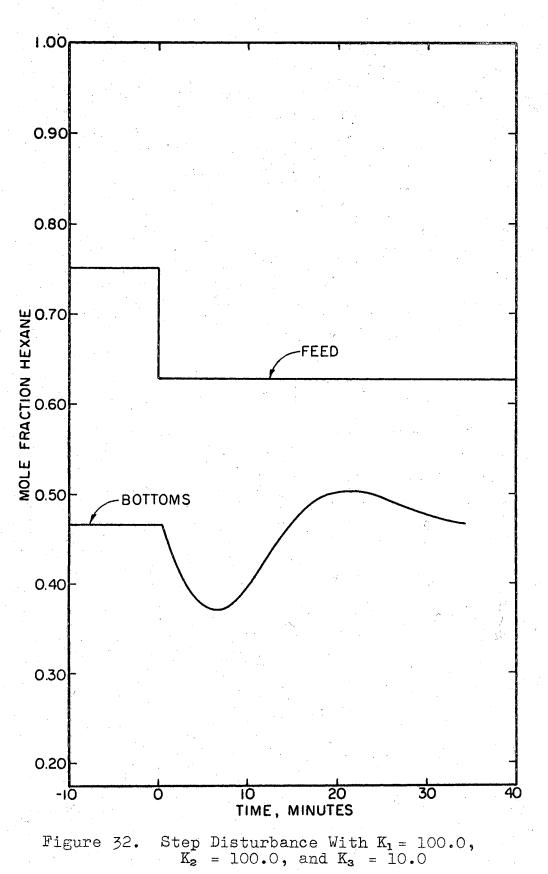


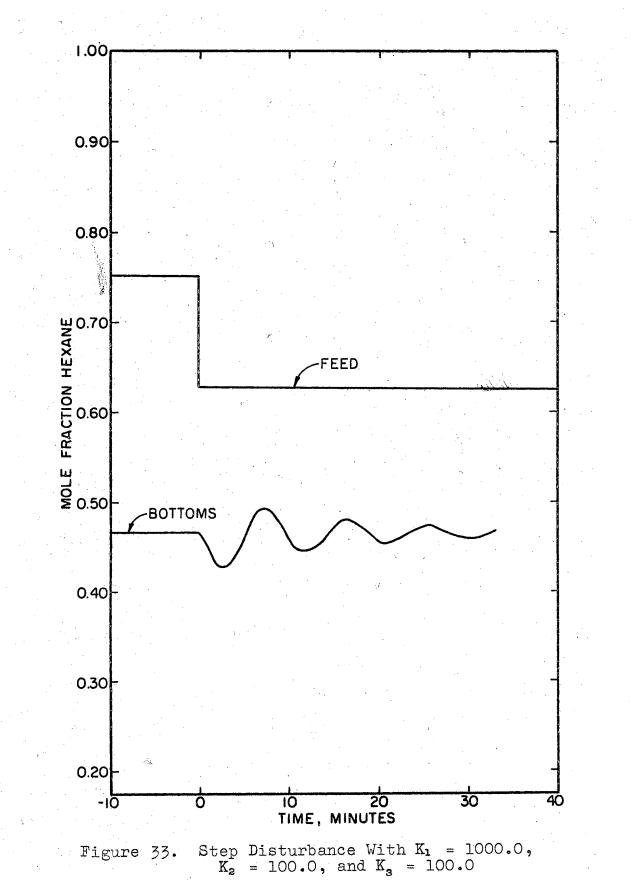


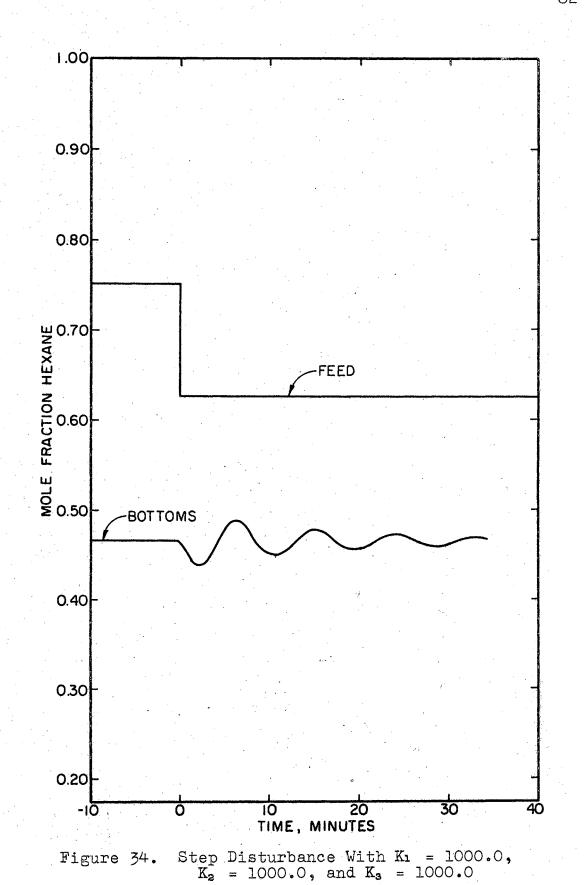


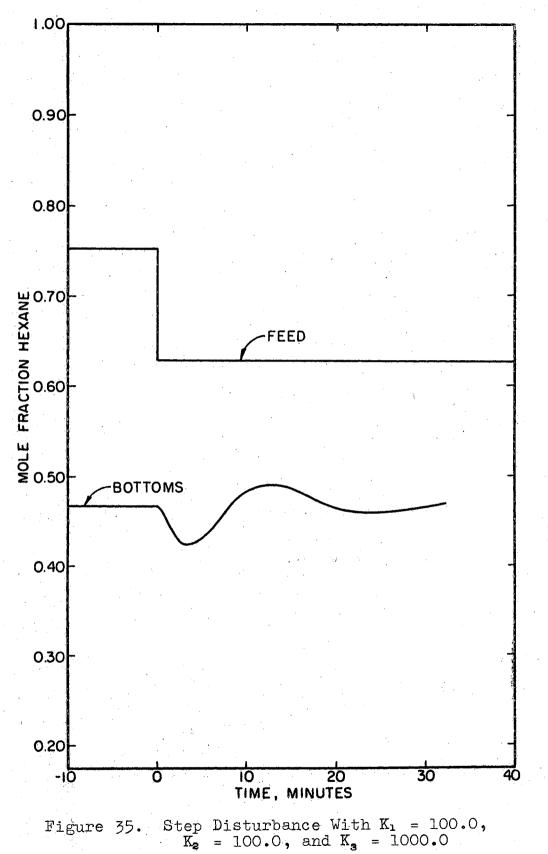












VITA Z

Larry E. Porter

Candidate for the Degree of

Master of Science

Thesis: AN EVALUATION OF THE FEEDBACK CONTROL OF A STRIPPING COLUMN MODEL

Major Field: Chemical Engineering

Biographical:

Personal Data: Born in El Reno, Oklahoma, November 10, 1944, the son of Mr. and Mrs. M. E. Porter.

Education: Attended elementary and high school in El Reno, Oklahoma; graduated from El Reno High School in 1963; attended Oklahoma State University and received the Bachelor of Science degree in 1967, with a major in Chemical Engineering; enrolled in Graduate College at Oklahoma State University in September, 1967; completed requirements for Master of Science degree in May, 1969. Membership in scholarly or professional societies includes Omega Chi Epsilon, Phi Eta Sigma, Sigma Tau, Pi Mu Epsilon, Phi Kappa Phi, and American Institute of Chemical Engineers.

Professional Experience: Summer employment in the Polyolefins Research division of E. I. Du Pont de Nemours, Inc., Orange, Texas, June to September, 1966 and 1967.